

# AMoN Site Characterization Study: Phase I Field Measurements

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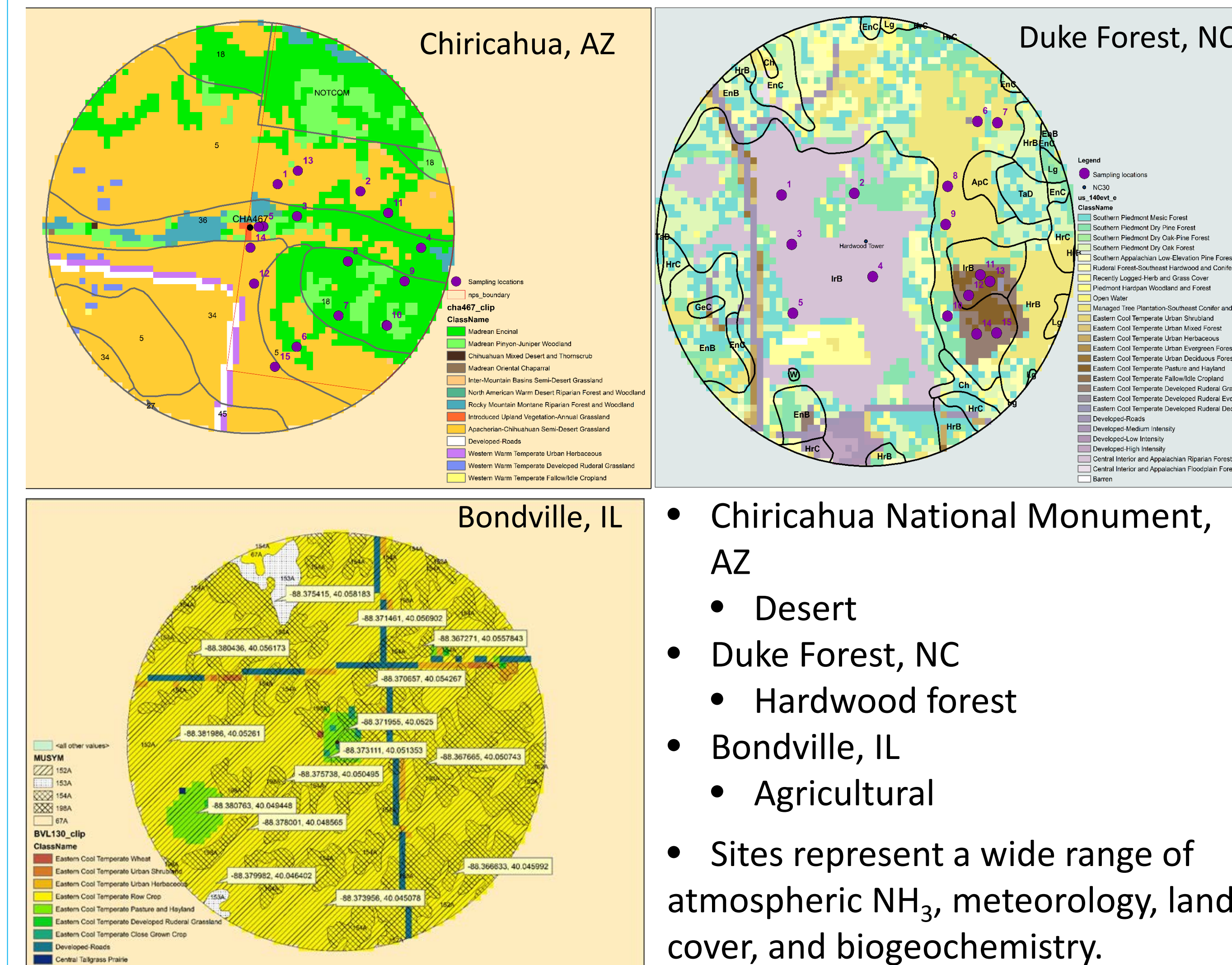
**Abstract:** Reduced inorganic nitrogen ( $\text{NH}_3 + \text{NH}_4^+$ ) is an increasingly important contributor to the total nitrogen deposition budget, yet the bi-directional nature of  $\text{NH}_3$  air-surface exchange makes incorporation of  $\text{NH}_3$  measurements into dry deposition schemes in field-scale and regional chemical transport models (CTMs) difficult. The purpose of this study is to develop a methodology for providing NADP with modeled  $\text{NH}_3$  fluxes using bi-weekly AMoN concentrations.  $\text{NH}_3$  fluxes derived from site specific  $\text{NH}_3$  measurements (AMoN) and surface parameterizations (i.e., compensation points) can provide “best” estimates of  $\text{NH}_3$  deposition for developing ecosystem specific deposition budgets and assessing sub-grid variability of  $\text{NH}_3$  fluxes in CTMs. This effort will therefore improve the total nitrogen deposition estimates provided by TDEP, which does not currently use the AMoN  $\text{NH}_3$  concentrations for dry deposition estimates.

This project consists of two phases:

- Phase I - Develop a database of soil and vegetation chemistry, micrometeorology, and surface physical characteristics at three pilot AMoN sites
- Phase II - Use datasets to parameterize and test a two-layer bi-directional  $\text{NH}_3$  flux model for implementation at AMoN sites
  - Assess model sensitivities to biogeochemical and meteorological inputs
  - Develop methodologies for use of time-integrated  $\text{NH}_3$  concentrations
  - Standardize model configuration for implementation across AMoN

Phase 1 field measurements began in the summer of 2017 and will continue through spring 2018. This paper describes field and laboratory methods and presents preliminary results of soil chemistry measurements.

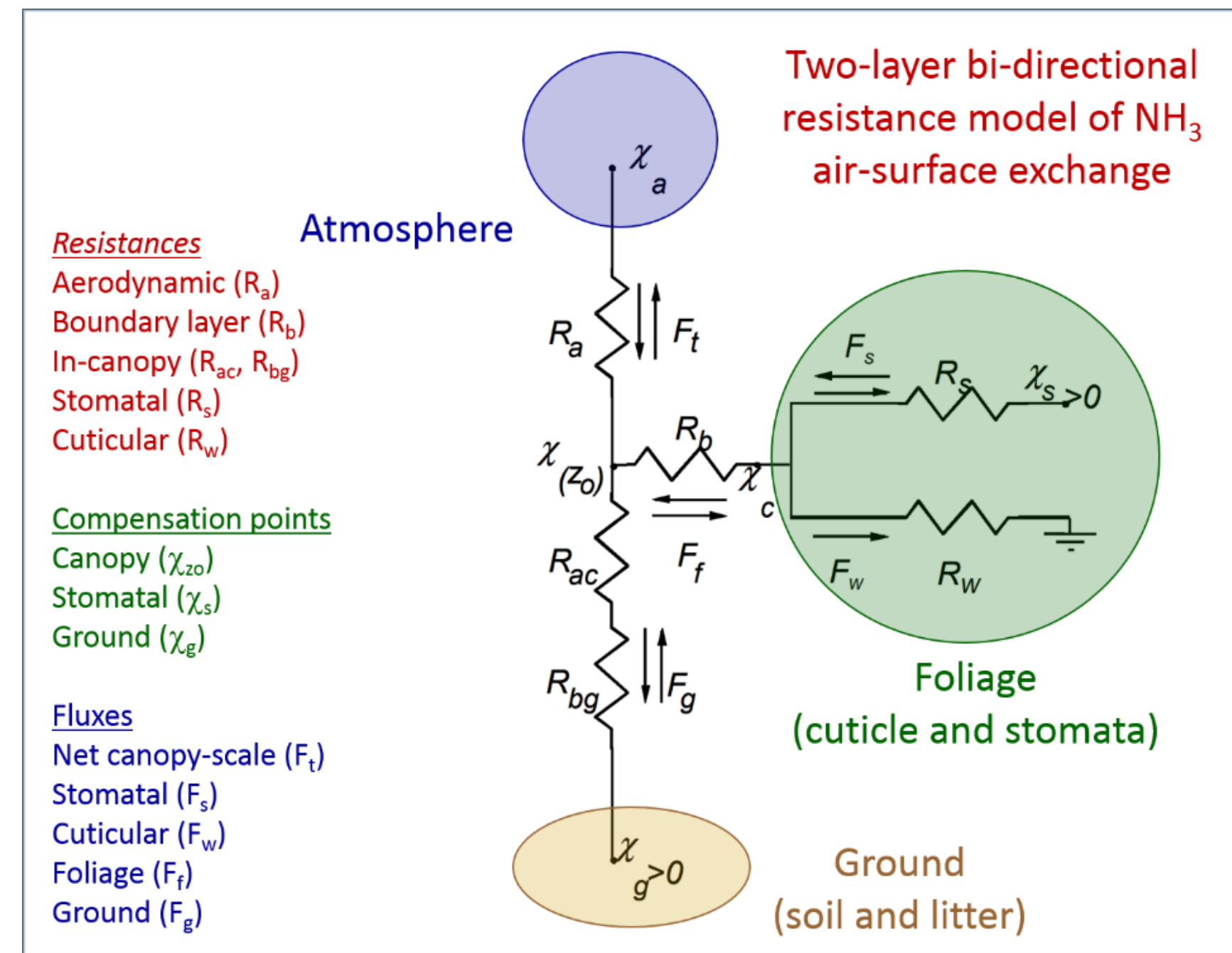
## Sites



**Figure 1:** Sampling locations around the Chiricahua National Monument (AZ98), Duke Forest (NC30), and Bondville (IL11) AMoN sites identified as black dots in the center of a sampling domain of 1 km radius. Maps include overlay of soil (NRCS Web Soil Survey) and vegetation types (LANDFIRE).

- Chiricahua National Monument, AZ
  - Desert
- Duke Forest, NC
  - Hardwood forest
- Bondville, IL
  - Agricultural
- Sites represent a wide range of atmospheric  $\text{NH}_3$ , meteorology, land cover, and biogeochemistry.

## Modeling Framework



**Figure 2:** Two-layer bidirectional  $\text{NH}_3$  flux model (see Massad et al., 2010)

- Modeling approach will follow bidirectional framework described by Massad et al., 2010.

$$\chi = \frac{161500}{T} \exp\left(-\frac{10380}{T}\right) \left(\frac{[\text{NH}_4^+]}{[\text{H}^+]}\right)$$

- Parameterizations will be consistent with CMAQ where possible.

- Soil, live vegetation, and leaf litter will be sampled seasonally to develop estimates of surface  $\text{NH}_3$  emission potentials ( $\Gamma$ ) from which  $\text{NH}_3$  compensation points ( $\chi$ ) will be derived.

## Methods

**Table 1:** Data collected during Phase 1.

Category	Parameter	Frequency
Micrometeorology	3D wind, solar radiation, RH, surface wetness, precipitation	Continuous; reported as hourly
Soil conditions	Moisture, temperature	Continuous; reported as hourly
Soil chemistry	$[\text{NH}_4^+]$ , $[\text{NO}_3^-]$ , pH	Seasonal, dominant soil types
Vegetation structure	LAI	Seasonal
Vegetation chemistry	Bulk leaf and litter analyzed for moisture, pH, soluble total [N], $[\text{NH}_4^+]$	Seasonal, litter and dominant vegetation
Atmospheric chemistry (CASTNET; NADP/AMoN)	$\text{NH}_4^+$ , $\text{NH}_3$ , $\text{HNO}_3$ , $\text{NO}_3^-$ , $\text{SO}_2$ , $\text{SO}_4^{2-}$ , $\text{Ca}^{2+}$ , $\text{Mg}^{2+}$ , $\text{Na}^+$ , $\text{K}^+$ , $\text{Cl}^-$	Weekly (CASTNET); bi-weekly (AMoN)
Wet deposition (NADP/NTN)	pH, specific conductance, $\text{NO}_3^-$ , $\text{NH}_4^+$ , $\text{SO}_4^{2-}$ , $\text{Ca}^{2+}$ , $\text{Mg}^{2+}$ , $\text{Na}^+$ , $\text{K}^+$ , $\text{Cl}^-$	Weekly

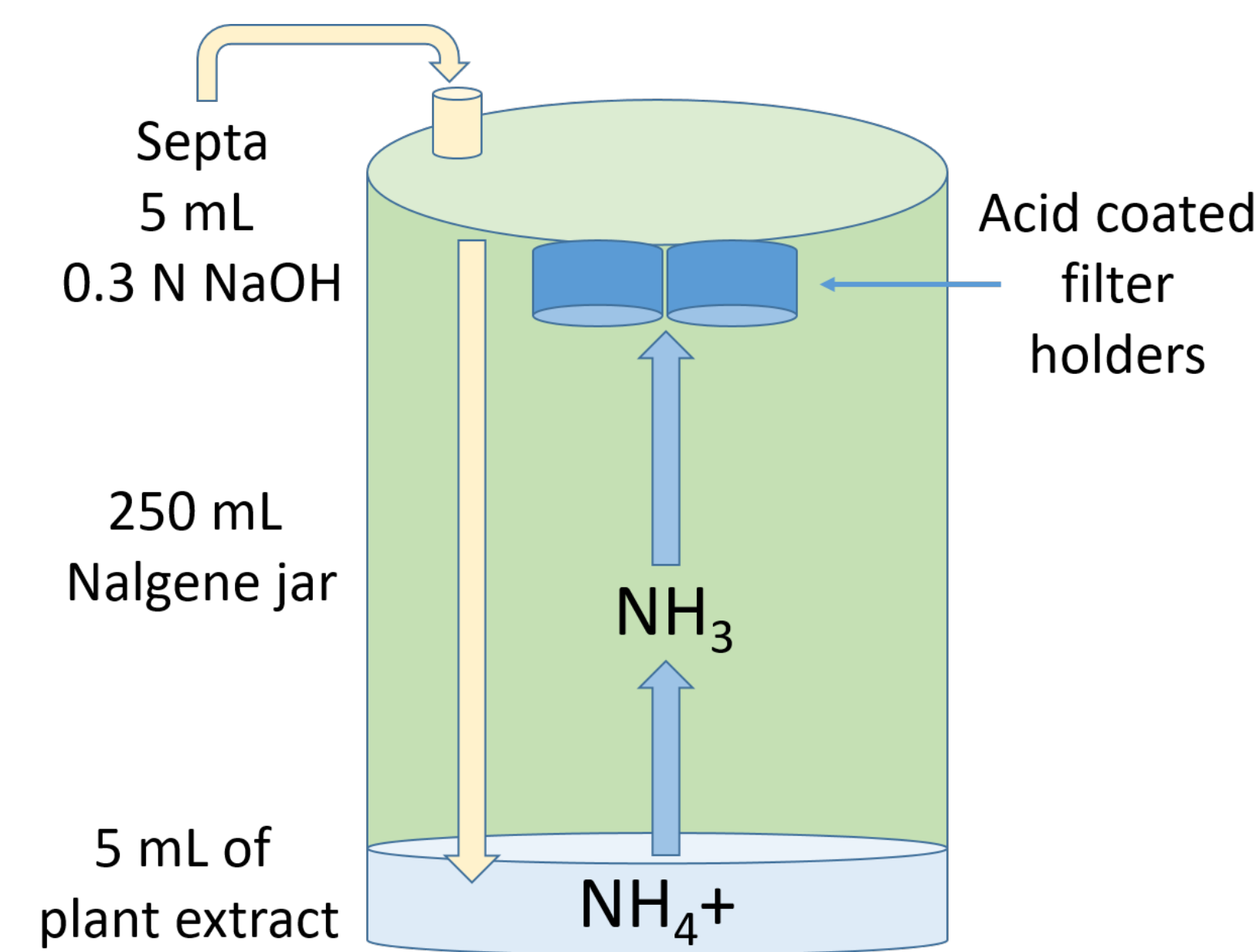
**Soils:** Two depths, N extractions with 0.01 M  $\text{CaCl}_2$ , pH in 1:1 water and 1:2  $\text{CaCl}_2$

- Measurements of  $\text{NH}_4^+$  and  $\text{H}^+$  used to calculate  $\Gamma_{\text{soil}}$

**Litter and Vegetation:** Samples

ground in liquid N, extracted in water

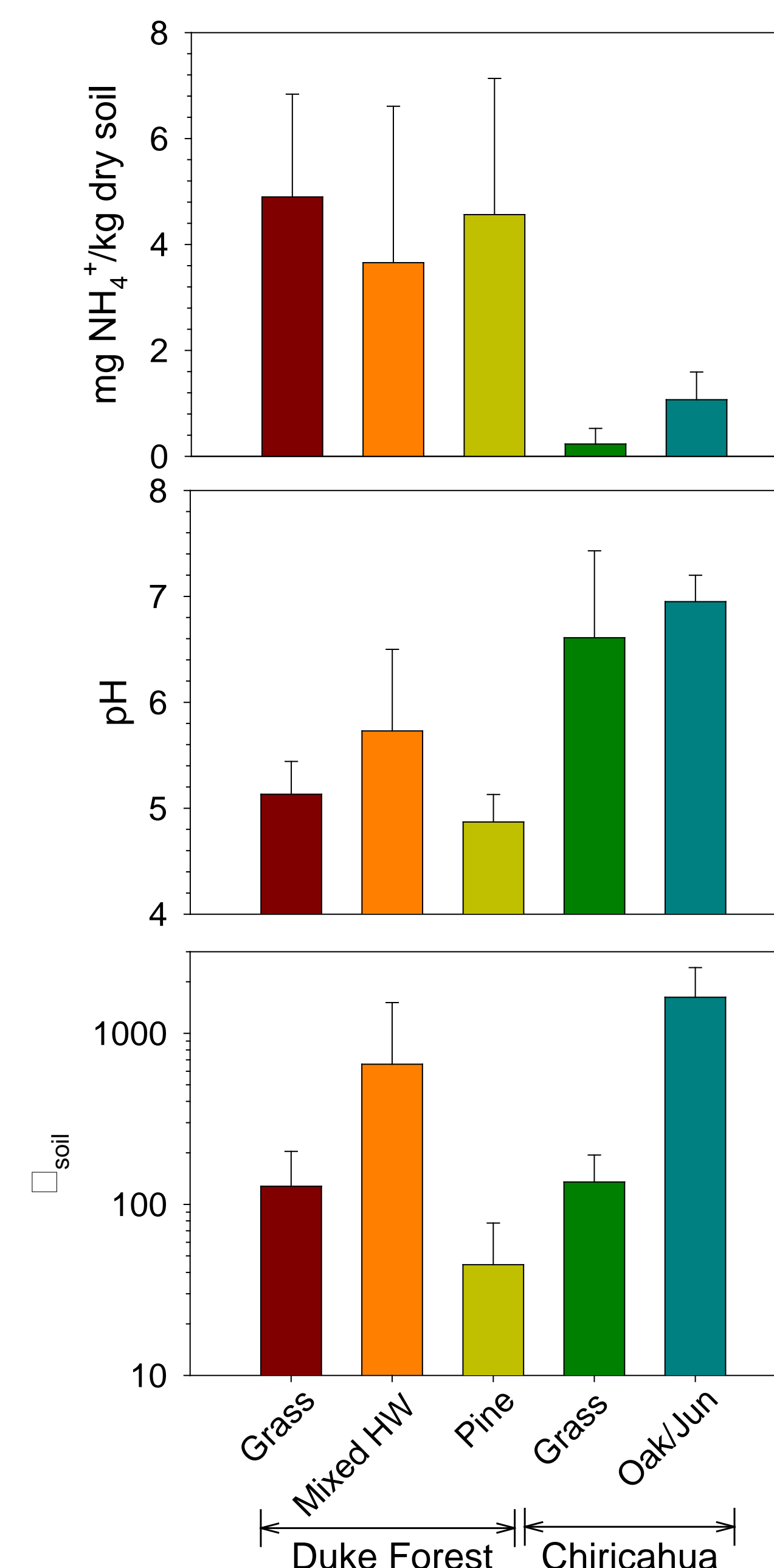
- Measurements of bulk soluble  $\text{NH}_4^+$  and  $\text{H}^+$  used to calculate  $\Gamma_{\text{stomatal}}$ ,  $\Gamma_{\text{litter}}$
- Mass of  $\text{NH}_4^+$  collected from extracts using sealed headspace diffusion technique (recovery > 98%)



**Figure 3:** Schematic of sealed headspace diffusion sampler for collection of  $\text{NH}_4^+$  from vegetation extracts.

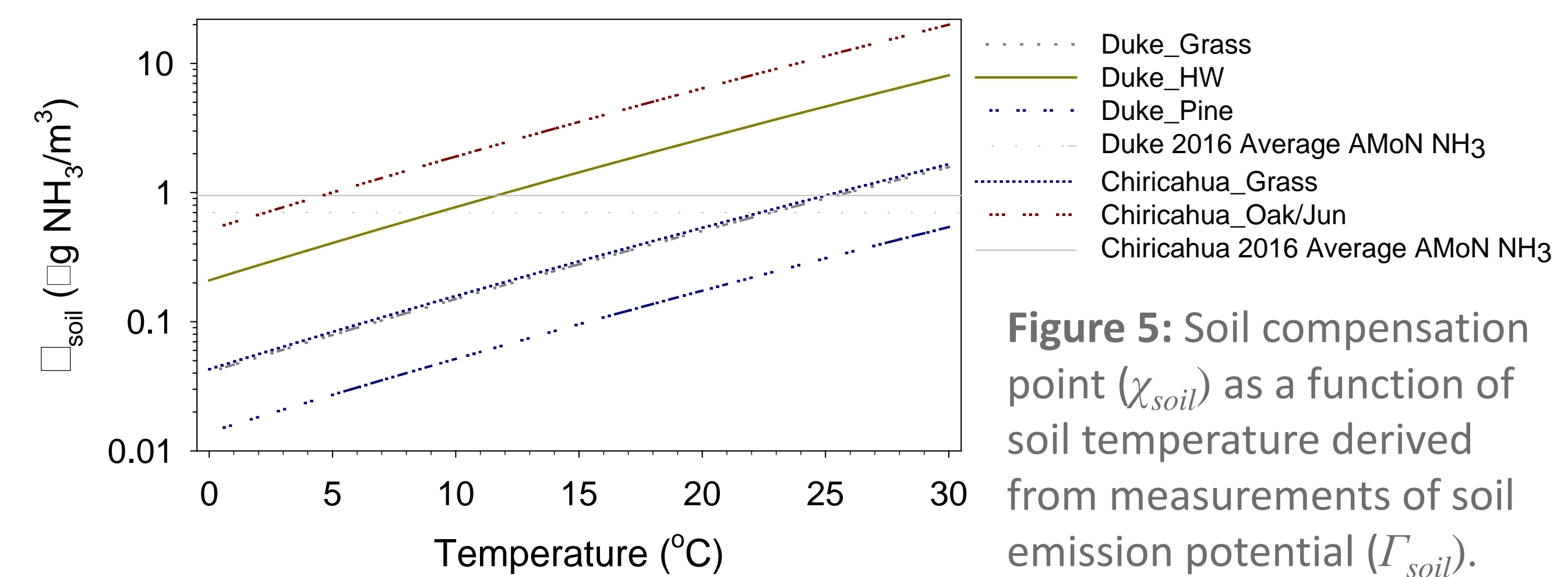
## Preliminary Results

*Soil chemistry – summer 2017*



- Concentrations of soil extractable  $\text{NH}_4^+$  much higher at forest site compared to desert site
- Higher pH observed in desert soils
- Lowest pH at Duke Forest observed in pine areas
- Variability in  $\text{NH}_4^+$  and pH result in large inter- and intra-site differences in soil emission potential ( $\Gamma_{\text{soil}}$ ).

**Figure 4:** Summer 2017 soil chemistry ( $\text{NH}_4^+$ , pH, and calculated  $\Gamma_{\text{soil}}$ ) at Duke Forest and Chiricahua. Data (mean and standard deviation) are summarized by primary vegetation at soil sampling location. HW= hardwood, Jun = juniper



**Figure 5:** Soil compensation point ( $\chi_{\text{soil}}$ ) as a function of soil temperature derived from measurements of soil emission potential ( $\Gamma_{\text{soil}}$ ).

- Variability in  $\Gamma_{\text{soil}}$  translates to large differences in soil compensation points ( $\chi_{\text{soil}}$ ).

## Next Steps

- Continue seasonal sampling of LAI and biogeochemistry through spring 2018
- Develop Eddy Pro processing routines for sonic data
- Presentation of more complete analysis of soil and vegetation chemistry at spring NADP meeting

**REFERENCES:** Massad, R.S.; Nemitz, E.; Sutton, M.A. (2010) Review and parameterization of bi-directional ammonia exchange between vegetation and the atmosphere. Atmos. Chem. & Phys. (10) 10359-10386.

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