



# Using Extractive FTIR to Measure N<sub>2</sub>O from Medium Heavy-Duty Diesel Vehicles

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## Purpose

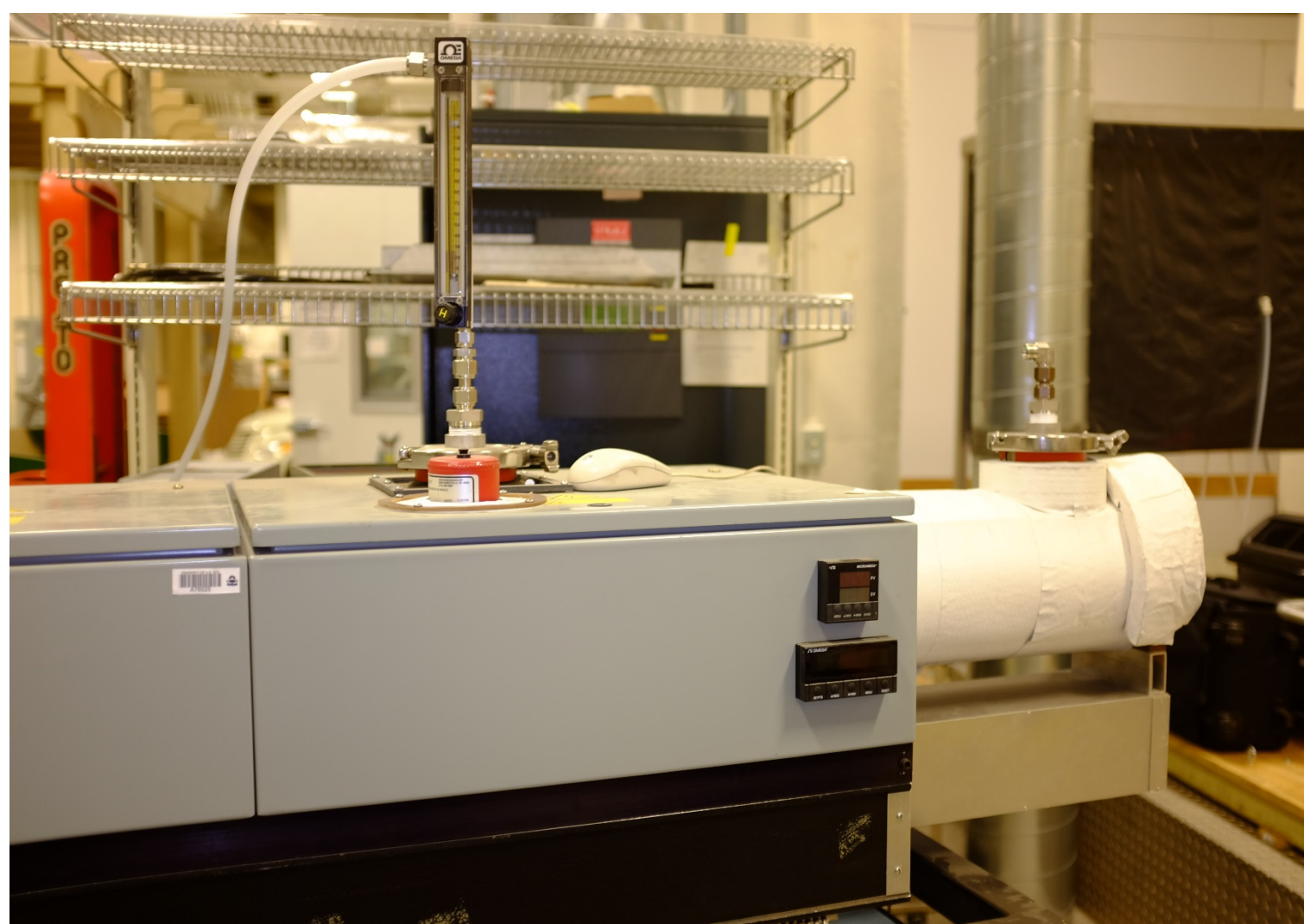
Two vehicles were tested to compare the emission effects of changing from Ultra Low Sulfur Diesel (ULSD). Two of these vehicles were tested on a chassis light-duty dynamometer because the dilution tunnel was equipped to collect samples for speciated SVOCs and VOCs.

Modal data allows emission concentrations to be compared to vehicle system parameters like air-fuel ratio (AFR), percent oxygen, exhaust temperature, and several more on real-time basis. This type of comparison makes it possible to begin to determine root causes of the analyte, a necessary step in determining potential means of controlling the emission rate. The FTIR was evaluated to whether it could produce real time modal data.

In order to verify that the FTIR was correctly time aligned and accurately measuring the diluted exhaust, comparisons were made to 40 CFR 1065 reference methods for CO<sub>2</sub> and NO<sub>x</sub>. It was observed that the FTIR modes of the CO<sub>2</sub> and NO<sub>x</sub> data correlated to those reference methods. This suggests that when a compound such as N<sub>2</sub>O was measured, its mode shape was correct and can be used in future studies comparing the N<sub>2</sub>O to other vehicle parameters. The goal was not to generate emission rates but to determine when the emission rates varied so that a statistical analysis might be applied to better understand root causes for vehicular emissions.

## FTIR

The FTIR system used for this study was an Industrial Monitor and Control Corporation (IMACC) (IMACC, Round Rock, TX) spectrometer equipped with a Micheleson interferometer, a zinc selenide beam splitter, a mercury cadmium telluride detector and a 12L, 1m multi-pass gas cell with gold-coated mirrors and a stainless steel coated body, with a path-length of 78 meters.



## Project Approach

The emissions of two medium heavy-duty diesel trucks, equipped with Diesel Particulate Filters (DPFs), were characterized under a variety of operating conditions as well as environmental conditions. One vehicle used a NO<sub>x</sub> Adsorber Catalyst (NAC) paired with a Diesel Oxidation Catalyst (DOC) and the other vehicle used a Selective Catalytic Reduction (SCR) system for control of nitrogen oxides (NO<sub>x</sub>). Both vehicles were tested with two different fuels [ultra-low sulfur diesel (ULSD) and biodiesel (B20)] and ambient temperatures (70°F and 20°F). Three driving cycles provided emissions estimates under changing operating conditions: (1) a cold start with low transients (CSLT), (2) the federal heavy-duty urban dynamometer driving schedule (HD-UDDS), and (3) a warm start with low transients (WSLT).

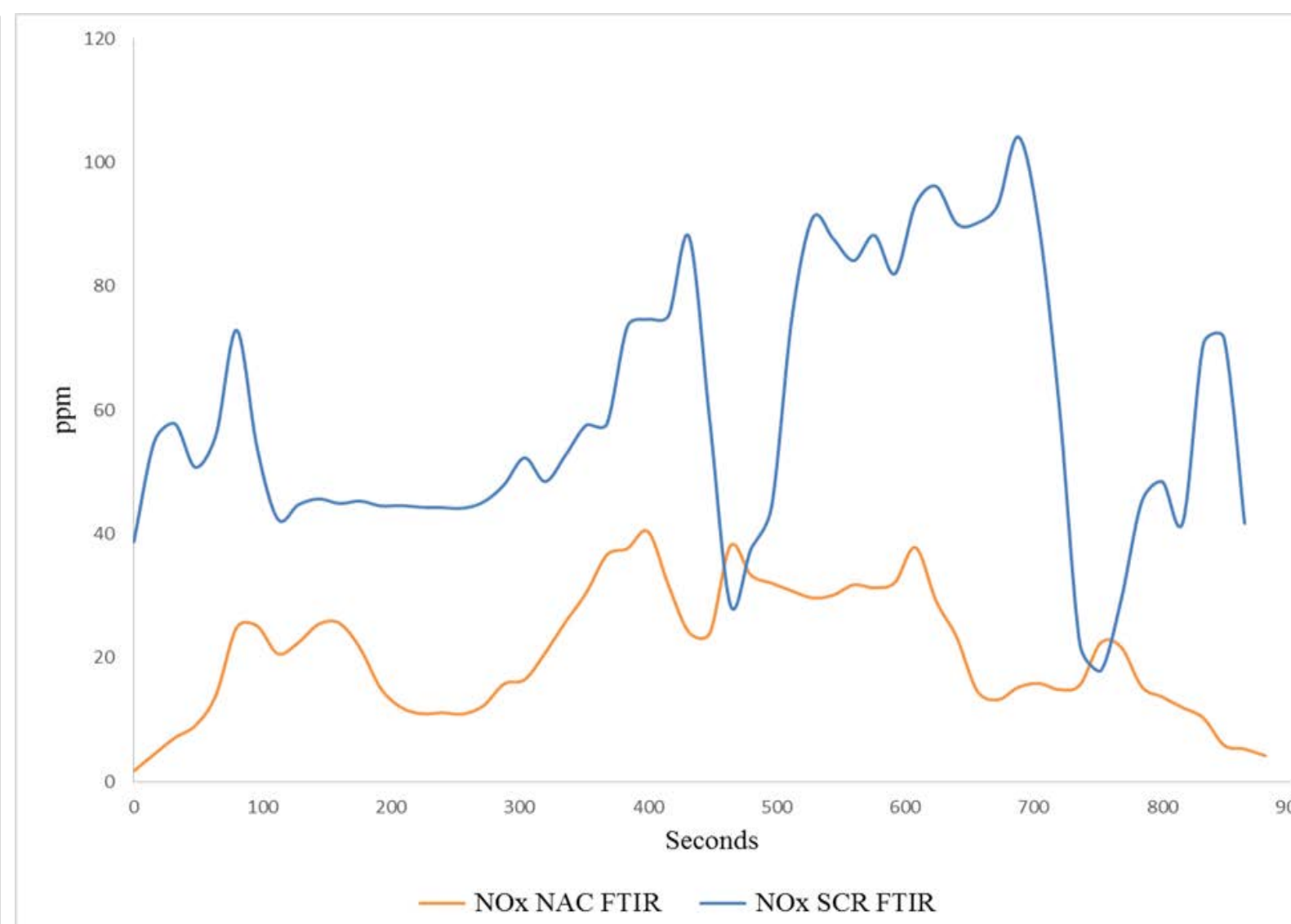
## SCR and NAC Comparisons

Figure 1 and Figure 2 present preliminary data from the study. Figure 1 presents NO<sub>x</sub> comparisons of the SCR and NAC vehicles at 20 °F using ULSD fuel under the HD-UDDS. Figure 2 presents N<sub>2</sub>O comparisons of the SCR and NAC technology for the same test. Both figures have concentration values on the y-axis and time on the x-axis. The changes in concentration values are related to changes in the speeds of the driving cycle. In Figure 1, the NO<sub>x</sub> emissions for the SCR are higher concentrations than those for the NAC. Figure 2 illustrates that the N<sub>2</sub>O values for the NAC were slightly higher, further parameters need to be investigated to explain these results.

Figure 1: NO<sub>x</sub> Comparison SCR and NAC Technology at 20 °F (ULSD Fuel)



Figure 2: N<sub>2</sub>O Comparison of SCR and NAC Technology at 20 °F (ULSD Fuel)



## Modal Comparisons

The modal data from the FTIR measurements of the CO<sub>2</sub> and NO<sub>x</sub> show that the FTIR had sufficient time resolution to suggest which portions of the driving cycle resulted in the highest emissions. That being the case, the FTIR provides valuable insight into the conditions that result in nitrous oxide.

