1	AIR QUALITY IMPACTS OF INCREASED USE OF ETHANOL UNDER THE
2	UNITED STATES' ENERGY INDEPENDENCE AND SECURITY ACT
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21	Submitted to the 2010 AAAR Air Pollution and Health Conference Special Issue
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1 Abstract

2

3 Increased use of ethanol in the United States fuel supply will impact emissions 4 and ambient concentrations of greenhouse gases, "criteria" pollutants for which the U.S. 5 EPA sets ambient air quality standards, and a variety of air toxic compounds. This paper 6 focuses on impacts of increased ethanol use on ozone and air toxics under a potential 7 implementation scenario resulting from mandates in the U.S. Energy Independence and Security Act (EISA) of 2007. The assessment of impacts was done for calendar year 8 9 2022, when 36 billion gallons of renewable fuels must be used. Impacts were assessed 10 relative to a baseline which assumed ethanol volumes mandated by the first renewable 11 fuels standard promulgated by U. S. EPA in early 2007. This assessment addresses both 12 impacts of increased ethanol use on vehicle and other engine emissions, referred to as 13 "downstream" emissions, and "upstream" impacts, i.e., those connected with fuel 14 production and distribution. Air quality modeling was performed for the continental 15 United States using the Community Multi-scale Air Quality Model (CMAQ), version 4.7. 16 Pollutants included in the assessment were ozone, acetaldehyde, ethanol, formaldehyde, 17 acrolein, benzene, and 1,3-butadiene. Results suggest that increased ethanol use due to 18 EISA in 2022 will adversely increase ozone concentrations over much of the U.S., by as 19 much as 1 ppb. However, EISA is projected to improve ozone air quality in a few 20 highly-populated areas that currently have poor air quality. Most of the ozone 21 improvements are due to increases in nitrogen oxides (NO_x) in volatile organic compound 22 (VOC)-limited areas. While there are some localized impacts, the EISA renewable fuel 23 standards have relatively little impact on national average ambient concentrations of most

1	air toxics, although ethanol concentrations increase substantially. Significant
2	uncertainties are associated with all results, due to limitations in available data. These
3	uncertainties are discussed in detail.

5 1. Introduction

6

7 In recent years, use of ethanol as a component of vehicle fuel has been heavily 8 promoted for a variety of reasons, including reduction of reliance on fossil fuel, reduction 9 in emissions that lead to global warming, and reduction in ambient concentrations of a 10 variety of air pollutants. Changes to the U. S. EPA's Renewable Fuel Standard program 11 (RFS2), as mandated by the U.S. Energy Independence and Security Act of 2007, will 12 result in large increases in the amount of ethanol used. Whether increased use of ethanol 13 made from various feedstocks and production pathways results in net increases or 14 decreases in greenhouse gases is currently the subject of considerable scientific debate 15 (Niven, 2005; Crutzen et al., 2008; Wakely et al., 2009). In addition, increased use of 16 ethanol will impact "criteria" pollutants (e.g., ozone) for which the U.S. Environmental 17 Protection Agency (U. S. EPA) has set National Ambient Air Quality Standards 18 (NAAQS), as well as those pollutants referred to as "air toxics" based on their potential 19 to cause adverse cancer and non-cancer health effects. This paper focuses on these two 20 sets of pollutants and how increased use of ethanol could potentially impact their 21 emissions and concentrations in the ambient air. Impacts of ethanol in fuel on vehicle 22 and nonroad engine emissions are addressed, as well as impacts associated with fuel 23 production and distribution. Concerns that increased use of ethanol could lead to higher

- volatile organic compound (VOC) emissions and elevated tropospheric ozone levels will
 be addressed in this paper (Jacobson, 2007; Smith et al., 2009).
- 3

4 Limitations in data present challenges in assessing impacts of increased ethanol 5 use. Most available engine emissions data for ethanol blends are from gasoline highway 6 vehicles using ten percent ethanol by volume (E10) blends with limited data on 85 7 percent ethanol (E85) blends. It has long been known that addition of 10 percent ethanol 8 by volume increases vehicle emissions of acetaldehyde substantially (U. S. EPA, 1993). 9 Most studies show reductions in emissions of carbon monoxide (CO), benzene, and 1,3-10 butadiene (U. S. EPA, 1993; U. S. EPA, 2007a), but there are other studies that do not 11 (Durbin et al., 2007). Inconsistencies across studies can result from differences in vehicle 12 emission control technologies, fuel composition, test cycle, and age of vehicles. Impacts 13 on emissions of VOCs and oxides of nitrogen (NO_x) are clearly dependent on vehicle 14 control technology (U. S. EPA, 2007a). Compared to gasoline without ethanol (E0), data 15 from recent E85 testing consistently show reductions in emissions of benzene and 1,3-16 butadiene, increases in formaldehyde, and very large increases in acetaldehyde (Yanowitz, 17 et al., 2009; Graham et al., 2008; Southwest Research Institute, 2007). Generally, data 18 also show CO reductions but are more equivocal for NO_x and VOC. When evaluating 19 impacts of ethanol on vehicle emissions, it is important to characterize direct emissions of 20 ethanol, which have the potential to cause adverse health effects and can also react in the 21 atmosphere to form acetaldehyde and peroxyacetyl nitrate (PAN). Other emissions 22 resulting from increased levels of ethanol in the fuel can also react in the atmosphere to 23 form ozone, organic aerosols, acetaldehyde and other pollutants. Accurate

characterization of emissions of these pollutants is critical in characterizing air quality
 impacts of increased ethanol use.

3

4 Along with the impacts of increased ethanol use on vehicle and other engine 5 emissions, there are "upstream" impacts, i.e., those connected with fuel production and 6 distribution. These impacts come from changes in agricultural processes, feedstock 7 transportation, and the production and distribution of renewable fuel. These processes 8 occur domestically and internationally. Emissions related to agricultural processes can 9 come from farm equipment, fertilizer production and application, pesticide production 10 and application, burning of crop residue, and fugitive dust from field tilling and related 11 activities. Feedstocks may be transported by rail, marine vessels, or trucks, and the 12 emissions associated with this transport must be estimated. In addition, there are 13 significant emissions associated with ethanol production facilities, as well as from the 14 production of energy used in the operation of these facilities. Key pollutants emitted by 15 these facilities and reported in U. S. EPA's Toxics Release Inventory and National 16 Emissions Inventory (NEI) include CO, particulate matter (PM), NO_x, sulfur oxides 17 (SO_x) , and VOCs. VOCs emitted in the largest quantity from ethanol plants include 18 ethanol and acetaldehyde (Brady and Pratt, 2007). However, displacement of gasoline 19 with ethanol can result in reduced emissions at petroleum refineries. Recent research 20 suggests that emissions associated with the transport and distribution of ethanol and 21 gasoline/ethanol blends may be significant as well (Wakeley et al., 2009). These 22 emissions result from combustion of fuels used in transport and distribution, as well as 23 evaporative and spillage emissions from storage and transfer activities. Since a pipeline

system does not exist for ethanol, increased ethanol use is likely to increase emissions
 associated with distribution of fuels.

3

4 Analyses done to date on impacts of increased ethanol use either model the 5 impacts from extreme cases or do not account for impacts from all sources across the life-6 cycle, or both. For instance, one recent analysis modeled emissions impacts, air quality 7 and potential health risks from changes in downstream sources due to 100 percent 8 penetration of E85 (Jacobson, 2007). While this study accounted for only impacts of 9 changes in vehicle and nonroad equipment emissions, subsequent work (Jacobson, 2009) 10 suggested downstream sources had a larger impact on mortality than upstream sources. 11 However, it should be noted that Jacobson used upstream emissions from Delucchi 12 (2006), which does not account for all upstream emission sources that can be impacted by 13 the increased use of ethanol. Also, while 100 percent penetration of E85 is not realistic, 14 such an extreme case can be used to scale impacts for different E85 penetration levels. 15 16 Qualitative inferences of the likely impacts of increased ethanol use can be 17 obtained from analysis of existing air quality monitoring data in locations where ethanol 18 use has increased. Much early work focused on air quality impacts of increased use of

19 ethanol from sugar cane in Brazil. Brazil is the only country in the world where a

20 nationwide, large-scale alcohol fuel program has been implemented. In 1997,

21 approximately 4 million automobiles ran on neat ethanol and approximately nine million

22 automobiles ran on a 22 percent ethanol-blended gasoline mixture (Grosjean et al.,

23 2002a). However, data from Brazil are limited by the lack of ambient monitoring data

prior to ethanol fuel use. Nonetheless, it is clear from data in Brazil that significant
increases in ambient acetaldehyde and PAN (which can be formed by photooxidation of
acetaldehyde) resulted from greater emissions of acetaldehyde and ethanol associated
with increased ethanol use (Andrade et al., 1998; Montero et al, 2001; Colon et al, 2001;
Grosjean et al., 2002b; Martins and Arbilla, 2003).

6

7 Monitoring and modeling studies of ethanol impacts resulting from U.S. winter 8 oxygenated fuel programs can also provide insights. A study in Albuquerque, New 9 Mexico, where 10 percent ethanol was used in a winter oxygenated fuel program, found 10 elevated levels of acetaldehyde and PAN (Gaffney et al., 1997). However, a similar 11 study in Denver, Colorado, found no elevation of acetaldehyde (Anderson, et al., 1994, 12 1996, 1997). One explanation hypothesized that photochemical production and 13 destruction of these compounds suppressed the emissions effect. Alkenes in particular 14 play a large role in acetaldehyde production (Altshuller 1991a, 1991b, 1993; Calvert et al., 15 2000). Furthermore, a modeling study in California found large increases in ethanol 16 concentration, but relatively small impacts on acetaldehyde with ten percent ethanol use 17 in winter (Allen et al., 1999).

18

19 Given that currently available data are only available from modeling and 20 monitoring studies with very different conditions than those likely to occur in the United 21 States under EISA, an assessment of likely impacts for a realistic scenario is needed. In 22 this paper, we present a comprehensive assessment of potential emission inventory and 23 air quality impacts that could result in 2022 with full implementation of requirements

1	under EISA, considering emission changes in all sources across the fuel lifecycle. This is
2	the first time that a full, 3-D modeling study has been performed using realistic future
3	emission scenarios and a full upstream and downstream inventory. This assessment
4	includes inventories for the following pollutants: VOCs, NOx, CO, particulate matter 10
5	microns in diameter and less (PM_{10}), particulate matter 2.5 microns in diameter and less
6	(PM _{2.5}), sulfur dioxide (SO ₂), ammonia (NH ₃), benzene, 1,3-butadiene, acrolein,
7	formaldehyde, acetaldehyde, and ethanol. Air quality modeling results are presented for
8	ozone and air toxics. While $PM_{2.5}$ concentrations will also be impacted, and these
9	impacts have been modeled (U. S. EPA, 2010a), air quality modeling results for this
10	pollutant are not discussed in this paper because of limitations in local-scale results,
11	resulting from an error in spatially allocating emissions from one PM _{2.5} emission source.
12	We compared modeled emissions and air quality under EISA to a base case scenario
13	which assumed renewable fuel volumes at the level mandated by U. S. EPA's 2007
14	Renewable Fuel Standard (RFS1; U. S. DOE, 2007; U. S. EPA, 2007b).
15	
16	2. Methodology
17	
18	Modeling was done for calendar year 2022, for two scenarios. The first was a
19	baseline scenario which assumed a volume of 7.5 billion gallons of renewable fuels, as
20	mandated by the U. S. EPA's 2007 renewable fuel standard (RFS1) rule (U. S. EPA,
21	2007b). The second scenario included impacts of EISA, which was implemented by the
22	RFS2 rule (U. S. EPA, 2010a). Based on EISA requirements and an U. S. EPA
23	assessment of production feasibility (U. S. EPA, 2009a), this scenario assumed that the

1	renewable fuel mandate of 36 billion gallons would be reached with 15 billion gallons of
2	corn ethanol, 16 billion gallons of cellulosic ethanol, and 3 billion gallons of imported
3	ethanol. The remainder of required renewable fuel volumes would be met with
4	renewable diesel and biodiesel fuel. Approximately 20 billion gallons of the ethanol will
5	be in the form of E85. U. S. EPA's final RFS2 rule implementing requirements of EISA
6	assumed that requirements would be met with lower volumes of cellulosic ethanol and
7	higher volumes of cellulosic diesel fuel (U. S. EPA, 2010a).
8	
9	2.1. Emission Inventory
10	
11	State-level light-duty highway vehicle exhaust emissions for all pollutants except
12	ethanol were estimated with a draft version of the Motor Vehicle Emissions Simulator
13	(MOVES) model (U. S. EPA, 2009b). MOVES contains significant updates to U. S.
14	EPA's previous highway vehicle emissions model, MOBILE6. Fuel properties used as
15	model inputs were based on data from refinery modeling to account for how gasoline
16	properties would be likely to change with increased use of ethanol. In modeling impacts
17	of RFS2, ethanol was assumed to be blended at either 10 percent (E10) or 85 percent
18	(E85) with gasoline, with no intermediate level blends. It must be emphasized that there
19	were significant limitations in emissions data quantifying the effects of E10 on advanced
20	technology vehicle emissions and the effects of E85 on all vehicles (U. S. EPA, 2010a).
21	U. S. EPA relied on unpublished data from a test program then in progress to quantify
22	E10 impacts; these data are discussed in U. S. EPA (2010a; pp. 520-521). Based on
23	these limited data, E10 use was assumed to decrease VOC emission rates by 7-10% and

1	increase NOx emissions by over 7%. However, some recent test data were not consistent
2	with these assumptions, and more data are needed to draw firm conclusions. Furthermore,
3	vehicles using E85 were assumed to have lower emissions of PM and CO, but this
4	assumption is also based on limited data (Southwest Research Institute, 2007;
5	Environment Canada, 2007; Graham et al., 2008). No reductions in VOC or NO_x
6	emissions were assumed for E85. Two studies (Environment Canada, 2007; Graham et
7	al., 2008) showed significant NO_x reductions with E85, but data from EPA's Certification
8	and Fuel Economy System showed a small impact. Overall, effects were not statistically
9	significant due to a large amount of variability in the data (U. S. EPA, 2010a). In
10	contrast to this modeling effort, Jacobson (2007) assumed significant NO_x effects with
11	E85. Also, E85 emissions were estimated in MOVES by applying multiplicative
12	adjustments to E0 emissions. These multiplicative adjustments did not vary with
13	temperature. Thus, emissions of vehicles running on E85 were assumed to change
14	proportionally to emissions of vehicles running on E0 in response to temperature. State-
15	level emissions from MOVES were allocated to counties using data from U. S. EPA's
16	National Mobile Inventory Model, v.2008 (NMIM2008; U. S. EPA, 2009c). A cost
17	effectiveness analysis was done to determine where E85 profit margins would be highest,
18	and higher levels of E85 use were assumed to occur in these areas. These areas of higher
19	E85 use include New England, much of the Midwest, the Pacific Northwest, and
20	Southern California (U. S. EPA, 2009a).
21	
22	Light-duty vehicle evaporative, heavy-duty vehicle, motorcycle, and nonroad

23 equipment emissions, except for ethanol, were estimated at the county level using

1	NMIM2008 (U. S. EPA, 2009c). Stationary source emissions were obtained by
2	procedures describe in the U. S. EPA Office of Air Quality Planning and Standards, 2005
3	Modeling Platform documentation (http://www.epa.gov/ttn/chief/emch/index.html). In
4	this platform, year 2022 projections were made from a year 2005 baseline. Impacts of
5	RFS2 on stationary source emissions, except for ethanol, were estimated using a U.S.
6	EPA spreadsheet model based on the GREET (Greenhouse Gases, Regulated Emissions,
7	and Energy Use in Transportation) model, originally developed by the U. S. Department
8	of Energy's Argonne National Laboratory (U. S. DOE, 2009, U. S. EPA, 2010a).
9	Improvements to GREET included a significant expansion of modeling agricultural
10	sector impacts, employing economic and agriculture models to consider factors such as
11	land-use impact, agricultural burning, fertilizer, pesticide use, livestock, crop allocation,
12	and crop exports. Other updates and enhancements included updated emission factors for
13	fuel and feedstock transport. In addition, the most recent version of GREET did not
14	include air toxics; thus, emission factors for benzene, 1,3-butadiene, acrolein,
15	formaldehyde, and acetaldehyde were added.
16	
17	2.2. Air Quality Modeling
18	
19	Air quality modeling was conducted for the 48 contiguous states, using the
20	Community Multi-scale Air Quality Model (CMAQ), version 4.7, with the Carbon Bond
21	05 (CB05) chemical mechanism (Byun and Schere, 2006, <u>http://www.cmaq-model.org</u>).
22	The modeling domain was made up of a large continental U.S. 36 kilometer (km) grid
23	and two 12 km grids (an Eastern U.S. and a Western U.S. domain), as shown in Figure 1.

1	The conditions from the 36 km coarse grid modeling were used as the initial/boundary
2	state for subsequent 12 km finer grid modeling. Since 12 km domains overlap, data from
3	the 12-km eastern domain data were used for all states from North Dakota to Texas
4	(except El Paso, TX) and east and data from the the 12-km western domain were used for
5	all states from Montana to New Mexico along with El Paso, TX, and west. A 12 km grid
6	size rather than a coarser grid was used for modeling, since finer grids are able to capture
7	variations in both VOC and NOx emissions and their effects on nonlinearities in ozone
8	photochemistry (Schere, 1988, Cohan et al., 2006). The modeling domain contained 14
9	vertical layers with the top of the modeling domain at about 16 km. The hourly results
10	were processed into daily maximum 8-hour ozone concentrations over the ozone season
11	(June-September), as well as seasonal and annual averages for air toxics.
12	
13	Emission inventories were prepared for input into CMAQ using the SMOKE
14	(Sparse Matrix Operator Kernel Emissions) emissions processing system
15	(http://www.smoke-model.org). SMOKE allocates emissions spatially and temporally,
16	using spatial allocation surrogates to allocate area and mobile source emissions to grid
17	cells and temporal activity profiles to allocate emissions to hour of day. Depending on
18	the source, spatial surrogates are based on types of land use, census data, employment
19	information, or other factors. SMOKE also speciates hydrocarbon and particulate matter
20	emissions into individual chemical compounds using a comprehensive set of speciation
21	profiles. Ethanol emissions were estimated in this speciation step.

1	While in general, county-level emissions of area and mobile sources were
2	spatially allocated to grid cells using surrogates from the SMOKE model, there were
3	some exceptions. Since increased use of ethanol results in higher emissions associated
4	with production and distribution, approaches had to be developed to allocate these
5	increased emissions to locations where plants were likely to be located. Increased corn
6	ethanol production was expected to occur at existing or planned plants, but likely
7	locations for cellulosic ethanol plants and associated activity were unknown. Emissions
8	from these plants were treated as area source emissions and allocated to counties based
9	on potential cellulosic feedstock sites across the U.S. that could economically justify the
10	construction of a cellulosic plant facility (U. S. EPA, 2010a). Emissions associated with
11	distribution were allocated based on an analysis by Oak Ridge National Research
12	Laboratory of projected ethanol transport modes, distances, and volumes transferred
13	under various ethanol volume scenarios (U. S. EPA, 2010a).
14	
15	The CMAQ simulations were done for the entire year of 2005, using
16	meteorological inputs developed by U. S. EPA. The meteorological inputs for air quality
17	modeling were derived from an offline simulation of the Pennsylvania State
18	University/National Center for Atmospheric Research Mesoscale Model, MM5 (Grell et
19	al., 1994). MM5 uses a limited-area, nonhydrostatic, terrain-following coordinate system
20	that solves the full set of physical and thermodynamic equations which govern
21	atmospheric motions. In this application, 34 vertical layers were used and the three-
22	dimensional analyses of temperature, moisture and wind field were nudged towards
23	observations. The simulated meteorological fields were found to closely match observed

synoptic patters and to compare well to observed values of meteorological parameters (U.
S. EPA, 2010b). The meteorological outputs from MM5 were processed to create modelready inputs for CMAQ using the Meteorology-Chemistry Interface Processor (MCIP)
version 3.4 (U. S. EPA, 1999). Boundary and initial species concentrations were
provided by a three-dimensional global atmospheric chemistry model, the GEOS-CHEM
model, run for 2005 with a grid resolution of 2.0 x 2.5 degrees (latitude-longitude) and 30
vertical layers (Le Sager et al., 2008).

8

9 In addition to modeling 2022 scenarios, modeling was done for calendar year 10 2005, for purposes of conducting an operational model performance evaluation using 11 2005 air quality monitoring data. An extensive discussion of this performance evaluation 12 can be found in U. S. EPA (2010b). The comparison included statistical assessment of 13 modeled versus monitored data paired in space and time. Results of this evaluation 14 indicated that performance was similar to that found in other recent U.S. EPA 15 applications, indicating that application of CMAQ using this modeling platform provides 16 a scientifically credible approach for assessing the impacts of future emissions changes 17 on air quality. However model performance for ozone, benzene, formaldehyde and 18 acetaldehyde was better than performance for 1,3-butadiene and acrolein, where there 19 was large bias and error. For butadiene, the approximate bias at the national level was 20 about -65% and the error was about 85%. For acrolein, the bias was about -90% with a 21 100% error. The bias and error may be due to several causes; chief among them are 22 emissions errors and ambient measurement errors. As a result, findings for 1,3-butadiene 23 and acrolein should be viewed as more uncertain.

2 **3. Results**

3

4 3.1. Nationwide Inventory Impacts

5

6 Table 1 presents 48 state emission inventories assuming renewable fuel volumes 7 mandated by RFS1 versus RFS2 in 2022, and the percent change in these emissions 8 under RFS2 relative to RFS1. Emissions are broken down into contributions from 9 upstream and downstream sources in the supplementary information, Table S1. Carbon 10 monoxide, benzene, and 1,3-butadiene are projected to decrease in 2022 with RFS2 11 renewable fuel volumes, while NOx, hydrocarbons (HC) and the other air toxics, 12 especially acetaldehyde, are projected to increase due to the impacts of ethanol. The 13 large estimated decrease in CO is due primarily to impacts of ethanol on exhaust 14 emissions from vehicles running on E10 and E85 and nonroad equipment running on E10. 15 The decrease in benzene is due primarily to displacement of gasoline with ethanol in the 16 fuel pool. Increases in NO_x result from a combination of increases in vehicle and 17 nonroad equipment emissions due to higher volumes of E10 gasoline, and increased 18 emissions from corn and cellulosic ethanol production. Small projected increases in 19 hydrocarbons are due to increases in vehicle and nonroad equipment emissions running 20 on E10, as well as emissions connected with renewable fuel production (countered by 21 decreases in emissions associated with gasoline production and distribution as ethanol 22 displaces some of the gasoline). Increases in SOx emissions are due to increases in 23 agricultural chemical production and transport, while substantial PM increases are also

1	associated with fugitive dust from agricultural operations. Ammonia emissions are
2	expected to increase slightly due to increased ammonia from fertilizer use. While PM
3	emissions from vehicles decrease as a result of increased E85 volumes, there are
4	localized increases associated with fugitive dust from agricultural operations, as well as
5	ethanol production and transport. Emissions of ethanol and acetaldehyde increase
6	substantially as a result of emissions from vehicles and engines and increased emissions
7	from ethanol production and transport.
8	
9	3.2. Air Quality Impacts
10	
11	3.2.1. Ozone
12	
13	Figure 2 presents changes in 8-hour ozone design values in 2022 between the
14	EISA scenario and the RFS1 scenario. An ozone design value is defined as the three-
15	year average of the annual, fourth-highest daily maximum 8-hour average ozone
16	concentration at a given monitor (Code of Federal Regulations, 40 CFR, Part 50,
17	Appendix P). To predict 8-hour ozone design values the CMAQ predictions were used in
18	a relative sense by combining the 2005 base year predictions with predictions from the
19	2022 scenarios and applying these modeled ratios (2022/2005) to existing ambient air
20	quality observations. To account for interannual variability in observed ozone levels, the
21	ambient air quality observations used here were an average of the three ozone design
22	values centered around the 2005 base year (i.e., 2003-2005, 2004-2006, 2005-2007).
23	Using this standard air quality modeling methodology (U. S. EPA, 2007c), it is projected

1	that achieving the required renewable fuel volumes by 2022 would adversely impact
2	ozone air quality over much of the U.S. However, ozone air quality improvements are
3	projected in a few highly-populated areas which currently have poor air quality.
4	
5	As can be seen in Figure 2, most of the counties where impact on design value
6	was calculated, especially those in the Midwest, see increases in their ozone design
7	values. The majority of these design value increases are more than 0.2 ppb but less than
8	0.5 ppb. However, there are some counties that will see 8-hour ozone design value
9	increases above 0.5 ppb; these counties are mainly in the eastern U.S. There are also
10	some counties that are projected to see 8-hour ozone design value decreases. The
11	counties with ozone design value decreases greater than 0.5 ppb are in Southern
12	California.
13	
14	Table 2 shows the average change in 2022 8-hour ozone design values for: (1) all
15	counties with 2005 baseline design values, (2) counties with 2005 baseline design values
16	that exceeded the 2008 ozone standard, (3) counties with 2005 baseline design values that
17	did not exceed the 2008 standard but were within ten percent of it, (4) counties with
18	2022 design values that exceeded the 2008 ozone standard, and (5) counties with 2022
19	design values that did not exceed the standard but were within ten percent of it. Many of
20	these statistics show an increase in ozone design values in 2022, but the magnitude of the
21	increase varies and there are some statistics which show a decrease in 8-hour ozone
22	design values as a result of increased ethanol usage.
22	

1	The projected ozone increases in some areas and decreases in other areas are
2	likely a result of the emissions changes due to the increased volumes of renewable fuels
3	combined with the photochemistry involved, differing emission rates of VOCs and NOx
4	in different areas of the country, and the meteorological conditions in different areas of
5	the country. When VOC levels are high relative to NOx (e.g., in areas with large
6	quantities of biogenic VOC emissions), ozone is reduced by NOx reductions and
7	increased by NOx increases. Such conditions are called "NOx-limited" or "NOx-
8	sensitive." The ozone increases seen in the southeastern U.S. and many of the other rural
9	areas are likely due to the fact that those areas are NOx-limited and the changes being
10	modeled are projected to increase NOx. However, when NOx levels are relatively high
11	and VOC levels relatively low, increases in NOx can decrease ozone, because NO reacts
12	directly with ozone and NO ₂ terminates radicals, forming nitric acid, which is removed
13	from the system or can form particulate nitrates. Such conditions are called "VOC-
14	limited" or "NO _x -saturated." Under these conditions, VOC reductions are effective in
15	reducing ozone, but NOx reductions can actually increase local ozone under certain
16	circumstances. The ozone decreases seen in southern California and some of the other
17	urban areas, like Cleveland and Miami, are likely due to the fact that these are VOC-
18	limited areas and they are projected to experience increases in NOx due to RFS2
19	requirements, both of which would result in ozone decreases. While RFS2 is projected to
20	lower ozone levels in southern California under our current assumptions of 2022
21	conditions, it is important to recognize that the State has plans (CARB, 2007) to
22	substantially lower NOx emissions in the future which may bring this area into a NOx-
23	limited regime. In that context, the NOx increases here run counter to State efforts and

1 may lead to ozone increases if the planned NOx reductions are achieved sometime in the
2 future.

3

4 Incremental impacts of emissions from vehicles running on E85 fuel in the RFS2 5 scenario can be made by comparing RFS2 results to an alternative baseline scenario 6 which consists largely of E10. Results of such a comparison are presented in the 7 supplementary information. The alternative baseline scenario assumed ethanol volumes 8 projected by the Department of Energy's Annual Energy Outlook (AEO) for 2007 9 (Energy Information Administration, 2007). Ethanol volumes for this alternative baseline 10 scenario relative to the RFS1 and RFS2 cases are provided in Table S2, and inventory 11 impacts in Table S3. Table S4 compares changes in 8-hour ozone design values for 12 RFS2 relative to the two baselines, and Figure S-1 depicts changes in design values for 13 RFS2 relative to the alternative baseline. These results suggest that incremental impact 14 of E85 use is to increase ozone design values with a similar spatial pattern of increases 15 and decreases as the comparison of RFS2 results to the RFS1 baseline. 16 17 3.2.2. Air Toxics 18

Results of this assessment indicate that, while there are some localized impacts,
the renewable fuel volumes required by RFS2 have relatively little impact on national
average ambient concentrations of the modeled air toxics. An exception is increased
ambient concentrations of ethanol.

1	The relatively small impact was particularly surprising for acetaldehyde, given the
2	increase in acetaldehyde emissions in Table 1. Annual percent changes in ambient
3	concentrations of acetaldehyde are less than one percent for most of the country (Figure
4	3). Several urban areas show decreases in ambient acetaldehyde concentrations ranging
5	from 1-10%, and some rural areas associated with new ethanol plants show increases in
6	ambient acetaldehyde concentrations ranging from 1-10% with the fuel volumes required
7	by RFS2. Annual absolute changes in ambient concentrations of acetaldehyde are
8	generally less than 0.1 μ g/m ³ . The largest increases in ambient acetaldehyde
9	concentrations with RFS2 volumes occur in areas associated with new ethanol plants.
10	This result is due to an increase in emissions of primary acetaldehyde and precursor
11	emissions from ethanol plants not included in the RFS1 baseline scenario. The relatively
12	small overall impact on acetaldehyde concentrations is likely attributable to the
13	complexity of the chemical formation of acetaldehyde. Most ambient acetaldehyde is
14	formed from secondary photochemical reactions of numerous precursor compounds, and
15	many photochemical mechanisms are responsible for this process. As discussed
16	previously, reductions in certain acetaldehyde precursors, primarily alkenes, may offset
17	the impacts of increases in direct emissions, and the acetaldehyde produced from
18	photochemical transformation of increased ethanol emissions. Analysis of data inputs
19	used in this assessment revealed a significant decrease in alkene levels estimated from
20	speciated emission profiles for ten percent ethanol gasoline versus non-ethanol gasoline;
21	these speciation profiles, which are used to break total hydrocarbons down to individual
22	constituent compounds, were applied to vapor losses associated with storage and
23	transport of gasoline. However, the profiles used may not be representative of real-world

1	conditions (U. S. EPA 2010a). A limited sensitivity analysis with more representative
2	profiles developed after modeling was completed, using newly collected data, suggested
3	acetaldehyde increases in many areas where decreases were projected.
4	
5	Modeling results do not show substantial impacts on ambient concentrations of
6	formaldehyde. As shown in Figure 4, most of the U.S. experiences a one percent or less
7	change. Decreases in ambient formaldehyde concentrations range between one and five
8	percent in a few urban areas. Increases range between 1-2.5% in some rural areas
9	associated with new ethanol plants; this result is due to increases in emissions of primary
10	formaldehyde and formaldehyde precursors from the new ethanol plants.
11	
12	Modeling results suggest that the renewable fuel volumes required by the RFS2
13	rule will lead to small nationwide decreases in ambient benzene concentrations. As
14	shown in Figure 5, decreases in ambient benzene concentrations range between 1-10%
15	across most of the country and can be higher in a few urban areas. Absolute changes in
16	ambient concentrations of benzene are all less than 0.2 μ g/m ³ . The modeling also shows
17	small increases and decreases in ambient concentrations of 1,3-butadiene in parts of the
18	U.S (Figure 6). Generally, decreases occur in some southern areas of the country and
19	increases occur in some northern areas and areas with high altitudes. Changes in absolute
20	concentrations of ambient 1,3-but adiene are generally less than 0.005 $\mu g/m^3$. The largest
21	1,3-butadiene increases appear in rural areas with cold winters and low ambient levels but
22	high contributions of emissions from snowmobiles. However, the snowmobile emission
23	factor data that were used to develop inventories were obtained from only three engines

1	and showed significant increases in 1,3-butadiene emissions with ten percent ethanol
2	gasoline; it is likely this increase is not representative of the fleet as a whole. Small
3	increases and decreases are also seen for acrolein (Figure 7). Decreases in acrolein
4	concentrations occur in some eastern and southern parts of the U.S., and increases occur
5	in some northern areas (due to 1,3-butadiene emissions from snowmobiles, because 1,3-
6	butadiene is an acrolein precursor) and areas associated with new ethanol plants.
7	
8	As expected, the renewable fuel volumes required by RFS2 will lead to
9	significant nationwide increases in ambient ethanol concentrations. Increases ranging
10	between 10 to 50 percent are seen across most of the country (Figure 8). The largest
11	increases (more than 100 percent) occur in urban areas with high amounts of onroad
12	emissions and in rural areas associated with new ethanol plants. Absolute increases in
13	ambient ethanol concentrations are above 1.0 ppb in some urban areas. However,
14	sensitivity tests using alternative profiles show that ethanol impacts could be reduced by
15	more than ten percent across much of the modeling domain (U. S. EPA, 2010b).
16	
17	4. Summary and Discussion
18	
19	While this assessment models realistic conditions which could result from
20	implementation of renewable fuel standards required by RFS2, actual levels of renewable
21	fuels are likely to be different from those modeled. Because increased use of ethanol
22	impacts emissions at many different points in the fuel life-cycle, the direction and
23	magnitude of air quality impacts for ozone and air toxics with the renewable fuel mandate

1 of the RFS2 rule will likely vary across the country. Results of this study support 2 concerns that increased use of ethanol may adversely impact ozone in many parts of the 3 United States, including many nonattainment areas, although some urban areas will see 4 reductions. In general, the ozone results are consistent with the findings of Jacobson 5 (2007). Study results do not suggest large impacts on air toxic concentrations across the 6 country, although there may be localized increases in some areas, particularly those 7 associated with fuel production. While increased use of ethanol will certainly result in 8 significantly increased emissions of acetaldehyde, these results do not show a significant 9 impact on ambient concentrations, due to reductions in acetaldehyde precursor emissions. 10 This result may seem to be inconsistent with some other studies, such as the Jacobson 11 (2007) study and monitoring studies in Brazil; however, these other studies involved 12 much higher levels of ethanol, and in the case of the Brazil studies, conditions not 13 relevant to the U.S. Also, while Jacobson (2007) showed increases in acetaldehyde and 14 formaldehyde since he modeled an extreme case of 100% penetration of E85, these 15 increases were offset by reductions in benzene and 1,3-butadiene. When changes in 16 population-weighted mixing ratios were weighted by cancer unit risk estimates, there was 17 little net impact on overall cancer risk, and Jacobson concluded that enhanced ozone risk 18 was a greater concern than air toxics with increased ethanol use. The results of this study 19 support that conclusion.

20

It should be emphasized that these results must be interpreted with the understanding that there are considerable limitations and uncertainties in inventories, atmospheric processes in CMAQ, and other aspects of the modeling process. A

1	discussion of these limitations and uncertainties can help elucidate directions for research
2	needed to better characterize potential air quality impacts. First, modeling assumed that
3	ethanol will make up 34 of the 36 billion gallon renewable fuel mandate, and that
4	approximately 20 billion gallons of this ethanol will be in the form of E85. The emission
5	impacts and air quality results would be different if, instead of E85, more non-ethanol
6	renewable fuels are used or mid-level ethanol blends are approved. In fact, U. S. EPA's
7	final RFS2 rule assumed E85 volumes are likely to be significantly lower than what we
8	assumed for this air quality assessment (U. S. EPA, 2010a). Assumptions about the types
9	of renewable fuels and ethanol blend levels are one of the largest sources of uncertainties
10	in this analysis.
11	
12	A second major source of uncertainty is limited emissions data for vehicles
13	running on E85, and for advanced technology vehicles running on lower level ethanol
14	blends. Thus estimated impacts on NO_x , VOC, and air toxics have considerable
15	uncertainty. As mentioned previously, emissions of vehicles running on E85 were
16	assumed to change proportionally to emissions of vehicles running on E0 in response to
17	temperature. However, if this assumption is not valid, as suggested by some limited
18	emissions data, modeled ozone formation at lower temperatures could be impacted
19	(Ginnebaugh et al., 2010).
20	
21	In addition, assumptions made about impacts of ethanol on NO_x emissions may

emissions with E10 and no change with E85. Previous EPA modeling analyses have

1 shown that reducing NOx emissions will have widespread ozone benefits (U. S. EPA, 2 2008). It was not within the scope of this analysis to conduct air quality modeling 3 simulations of various ethanol/NOx combinations, but it is expected that the ozone 4 response in some locations is highly sensitive to these assumptions. 5 6 Clearly, emission testing of advanced technology vehicles on a variety of ethanol 7 blends is needed to lower the uncertainty associated with this analysis. Other sources of 8 uncertainty are estimates of emissions from ethanol plants, particularly cellulosic ethanol 9 plants, and emissions associated with storage and transport of fuel. All of these estimates 10 are based on very limited data. As mentioned previously, there is no way of knowing 11 where future plants will be located, and siting is based on economic assumptions. 12 Moreover, there are numerous assumptions about land use changes that impact 13 inventories for upstream sources and consequently can impact air quality modeling 14 results. In addition to assumptions that affect emission inventories, limitations in data 15 used to develop hydrocarbon speciation profiles introduce uncertainty into modeling 16 results, as discussed previously. We are currently collecting updated emission profiles 17 for different biofuel mixtures, based on ongoing measurements with new engines under 18 varied driving cycles. With these profiles we will test the sensitivity of the ozone, PM2.5 19 and HAPs results in this paper to variations in VOC profiles and NOx uncertainties. 20 21 Another lesser source of uncertainty is the photochemical mechanisms used in the 22 air quality model. Ozone is formed entirely through thousands of photochemical

23 reactions involving all emitted VOCs and NO_x and its representation in air quality models

1	must be highly condensed. Other pollutants such as acetaldehyde, formaldehyde, and
2	acrolein have substantial contributions from atmospheric chemical reactions.
3	Formaldehyde and acetaldehyde are produced as byproducts of almost every VOC that
4	reacts. Since secondarily formed pollutants can result from many different reaction
5	pathways, they are affected by uncertainties associated with any pathway, as well as
6	uncertainties in the emissions of their precursors.
7	
8	Acknowledgements
9	
10	The authors would like to thank Kathryn Sargeant, John Koupal, Megan
11	Beardsley, Patricia Rowley, Aron Butler, and Julie Boldevich for their contributions to
12	this work.
13	
14	Disclaimer
15	
16	This paper has been reviewed in accordance with U.S. EPA peer and
17	administrative review policies and approved for presentation and publication. This work
18	does not necessarily reflect U. S. EPA policies or views. Mention of trade names or
19	commercial products does not constitute endorsement or recommendation for their use.
20	
21	References
22	

1	Allen P., Bradley R., Croes B. E., DaMassa J., Effa R., Fuentes M., Hebert A., Luo D.,
2	Vincent R., Woodhouse L., Yang E., 1999. Analysis of the Air Quality Impacts of the
3	Use of Ethanol in Gasoline. California Air Resources Board, November 18, 1999.
4	http://www.arb.ca.gov/fuels/gasoline/ethanol/ethfate/Report.doc
5	
6	Altshuller A. P., 1993. Production of Aldehydes as Primary Emissions and from
7	Secondary Atmospheric Reactions of Alkenes and Alkanes During the Night and Early
8	Morning Hours. Office of Research and Development, Research Triangle Park, North
9	Carolina. Report No. EPA600/J-93/467.
10	
11	Altshuller A. P., 1991a. Chemical reactions and transport of alkanes and their products in
12	the troposphere. Journal of Atmospheric Chemistry 12, 19-61.
13	
14	Altshuller A. P., 1991b. Estimating product yields of carbon-containing products from
15	the atmospheric pliotooxidation of ambient air alkenes. Journal of Atmospheric
16	Chemistry 13, 131-154.
17	
18	Anderson L.G., Wolfe P., Barrell R. A., and Lanning J. A., 1994. The Effects of
19	Oxygenated Fuels on the Atmospheric Concentrations of Carbon Monoxide and
20	Aldehydes in Colorado, in: Sterrett F., (Ed.) Alternative Fuels and the Environment,
21	Lewis Publishers, Boca Raton, Florida, pp. 75-103.
22	

1	Anderson L. G., Lanning J. A., Barrell R., Miyagishima J., Jones, R., 1996. Sources and
2	sinks of formaldehyde and acetaldhde: an analysis of Denver's ambient concentration
3	data. Atmospheric Environment 12, 2113-2133.
4	
5	Anderson L.G., Lanning J.A., Wilkes E., Wolfe P., Jones R. H., 1997. Effects of using
6	oxygenated fuels on carbon monoxide, formaldehyde and acetaldehyde concentrations in
7	Denver. Paper 97-RP139.05, Air & Waste Management Association 90th Annual
8	Meeting & Exhibition, June 8-13, 1997, Toronto, Ontario, Canada.
9	
10	de Andrade J. B., de Andrade M. V., Pinheiro H. L. C., 1998. Atmospheric levels of
11	formaldehyde and acetaldehyde and their relationship with the vehicular fleet
12	composition in Salvador, Bahia, Brazil. Journal of the Brazilian Chemical Society 9(3),
13	219-223.
14	
15	Brady D., Pratt G. C., 2007. Volatile organic compound emissions from dry mill fuel
16	ethanol production. Journal of the Air and Waste Management Association 57, 1091-
17	1102.
18	
19	Colon M., Pleil J. D., Hartlage T. A., Guardani M. L., Martins, M. H., 2001. Survey of
20	volatile organic compounds associated with automotive emissions in the urban airshed of
21	Sao Paulo, Brazil. Atmospheric Environment 35, 4017–4031.
22	

1	Byun D., Schere K. L., 2006. Review of the Governing Equations, Computational
2	Algorithms, and Other Components of the Models-3 Community Multiscale Air Quality
3	(CMAQ) Modeling System. Applied Mechanics Reviews 59, 51-77.
4	
5	California Air Resources Board, 2007. Status Report on the State Strategy for
6	California's 2007 State Implementation Plan (SIP) and Proposed Revision to the SIP
7	Reflecting Implementation of the 2007 Strategy.
8	http://www.arb.ca.gov/planning/sip/meetings/sipupdatereport.pdf
9	
10	Calvert J. G., Atkinson R., Kerr J. A., Madronich S., Moortgat G. K., Wallington T. J.,
11	Yarwood G., 2000. The Mechanisms of Atmospheric Oxidation of the Alkenes. Oxford
12	University Press: New York/Oxford, p vii+552.
13	
14	Cohan D. S., Hu Y., Russell A. G., 2006. Dependence of ozone sensitivity analysis on
15	grid resolution. Atmospheric Environment 40, 126-135.
16	
17	Crutzen P. J., Mosier A. R., Smith K. A. Winiwarter, W., 2008. N2O release from agro-
18	biofuel production negates global warming reduction by replacing fossil fuels.
19	Atmospheric Chemistry and Physics Discussions 8, 11191-11205.
20	
21	Delucchi M, 2006. Lifecycle Analyses of Biofuels: Draft Report. Institute of
22	Transportation Studies, University of California, Davis, CA, Report No. UCD-ITS-RR-
23	06-08. http://www.its.ucdavis.edu/publications/2006/UCD-ITS-RR-06-08.pdf.

1	
2	Durbin T. D., Miller J. W., Younglove T., Huai T., Cocker K., 2007. Effects of fuel
3	ethanol content and volatility on regulated and unregulated exhaust emissions for the
4	latest technology gasoline vehicles. Environmental Science and Technology 41, 4059-
5	4064.
6	
7	Energy Information Administration, 2007. Annual Energy Outlook 2007 with Projections
8	to 2030. U.S. Department of Energy, Report # DOE/EIA-0383.
9	http://www.eia.doe.gov/oiaf/aeo/index.html
10	
11	Environment Canada, 2007. Comparison of Emissions from Conventional and Flexible
12	Fuel Vehicles Operating on Gasoline and E85 Fuels. ERM Report No. 05-039,
13	Emissions Research Division.
14	
15	Gaffney J. S., Marley N. A., Martin R. S., Dixon R. W., Reyes L. J., Popp C. J., 1997.
16	Potential air quality effects of using ethanol-gasoline fuel blends: A field study in
17	Albuquerque, New Mexico. Environmental Science and Technology 31, 3053 - 3061.
18	
19	Graham L., Belisle S., Baas C., 2008. Emissions from light duty gasoline vehicles
20	operating on low blend ethanol gasoline and E85. Atmospheric Environment 42, 4498-
21	4516.

1	Ginnebaugh D. L., Liang J., Jacobson M. Z., 2010. Examining the temperature
2	dependence of ethanol (E85) versus gasoline emissions on air pollution with a largely-
3	explicit chemical mechanism. Atmospheric Environment 44, 1192-1199.
4	
5	Grell G., Dudhia J., Stauffer D., 1994. A Description of the Fifth-Generation Penn
6	State/NCAR Mesoscale Model (MM5), National Center for Atmospheric Research,
7	Boulder CO, Report No. NCAR/TN-398+STR.
8	
9	Grosjean D, Grosjean E, Moreira L. F. R., 2002a. Speciated ambient carbonyls in Rio de
10	Janeiro, Brazil. Environmental Science and Technology 36, 1389-1395.
11	
12	Grosjean E., Grosjean D., Woodhouse L., Yang Y., 2002b. Peroxyacetyl nitrate and
13	peroxypropionyl nitrate in Porto Alegre, Brazil. Atmospheric Environment 36, 2405-
14	2419.
15	
16	Jacobson M. Z, 2007. Effects of ethanol (E85) versus gasoline vehicles on Cancer and
17	Mortality in the United States. Environmental Science and Technology 41, 4150-4157.
18	
19	Jacobson M. Z., 2009. Review of energy solutions to global warming, air pollution, and
20	energy security. Energy and Environmental Science 2, 148-173.
21	

1	Le Sager P., Yantosca B., Carouge C., 2008. GEOS-Chem v*-02-01 Online User's
2	Guide. School of Engineering and Applied Sciences, Harvard University, Cambridge,
3	MA. http://acmg.seas.harvard.edu/geos/doc/man/index.html
4	
5	Martins E. M., Arbilla G., 2003. Computer modeling study of ethanol and aldehyde
6	reactivities in Rio de Janeiro urban air. Atmospheric Environment 37, 1715–1722.
7	
8	Montero L., Vasconcellos P. C., Souza S. R., Pires M. A. F., Sanchez-Ccoyllo O. R.,
9	Andrade M. F., Carvalho L. R. F., 2001. Measurements of atmospheric carboxylic acids
10	and carbonyl compounds in Sao Paulo City, Brazil. Environmental Science and
11	Technology 35, 3071-3081.
12	
13	Niven R., 2005. Ethanol in gasoline: environmental impacts and sustainability review
14	article. Renewable and Sustainable Energy Reviews 9, 535-555.
15	
16	Smith K. R., Jerrett M., Anderson H. R., Burnett R. T., Stone V., Derwent R., Atkinson R.
17	W., Cohen A., Shonkoff S. B., Krewski D., Pope C. A., Thun M. J., Thurston G., 2009.
18	Public health benefits of strategies to reduce greenhouse-gas emissions: health
19	implications of short-lived greenhouse pollutants. Lancet, 374 (9707), 2091-2103
20	
21	Schere K., 1988. Modeling ozone concentrations. Environmental Science and
22	Technology 22, 488-495.
23	

1	Southwest Research Institute, 2007. Flex Fuel Vehicles (FFVs) VOC/PM Cold
2	Temperature Characterization When Operating on Ethanol (E10, E70, E85). Prepared for
3	U. S. Environmental Protection Agency. Docket EPA-HQ-OAR-2005-0161 at
4	www.regulations.gov.
5	
6	U. S. Department of Energy, 2009. Greenhouse Gases Regulated Emissions and Energy
7	Use in Transportation (GREET) Model. Argonne National Laboratory.
8	http://www.transportation.anl.gov/modeling_simulation/GREET
9	
10	U. S. Department of Energy, 2007. Annual Energy Outlook 2007 with Projections to
11	2030. Energy Information Administration, Washington, DC, Report No. DOE/EIA-0383.
12	http://www.eia.doe.gov/oiaf/archive/aeo07/index.html
13	
14	U. S. Environmental Protection Agency. 2010a. Final Regulatory Impact Analysis:
15	Changes to Renewable Fuel Standard Program. Office of Transportation and Air Quality,
16	Ann Arbor, MI, Report No. EPA-420-R-10-006.
17	http://www.epa.gov/otaq/renewablefuels/index.htm
18	
19	U. S. Environmental Protection Agency. 2010b. Air Quality Modeling Technical
20	Support Document: Changes to the Renewable Fuel Standard Program. Office of Air
21	Quality Planning and Standards, Research Triangle Park, NC, Report No. EPA 454/R-10-
22	001. http://www.epa.gov/otaq/renewablefuels/index.htm
23	

1	U. S. Environmental Protection Agency. 2009a. Draft Regulatory Impact Analysis:
2	Changes to Renewable Fuel Standard Program. Pp. 274-275. Office of Transportation
3	and Air Quality, Ann Arbor, MI, Report No. EPA-420-D-09-001.
4	http://www.epa.gov/otaq/renewablefuels/420d09001.pdf
5	
6	U. S. Environmental Protection Agency, 2009b. MOVES2010.
7	http://www.epa.gov/otaq/models/moves
8	
9	U. S. Environmental Protection Agency, 2009c. National Mobile Inventory Model.
10	http://www.epa.gov/otaq/nmim.htm
11	
12	U. S. Environmental Protection Agency, 2008. Final Ozone NAAQS Regulatory Impact
13	Analysis. Office of Air Quality Planning and Standards, RTP, NC, Report No. EPA-
14	452/R-08-003. http://www.epa.gov/ttn/ecas/regdata/RIAs/452_R_08_003.pdf
15	
16	U. S. Environmental Protection Agency, 2007a. Regulatory Impact Analysis:
17	Renewable Fuel Standard Program, Office of Transportation and Air Quality, Ann Arbor,
18	MI, Report No. EPA420-R-07-004. http://www.epa.gov/otaq/renewablefuels/index.htm
19	
20	U. S. Environmental Protection Agency, 2007b. Regulation of fuels and fuel additives:
21	renewable fuel standard program; final rule. Federal Register, 72(83), 23900-24013.
22	http://www.epa.gov/otaq/renewablefuels/rfs-finalrule.pdf
23	

1	U. S. Environmental Protection Agency, 2007c. Guidance on the Use of Models and		
2	Other Analyses for Demonstrating Attainment of Air Quality Goals for Ozone, $PM_{2.5}$,		
3	and Regional Haze, Office of Air Quality Planning and Standards, Research Triangle		
4	Park, NC, Report No. EPA -454/B-07-002.		
5	http://www.epa.gov/scram001/guidance/guide/final-03-pm-rh-guidance.pdf		
6			
7	U. S. Environmental Protection Agency, 1999. Byun, D.W., and Ching, J.K.S. (Eds.),		
8	Science algorithms of EPA Models-3 Community Multiscale Air Quality (CMAQ)		
9	modeling system, Office of Research and Development, Research Triangle Park, NC,		
10	Report No. EPA/600/R-99/030. http://www.cmascenter.org		
11			
12	U. S. Environmental Protection Agency, 1993. Motor Vehicle-Related Air Toxics Study,		
13	Office of Mobile Sources, Emission Planning and Strategies Division, Ann Arbor, MI,		
14	Report No. EPA420-R-93-005. <u>http://www.epa.gov/otaq/regs/toxics/tox_archive.htm</u>		
15			
16	Wakeley H. L., Hendrickson C. T., Griffin M. W., Matthews H. S., 2009. Economic and		
17	environmental transportation effects of large-scale ethanol production and distribution in		
18	the United States. Environmental Science and Technology 43, 2228-2233.		
19 20	Woods & Poole Economics Inc., 2008. Population by Single Year of Age CD. CD-ROM.		
21	Woods & Poole Economics, Inc. Washington, D.C.		
22			
23	Yanowitz J., McCormick R. L., 2009. Effect of E85 on tailpipe emissions from light-		
24	duty vehicles. Journal of the Air and Waste Management Association 59, 182-282.		

Table 1

2 Nationwide emission inventories for 2022, RFS2 versus RFS1.

De llesse of	US Total RFS1	US Total RFS2	RFS2 vs RFS1	
Pollutant	Annual Short Tons	Annual Short Tons	Percent Change	
NOx	11,415,147	11,781,115	3.21%	
HC	10,292,785	10,412,658	1.16%	
PM10	11,999,983	12,068,629	0.57%	
PM2.5	3,371,024	3,389,223	0.54%	
CO	51,631,075	47,011,171	-8.95%	
Benzene	226,683	217,021	-4.26%	
1,3-Butadiene	14,458	14,264	-1.34%	
Acetaldehyde	58,405	65,722	12.53%	
Formaldehyde	140,156	140,330	0.12%	
Acrolein	6,399	6,477	1.23%	
Ethanol	457,071	906,719	98.37%	
SO2	8,878,706	8,936,086	0.65%	
NH3	4,213,048	4,213,189	0.00%	

1 Table 2.

Average Change in Projected Future Year 8-hour Ozone Design Value, RFS2 versus 2

- RFS1. 3
- 4

AVERAGE ^a	NUMBER	2020	CHANGE IN
	OF US	POPULATION ^b	2022 DESIGN
	COUNTIES		VALUE (PPB)
			RFS2-RFS1
All	678	238,378,000	0.46
All, population-weighted	678	238,378,000	0.28
Counties whose 2005 base year is			
violating the 2008 8-hour ozone			
standard	389	174,967,000	0.44
Counties whose 2005 base year is			
violating the 2008 8-hour ozone			
standard, population-weighted	389	174,967,000	0.26
Counties whose 2005 base year is			
within 10 percent of the 2008 8-hour			
ozone standard	208	43,172,000	0.52
Counties whose 2005 base year is			
within 10 percent of the 2008 8-hour			
ozone standard, population-weighted	208	43,172,000	0.35
Counties whose 2022 RFS2 control case			
is violating the 2008 8-hour ozone			
standard	26	41,017,000	0.04
Counties whose 2022 RFS2 control case			
is violating the 2008 8-hour ozone			
standard, population-weighted	26	41,017,000	-0.14
Counties whose 2022 RFS2 control case			
is within 10% of the 2008 8-hour ozone			
standard	110	61,618,000	0.34
Counties whose 2022 RFS2 control case			
is within 10% of the 2008 8-hour ozone			
standard, population-weighted	110	61,618,000	0.31

Notes:

^a Changes are only calculated for counties with a valid base year design value ^b Woods & Poole Economics, Inc. (2001). Population by Single Year of Age CD.

1	Fig. 1	. Map	of the	CMAO	modeling	domain.
	0			· ·	0	

RFS2 scenario and the RFS1 scenario.

5	
6	Fig. 3. Annual percent changes in ambient concentrations of acetaldehyde between the
7	RFS2 scenario and the RFS1 scenario.
8	
9	Figure 4. Annual percent changes in ambient concentrations of formaldehyde between
10	the RFS2 scenario and the RFS1 scenario.
11	
12	Figure 5. Annual percent changes in ambient concentrations of benzene between the
13	RFS2 scenario and the RFS1 scenario.
14	
15	Figure 6. Annual percent changes in ambient concentrations of 1,3-butadiene between
16	the RFS2 scenario and the RFS1 scenario.
17	
18	Figure 7. Annual percent changes in ambient concentrations of acrolein between the
19	RFS2 scenario and the RFS1 scenario.
20	
21	Figure 8. Annual percent changes in ambient concentrations of ethanol between the
22	RFS2 scenario and the RFS1 scenario.
23	

Fig. 2. Changes in 2022 8-hour ozone design values for nonattainment areas between the

Figure





















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