

# Emergency Evaluation of Struvite Application as an Alternative to Conventional Phosphorus Fertilizer for Crop Production

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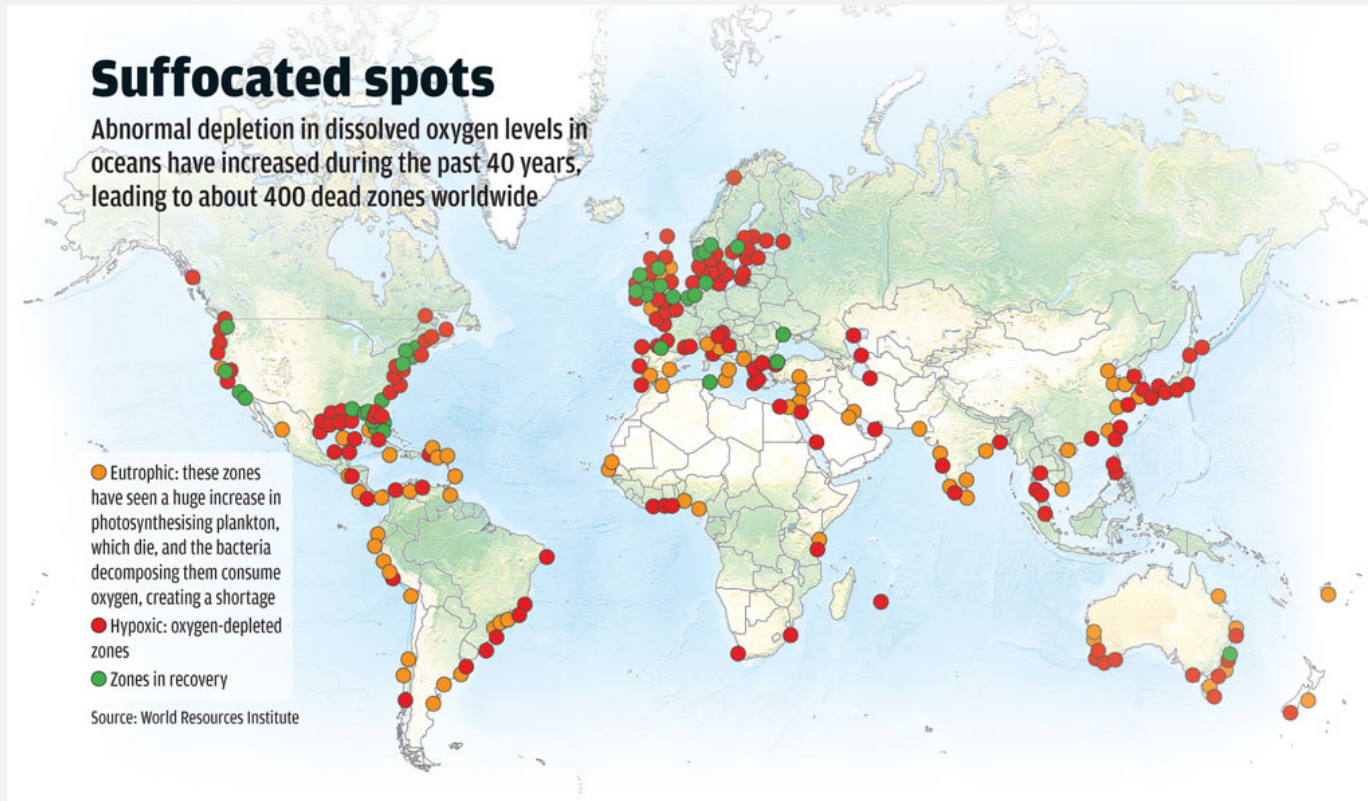
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# Importance of Nutrient Management



Source: World Resources Institute, 2015

- Eutrophication - enrichment of an ecosystem with chemical nutrients, typically compounds containing nitrogen (N), phosphorus (P), or both.
- Clean Water Act (CWA) requires wastewater treatment plants (WWTPs) to reduce nutrient discharge levels to prevent eutrophication

# Study Objectives and Approach

- Aims to address
  - 1) how regulations drive system changes;
  - 2) how conventional systems can be transitioned to more cost effective and sustainable alternatives using nutrient management.
- Influent wastewater flow and nutrient levels, capital, and operational data were collected from previous nutrient removal studies and for nutrient recovery from Ostara Nutrient Recovery Technologies, Inc.
- Use emergy accounting to provide system analysis
- All UEVs used and given hereafter (including those referenced in the text) were normalized to the **1.20 E25 sej/yr (solar emjoules/year)** global emergy baseline (Brown et al., 2016)

# Nutrient Recovery and Benefits

- Nutrient recovery - practice of recovering nutrients (N and P) from wastewater and converting them into an environmental friendly fertilizer
- Industrial phosphate ( $\text{PO}_4^{3-}$ ) fertilizers - manufactured using  $\text{PO}_4^{3-}$  rock (non-renewable resource)
- Nutrient recovery provides a self-sustainable solution to WWTPs
  - revenue generation from fertilizers
  - reduces fouling of equipment with involuntary precipitation of struvite
  - helps meet discharge limits
- $\text{PO}_4^{3-}$  precipitation from wastewater is less energy intensive and economical compared to manufacture of phosphate fertilizers

# Struvite Formation and Production

- Recovered from municipal wastewater (MWW)/urine source - slow-release mineral fertilizer given by the simplified equation

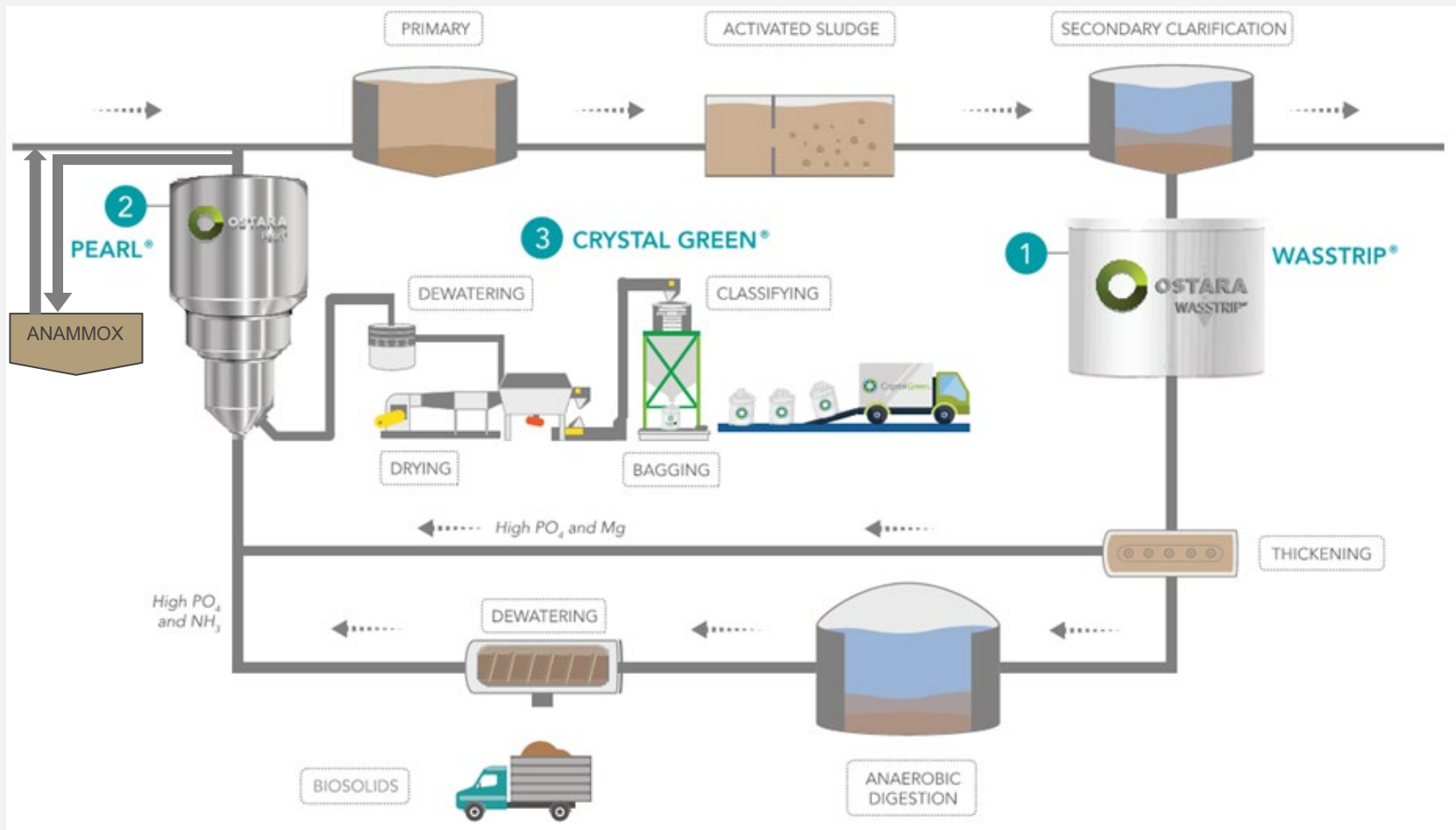


Magnesium Ammonium Phosphate

- Methods of struvite recovery from MWW have been under development, this study cites WASSTRIP™ and PEARL® process by Ostara Nutrient Recovery Technologies, Inc.
- Marketed fertilizer - 5% N, 28%  $\text{PO}_4^{3-}$ , and 0% potash, with 16.6% MgO (10% Mg)



# Nutrient Recovery Technology Considered

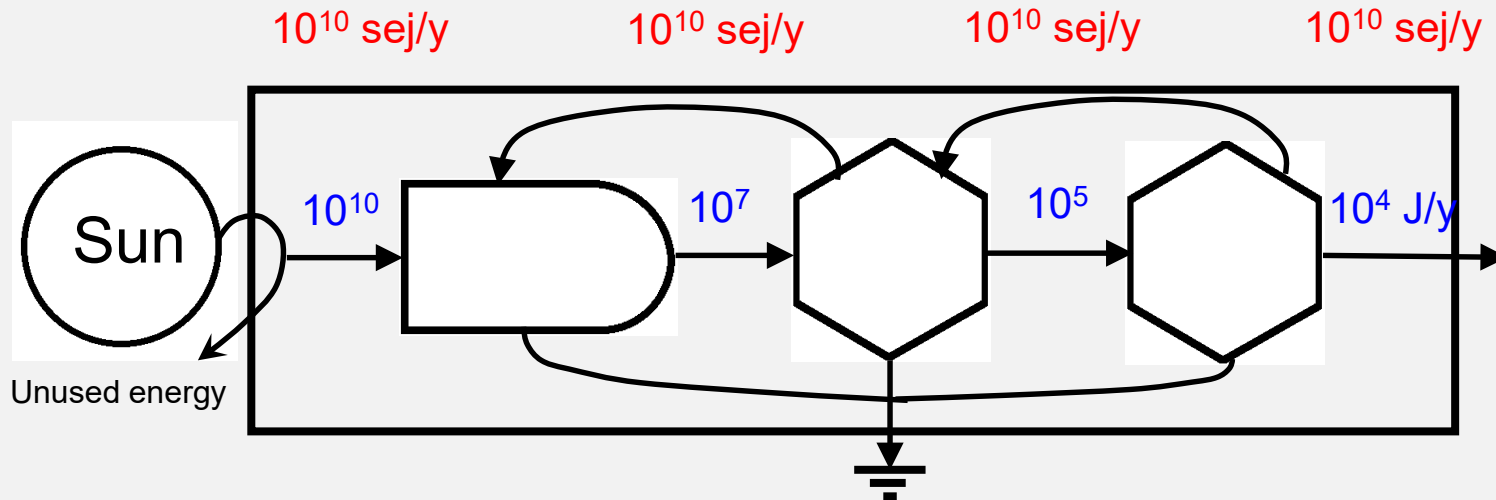


PEARL® process by Ostara Nutrient Recovery Technologies, Inc, 2016

- In addition to P precipitation, partial nitrification anammox was considered for nitrogen reduction in the nutrient recovery alternative.

# What is Energy

- Available energy of any kind previously used both directly and indirectly to make another form of energy, product or service (H.T. Odum, 1996)



# Unit Energy Value (UEV)

- Material (per mass) – specific energy

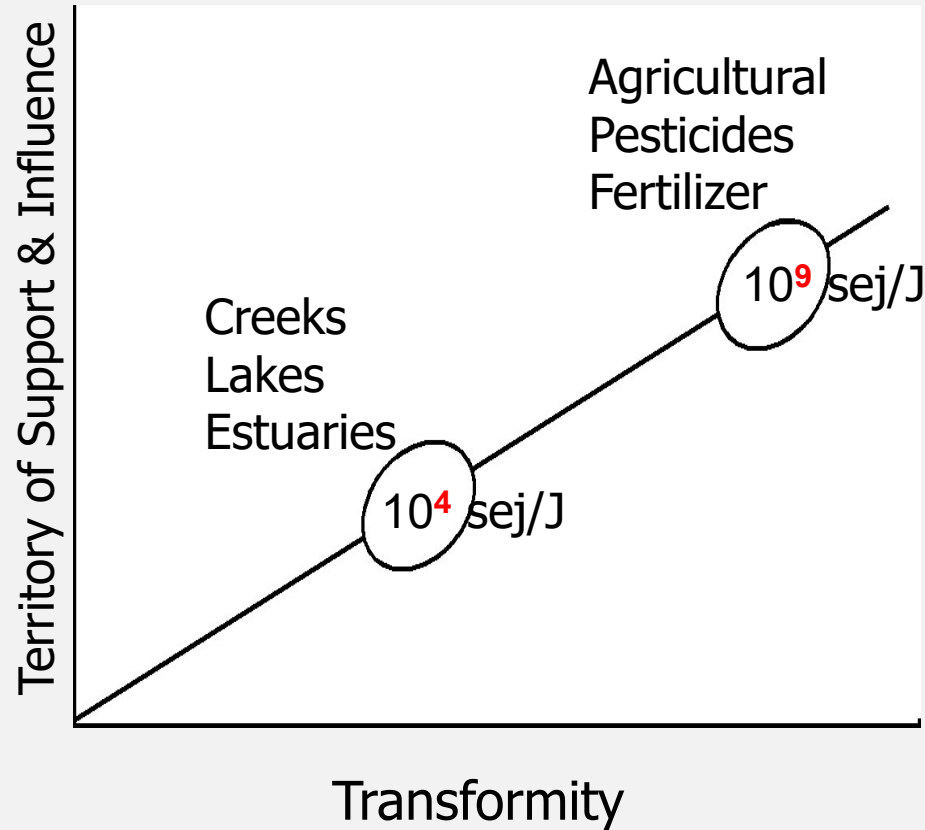
$$\frac{\text{total energy input}}{\text{mass output}} = \text{sej/g}$$

- Energy (per joule) –Transformity

$$\frac{\text{total energy input}}{\text{energy output}} = \text{sej/J}$$



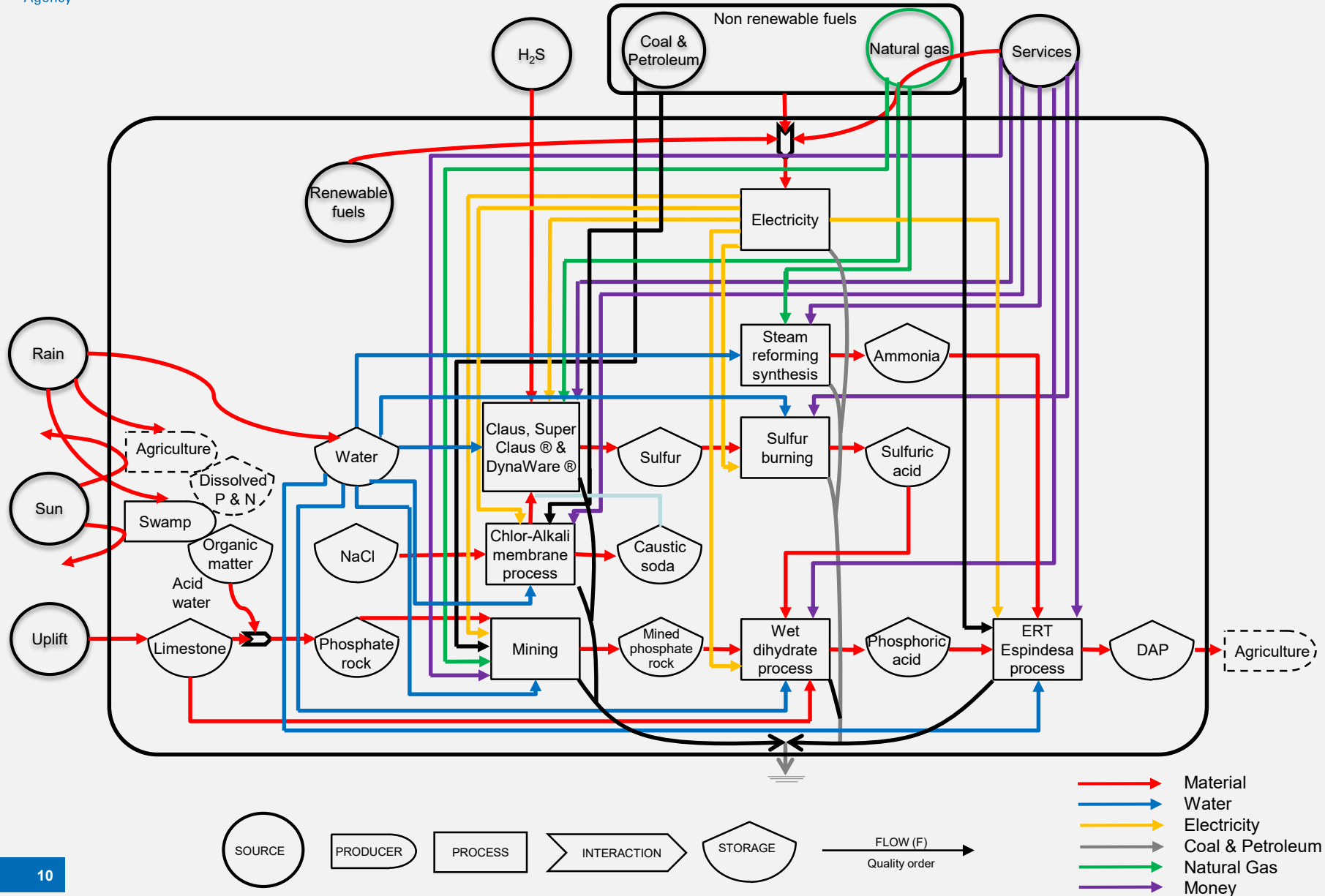
# Transformity



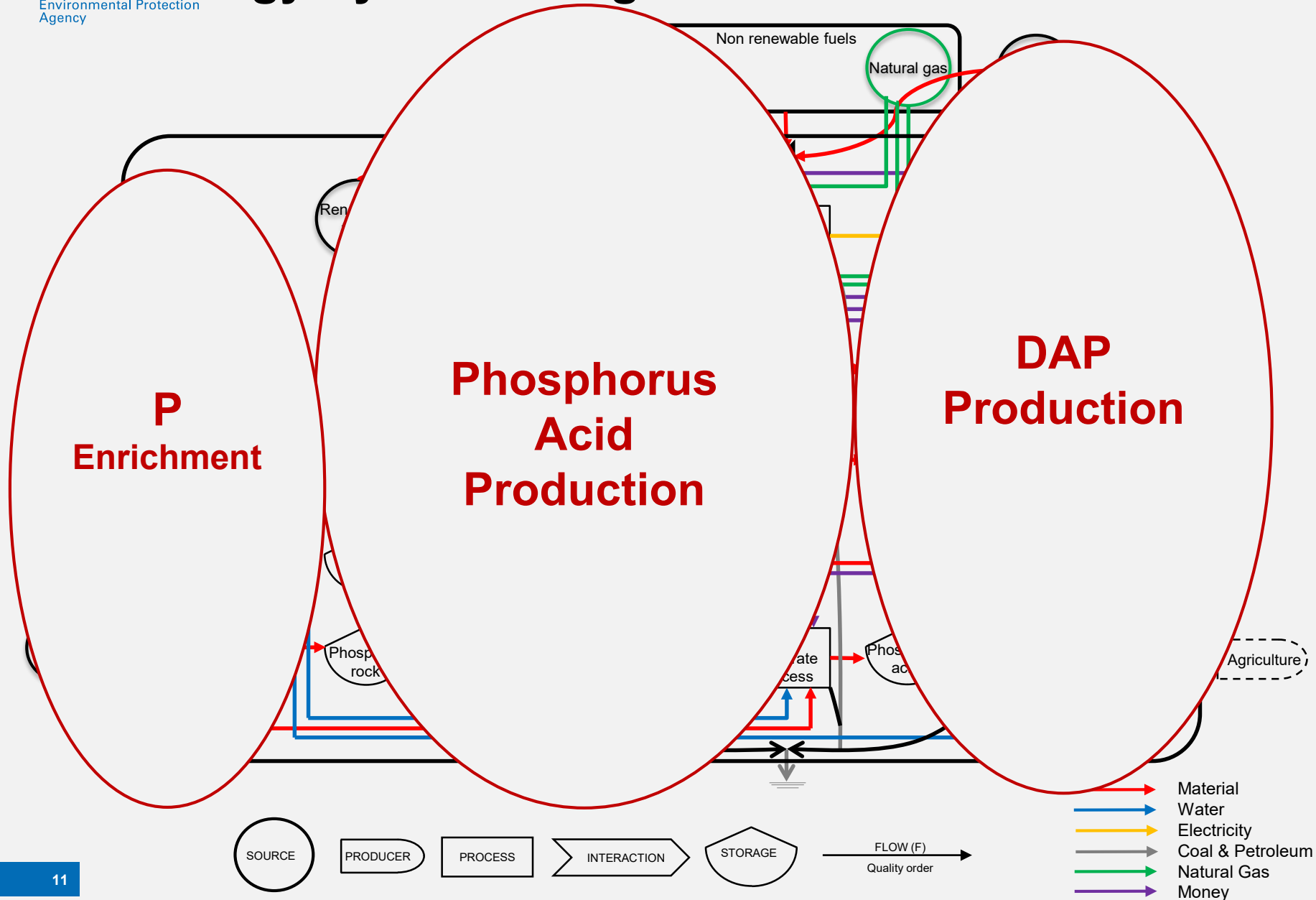
- High transformity = high hierarchical order
- High transformity = high territory of influence
- High transformity = more emergy required to make product flow
- High transformity = less efficient

Transformity measures energy quality

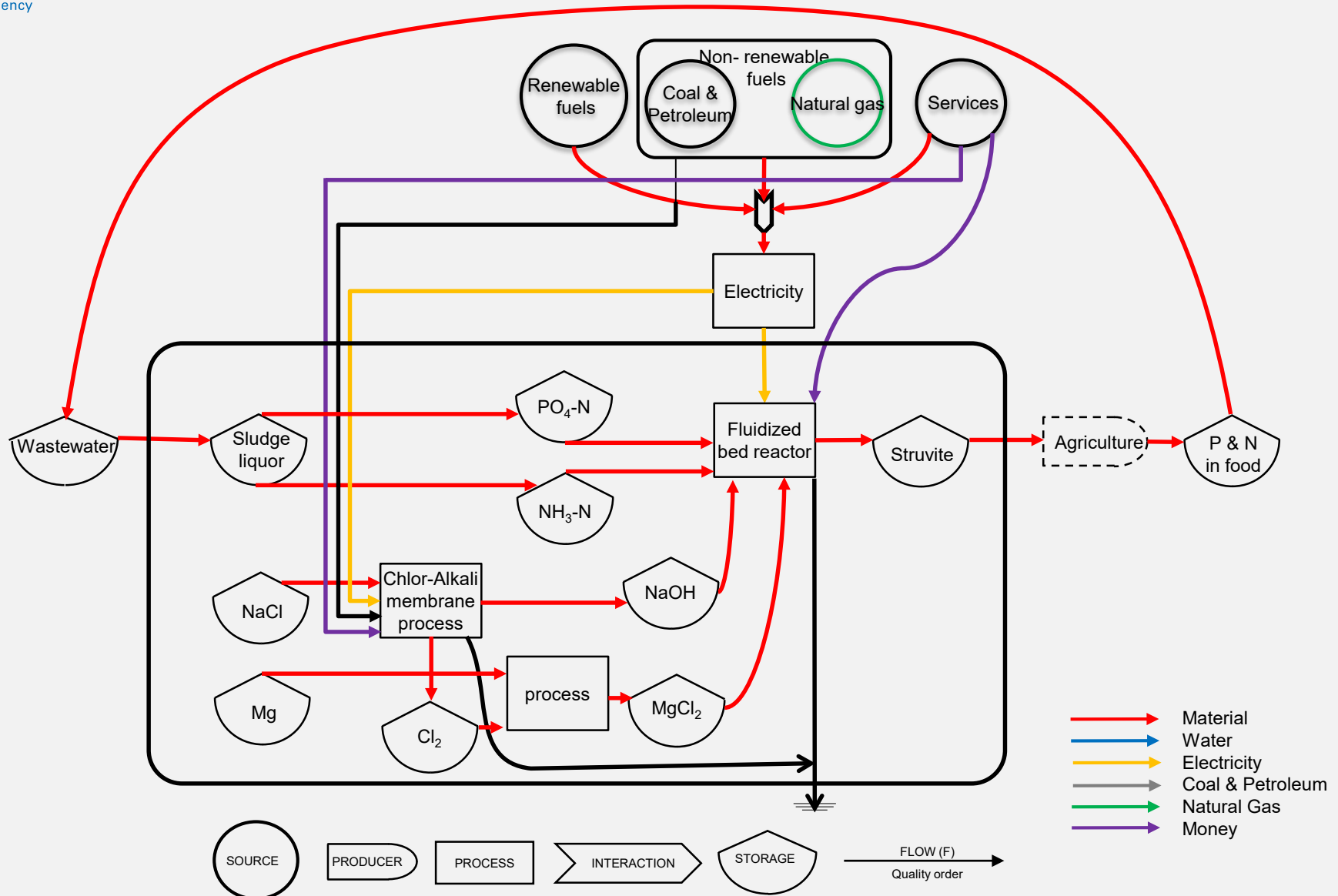
# Energy Systems Diagram for DAP Production



# Energy Systems Diagram for DAP Production



# Energy Systems Diagram for Nutrient Recovery



External forcing functions (circles) provide inflow energy materials and information to the producers (bullet-shape symbols). Internal storages (tank symbols) and economic and social subsystems (boxes) are shown

# Results of Traditional Fertilizer Vs. Nutrient Recovery

## Diammonium Phosphate (DAP)

Chemical formula: $(\text{NH}_4)_2\text{HPO}_4$ Composition: 18% N, 46% $\text{P}_2\text{O}_5$ (20% P)					
Note	Description	Data	Unit	UEV (sej/unit)	EMERGY (E sej/yr)
Infrastructure input					
*	Capital	1.14E+01	\$	2.02E+12	2.31E+13
Operational inputs per year (2013)					
1	Materials				
1a	Phosphate Rock	1.50E+06	g	3.61E+09	5.40E+15
1b	Ammonia	1.44E+05	g	6.48E+09	9.35E+14
1c	Sulfur	3.97E+05	g	9.50E+10	3.77E+16
1d	Limestone	3.02E+04	g	2.20E+08	6.65E+12
2	Energy				
2a	Electricity	1.16E+08	J	7.26E+05	7.85E+12
2b	Fuels	4.34E+08	J	6.13E+05	4.01E+13
3	Services	5.12E+02	\$	2.02E+12	1.04E+15
4	Water	3.56E+01	m <sup>3</sup>	8.22E+11	1.23E+13
	<b>Total EMERGY</b>				<b>5.03E+16</b>
5	Transformity	w/o capital invest		5.03E+10	sej/g DAP
		with capital invest		5.03E+10	sej/g DAP
		w/o capital invest		1.18 E+10	sej/g P

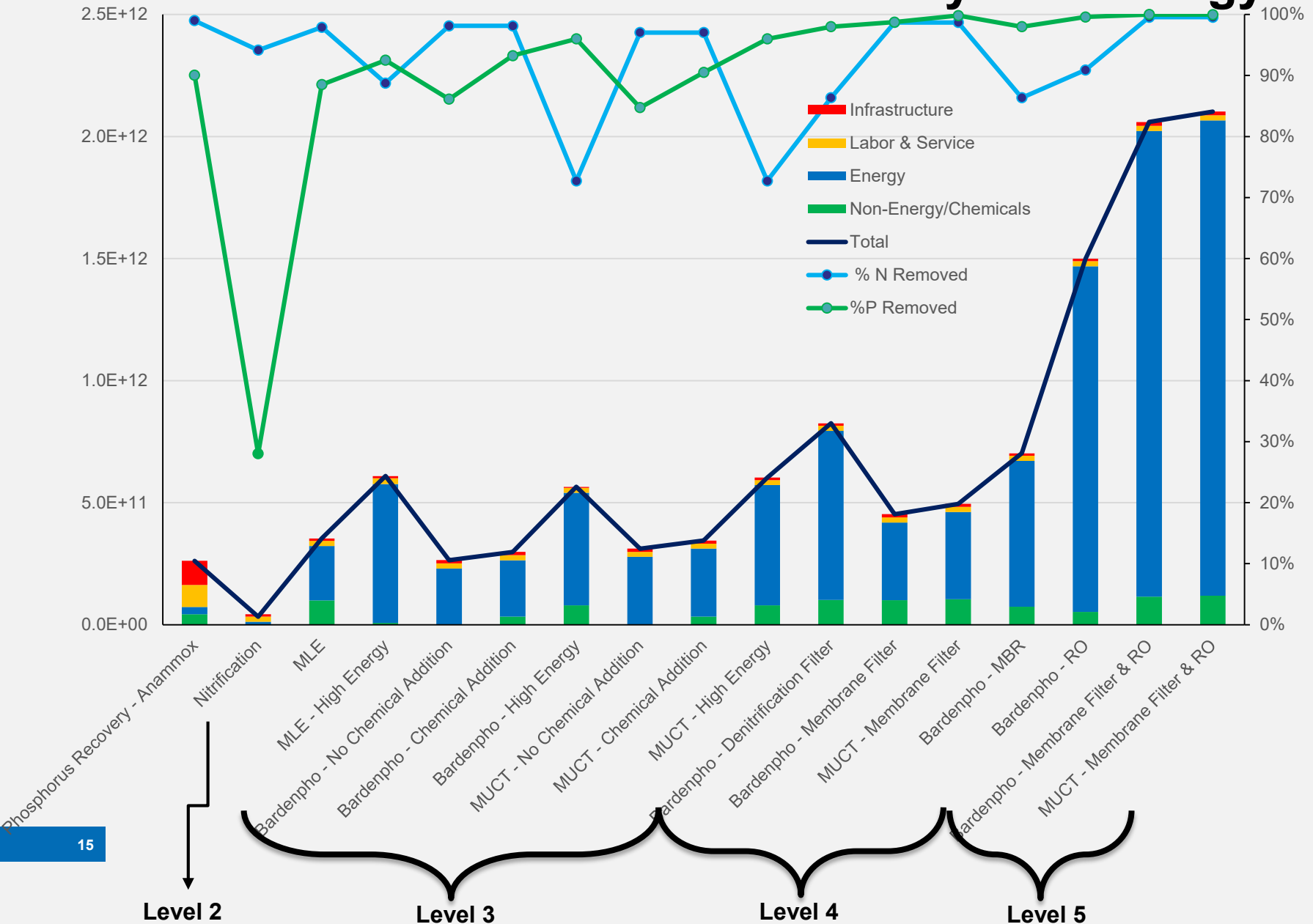
## Struvite

Chemical Formula: Crystal Green®, $\text{NH}_4\text{MgPO}_4 \cdot 6\text{H}_2\text{O}$ (5-28-0 +10% Mg)					
Note	Description	Data	Unit	UEV (sej/unit)	EMERGY (E sej/yr)
Infrastructure input					
*	Capital	2.47E+02	\$	2.02E+12	5.01E+14
Operational inputs per year (2013)					
1	Materials				
1a	Phosphate, eq. to elemental phosphorus ( $\text{PO}_4\text{-P}$ )	1.40E+05	g		0.00E+00
1b	Ammonia, equivalent to elemental Nitrogen ( $\text{NH}_3\text{-N}$ )	2.10E+05	g		0.00E+00
1c	Sodium hydroxide (NaOH)	4.90E+04	g	4.14E+09	2.03E+14
1d	Magnesium chloride ( $\text{MgCl}_2$ ) as Mg	1.47E+05	g	4.34E+10	6.38E+15
2a	Electricity	6.40E+08	J	2.21E+05	1.41E+14
3	Services	5.33E+01	\$	2.02E+12	1.08E+14
4	Wastewater	2.63E+02	g	3.26E+05	8.56E+07
	<b>Total EMERGY</b>				<b>7.10E+15</b>
5	Transformity	w/o capital invest		7.10E+09	sej/g CG
		with capital invest		7.60E+09	sej/g CG
		w/o capital invest		8.96 E+08	sej/g P

# Processes Considered for the Study

Treatment Level (Effluent Limits)	Nutrient Removal/Recovery Process	Energy (kWh/m <sup>3</sup> )	Influent Ammonia (mg/L as NH <sub>3</sub> -N)	Influent P (mg/L as P)
Recovery	Phosphorus Recovery - Anammox	0.14	20	7
Level 2 (TN – 8 mg/L, TP – 1 mg/L)	Nitrification	0.23	24	10
Level 3 (TN – 4-8 mg/L, TP – 0.1-0.3 mg/L)	MLE	0.28	23	8
	MLE - High Energy	0.59	32	8
	Bardenpho - No Chemical Addition	0.29	23	8
	Bardenpho - Chemical Addition	0.29	23	8
	Bardenpho - High Energy	0.58	22	5
	MUCT - No Chemical Addition	0.35	23	8
	MUCT - Chemical Addition	0.35	23	8
	MUCT - High Energy	0.56	22	5
Level 4 (TN – 3 mg/L, TP – 0.1 mg/L)	Bardenpho - Denitrification Filter	0.53	22	5
	Bardenpho - Membrane Filter	0.4	23	8
	MUCT - Membrane Filter	0.45	23	8
	Bardenpho - MBR	0.53	22	5
Level 5 (TN - <2 mg/L, TP<0.02 mg/L)	Bardenpho - RO	0.60	22	5
	Bardenpho - Membrane Filter & RO	2.4	23	8
	MUCT - Membrane Filter & RO	2.45	23	8

# Total Energy Comparison between Different Nutrient Removal and Recovery Technology

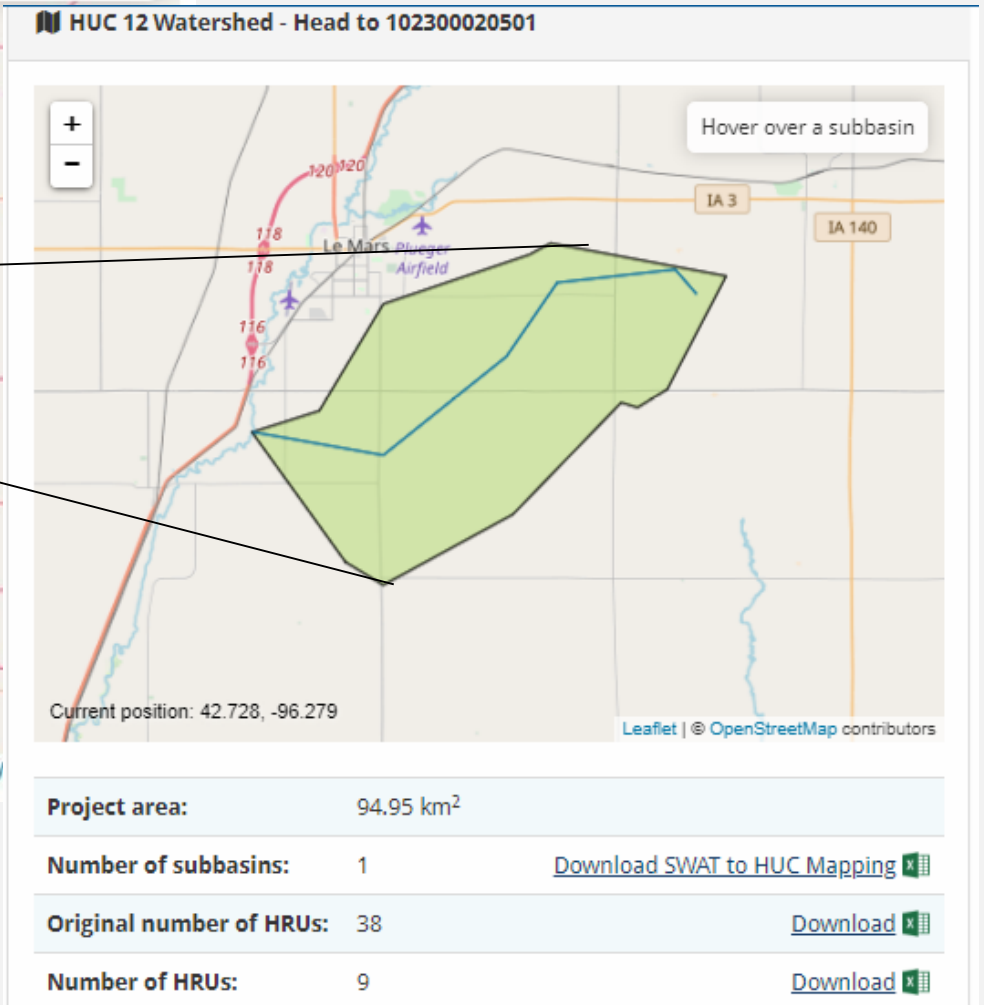
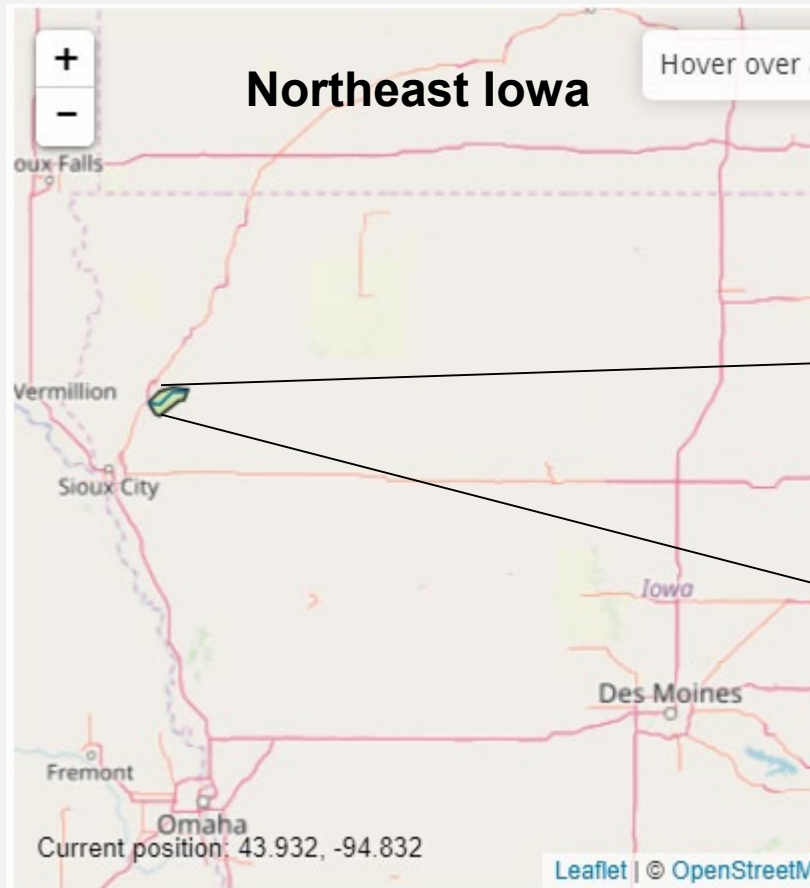


# HAWQS/SWAT Tool and Study Area

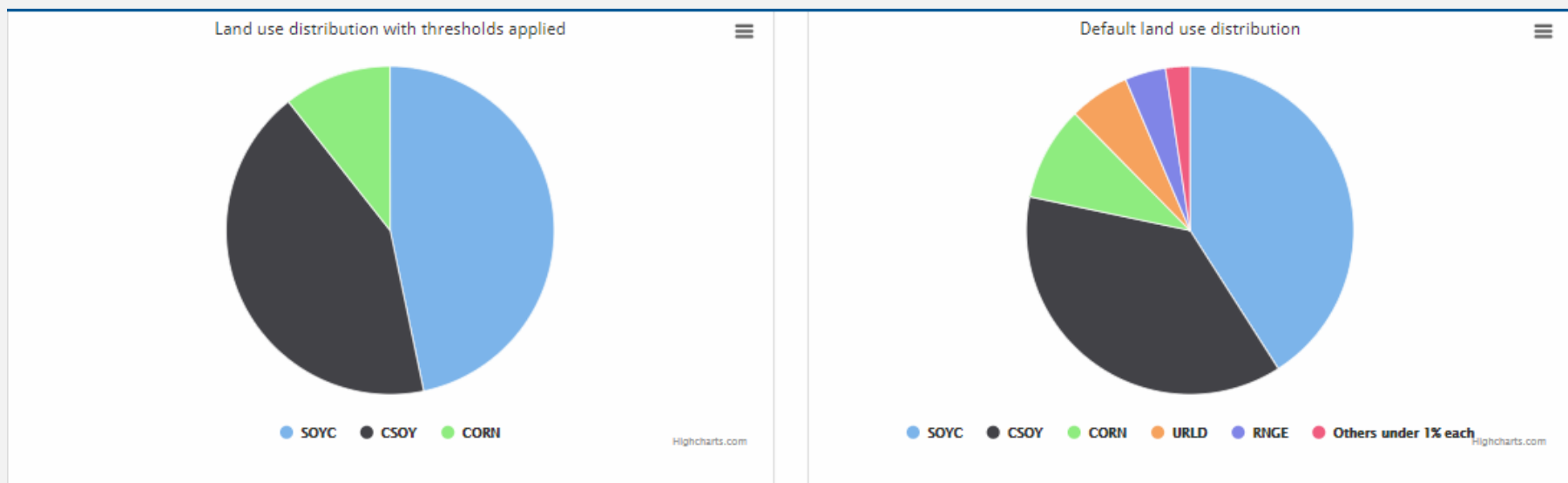
- Hydrologic and Water Quality System (HAWQS) is a web-based interactive water quantity and water quality modeling system that employs Soil and Water Assessment Tool (SWAT) as its core modeling engine
- Enables use of SWAT to simulate the effects of management practices based on an extensive array of crops, fertilizers, soils, natural vegetation types, land uses, and climate change scenarios



# HAWQS/SWAT Tool and Study Area



# Hydraulic Response Unit (HRUs) Chosen



Land use distribution comparison

With thresholds applied			Original	
Land Use	Area	% of Total Area	Area	% of Total Area
SOYC	44.37 km <sup>2</sup>	46.73 %	38.88 km <sup>2</sup>	40.95 %
CSOY	40.48 km <sup>2</sup>	42.63 %	35.47 km <sup>2</sup>	37.35 %
CORN	10.10 km <sup>2</sup>	10.64 %	8.85 km <sup>2</sup>	9.32 %

Soils and slope classes after applying thresholds

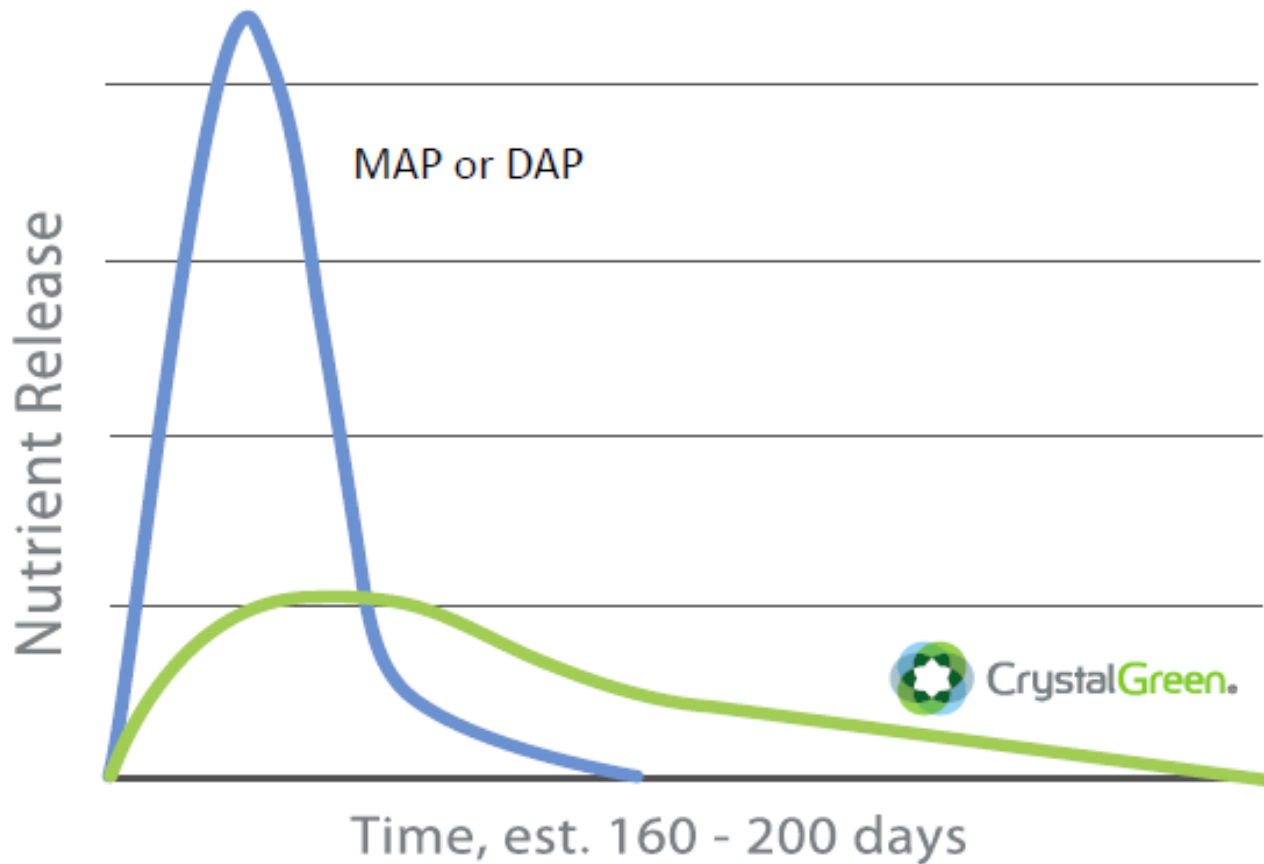
Soil Type	Area	% of Total Area
IA091	45.15 km <sup>2</sup>	47.54 %
IA092	41.28 km <sup>2</sup>	43.47 %
IA098	8.53 km <sup>2</sup>	8.99 %

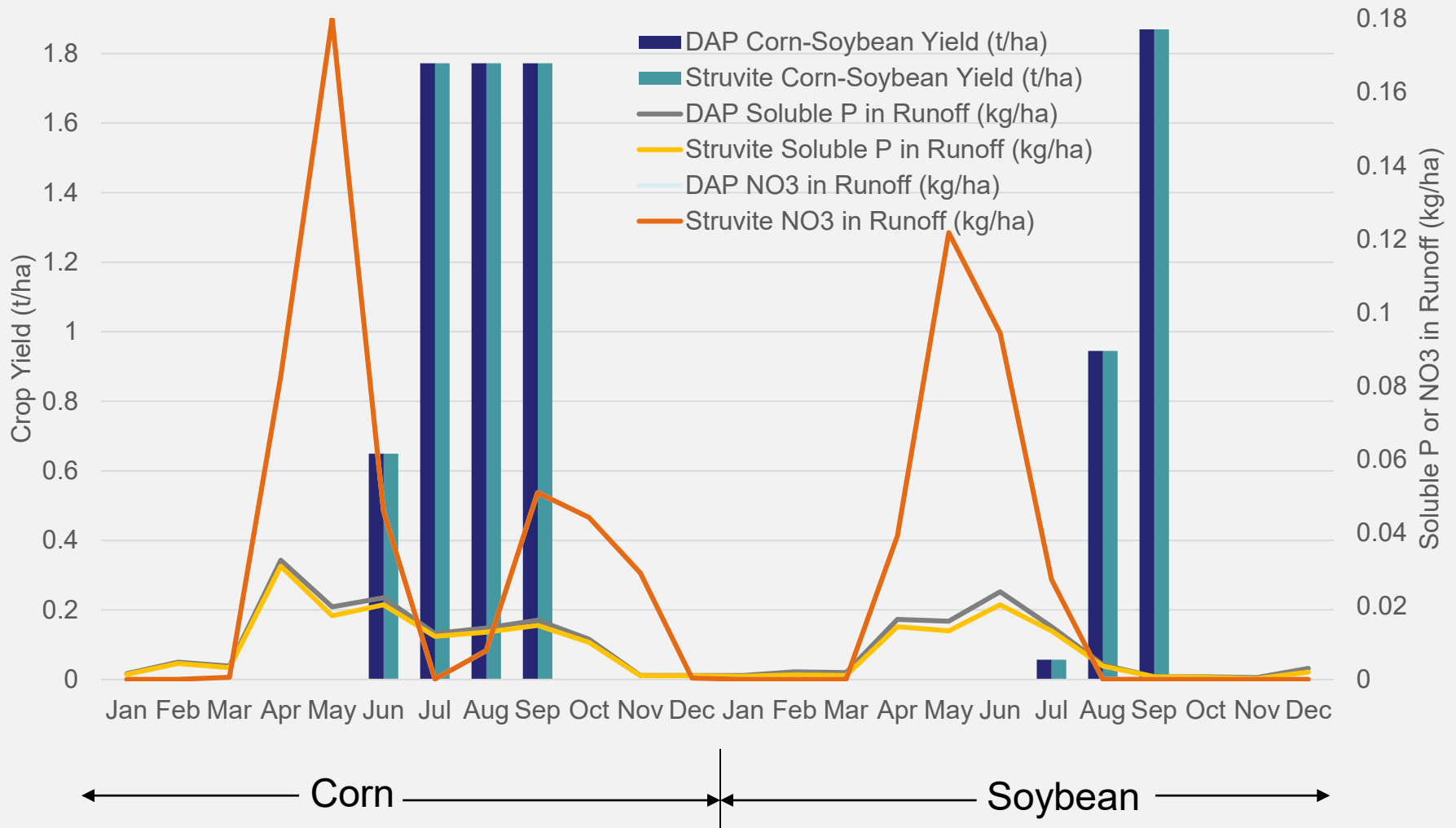
Slope Class	Area	% of Total Area
0-1	94.95 km <sup>2</sup>	100.00 %

**Crop rotation – Soybean-Corn (SOYC), Soil Type – IA-091 (Loess Ridges/Clay Paleosol), Slope >1%, HRU Area - 3.5 km<sup>2</sup>**

# DAP and Struvite characteristics and quantity applied

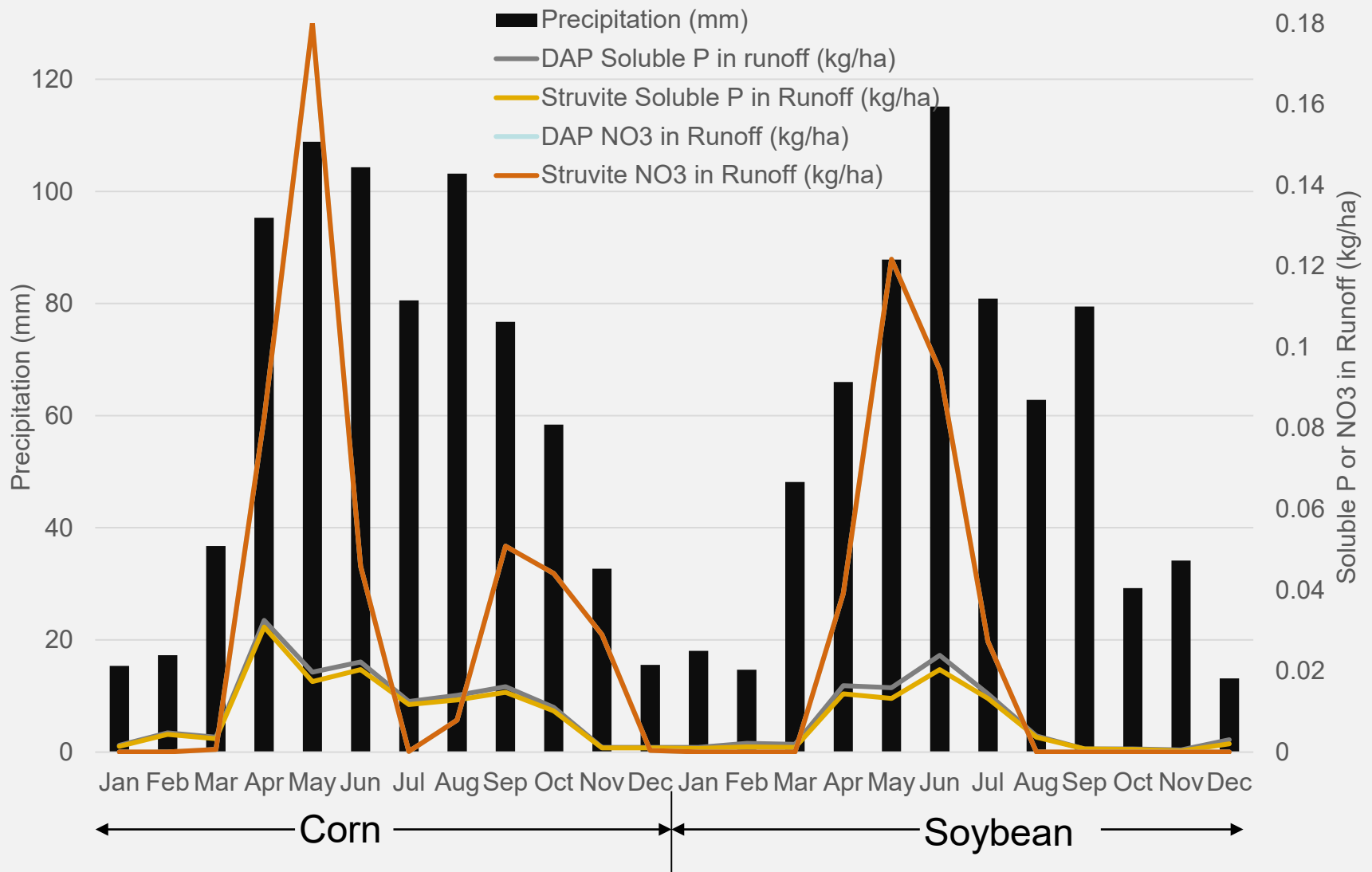


# Nitrogen and Phosphorus in Runoff and Crop Yield Results (one-cycle of crop rotation)



- Simulation results shown here is only for the two year period, long term evaluation may indicated less Struvite demand with less loss via runoff

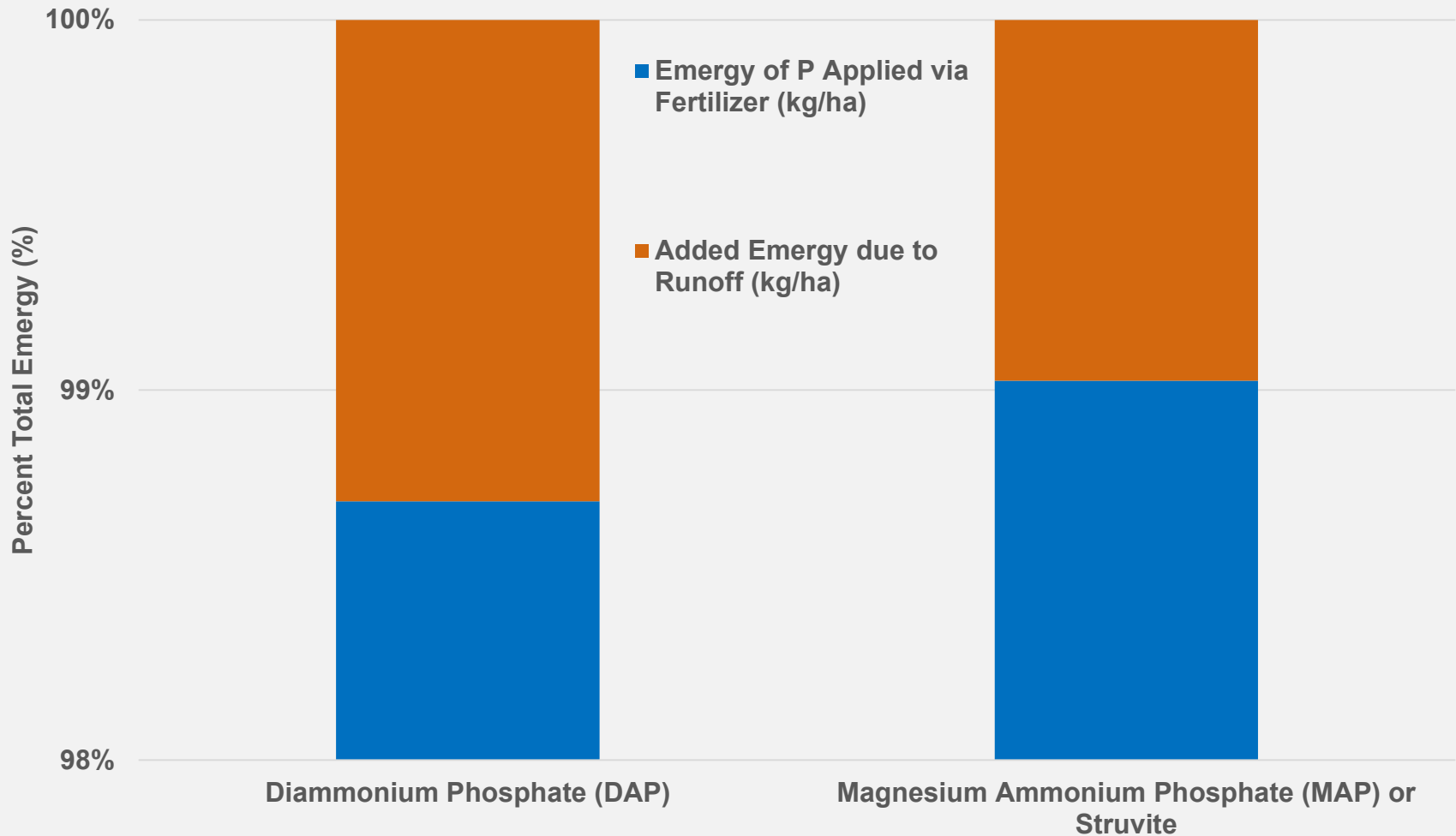
# Nitrogen and Phosphorus in Runoff and Crop Yield Results (one-cycle of crop rotation)



# Total Energy of DAP vs. Struvite via Field Application

Parameter	Diammonium Phosphate (DAP)	Magnesium Ammonium Phosphate (MAP) or Struvite
Crop Type/Rotation	Corn-Soybean	Corn-Soybean
Fertilizer Quantity (kg/ha)	100	170
P Applied via Fertilizer (kg P/ha)	18	20.4
Emergy of P Applied via Fertilizer (sej/kg P applied)	1.81E+14	1.96E+13
Runoff P (kg/ha)	0.24	0.20
Added Emergy due to Runoff (sej/kg P in runoff)	2.39E+12	1.93E+11
Runoff N (kg/ha)	0.72	0.72
Crop Yield (t/ha)	8.838	8.838
P Required (kg/t of Crop)	2.06	2.33
UEV of Fertilizer (sej/kg P)	1.01E+13	9.60E+11
Transformity/Yield (sej/t)	2.07E+13	2.24E+12

# Total Emergy of DAP vs. Struvite via Field Application



# Results and Discussions

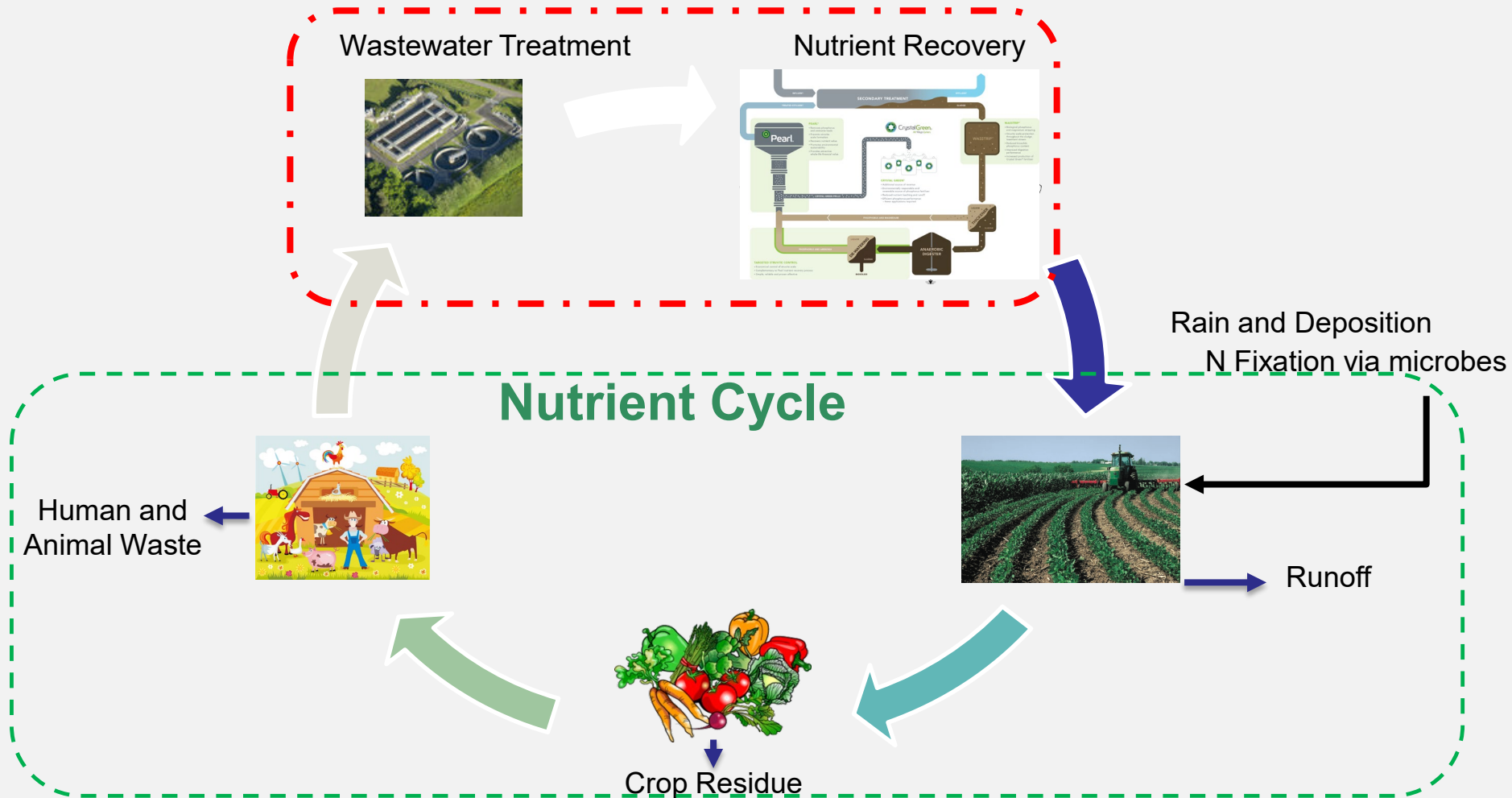
- Stringent nutrient reduction regulations lead to trade-offs that need further evaluation to choose the most sustainable treatment alternative
- Emergy analysis justifies nutrient recovery from wastewater sludge and provides sound economic and ecological comparison of removal and recovery treatment alternative independent of perceived monetary value
- Application of Struvite as a replacement/substitution to traditional phosphate fertilizers for crop growth over a long period of time can lead to substantial phosphorus and overall emergy reduction.
- DAP with an order of magnitude higher total emergy relative to struvite, displays a bigger environmental 'footprint'.



## Selected References

- Talboys, P.J., Heppell, J., Roose, T., Healey, J.R., Jones, L. D., Withers, P.J. A. (2016). *Struvite: a slow-release fertilizer for sustainable phosphorus management?* Plant Soil 401:1090-123.
- USEPA (2017). Hydraulic and Water Quality System – A National Watershed and Water Quality Assessment Tool (HAWQS v1.0). Spatial Sciences Laboratory, Texas A&M AgriLife Research and Office of Water, Immediate Office, USEPA, Washington, DC.
- Eastern Research Group, Inc. (2018). *Life Cycle and Cost Assessments of Nutrient Removal Technologies in Wastewater Treatment Plants*, Report Prepared for U.S.EPA (draft).
- Arden, S., Ma, X. and Brown, M. (2018) *Holistic Analysis of Urban Water Systems in the Greater Cincinnati Region: (2) Resource Use Profiles by Emergy Accounting Approach*. Submitted to Environmental Science and Technology (ES&T).
- Rahman, M.S., Eckelman, J.M., Onnis-Hayden, A. and Gu, A.Z.(2016) *Life-Cycle Assessment of Advanced Nutrient Removal Technologies for Wastewater Treatment*. Environmental Science and Technology, 50, pp 3020 - 3030
- Foley, J., de Haas, D., Hartley, K. and Lant, P. (2010) *Comprehensive life cycle inventories of alternative wastewater treatment systems*. Water Research, 44, pp 1654 – 1666.
- Odum, H.T. Environmental accounting. John Wiley & Sons: New York, 1996.
- Brown, M. T., Campbell, D. E., De Vilbiss, C., Ulgiati, S. (2016) *The Geobiosphere Emergy Baseline: A Synthesis*. Ecological Modelling.
- Fux, C. and Siegrist, H. (2004). *Nitrogen removal from sludge digester liquids by nitrification/denitrification or partial nitrification/anamox: environmental and economical considerations*. Water Science and Technology. 10, pp. 19-26

# Future or Continued Work



Account for the benefits of nutrient recovery via efficient use of the struvite fertilizer and the flow of N and P nutrients in the food system, the economic, environmental and societal benefits of struvite recovery would be more perceptible.

# Acknowledgements

- Safe and Sustainable Water Resources National Research Program in the EPA's Office of Research and Development
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# Thank you! Questions?

