

# **Oceanic methane concentrations in three Mexican regions**

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## The role of methane as green-house gas

Methane (CH<sub>4</sub>) is a important atmospheric gas with a potent role as an infrared trapping molecule. It is about 300 times more effective for trapping IR radiation than CO2 on a molecule per molecule basis.

Methane concentration has been slowly increasing and it has duplicated its concentration over the last 200 years (Figure 1).

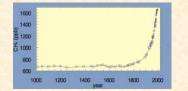


Figure 1. Atmospheric methane concentration vs. time (Lowe and Etheridge, 2001)

#### Methane in the oceans

Although the CH<sub>4</sub> contribution to the atmosphere by the oceans is relatively minor (< 2%; Reeburgh, 2007), it has a large uncertainty and it has been stated that highly productive coastal regions are responsible for as much as 75% of the total CH4 oceanic emissions to the atmosphere (Bange et al., 1998).

Two facts make the study of CH4 in the oceans a necessity. First, its concentration appears to be related to hypoxic and anoxic waters around the world. The distribution of waters with low oxygen is increasing due to anthropogenic processes (Naqvi et al., 2009). Second, there is a risk of destabilization of the large deposits of CH<sub>4</sub> hydrates that are trapped in many oceans around the world (Westbrook et al., 2009).

There are other activities at regional scale that have proven to be of significant importance as contributors to the CH4 budget such as (i.e. rice paddies and dairy farms), however its contribution to global budgets and new sources are still unknown.

The origin of CH4 in the ocean is mainly from two sources: hydrothermal and biogenic. In surface ocean waters, there is a slight supersaturation of CH4 with respect to atmospheric concentration. However, the origin and presence of CH<sub>4</sub> in oxygenated waters is still in debate, this is referred as to "the methane paradox". The presence of CH4 in oxygenated waters has been explained after being released from anoxic microenvironments such as organic particulate matter, from digestive tracts of zooplankton and from its production in pore water from sediments, particularly in shallow basins.

#### Sampling areas

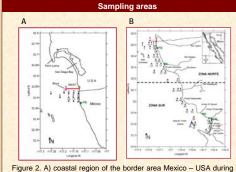
We have measured the dissolved methane concentration in three different areas in coastal waters off Baja California. Each area has a particular characteristic and, consequently, all the CH4 fluxes are related to different sources and processes.

#### The study areas are:

1) The coastal Pacific region near the international border and from this area to Punta Banda, Baja California representing the Mexican sector of the Southern California Bight, SCBMex (Figure 2).

2) The IMECOCAL region, from the port of Ensenada to the south of Punta Santa Eugenia (Figure 3).

3) Northern part of the Gulf of California, sampled during the "sill's" 3rd leg campaign (Figure 4).



rainy event and, B) SCBMex after a rainy event.

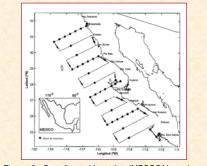


Figure 3. Sampling grid at the IMECOCAL region. Methane was collected during the months of April and July, 2006

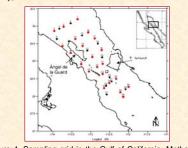


Figure 4. Sampling grid in the Gulf of California. Methane was collected during the months of April and July, 2004

# Methods and Materials

Water samples were collected using Niskin bottles at different standard depths and transferred to amber glass bottles, poisoned with 1.5 ml of HgCl<sub>2</sub> to avoid biological activity. All samples were hermetically sealed and stored in a cool dark place until further analysis.

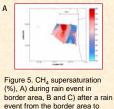
For CH<sub>4</sub> extraction, we used the gas-water phase equilibrium method (McAulliffe, 1971; Capasso e Inguaggiato, 1998) at standard temperature (25 °C) and pressure (1 atm). We used He zero grade as inert gas phase.

The quantification of CH<sub>4</sub> was carried out in a HP-6890 Gas Cromatograph equiped with FID detector.

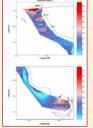
An external calibration was used with 7 different gas concentration mixtures (CH<sub>4</sub>:He) using a NIST traceable CH<sub>4</sub> standard (Scott-Marrin Inc.).

### Results and discussions

The results of CH<sub>4</sub> supersaturation (%) at the border coastal area (Figure 5) after a rain event showed a higher values apparently from the high loads of organic material acting as anoxic micro environments.

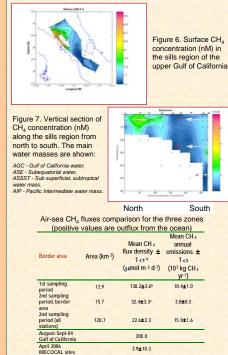


Punta Banda (SCBMex)



In the Gulf of California, the largest concentration measured occurred around 50 m depth. Surprisingly, the Gulf of California water mass appears to transport the largest methane concentrations in the area in a southward direction (Figure 7).

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Take home message

0.14±1.03

July 200

**IMECOCAL** site

The largest CH4 fluxes were measured at the Gulf of California. This is expected due to the larger productivity characteristic in this region. In the border area, CH4 was used as a tracer of wastewater inflows to the coast. The constant particulate organic material discharged in the wastewater flows acts as suitable microenvironments for local CH<sub>4</sub> production in oxygenated waters of the region.

# References

Bange et al., 1998. Methane in surface waters of the Arabian sea. Geophys. Res. Lett., 25:3547-3550

23.5977-5500. Capasso, G. & Inguaggiato, S. 1998. A simple method for the determination of dissolved gases in natural waters. An application to water from Vulcano Island. *Applied Geochem.* 13:631-642.

13:631-642. Lowe, D., and Ethenidge. 2001. Antarctic ice:the world's air musseum. Water and Atmosphere. 9(1). McAulifie, C.C. 1971. GC determination of solutes by multiple phase equilibration. Chem. Technol. 1, 46-51. Reekurgh, W. S. 2007. Oceanic methane biogeochemistry, Chem. Rev., 107, 486-513. Weis, 1970. The solubility of Nitrogen, oxygen and argon on water and seawater. Deep-Sea Research 17:721-735. Westbrook, G. K., et al., 2009. Escape of methane gas from the seabed along the West Splisbergen continental Margin. Geptys. Res. Lett., 36. L15608.

