

Supporting Information

A Computational Study of Acid Catalyzed Aerosol Reactions of Atmospherically Relevant Epoxides

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1. Theory

Equation 1 of the main text illustrates that $\Delta G_{act}^o(T)$ is fundamental to computing a rate constant and requires that the free energy surface (FES) of a solute/solvent system is properly defined. The solvation of a solute may be approximated using either explicit (discrete) or implicit (continuum) solvation models.¹ In explicit solvation models, solute and solvent molecules are both included in calculations in order to account for specific solute-solvent interactions. However, accounting for all intra- and intermolecular interactions accurately may be computationally costly due to the large number of degrees of freedom.² Alternatively, implicit solvation models remove explicit solvent degrees of freedom by modeling the solvent as a continuous homogeneous and isotropic medium characterized by a dielectric constant ϵ .³⁻⁵ This model may be hampered by first solvation shell effects that are sensitive to the size of the solute cavity and specific intermolecular interactions such as hydrogen bonding.⁶ Hybrid models that incorporate some explicit solvent molecules around a solute (thereby forming a supersolute) immersed in a solvent bath continuum are appealing because they may avoid the disadvantages of either purely explicit or implicit models.⁷ In this work, a hybrid model was used to describe a variety of epoxide ring opening reactions in the condensed phase which is described in detail in the computational methods section.

When a continuum model is used, the FES or statistical mechanical potential of mean force,⁸ $W(\mathbf{R})$ for an N atom solute (or supersolute) with fixed 3N-6 nuclear coordinates at $\mathbf{R}$ is given by,

$$W(\mathbf{R}) = V(\mathbf{R}) + \Delta G_S^*(\mathbf{R}) \quad (\text{S1})$$

where $V(\mathbf{R})$ denotes the gas phase potential energy surface and $\Delta G_S^*(\mathbf{R})$ is the free energy of solvation with nuclei clamped and fixed standard state concentrations. The choice of a particular standard state is made by computational convenience or by following convention (standard state solute concentrations of 1 mol/L are used in this work). While $\Delta G_S^*(\mathbf{R})$ depends on the choice of standard state, the rate

constant does not because the standard state concentrations also appear in the denominator of the pre-exponential term in Equation 1 which has a cancelling effect as will be shown in the discussion that follows.

A hierarchy of theoretical treatments exists in finding minima and saddle points on the liquid phase FES. The simplest and most approximate method is deemed the separable-equilibrium solvation (SES) approximation where stationary and saddle points are found on $V(\mathbf{R})$ which are then solvated.⁹ Free energies in the gas phase are trivial to calculate within the rigid-rotor and harmonic oscillator approximations.⁸ In this manner, gas phase reaction and solvation free energies have been combined in thermodynamic cycles invoking the SES to calculate pK_a 's of different acids.¹⁰ Another treatment denoted the equilibrium solvation path (ESP) method optimizes structures on $W(\mathbf{R})$ in the presence of equilibrium solvation forces from a continuum model.¹¹ ESP may be further extended by including additional solvent degrees of freedom into $\mathbf{R}$ thereby taking into account nonequilibrium solvation (NES) contributions.¹² This extension may be necessary given that solvents do not respond infinitely fast to changes along a reaction coordinate. All calculations in this work were carried out using the ESP method and NES effects were not considered.

It is important to understand that the FES from Equation 2 pertains to solutes with fixed nuclei at $\mathbf{R}$. According to Ben-Naim,^{13, 14} an additional entropy term to the molar Gibbs free energy (chemical potential) is required to allow the species to move around in the solution volume. The molar Gibbs free energy of a species i in solution may be expressed as,

$$G_i = G_i^* - TS_{lib,i} \quad (\text{S2})$$

The first term G_i^* is the chemical potential of a species with fixed nuclei in solution (pseudochemical potential). The second term has been called the liberation free energy¹³ and contains the absolute temperature T and the entropy of liberation $S_{lib,i}$ for species i ,

$$S_{lib,i} = -R \ln(C_i N_A \Lambda_i^3) \quad (S3)$$

where R is the gas constant, C_i is the concentration of i in moles per unit volume, N_A is Avogadro's number and Λ_i is the thermal de Broglie wavelength defined by,

$$\Lambda_i = h(2\pi m_i k_B T)^{-1/2} \quad (S4)$$

in which m_i is the mass of species i . The $\Delta G_{act}^o(T)$ may now be computed using Equations 3 – 5 and the following,

$$\Delta G_{act}^o(T) = \Delta G_{act}^*(T) - T \Delta S_{lib,act}(T) \quad (S5)$$

where $\Delta G_{act}^*(T)$ is the difference in internal free energy between the transition state and reactant complexes (pseudochemical potential energy difference). In the calculations presented, $\Delta G_{act}^*(T)$ is approximated using the zero point inclusive electronic energies with vibrational frequencies calculated in solution using a continuum model.¹¹ $\Delta S_{lib,act}$ is the difference in liberation entropy between the transition state and separate reactant species. It contains the standard state concentrations (see Equation 4) which cancel the concentration term in the denominator of Equation 1 so that the rate constant does not depend on the choice of standard state. Equations 1 and 6 provide a means to estimate liquid phase rate constants for atmospherically relevant epoxide reactions.

References

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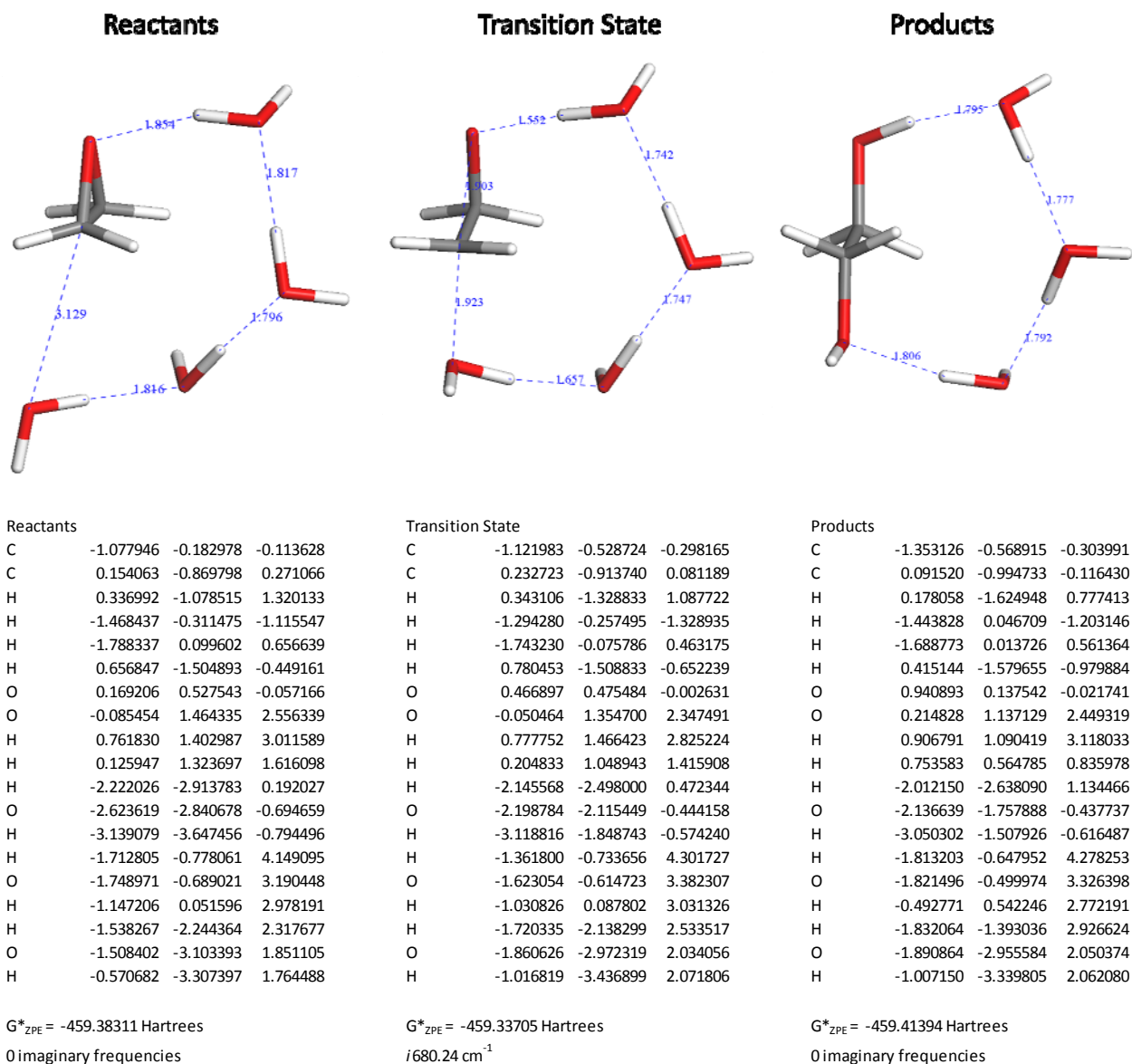
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2. Movies of Oxirane Hydrolysis

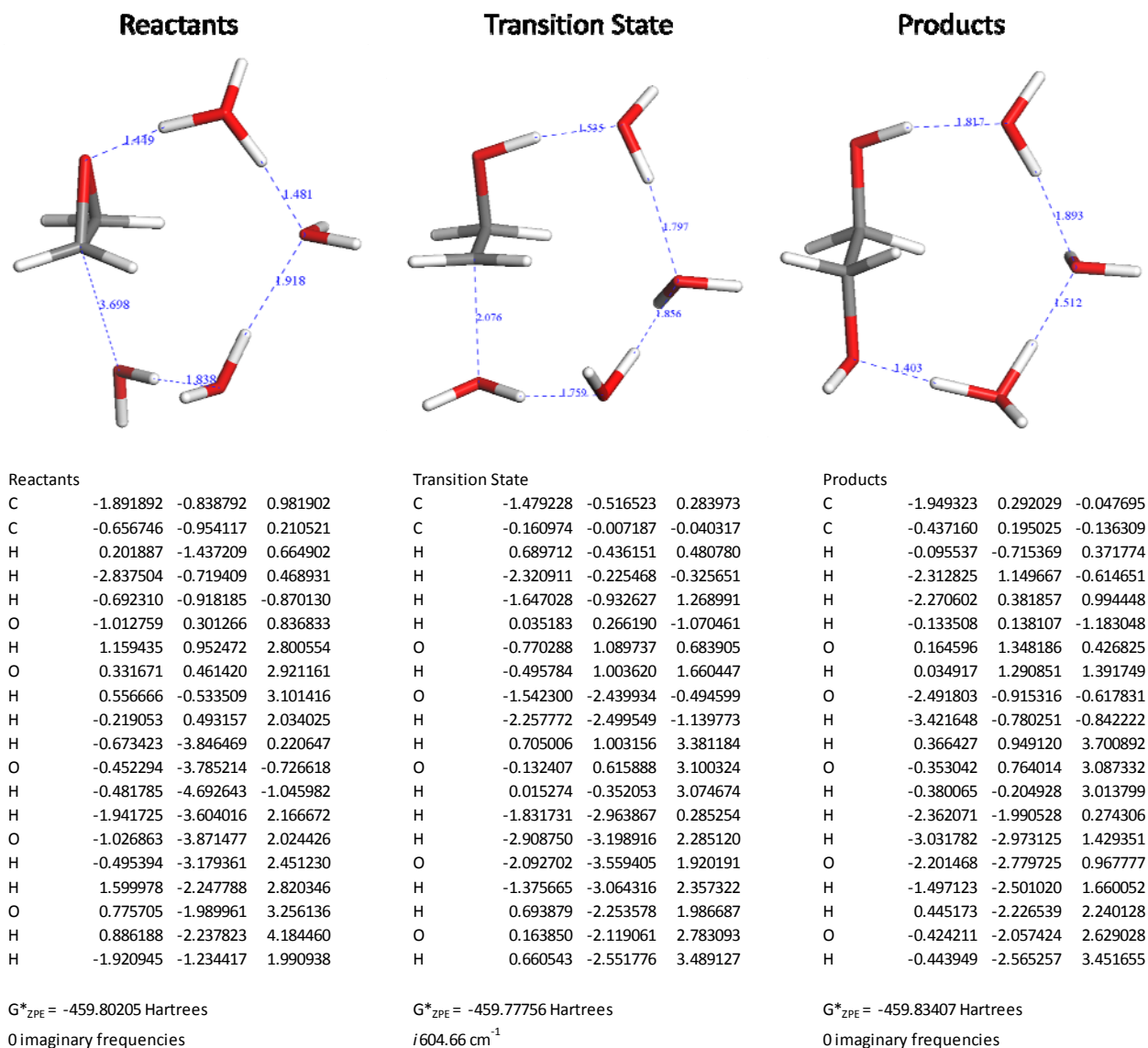
Attached are three movie files of oxirane hydrolysis using three different acids: H_2O (neutral), H_3O^+ and HCOOH . These movies were constructed using the minimum energy pathway obtained from a nudged elastic band (NEB) calculation at the M062x/6-31+G* level of theory. Intermolecular distances are in units of Angstroms.

3. Structures, Coordinates and Zero Point Corrected Energies of Reactive Epoxides (all calculations in water continuum unless indicated otherwise)

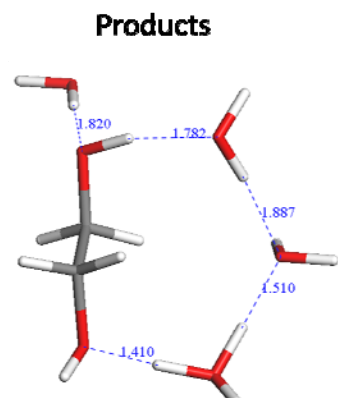
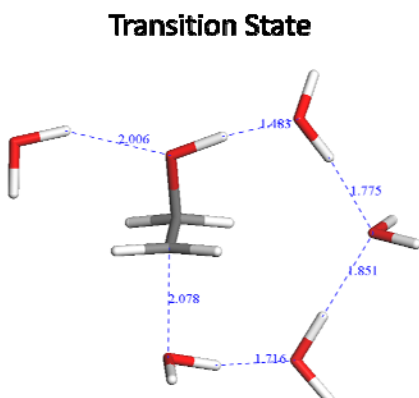
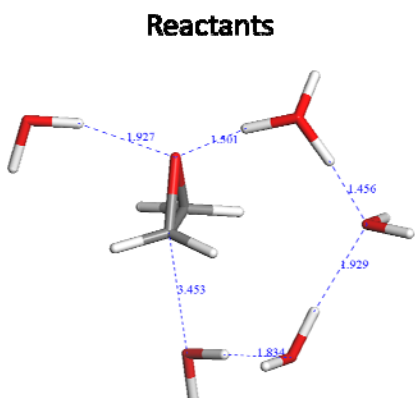
1. Oxirane hydrolysis (4 H₂O)



2. Oxirane hydrolysis (H_3O^+ , 3 H_2O)



3. Oxirane hydrolysis (H_3O^+ , 4 H_2O)



Reactants

C	-1.771355	-0.720048	1.039845
C	-0.52867	-0.721169	0.272735
H	0.334813	-1.247258	0.665257
H	-2.712498	-0.543015	0.533282
H	-0.558598	-0.541623	-0.794531
O	-0.90337	0.439538	1.056176
H	1.240429	0.887408	3.160823
O	0.403770	0.398587	3.209487
H	0.613777	-0.620178	3.299780
H	-0.110734	0.522536	2.328229
H	-0.908789	-3.588604	0.092102
O	-0.894234	-3.457659	-0.873775
H	-0.876046	-4.347063	-1.241025
H	-1.880062	-3.619337	2.177538
O	-0.978640	-3.829989	1.908701
H	-0.423938	-3.191010	2.384864
H	1.662358	-2.276638	2.892655
O	0.824622	-2.060213	3.325650
H	0.904412	-2.398310	4.228482
H	-1.808254	-1.245413	1.987213
H	-2.639162	1.329450	-1.315602
O	-2.229885	2.086184	-0.882287
H	-1.720843	1.693674	-0.157355

$G^*_{\text{ZPE}} = -536.21732$ Hartrees

0 imaginary frequencies

Transition State

C	-1.427475	-0.075849	0.361585
C	-0.002114	-0.323837	0.242457
H	0.433042	-1.0884	0.879953
H	-1.907842	0.568475	-0.358765
H	0.445048	-0.245608	-0.742448
O	-0.073936	0.971416	0.888883
H	0.914759	0.766737	3.752851
O	0.082801	0.494713	3.347359
H	0.051614	-0.482822	3.420307
H	0.029459	0.835898	1.905296
H	-1.990903	-2.442875	0.017216
O	-2.340466	-1.710237	-0.541112
H	-3.284673	-1.656226	-0.347516
H	-1.573443	-4.284005	1.337225
O	-1.100211	-3.501879	1.032769
H	-0.777500	-3.064539	1.841983
H	0.791686	-2.631064	3.452862
O	-0.090992	-2.251087	3.355834
H	-0.580914	-2.568629	4.125148
H	-1.919029	-0.298748	1.298463
H	-0.657260	1.903278	-1.899436
O	-0.244663	2.660097	-1.468707
H	-0.066207	2.346323	-0.571944

$G^*_{\text{ZPE}} = -536.19397$ Hartrees

$i606.47 \text{ cm}^{-1}$

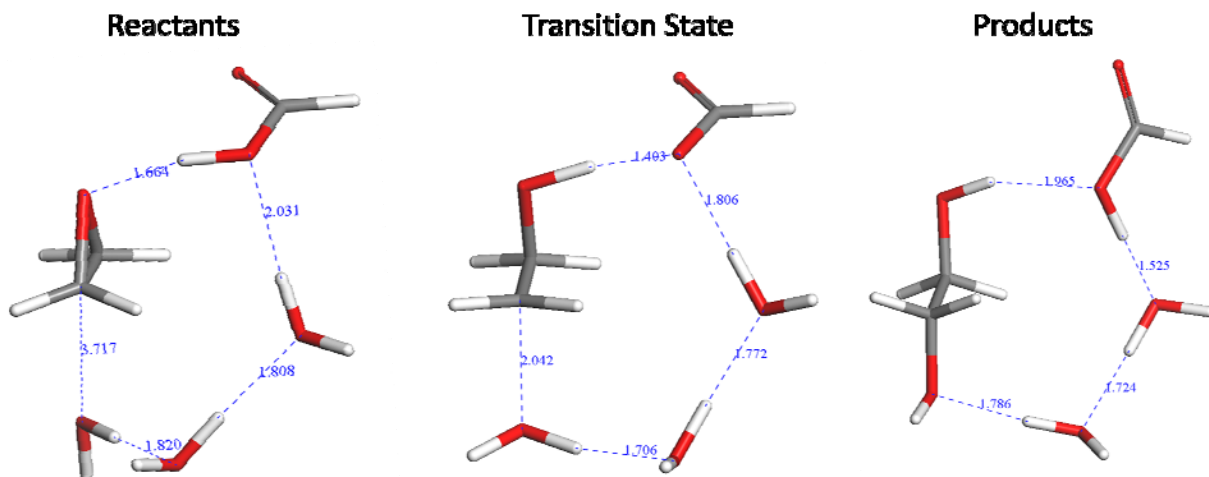
Products

C	-2.016031	0.366474	-0.003042
C	-0.500973	0.41372	-0.044154
H	-0.086098	-0.48732	0.423197
H	-2.441589	1.216436	-0.537416
H	-2.371607	0.370708	1.031505
H	-0.156281	0.45669	-1.07865
O	-0.035940	1.583111	0.615155
H	-0.173872	1.443625	1.573988
O	-2.427673	-0.852896	-0.649447
H	-3.365226	-0.796664	-0.876549
H	0.176749	1.003452	3.824608
O	-0.517517	0.784115	3.193359
H	-0.444478	-0.176351	3.059102
H	-2.194208	-1.964411	0.186854
H	-2.777765	-3.082204	1.258343
O	-1.963907	-2.772629	0.832159
H	-1.305217	-2.469528	1.559547
H	0.588428	-2.111804	2.253692
O	-0.310192	-1.999727	2.593389
H	-0.351420	-2.530852	3.400517
H	2.752416	2.211691	-0.833662
O	2.680044	1.706950	-0.017541
H	1.731418	1.725832	0.205223

$G^*_{\text{ZPE}} = -536.25111$ Hartrees

0 imaginary frequencies

4. Oxirane hydrolysis (HCOOH , 3 H_2O)



Reactants

C	-1.746200	-0.912191	1.095815
C	-0.519897	-0.968871	0.301335
H	0.403138	-1.292735	0.773111
H	-2.706721	-0.984760	0.599947
H	-0.584273	-1.075958	-0.774020
O	-1.017144	0.290918	0.791295
C	1.824738	0.832854	2.479422
O	0.583903	0.576566	2.869186
H	-0.038948	0.641776	2.091143
H	-0.834802	-3.924565	0.171113
O	-0.553811	-3.901788	-0.763050
H	-0.449520	-4.826671	-1.008112
H	-2.211410	-3.618594	2.024668
O	-1.327256	-3.987714	1.921830
H	-0.738423	-3.383641	2.414541
H	0.450442	-1.428951	3.162348
O	0.497602	-2.388336	3.281309
H	0.485732	-2.519200	4.236264
H	-1.710607	-1.192407	2.143232
O	2.137044	1.134462	1.352062
H	2.537530	0.738434	3.305208

$G_{\text{ZPE}}^* = -572.69717$ Hartrees
0 imaginary frequencies

Transition State

C	-1.596106	-0.332176	0.294125
C	-0.275975	0.163347	-0.047710
H	0.576744	-0.346276	0.395366
H	-2.451948	0.050337	-0.240872
H	-0.112401	0.497467	-1.066355
O	-0.794707	1.218516	0.786543
C	1.319563	0.911151	3.155152
O	0.085940	0.604285	2.998120
H	-0.423669	1.036671	1.764416
H	-1.862054	-2.766011	0.126663
O	-1.783638	-2.141470	-0.634630
H	-2.636068	-2.164667	-1.087036
H	-2.726354	-3.642158	1.997953
O	-1.846491	-3.614442	1.606353
H	-1.306921	-3.074970	2.220863
H	-0.166108	-1.172886	3.198970
O	-0.210422	-2.146800	3.258411
H	-0.335498	-2.338171	4.194381
H	-1.735937	-0.805461	1.257343
O	1.979329	1.638586	2.408647
H	1.801472	0.470838	4.044292

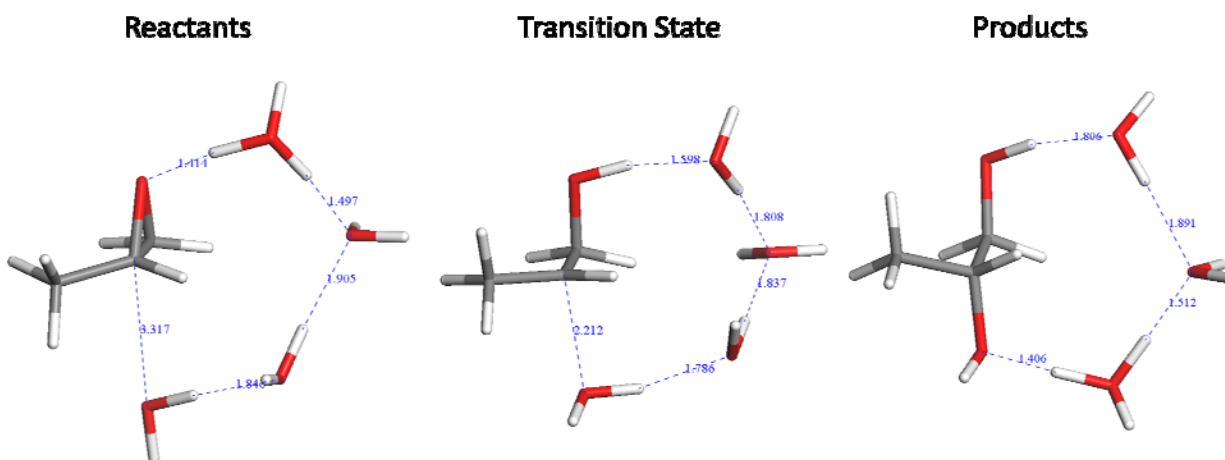
$G_{\text{ZPE}}^* = -572.66331$ Hartrees
 $i637.16 \text{ cm}^{-1}$

Products

C	-2.036471	0.369868	-0.188627
C	-0.527212	0.230450	-0.121912
H	-0.259226	-0.634200	0.496712
H	-2.302872	1.232013	-0.805468
H	-2.446591	0.512609	0.817314
H	-0.128374	0.071646	-1.125513
O	0.065316	1.417004	0.387647
H	-0.072229	1.414010	1.345073
O	-2.547516	-0.832198	-0.769841
H	-3.488869	-0.718775	-0.942647
C	0.676101	0.923157	3.944530
O	-0.268597	0.557518	3.102658
H	-2.304667	-2.167356	0.391355
H	-2.943843	-3.038626	1.493773
O	-2.103121	-2.850199	1.062156
H	-0.940168	-2.295953	2.207279
H	-0.293076	-0.457570	3.005880
O	-0.258952	-1.973284	2.841581
H	-0.434088	-2.429206	3.673124
O	0.915715	2.085413	4.183620
H	1.225548	0.094714	4.410731

$G_{\text{ZPE}}^* = -572.72951$ Hartrees
0 imaginary frequencies

5. 1,2-epoxypropane hydrolysis (H_3O^+ , 3 H_2O)



Reactants

C	-1.716832	-0.816314	0.597024
C	-0.338260	-0.968325	0.134978
H	0.375640	-1.511084	0.746147
H	-0.124724	-0.885588	-0.923949
O	-0.772662	0.285528	0.722932
H	0.765240	1.010511	3.167258
O	0.083464	0.333876	3.035098
H	0.499154	-0.598484	3.170502
H	-0.281122	0.390802	2.044852
H	-0.963669	-3.729131	0.205371
O	-0.984616	-3.748786	-0.768737
H	-1.039297	-4.684115	-0.990835
H	-1.819568	-3.460213	2.351482
O	-0.954888	-3.761734	2.050800
H	-0.322714	-3.144688	2.455527
H	1.820640	-2.140006	2.837489
O	0.970547	-2.015346	3.281444
H	1.114735	-2.279638	4.200183
H	-1.957563	-1.245531	1.565372
C	-2.850614	-0.582878	-0.348318
H	-3.585803	0.092271	0.093481
H	-3.342487	-1.534827	-0.559271
H	-2.486376	-0.158141	-1.285306

$G^*_{\text{ZPE}} = -499.08620$ Hartrees

0 imaginary frequencies

Transition State

C	-1.831112	-0.284744	0.016482
C	-0.378477	-0.335076	0.066982
H	0.068036	-1.074253	0.727779
H	0.141914	-0.214417	-0.877983
O	-0.578248	0.942104	0.712807
H	-0.147910	0.719778	3.813670
O	-0.561118	0.130900	3.171429
H	0.002271	-0.667278	3.144275
H	-0.573116	0.758596	1.701796
H	-1.965881	-2.759455	0.248327
O	-2.115323	-2.377113	-0.642253
H	-3.049256	-2.524314	-0.835798
H	-2.217118	-2.404111	2.454791
O	-1.636611	-3.014437	1.985408
H	-0.736997	-2.764865	2.264548
H	1.616940	-2.192984	2.231090
O	0.903164	-2.211886	2.880120
H	1.267095	-2.660673	3.652571
H	-2.353151	-0.560179	0.926165
C	-2.573191	0.406779	-1.055975
H	-2.963019	1.349724	-0.665742
H	-3.437630	-0.201115	-1.328554
H	-1.947555	0.590195	-1.928101

$G^*_{\text{ZPE}} = -499.06383$ Hartrees

$i511.18 \text{ cm}^{-1}$

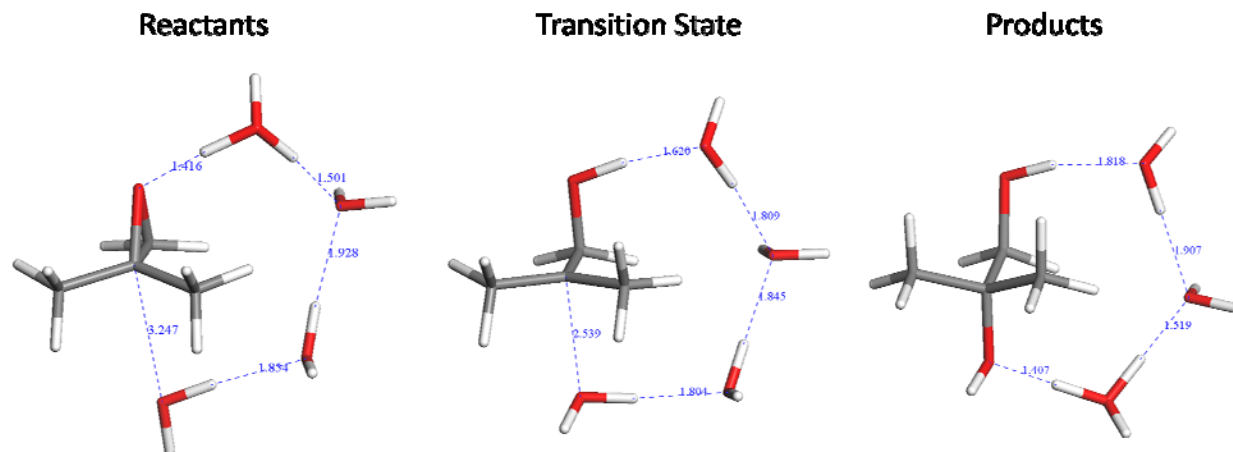
Products

C	-2.019340	0.256148	-0.107330
C	-0.508801	0.202409	-0.294537
H	-0.116661	-0.708933	0.173794
H	-2.248600	0.313837	0.962730
H	-0.279388	0.166566	-1.362563
O	0.115339	1.354850	0.241800
H	0.055426	1.285734	1.213406
O	-2.513055	-1.010452	-0.608945
H	-3.476841	-0.970057	-0.682758
H	0.612126	0.957137	3.466779
O	-0.189394	0.837867	2.945475
H	-0.350742	-0.120645	2.948654
H	-2.150870	-2.097811	0.205223
H	-2.582282	-3.446012	1.070152
O	-1.821182	-2.901698	0.816061
H	-1.375568	-2.542015	1.667261
H	0.193930	-2.396981	2.972307
O	-0.675378	-1.982207	2.884392
H	-1.151142	-2.211734	3.694629
C	-2.671897	1.396541	-0.855220
H	-3.755322	1.376186	-0.714673
H	-2.448860	1.325502	-1.922800
H	-2.297955	2.349841	-0.478936

$G^*_{\text{ZPE}} = -499.11641$ Hartrees

0 imaginary frequencies

6. 2-methyl-1,2-epoxypropane hydrolysis (H_3O^+ , 3 H_2O)



Reactants

C	-1.826169	-0.754509	0.571840
C	-0.425873	-0.981853	0.210026
H	0.195892	-1.593482	0.857258
H	-0.113944	-0.887310	-0.823303
O	-0.816767	0.281213	0.810445
H	0.860292	1.053747	3.120483
O	0.173124	0.370883	3.071927
H	0.606362	-0.553481	3.206204
H	-0.272915	0.387634	2.114008
H	-0.976976	-3.797419	0.321324
O	-1.281452	-3.732988	-0.602056
H	-1.336537	-4.646594	-0.901126
H	-1.301153	-4.032457	2.554669
O	-0.459839	-4.045496	2.083926
H	0.043540	-3.308986	2.468242
H	1.992303	-2.034402	2.890565
O	1.126582	-1.957638	3.315067
H	1.267076	-2.212870	4.237080
C	-2.813204	-0.382489	-0.495825
H	-3.574567	0.287210	-0.089391
H	-3.308166	-1.287435	-0.855373
H	-2.314698	0.104109	-1.334932
C	-2.390514	-1.365210	1.824039
H	-3.115775	-0.683929	2.275693
H	-1.603050	-1.586983	2.547326
H	-2.904973	-2.295054	1.570186

$G^*_{\text{ZPE}} = -538.37024$ Hartrees

0 imaginary frequencies

Transition State

C	-1.898437	-0.084176	0.171993
C	-0.439285	-0.182843	0.068947
H	0.024358	-0.931570	0.709670
H	-0.039217	-0.174969	-0.940133
O	-0.397614	1.136571	0.640680
H	0.709404	1.082753	3.568361
O	-0.056885	0.641123	3.184318
H	0.161001	-0.312214	3.191797
H	-0.257889	1.036344	1.626243
H	-1.698960	-2.890744	0.015425
O	-1.850228	-2.417936	-0.827045
H	-2.777229	-2.573717	-1.040060
H	-2.212493	-3.499854	2.112303
O	-1.383019	-3.618390	1.635463
H	-0.738177	-3.071163	2.120670
H	1.364650	-2.155053	2.551664
O	0.525792	-2.075813	3.023628
H	0.682382	-2.489010	3.882266
C	-2.662083	0.576217	-0.897366
H	-3.382394	1.284110	-0.484722
H	-3.238238	-0.221471	-1.383089
H	-2.016284	1.042949	-1.638647
C	-2.585504	-0.477291	1.415345
H	-3.018253	0.412926	1.881651
H	-1.932811	-1.001484	2.110556
H	-3.430655	-1.113431	1.131800

$G^*_{\text{ZPE}} = -538.35486$ Hartrees

$i/302.29 \text{ cm}^{-1}$

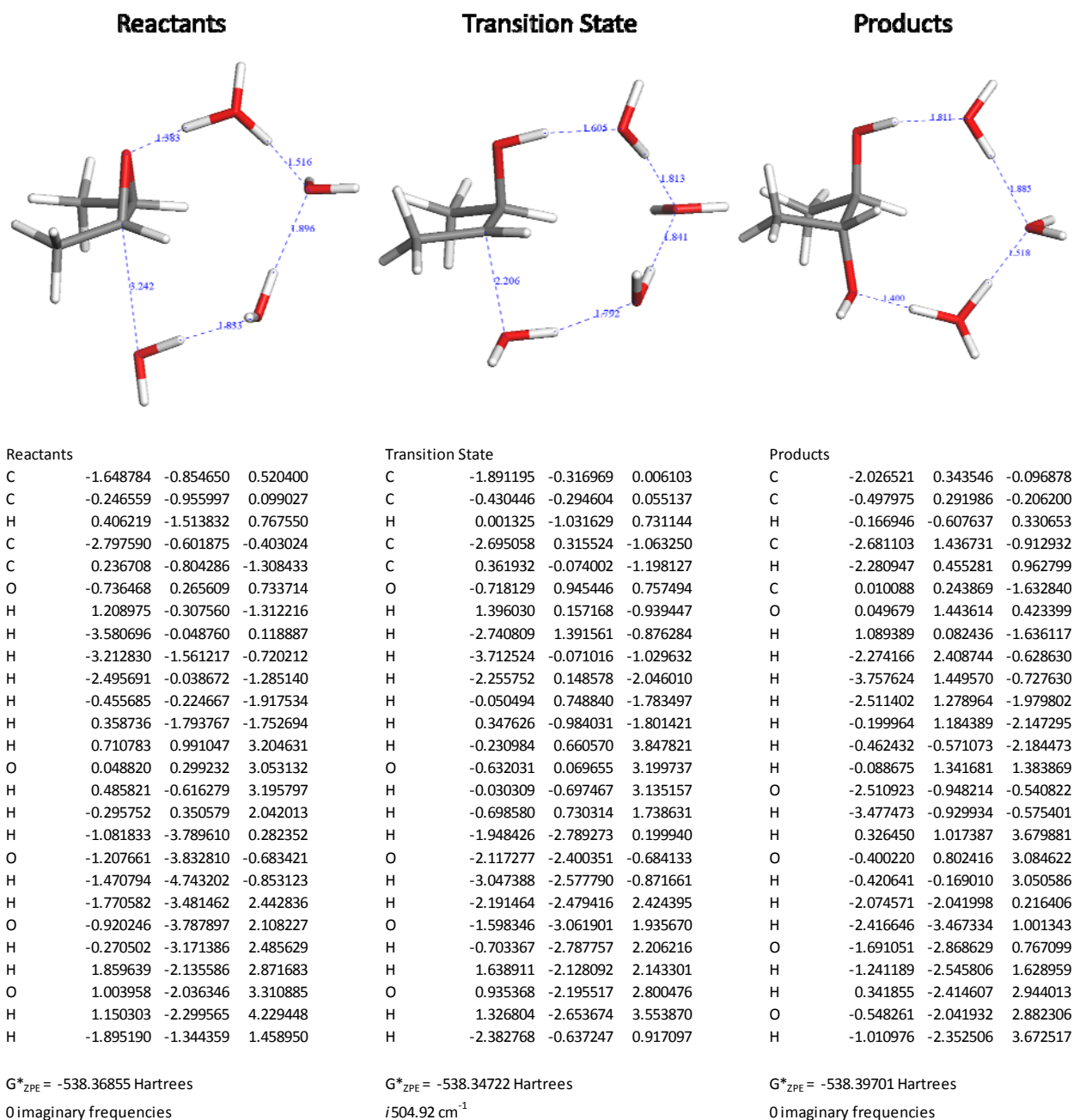
Products

C	-2.092737	0.283709	-0.142182
C	-0.564430	0.193377	-0.112500
H	-0.270576	-0.673665	0.492802
H	-0.200694	0.041989	-1.131742
O	0.029094	1.374720	0.393572
H	0.047457	1.292773	1.364706
O	-2.492935	-1.016557	-0.665403
H	-3.442280	-1.003237	-0.855181
H	0.891976	0.725675	3.497264
O	0.022391	0.698987	3.082642
H	-0.176763	-0.247688	2.985227
H	-2.203013	-2.087363	0.200141
H	-2.762490	-3.265974	1.218712
O	-1.949520	-2.896752	0.841065
H	-1.348907	-2.575340	1.607941
H	0.483795	-2.427274	2.518896
O	-0.412392	-2.120102	2.713526
H	-0.648053	-2.531272	3.556497
C	-2.571810	1.357726	-1.101454
H	-3.663773	1.355712	-1.157193
H	-2.162990	1.189909	-2.100350
H	-2.249146	2.339417	-0.750589
C	-2.676016	0.460871	1.250497
H	-3.767898	0.446347	1.203821
H	-2.362915	1.418920	1.670112
H	-2.337094	-0.339566	1.914113

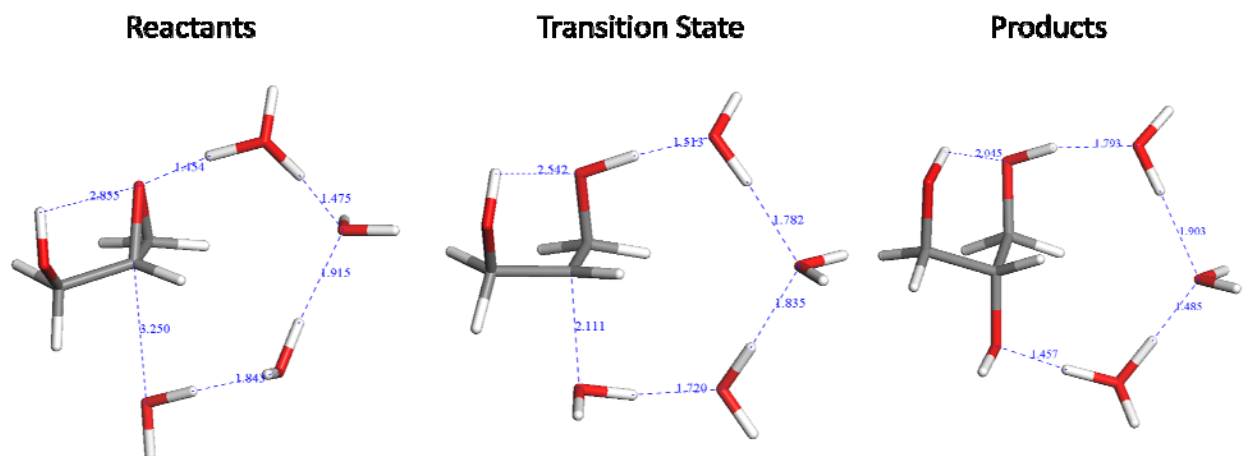
$G^*_{\text{ZPE}} = -538.39975$ Hartrees

0 imaginary frequencies

7. *cis*-2,3-epoxybutane hydrolysis (H_3O^+ , 3 H_2O)



8. 2,3-epoxypropan-1-ol hydrolysis (H_3O^+ , 3 H_2O) – nucleophile adds to secondary carbon



Reactants

C	-1.631152	-0.797998	0.557479
C	-0.249655	-0.991278	0.124258
H	0.427899	-1.551470	0.760012
H	-0.009193	-0.935416	-0.930478
O	-0.665548	0.278656	0.683089
H	0.828197	1.008333	3.184897
O	0.143719	0.337523	3.036895
H	0.547273	-0.606924	3.178292
H	-0.192397	0.400452	2.052637
H	-1.072996	-3.720322	0.256952
O	-1.121924	-3.744697	-0.716186
H	-1.361416	-4.653661	-0.924740
H	-1.848266	-3.382659	2.413731
O	-1.001207	-3.717884	2.098701
H	-0.341215	-3.114349	2.477924
H	1.838335	-2.150439	2.852693
O	0.990928	-2.008312	3.296820
H	1.130383	-2.269541	4.217341
H	-1.927740	-1.202536	1.521469
C	-2.725556	-0.540182	-0.431567
O	-3.721144	0.320603	0.103064
H	-3.214881	-1.486948	-0.666055
H	-2.300256	-0.126012	-1.350770
H	-3.329685	1.194168	0.220320

$G^*_{\text{ZPE}} = -574.30311$ Hartrees

0 imaginary frequencies

Transition State

C	-1.714979	-0.206507	0.093728
C	-0.304770	-0.406597	-0.197959
H	0.234383	-1.119446	0.420288
H	-0.010737	-0.411456	-1.242673
O	-0.293451	0.935502	0.337938
H	1.131441	0.905312	3.032794
O	0.214806	0.705483	2.808173
H	0.081402	-0.235632	3.045720
H	-0.061282	0.883064	1.331332
H	-1.951747	-2.712566	0.139699
O	-2.460292	-2.105821	-0.447248
H	-3.380637	-2.171934	-0.162818
H	-1.181698	-4.408830	1.477290
O	-0.849894	-3.566364	1.147387
H	-0.645288	-3.039658	1.942363
H	0.581722	-2.339327	3.753840
O	-0.217933	-1.962687	3.364874
H	-0.891639	-2.049722	4.051395
H	-2.035874	-0.361190	1.118393
C	-2.594826	0.610115	-0.810734
O	-3.136312	1.719927	-0.121675
H	-3.433500	-0.006525	-1.130638
H	-2.026175	0.921096	-1.691021
H	-2.410920	2.319451	0.091836

$G^*_{\text{ZPE}} = -574.27668$ Hartrees

$i/574.96 \text{ cm}^{-1}$

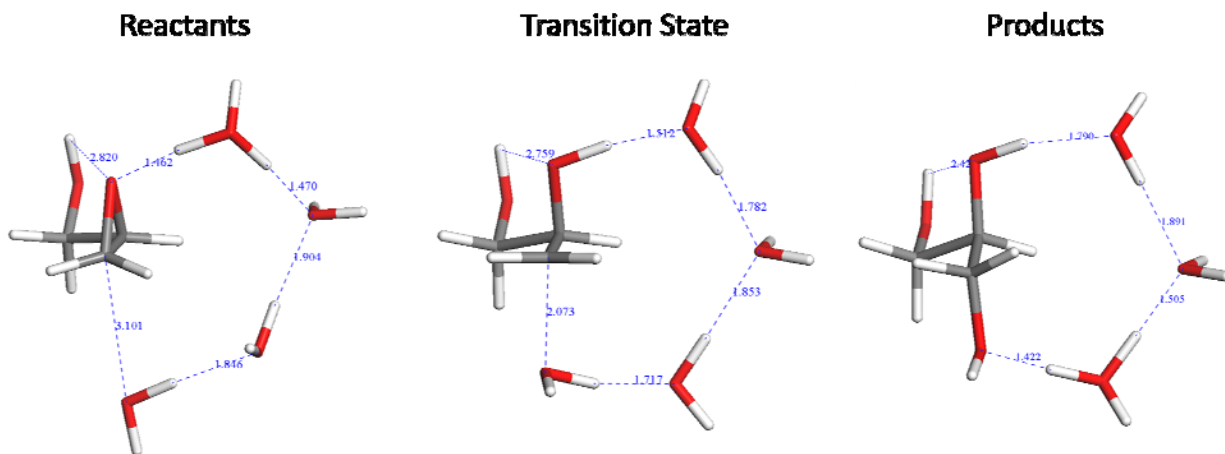
Products

C	-2.007090	0.223618	-0.093353
C	-0.497062	0.116680	-0.268552
H	-0.121506	-0.761684	0.269078
H	-2.258617	0.304830	0.970759
H	-0.271271	-0.003664	-1.330796
O	0.148145	1.299955	0.178672
H	0.135492	1.273857	1.155875
O	-2.549904	-0.999191	-0.622760
H	-3.515808	-0.942221	-0.637747
H	0.700669	0.949242	3.400457
O	-0.107127	0.858351	2.882966
H	-0.310276	-0.091863	2.897705
H	-2.184607	-2.133840	0.215245
H	-2.627709	-3.440715	1.104324
O	-1.862345	-2.910959	0.832008
H	-1.400336	-2.528899	1.674681
H	0.173145	-2.378066	2.934268
O	-0.692605	-1.955154	2.847309
H	-1.160916	-2.157074	3.669470
C	-2.598956	1.410504	-0.844871
H	-3.686434	1.311516	-0.878103
H	-2.218923	1.416148	-1.873318
O	-2.313699	2.637095	-0.188839
H	-1.355156	2.665192	-0.058628

$G^*_{\text{ZPE}} = -574.33287$ Hartrees

0 imaginary frequencies

9. 2,3-epoxypropan-1-ol hydrolysis (H_3O^+ , 3 H_2O) – nucleophile adds to primary carbon



Reactants

C	-1.635540	-0.947777	0.506444
C	-0.224755	-1.032603	0.137138
H	0.444760	-1.580193	0.796910
H	-2.374193	-0.747701	-0.260615
C	0.221697	-0.848510	-1.280518
O	-0.751161	0.171002	0.754493
O	1.478831	-0.189435	-1.346045
H	-0.541335	-0.294579	-1.835985
H	0.351058	-1.829012	-1.740686
H	0.675330	0.930168	3.279582
O	0.039181	0.215000	3.122239
H	0.518408	-0.697143	3.244966
H	-0.303224	0.272533	2.142703
H	-1.172065	-3.812568	0.377444
O	-1.411913	-3.849111	-0.566283
H	-1.471771	-4.789117	-0.765635
H	-1.666740	-3.501763	2.594695
O	-0.845659	-3.809231	2.194078
H	-0.165849	-3.204992	2.535038
H	1.961800	-2.092553	2.905131
O	1.103346	-2.042079	3.347591
H	1.263262	-2.299864	4.265786
H	-1.975588	-1.448757	1.405501
H	1.358197	0.715666	-1.035872

$G^*_{\text{ZPE}} = -574.30335$ Hartrees

0 imaginary frequencies

Transition State

C	-1.763803	-0.274480	0.192577
C	-0.324584	-0.402385	0.047326
H	0.190026	-1.025375	0.776559
H	-2.320447	0.228129	-0.584257
C	0.269040	-0.442865	-1.331905
O	-0.500040	0.954350	0.533264
O	1.649742	-0.124869	-1.303362
H	-0.281855	0.238156	-1.988263
H	0.175607	-1.458818	-1.718980
H	0.616567	1.165292	3.368010
O	-0.175865	0.729969	3.032188
H	-0.036402	-0.229773	3.175312
H	-0.347008	0.941811	1.545125
H	-2.022649	-2.715392	0.095604
O	-2.515316	-2.085083	-0.479924
H	-3.440254	-2.147557	-0.210223
H	-1.215706	-4.381393	1.462346
O	-0.898687	-3.552508	1.086752
H	-0.579610	-3.028550	1.844491
H	1.087404	-2.190629	3.122568
O	0.143721	-2.002833	3.207131
H	-0.103202	-2.356950	4.070899
H	-2.206794	-0.425417	1.166958
H	1.741204	0.806386	-1.069626

$G^*_{\text{ZPE}} = -574.28003$ Hartrees

$i/600.16 \text{ cm}^{-1}$

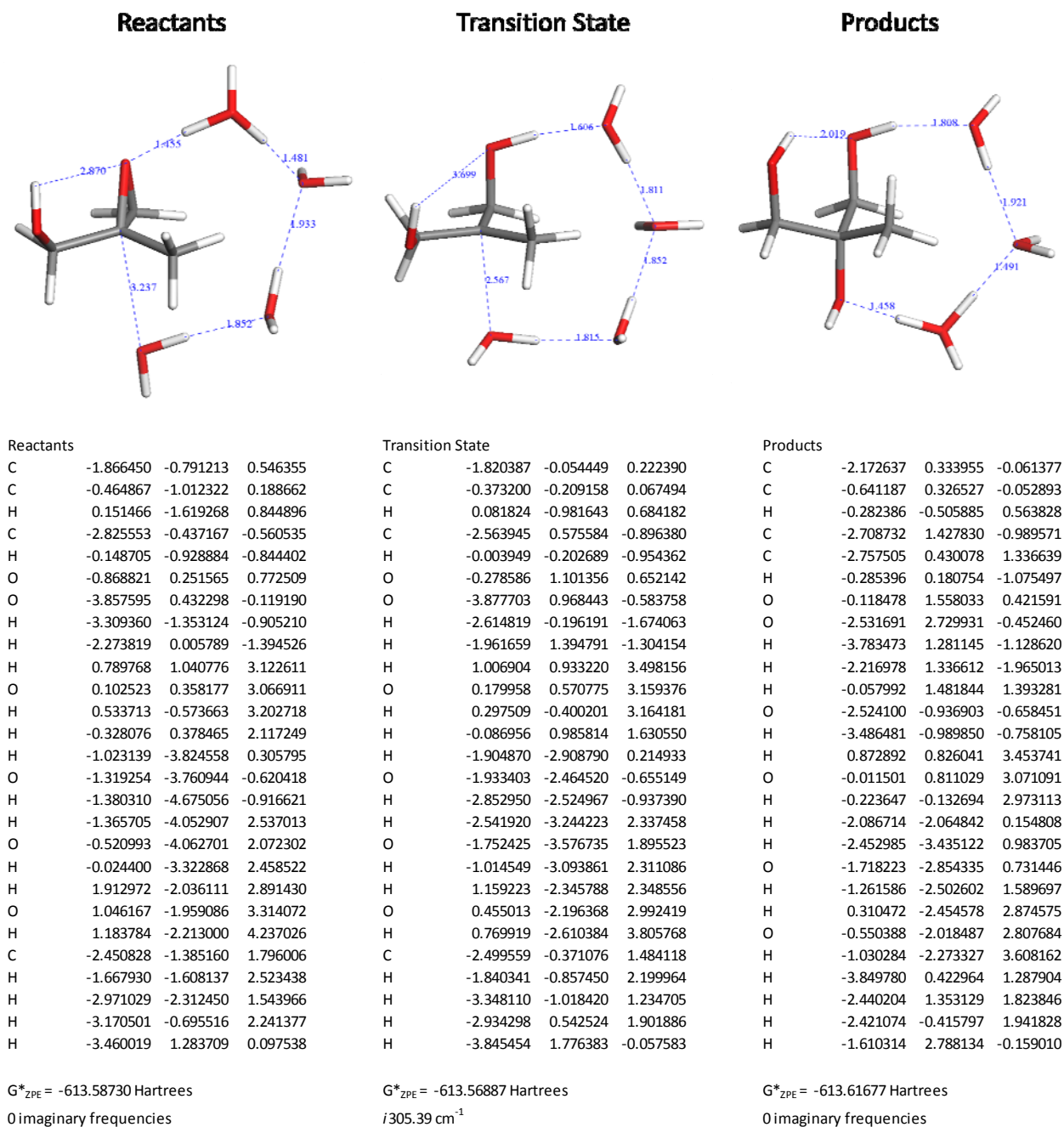
Products

C	-2.022005	0.324945	-0.039789
C	-0.503010	0.290049	-0.101301
H	-0.136372	-0.587822	0.448009
H	-2.407407	1.143011	-0.652659
H	-2.346520	0.460300	0.995268
C	0.004003	0.230645	-1.528008
O	0.009819	1.477465	0.483011
O	1.422509	0.145247	-1.568282
H	-0.342823	1.120574	-2.067150
H	-0.383442	-0.656586	-2.028777
H	-0.033847	1.364658	1.452353
O	-2.525185	-0.922773	-0.550825
H	-3.489496	-0.884753	-0.599802
H	0.534045	0.909526	3.662886
O	-0.261141	0.788202	3.132106
H	-0.372259	-0.175308	3.063509
H	-2.097660	-2.050674	0.201509
H	-2.428286	-3.479465	0.968730
O	-1.710076	-2.867462	0.745709
H	-1.269574	-2.541050	1.615411
H	0.290480	-2.430372	2.931603
O	-0.593023	-2.042562	2.864521
H	-1.066017	-2.346609	3.651423
H	1.769576	0.905219	-1.085789

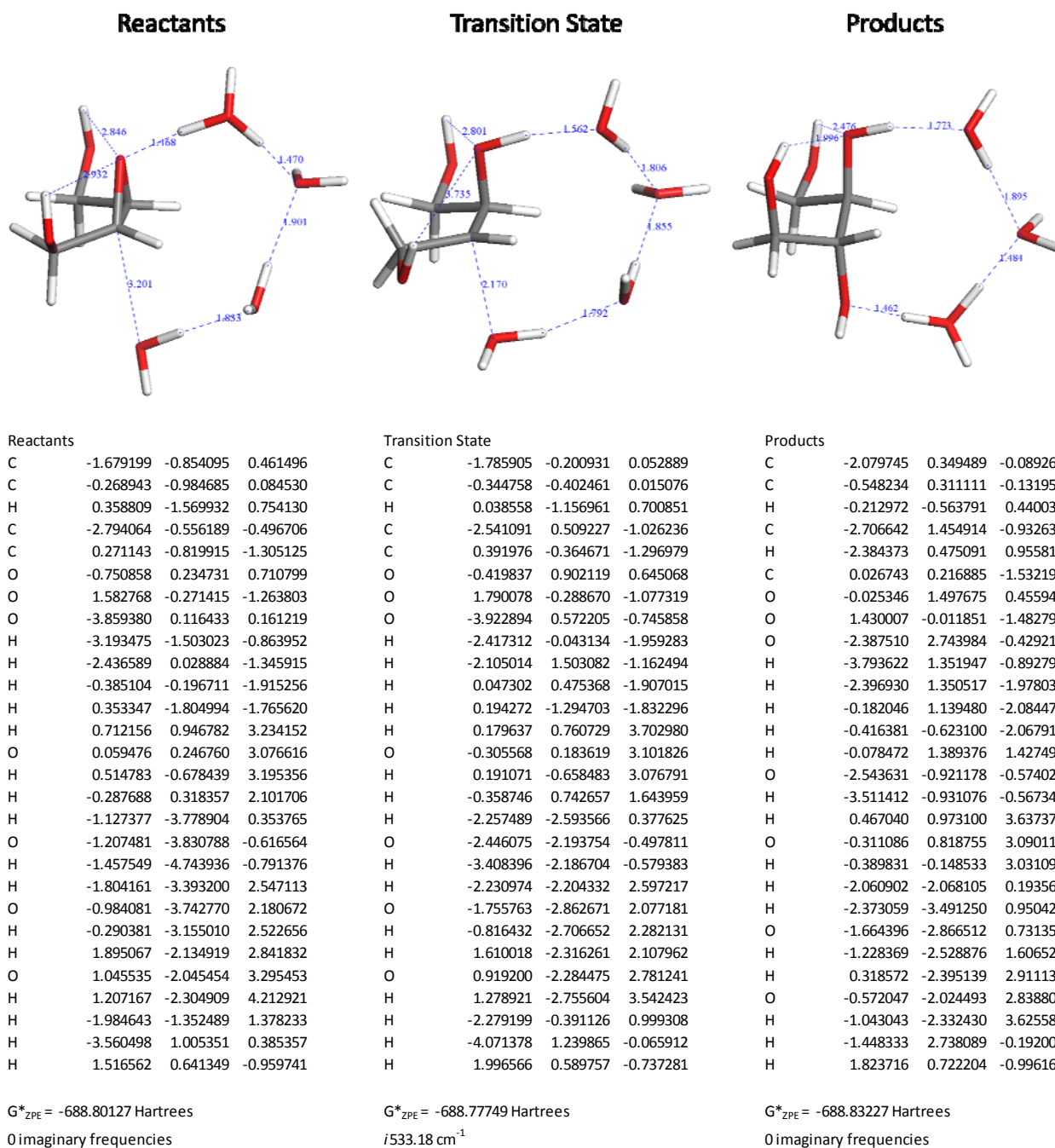
$G^*_{\text{ZPE}} = -574.33532$ Hartrees

0 imaginary frequencies

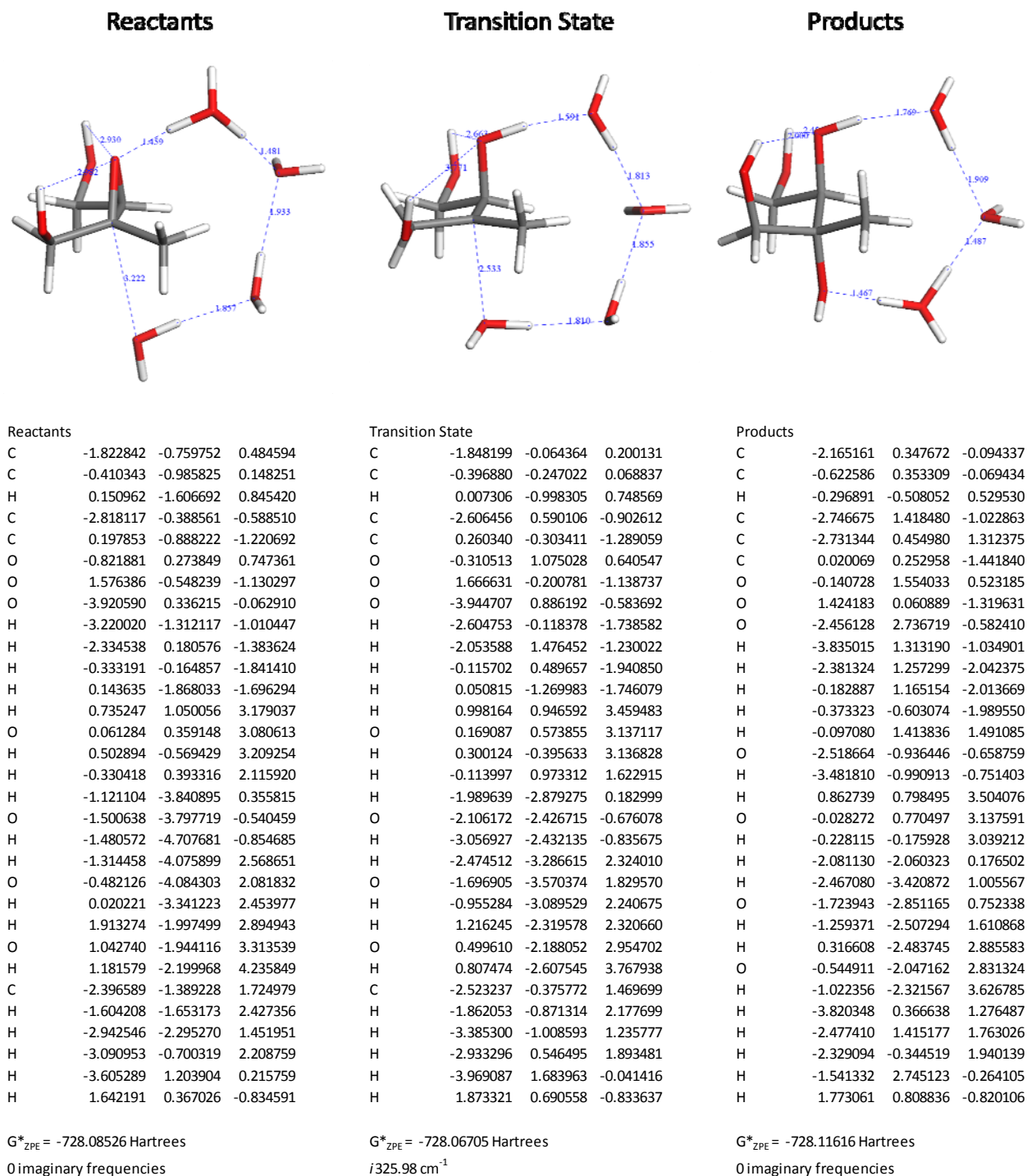
10. 2-methyl-2,3-epoxypropan-1-ol hydrolysis (H_3O^+ , 3 H_2O)



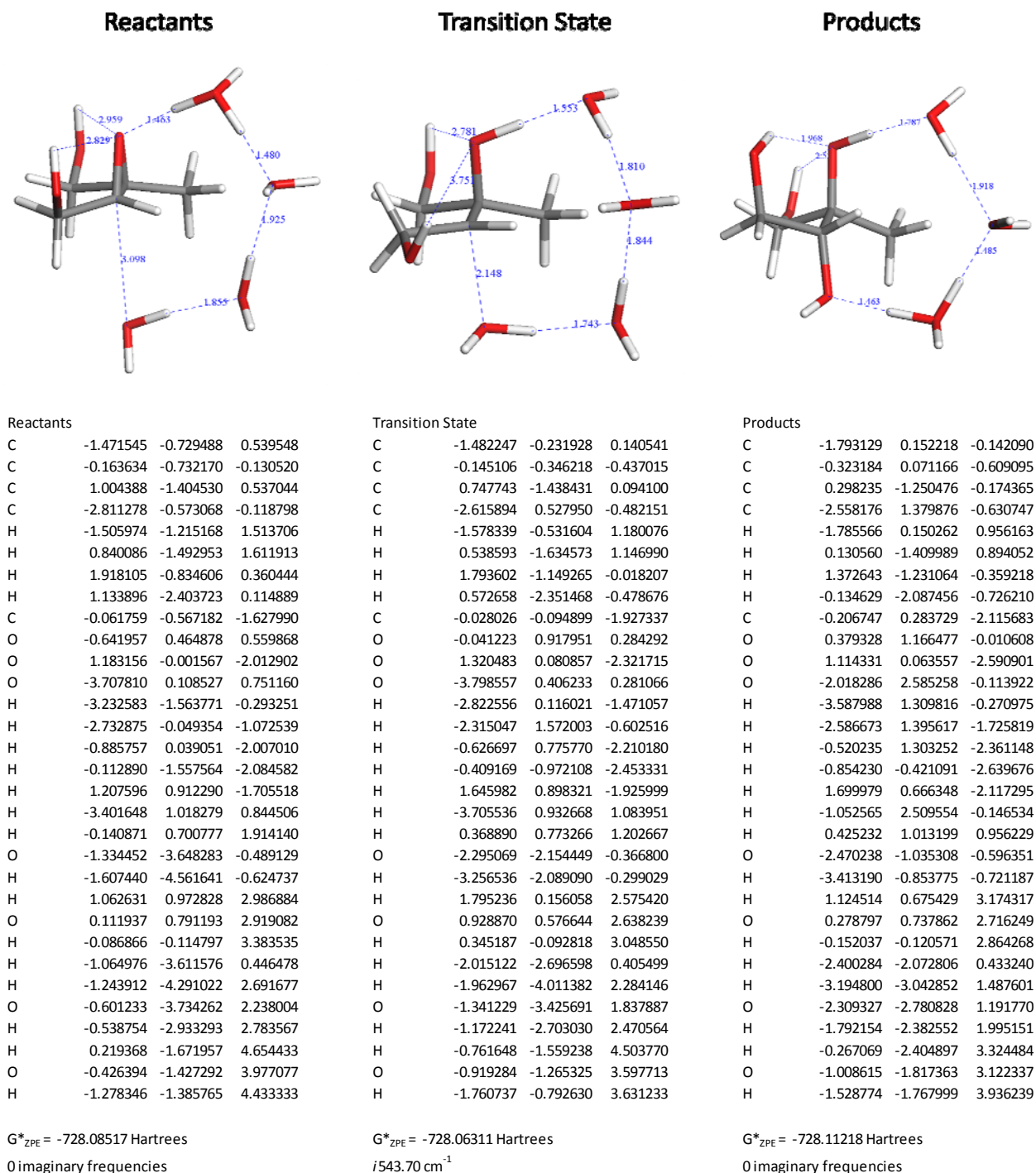
11. *cis*-2,3-epoxybutane-1,4-diol hydrolysis (H_3O^+ , 3 H_2O)



12. β -IEPOX hydrolysis (H_3O^+ , 3 H_2O) – nucleophile adds to tertiary carbon

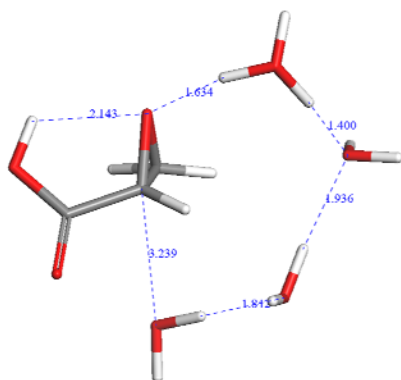


13. β -IEPOX hydrolysis (H_3O^+ , 3 H_2O) – nucleophile adds to secondary carbon



14. 2,3-epoxypropanoic acid hydrolysis (H_3O^+ , 3 H_2O)

Reactants



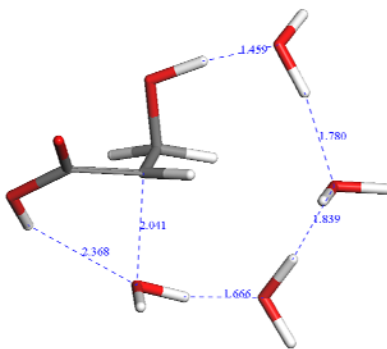
Reactants

C	-1.656542	-0.770511	0.587867
C	-0.278945	-0.958561	0.112233
H	0.395008	-1.541671	0.730108
H	-0.069865	-0.892387	-0.948758
O	-0.702446	0.285247	0.699584
H	0.896367	1.002301	3.318707
O	0.203687	0.342682	3.155601
H	0.601749	-0.640737	3.238346
H	-0.139520	0.460000	2.223235
H	-1.078110	-3.697068	0.246120
O	-1.122828	-3.682031	-0.727301
H	-1.392127	-4.573096	-0.972920
H	-1.845238	-3.402924	2.415856
O	-0.993926	-3.712093	2.085870
H	-0.344482	-3.096760	2.462240
H	1.848571	-2.116195	2.835893
O	1.003952	-1.981198	3.288955
H	1.140014	-2.277895	4.199954
H	-1.972936	-1.227377	1.518516
C	-2.744822	-0.544296	-0.415287
O	-2.621292	0.522371	-1.199797
O	-3.676314	-1.301730	-0.524913
H	-1.844446	1.048788	-0.945070

$G^*_{\text{ZPE}} = -648.36333$ Hartrees

0 imaginary frequencies

Transition State



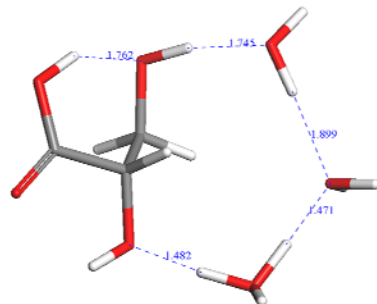
Transition State

C	-1.564324	-0.128764	0.353057
C	-0.150717	-0.184335	0.016692
H	0.455055	-0.884835	0.583929
H	0.137527	-0.075017	-1.024411
O	-0.300249	1.099601	0.663578
H	1.213338	0.788710	3.244330
O	0.273315	0.682717	3.054498
H	0.074616	-0.271730	3.164278
H	-0.036051	0.995949	1.663888
H	-1.763120	-2.514248	-0.043539
O	-2.360889	-1.830116	-0.445592
H	-3.242186	-1.990956	-0.080567
H	-0.823783	-4.296433	0.901441
O	-0.560648	-3.389614	0.706822
H	-0.476356	-2.951321	1.574878
H	-0.159876	-2.455161	3.846178
O	-0.528029	-1.945418	3.113123
H	-1.470169	-1.869011	3.312640
H	-1.872555	-0.395776	1.356393
C	-2.537713	0.721781	-0.439744
O	-2.509868	0.620285	-1.760664
O	-3.290062	1.473451	0.118887
H	-1.979648	-0.136991	-2.056690

$G^*_{\text{ZPE}} = -648.32922$ Hartrees

$i615.03 \text{ cm}^{-1}$

Products



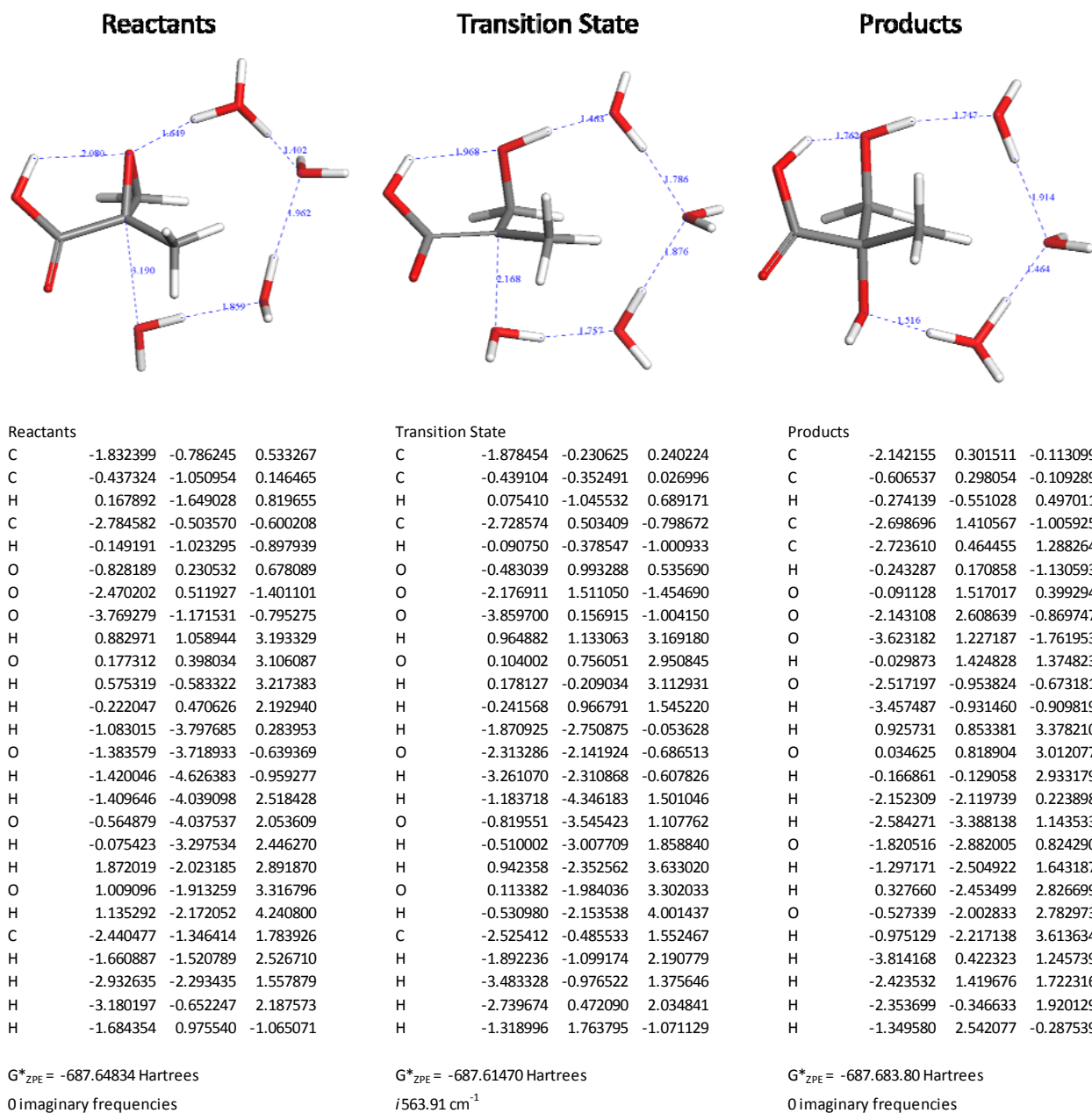
Products

C	-1.878476	0.118513	-0.195903
C	-0.356497	0.171088	-0.342381
H	0.071698	-0.698458	0.163741
H	-2.139710	0.232570	0.865020
H	-0.083397	0.140088	-1.398343
O	0.139694	1.381928	0.208087
H	0.137194	1.279343	1.185098
O	-2.343380	-1.128958	-0.682878
H	-3.191338	-0.991206	-1.137086
H	0.768973	0.892337	3.357691
O	-0.042483	0.784028	2.849121
H	-0.217165	-0.172171	2.852067
H	-2.389204	-2.149416	0.391092
H	-3.196722	-2.865652	1.610263
O	-2.333995	-2.824857	1.168792
H	-1.620071	-2.504319	1.852230
H	0.266634	-2.455336	2.558102
O	-0.576869	-2.034816	2.776607
H	-0.774984	-2.307671	3.683436
C	-2.578547	1.240437	-0.944431
O	-3.565911	1.037629	-1.612383
O	-2.065050	2.452024	-0.803661
H	-1.205987	2.389660	-0.320126

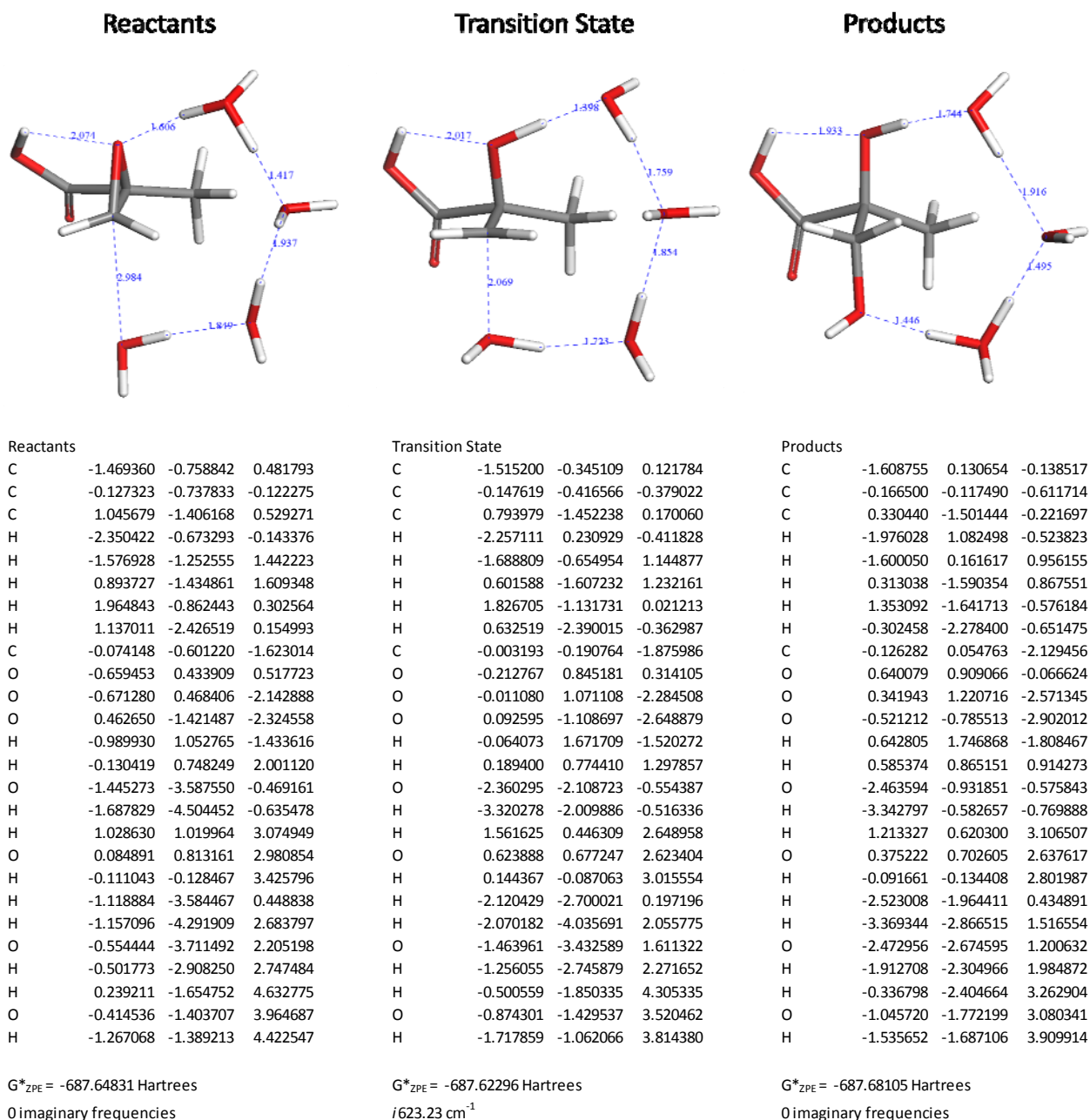
$G^*_{\text{ZPE}} = -648.40041$ Hartrees

0 imaginary frequencies

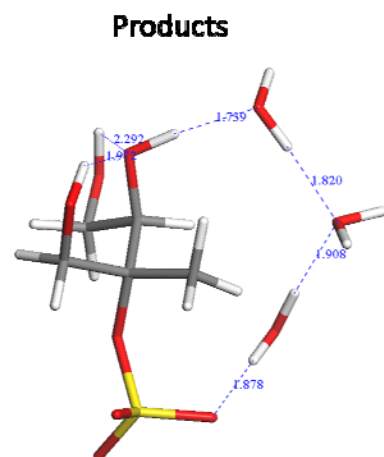
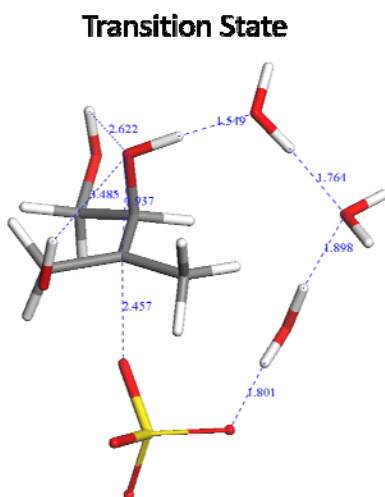
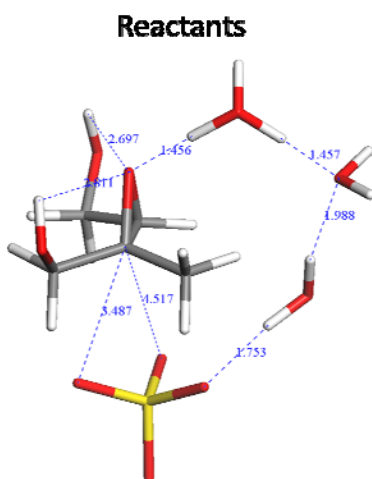
15. MAE hydrolysis (H_3O^+ , 3 H_2O) – nucleophile adds to tertiary carbon



16. MAE hydrolysis (H_3O^+ , 3 H_2O) – nucleophile adds to primary carbon



17. β -IEPOX reaction with sulfate (H_3O^+ , $2 \text{H}_2\text{O}$, SO_4^{2-})



Reactants

C	-2.009447	-0.719267	0.672294
C	-0.790891	-1.358841	0.149745
H	-0.280907	-2.032269	0.837019
C	-3.072123	-0.163018	-0.243834
C	-0.410195	-1.537803	-1.292188
O	-0.752051	0.021931	0.607960
O	0.983707	-1.289093	-1.452010
O	-3.650260	1.017082	0.300161
H	-3.873876	-0.897665	-0.329257
H	-2.670241	0.036817	-1.237926
H	-0.989919	-0.885963	-1.947256
H	-0.586415	-2.575168	-1.573158
H	1.296032	0.781775	2.632793
O	0.427187	0.377061	2.779664
H	0.550690	-0.542900	3.252207
H	-0.047775	0.239698	1.863386
H	-1.049206	-4.121236	1.857689
O	-0.227954	-4.197349	2.386897
H	-0.094353	-3.336038	2.805403
H	1.580382	-2.095952	4.048221
O	0.659285	-1.848608	3.888766
H	0.228779	-1.864958	4.754336
C	-2.453240	-1.029977	2.076919
H	-1.640004	-1.487548	2.644819
H	-3.294916	-1.723704	2.047240
H	-2.774364	-0.116378	2.580965
H	-2.978548	1.709359	0.295579
H	1.141691	-0.354217	-1.274736
O	-1.162221	-4.873299	-0.886967
S	-2.567805	-4.551707	-0.472339
O	-2.571332	-4.174325	0.990423
O	-3.089840	-3.400127	-1.278141
O	-3.442611	-5.750903	-0.672880

$G^*_{\text{ZPE}} = -1351.01885$ Hartrees

0 imaginary frequencies

Transition State

C	-2.077569	-0.191765	0.472201
C	-0.676660	-0.477334	0.121638
H	-0.228507	-1.259515	0.736247
C	-2.938461	0.609918	-0.448335
C	-0.208811	-0.551322	-1.309999
O	-0.447305	0.831140	0.686748
O	1.207805	-0.473595	-1.341759
O	-4.056195	1.191342	0.184620
H	-3.307826	-0.098737	-1.195106
H	-2.331076	1.358385	-0.962492
H	-0.653290	0.244202	-1.913613
H	-0.492870	-1.517146	-1.724651
H	1.249010	0.760659	3.238201
O	0.322829	0.524145	3.109809
H	0.264235	-0.432972	3.313181
H	-0.127565	0.729131	1.642251
H	-1.658224	-3.811350	1.353311
O	-0.692542	-3.926848	1.448808
H	-0.426788	-3.285376	2.125362
H	0.907312	-2.557723	3.956218
O	0.113114	-2.172550	3.565251
H	-0.591520	-2.356111	4.199596
C	-2.572435	-0.527383	1.819374
H	-1.882617	-1.172713	2.360992
H	-3.548043	-1.008652	1.702538
H	-2.755019	0.397438	2.375213
H	-3.754883	1.927487	0.731034
H	1.466202	0.410625	-1.056136
O	-4.147654	-4.176601	-1.106688
S	-3.803172	-3.101068	-0.129081
O	-3.450029	-3.716358	1.202016
O	-2.604989	-2.329067	-0.617933
O	-4.959968	-2.166268	0.042736

$G^*_{\text{ZPE}} = -1351.00333$ Hartrees

$i354.97 \text{ cm}^{-1}$

7 -- 21 bond distance fixed

22 -- 31 bond distance fixed

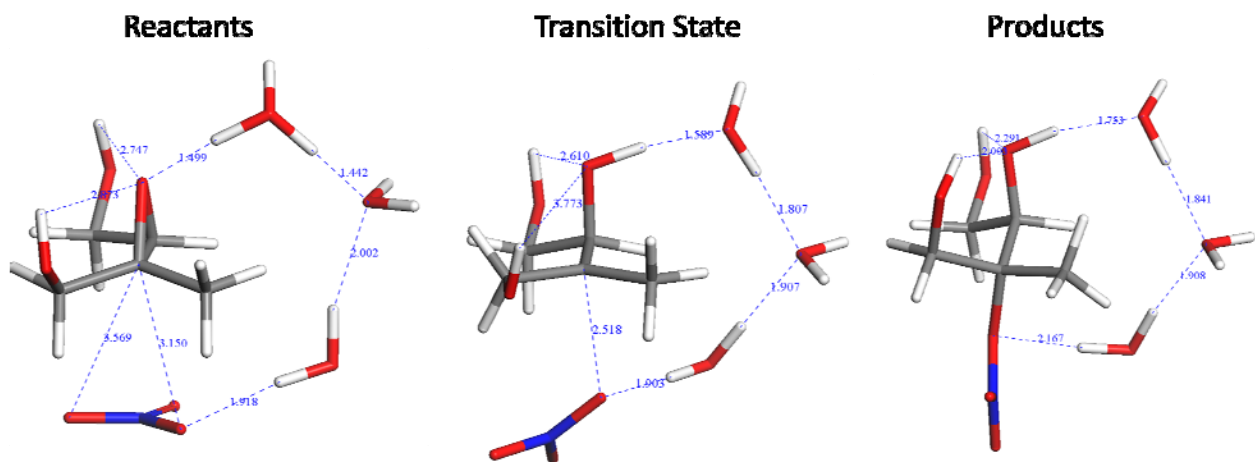
Products

C	-2.353677	0.251273	0.176541
C	-0.824651	0.060259	0.046533
H	-0.533101	-0.789659	0.675944
C	-2.828935	1.590765	-0.411436
C	-2.750853	0.100552	1.636959
C	-0.335404	-0.205820	-1.369911
O	-0.160490	1.235757	0.502281
O	1.080932	-0.347100	-1.370683
O	-2.493967	2.702524	0.405718
H	-3.915986	1.565731	-0.485155
H	-2.419664	1.713684	-1.419449
H	-0.630718	0.611387	-2.037159
H	-0.746763	-1.138899	-1.749120
H	0.116377	1.091775	1.431666
H	1.532370	0.500738	3.086391
O	0.579533	0.639183	3.045242
H	0.188387	-0.240417	3.207169
H	-2.402829	-3.029745	0.978535
O	-1.501476	-3.344518	1.153575
H	-1.193144	-2.813671	1.905652
H	-0.129940	-2.423614	4.051796
O	-0.623331	-1.851355	3.451243
H	-1.456948	-1.673891	3.905293
H	-3.802289	0.340056	1.801083
H	-2.145730	0.786874	2.232938
H	-2.552158	-0.915971	1.976566
H	-1.540308	2.642970	0.562386
H	1.440587	0.431244	-0.927373
O	-5.311673	-0.434389	-0.151366
S	-4.349057	-1.492901	-0.472331
O	-2.899475	-0.784303	-0.694076
O	-4.156894	-2.455778	0.628534
O	-4.561804	-2.133535	-1.770600

$G^*_{\text{ZPE}} = -1351.06183$ Hartrees

0 imaginary frequencies

18. β -IEPOX reaction with nitrate (H_3O^+ , $2 \text{H}_2\text{O}$, NO_3^-)



Reactants

C	-1.929720	-1.043776	0.497842
C	-0.482641	-1.325463	0.420275
H	-0.083241	-1.963784	1.208886
C	-2.716345	-0.629639	-0.722606
C	0.407747	-1.202972	-0.786946
O	-0.953383	-0.052739	0.934565
O	1.678964	-0.700322	-0.387690
O	-3.763725	0.269270	-0.382974
H	-3.186639	-1.521429	-1.140583
H	-2.067826	-0.185856	-1.479189
H	-0.032955	-0.557604	-1.547919
H	0.583878	-2.189204	-1.215514
H	-1.405685	-4.296254	1.511382
O	-1.018648	-4.526452	2.371464
H	-0.733803	-3.690884	2.767389
C	-2.739678	-1.616388	1.629756
H	-2.090166	-1.925217	2.449867
H	-3.307838	-2.478729	1.280100
H	-3.438683	-0.863529	1.999432
H	-3.365023	1.104611	-0.112430
H	1.567936	0.220194	-0.121860
N	-1.313333	-4.008636	-1.060385
O	-1.523579	-3.420786	-2.132748
O	-2.225019	-4.098811	-0.211391
O	-0.195872	-4.494083	-0.817933
H	0.391938	0.561446	3.536020
O	-0.422778	0.052256	3.401073
H	-0.251695	-0.940338	3.693161
H	0.874555	-2.527699	4.200718
O	-0.059905	-2.320474	4.062491
H	-0.499512	-2.524315	4.899524
H	-0.659854	0.066692	2.399610

$G^*_{\text{ZPE}} = -932.10963$ Hartrees
0 imaginary frequencies

Transition State

C	-2.021339	-0.050280	0.381173
C	-0.610719	-0.469642	0.279584
H	-0.334589	-1.208191	1.032750
C	-2.666953	0.629009	-0.768279
C	0.003452	-0.796723	-1.062313
O	-0.262446	0.855998	0.712916
O	1.411616	-0.881062	-0.926406
O	-3.919863	1.198041	-0.478543
H	-2.816863	-0.147940	-1.530692
H	-1.966029	1.355989	-1.191931
H	-0.268194	-0.049857	-1.813226
H	-0.347287	-1.776989	-1.390482
H	-1.295909	-3.364469	0.764589
O	-0.611346	-3.799701	1.298041
H	-0.531774	-3.247620	2.092560
C	-2.727660	-0.165424	1.659980
H	-2.155223	-0.720116	2.400803
H	-3.682176	-0.660884	1.451761
H	-2.986329	0.834784	2.023461
H	-3.784633	2.015766	0.015290
H	1.745180	0.000346	-0.719427
N	-2.761051	-2.900285	-1.321033
O	-3.550782	-2.391597	-2.126521
O	-2.666526	-2.422690	-0.161558
O	-2.053932	-3.864074	-1.640945
H	1.064087	0.715564	3.515351
O	0.153581	0.541913	3.249354
H	0.005335	-0.409718	3.423102
H	0.240038	-2.582464	4.260938
O	-0.378069	-2.160285	3.651303
H	-1.234596	-2.206681	4.094666
H	-0.070540	0.807979	1.699148

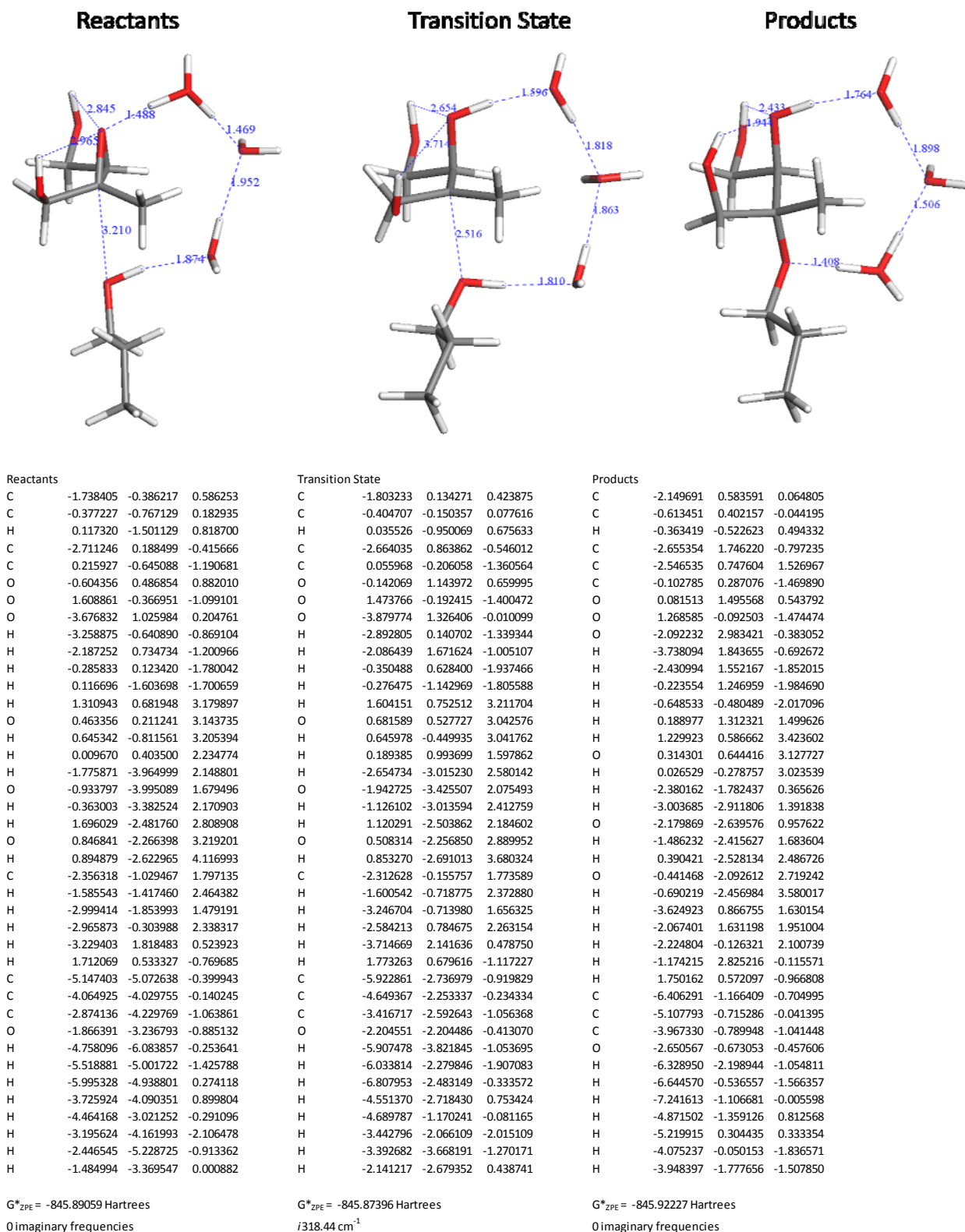
$G^*_{\text{ZPE}} = -932.08996$ Hartrees
 $i255.44 \text{ cm}^{-1}$

Products

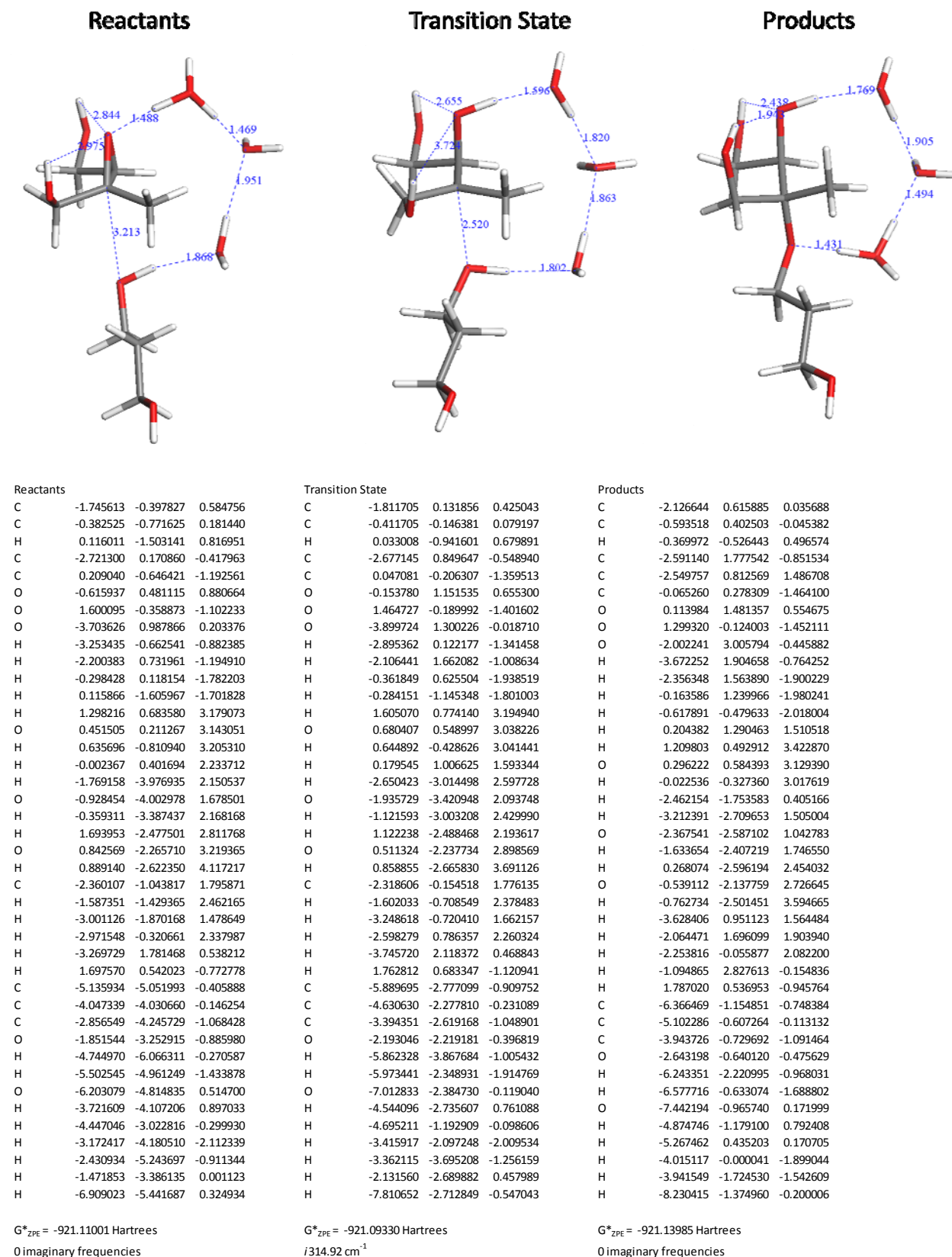
C	-2.284392	0.309050	-0.089755
C	-0.736019	0.220468	-0.075505
H	-0.444907	-0.526184	0.672643
C	-2.784251	1.537124	-0.870247
C	-2.808672	0.238100	1.336732
C	-0.097556	-0.161739	-1.406140
O	-0.219103	1.488385	0.300446
O	1.317344	-0.138165	-1.276654
O	-2.667603	2.718400	-0.095558
H	-3.841658	1.417462	-1.101561
H	-2.232536	1.628891	-1.810481
H	-0.416922	0.523649	-2.198334
H	-0.370792	-1.177883	-1.686353
H	0.028215	1.433925	1.247474
H	1.268043	0.867720	3.117991
O	0.332164	0.959890	2.907085
H	-0.035420	0.059502	2.989153
H	-1.568950	-2.581417	0.030906
O	-1.304380	-3.167298	0.749332
H	-1.126420	-2.577645	1.501261
H	-0.180640	-2.180302	3.690184
O	-0.735698	-1.638211	3.115732
H	-1.578164	-1.564747	3.581783
H	-3.829086	0.605222	1.427387
H	-2.164619	0.849631	1.972379
H	-2.756080	-0.794567	1.690981
H	-1.736119	2.788509	0.157875
H	1.554826	0.734389	-0.938205
N	-3.906717	-1.357686	-0.882167
O	-4.779802	-0.667794	-0.437970
O	-2.618121	-0.907402	-0.859186
O	-4.009997	-2.445650	-1.378781

$G^*_{\text{ZPE}} = -932.12545$ Hartrees
0 imaginary frequencies

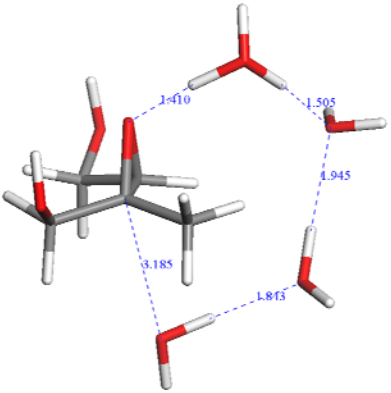
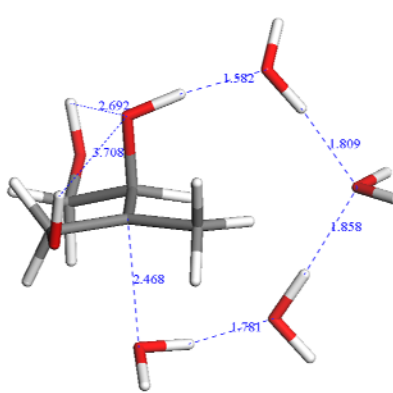
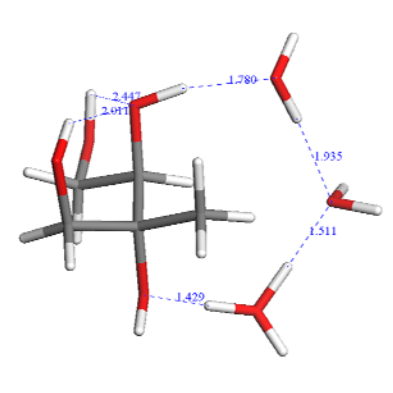
19. β -IEPOX reaction with propanol (H_3O^+ , 2 H_2O , $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$)



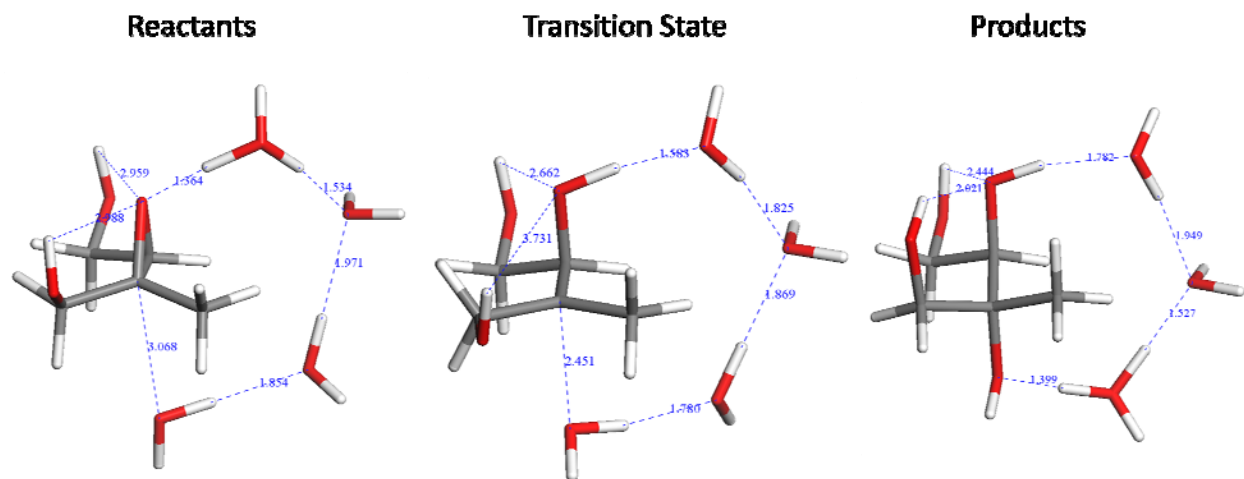
20. β -IEPOX reaction with propane-1,4-diol (H_3O^+ , 2 H_2O , $\text{HOCH}_2\text{CH}_2\text{CH}_2\text{OH}$)



21. β -IEPOX hydrolysis (ethanol continuum, H_3O^+ , 3 H_2O)

Reactants				Transition State				Products			
											
Reactants				Transition State				Products			
C	-1.844545	-0.749099	0.463709	C	-1.885084	0.003651	0.202233	C	-2.168652	0.351345	-0.088489
C	-0.442846	-1.023559	0.124145	C	-0.462405	-0.297517	-0.008633	C	-0.625934	0.351133	-0.058686
H	0.094841	-1.674201	0.812626	H	-0.078296	-1.060810	0.670301	H	-0.302256	-0.494235	0.565291
C	-2.828046	-0.321993	-0.599257	C	-2.645278	0.735393	-0.851285	C	-2.745949	1.423858	-1.018812
C	0.177209	-0.925474	-1.239249	C	0.113078	-0.451475	-1.396780	C	-2.740161	0.459290	1.315618
O	-0.803301	0.246539	0.740623	O	-0.243967	1.025347	0.518216	C	0.028924	0.212284	-1.422849
O	1.563326	-0.630606	-1.132207	O	1.527394	-0.451596	-1.323676	O	-0.147704	1.566024	0.503044
O	-3.896600	0.435725	-0.055532	O	-3.902072	1.198873	-0.430435	O	1.431897	0.041747	-1.279314
H	-3.267505	-1.224362	-1.030822	H	-2.800292	0.018320	-1.664744	O	-2.483359	2.738364	-0.560058
H	-2.326396	0.236241	-1.391378	H	-2.018658	1.541359	-1.244588	H	-3.833127	1.306482	-1.049871
H	-0.327329	-0.177365	-1.853319	H	-0.239740	0.341992	-2.061942	H	-2.364524	1.276293	-2.035103
H	0.093533	-1.898021	-1.726407	H	-0.195804	-1.416651	-1.799165	H	-0.180216	1.101271	-2.028434
H	0.589023	1.089085	3.241320	H	1.208569	0.932379	3.284519	H	-0.347911	-0.667638	-1.944707
O	-0.006544	0.341221	3.074712	O	0.334167	0.634748	3.006595	H	-0.081779	1.445587	1.471222
H	0.492700	-0.544064	3.234076	H	0.292880	-0.319139	3.216214	O	-2.527126	-0.933870	-0.653687
H	-0.353976	0.361576	2.072619	H	0.010086	0.937626	1.488126	H	-3.485980	-0.972769	-0.791671
H	-1.102102	-3.862317	0.367359	H	-1.927725	-2.802623	0.008528	H	0.823633	0.884235	3.574059
O	-1.512317	-3.763116	-0.510407	O	-2.401841	-2.254927	-0.648266	O	-0.020686	0.778512	3.120901
H	-1.521565	-4.652643	-0.878860	H	-3.334335	-2.477191	-0.546778	H	-0.164483	-0.179955	3.063371
H	-1.018723	-4.375955	2.672674	H	-0.971137	-4.450481	1.467351	H	-2.123471	-2.041067	0.155318
O	-0.357507	-4.073486	2.039529	O	-0.821863	-3.538515	1.194511	H	-2.495801	-3.432862	0.986123
H	0.092367	-3.333574	2.477597	H	-0.540995	-3.065414	1.997909	O	-1.764830	-2.845071	0.737536
H	1.988527	-1.960141	2.995988	H	0.899339	-2.496669	3.736680	H	-1.279546	-2.524258	1.585700
O	1.093025	-1.918209	3.360792	O	0.063244	-2.096619	3.463520	H	0.356371	-2.493879	2.822262
H	1.176745	-2.190608	4.285759	H	-0.533327	-2.232415	4.211523	O	-0.511775	-2.065469	2.803499
C	-2.440476	-1.370184	1.697132	C	-2.505726	-0.250127	1.514500	H	-0.953767	-2.345519	3.617925
H	-1.660677	-1.706562	2.382877	H	-1.879362	-0.875346	2.149917	H	-3.830248	0.381804	1.277036
H	-3.048516	-2.231282	1.410348	H	-3.477075	-0.720687	1.340966	H	-2.480490	1.415962	1.770840
H	-3.084112	-0.651508	2.207802	H	-2.718021	0.707389	2.001951	H	-2.348195	-0.344113	1.945268
H	-3.560059	1.303159	0.197872	H	-3.787089	2.020070	0.062270	H	-1.560903	2.771450	-0.266523
H	1.659691	0.289849	-0.862180	H	1.823248	0.439892	-1.104341	H	1.771728	0.817292	-0.816520
$G^*_{\text{ZPE}} = -728.08270$ Hartrees				$G^*_{\text{ZPE}} = -728.06772$ Hartrees				$G^*_{\text{ZPE}} = -728.11281$ Hartrees			
0 imaginary frequencies				$i/333.94 \text{ cm}^{-1}$				0 imaginary frequencies			

22. β -IEPOX hydrolysis (1-octanol continuum, H_3O^+ , 3 H_2O)



Reactants

C	-1.826024	-0.769143	0.481995
C	-0.422855	-1.024445	0.140973
H	0.129504	-1.650685	0.840467
C	-2.822034	-0.370605	-0.579756
C	0.195965	-0.944774	-1.223795
O	-0.797780	0.255261	0.736000
O	1.573951	-0.615913	-1.118617
O	-3.884201	0.397431	-0.041837
H	-3.261644	-1.286049	-0.981842
H	-2.330498	0.166868	-1.392413
H	-0.324531	-0.226023	-1.859053
H	0.132489	-1.932293	-1.682998
H	0.624660	1.139717	3.180770
O	0.017975	0.397612	3.034118
H	0.488278	-0.486227	3.231550
H	-0.352937	0.389733	2.018239
H	-1.167241	-3.811811	0.326930
O	-1.606313	-3.658317	-0.527495
H	-1.616261	-4.518824	-0.958696
H	-0.906492	-4.537895	2.595656
O	-0.346974	-4.064418	1.970328
H	0.046330	-3.336538	2.474900
H	1.947752	-2.006858	3.086553
O	1.038413	-1.907872	3.401978
H	1.055604	-2.179269	4.330781
C	-2.403360	-1.375262	1.730942
H	-1.613086	-1.669154	2.424580
H	-2.981065	-2.262734	1.463397
H	-3.071218	-0.665367	2.221837
H	-3.549995	1.274572	0.176991
H	1.651433	0.324358	-0.922539

$G^*_{\text{ZPE}} = -728.07001$ Hartrees

0 imaginary frequencies

Transition State

C	-1.924842	0.006508	0.213233
C	-0.505100	-0.329401	0.065389
H	-0.159557	-1.079365	0.780095
C	-2.633028	0.727829	-0.885774
C	0.122400	-0.533123	-1.294379
O	-0.278609	1.005790	0.558507
O	1.531701	-0.514490	-1.167429
O	-3.903383	1.203883	-0.526738
H	-2.757665	0.001875	-1.696019
H	-1.984225	1.525253	-1.260595
H	-0.212357	0.227439	-2.005561
H	-0.162306	-1.516276	-1.669374
H	0.871683	1.142402	3.493471
O	0.148547	0.655847	3.080843
H	0.313195	-0.289519	3.261158
H	-0.074112	0.947157	1.541282
H	-2.044783	-2.804560	0.070190
O	-2.451874	-2.239658	-0.615419
H	-3.385152	-2.480015	-0.632942
H	-0.673302	-4.390797	1.211684
O	-1.134861	-3.565848	1.396860
H	-0.580713	-3.088313	2.037727
H	1.312113	-2.387408	3.210764
O	0.390425	-2.112264	3.301074
H	0.119741	-2.428364	4.173008
C	-2.600369	-0.193381	1.508079
H	-2.012088	-0.803404	2.191262
H	-3.572054	-0.654226	1.312577
H	-2.814774	0.782315	1.956533
H	-3.809448	2.040747	-0.057340
H	1.809401	0.388396	-0.973340

$G^*_{\text{ZPE}} = -728.05748$ Hartrees

$i/338.86 \text{ cm}^{-1}$

$i/18.12 \text{ cm}^{-1}$

7 -- 24 bond distance fixed

Products

C	-2.165997	0.348428	-0.088413
C	-0.623401	0.346843	-0.061846
H	-0.297300	-0.501110	0.558612
C	-2.745113	1.421668	-1.016463
C	-2.735750	0.454029	1.315833
C	0.031918	0.209441	-1.426316
O	-0.145447	1.560125	0.499218
O	1.433644	0.043928	-1.279413
O	-2.491374	2.732936	-0.550974
H	-3.831986	1.300259	-1.050528
H	-2.361485	1.278937	-2.032744
H	-0.180430	1.097317	-2.032275
H	-0.341341	-0.671735	-1.948710
H	-0.084056	1.444304	1.467745
O	-2.526679	-0.938713	-0.655361
H	-3.479312	-0.964098	-0.832123
H	0.804439	0.929025	3.594718
O	-0.019570	0.777535	3.118620
H	-0.131319	-0.185134	3.087281
H	-2.145927	-2.020723	0.145391
H	-2.508334	-3.435796	0.977223
O	-1.788793	-2.829189	0.743368
H	-1.304157	-2.521351	1.590574
H	0.358802	-2.511591	2.814528
O	-0.506857	-2.078015	2.815301
H	-0.929757	-2.349024	3.642436
H	-3.825831	0.375423	1.281088
H	-2.478255	1.411715	1.769679
H	-2.340785	-0.345774	1.948364
H	-1.566736	2.775510	-0.266491
H	1.770870	0.827055	-0.828031

$G^*_{\text{ZPE}} = -728.09922$ Hartrees

0 imaginary frequencies