

CHAPTER 9

INTEGRATIVE SYNTHESIS

Table of Contents

	<u>Page</u>
9. INTEGRATIVE SYNTHESIS	9-1
9.1 INTRODUCTION	9-1
9.2 SYNTHESIS OF AVAILABLE INFORMATION ON PM-RELATED HEALTH EFFECTS	9-2
9.2.1 Does the Newly Available Information Continue to Support Consideration of Fine and Coarse Particles as Separate Subclasses of PM Pollution?	9-4
9.2.1.1 Key Points from Previous PM NAAQS Reviews	9-4
9.2.1.2 Integration of New Information	9-6
9.2.1.2.1 Physics and chemistry considerations	9-6
9.2.1.2.2 Exposure-related considerations	9-9
9.2.1.2.3 Dosimetric considerations	9-13
9.2.1.3 Summary and Conclusions	9-15
9.2.2 How Does the Newly Available Information Inform Our Judgments about the Strengths and Limitations of the Epidemiologic Evidence for Health Effects Related to Ambient Fine and Coarse Thoracic PM, Acting Alone and/or in Combination With Other Pollutants?	9-17
9.2.2.1 Key Points from 1996 Integrative Synthesis	9-18
9.2.2.2 Integration of New Information	9-19
9.2.2.2.1 Strength of epidemiologic evidence on health effects associations with PM	9-20
9.2.2.2.2 Assessment of robustness of associations for epidemiologic studies	9-29
9.2.2.2.3 Assessment of consistency in epidemiologic study results	9-33
9.2.2.2.4 Lag period between exposure and effect	9-35
9.2.2.2.5 Form of concentration-response function	9-37
9.2.2.2.6 Intervention studies	9-38
9.2.2.2.7 Summary and conclusions	9-39
9.2.3 How Does Newly Available Information Inform Assessment of Biological Plausibility and Coherence of Health Effects Attributed to Ambient Fine and Coarse Thoracic PM and/or Their Components?	9-40
9.2.3.1 Key Points from 1996 Integrative Synthesis	9-40
9.2.3.2 Integration of New Evidence	9-41
9.2.3.2.1 Chemical components and source categories associated with health effects in epidemiologic studies	9-41
9.2.3.2.2 Approaches to experimental evaluation of PM health effects	9-45
9.2.3.2.3 Interspecies comparisons of experimental results	9-49

Table of Contents
(cont'd)

		<u>Page</u>
	9.2.3.2.4 General overview of toxicologic findings . . .	9-54
	9.2.3.2.5 Links between specific PM components/ characteristics and health effects	9-60
	9.2.3.2.6 Mechanisms of action	9-70
	9.2.3.2.7 Inhaled particles as potential carriers of toxic agents	9-73
	9.2.3.2.8 Coherence of evidence	9-80
	9.2.3.2.9 Summary and conclusions	9-85
9.2.4	How Does Newly Available Information Inform Our Understanding of Subpopulations Potentially Susceptible to PM-Related Health Effects?	9-86
	9.2.4.1 Key Points from 1996 Integrative Synthesis	9-86
	9.2.4.2 Integration of Newly Available Information	9-87
	9.2.4.2.1 Preexisting disease as a risk factor for particulate matter health effects	9-87
	9.2.4.2.2 Age-related at-risk population groups: the elderly and children	9-89
	9.2.4.2.3 Genetic susceptibility	9-91
	9.2.4.2.4 Gender	9-92
	9.2.4.2.5 Socioeconomic status	9-92
	9.2.4.2.6 Enhanced vulnerability due to heightened exposure levels	9-94
	9.2.4.3 Summary and Conclusions	9-95
9.2.5	What Does the Newly Available Information Imply With Regard to Potential Public Health Impacts of Human Exposures to Ambient PM in the United States?	9-95
	9.2.5.1 Key Points from 1996 Integrative Synthesis	9-95
	9.2.5.2 Integration of New Information	9-95
	9.2.5.2.1 Magnitude of susceptible groups	9-95
	9.2.5.2.2 Evidence of new endpoints and potentially susceptible groups	9-96
	9.2.5.2.3 Impact on life-expectancy	9-100
	9.2.5.3 Summary and Conclusions	9-101
9.3	SYNTHESIS OF AVAILABLE INFORMATION ON PM-RELATED WELFARE EFFECTS	9-102
	9.3.1 Airborne Particle Effects on Visibility	9-102
	9.3.1.1 How Does Newly Available Information Inform Our Understanding of How Ambient PM and Its Major Constituents Affect Visibility?	9-102
	9.3.1.2 How Does Newly Available Information Inform Our Understanding of How the Public Values Improvements in Visibility, Especially in Urban Areas?	9-105

Table of Contents
(cont'd)

		<u>Page</u>
9.3.2	Effects of Ambient PM on Vegetation and Ecosystems	9-106
9.3.2.1	What Are the Direct and Indirect Effects of Ambient PM?	9-106
9.3.2.2	What are the Components in Ambient PM that are Major Ecosystem Stressors?	9-107
9.3.2.2.1	Nitrogen	9-108
9.3.2.2.2	Acidification from PM Deposition	9-113
9.3.2.3	How can Exposures of Concern for Ecosystem Stressor Components of PM be Characterized?	9-114
9.3.2.4	Summary and Conclusions	9-115
9.3.3	What Does the Available Information Indicate About the Relationships Between Atmospheric PM and Climate Change Processes?	9-116
9.3.3.1	Key Points from 1996 PM AQCD	9-116
9.3.3.2	Integration of New Information	9-116
9.3.4	What Does the Available Information Indicate About the Effects on Man-Made Materials Associated With Ambient PM and its Major Constituents?	9-118
9.3.4.1	Key Points from 1996 PM AQCD	9-118
9.3.4.2	Integration of New Information	9-118
REFERENCES	9-120
APPENDIX 9A	Key Quantitative Estimates of Relative Risk for Particulate Matter-Related Health Effects Based on Epidemiologic Studies of U.S. and Canadian Cities	9A-1

List of Tables

<u>Number</u>		<u>Page</u>
9-1	Comparison of Ambient Particles, Fine Particles (ultrafine plus accumulation-mode) and Coarse Particles	9-10
9-2	Exposure-Related Relationships for Particle Size Fractions	9-12
9-3	Particulate Matter Characteristics, Components, or Source Categories Shown to be Associated with Mortality in U.S., Canadian, or European Epidemiologic Studies	9-43
9-4	CAPS: Rat and Human Inhalation Study Comparisons	9-53
9-5	Prevalence of Selected Cardiorespiratory Disorders by Age Group and by Geographic Region, 2000 (reported as percent or numbers of cases in millions)	9-97
9-6	Number of Acute Respiratory Conditions per 100 Persons per Year, by Age: United States, 1996	9-98
9A-1	Estimated Total, Cardiovascular, and Respiratory Mortality Effect Sizes per Increments in 24-h Concentrations of PM ₁₀ , PM _{2.5} , and PM _{10-2.5} from U.S. and Canadian Studies	9A-2
9A-2	Cardiovascular and Respiratory-Related Morbidity Effect Size Estimates per Increment in 24-h Concentrations of PM ₁₀ , PM _{2.5} , and PM _{10-2.5} in U.S. and Canadian Studies	9A-11
9-A3	Effect Estimates per Increments in Long-Term Mean Levels of Fine and Coarse Fraction Particle Indicators from U.S. and Canadian Studies	9A-22

List of Figures

<u>Number</u>		<u>Page</u>
9-1	Volume size distribution, measured in traffic, showing fine and coarse particles and the Aitken and accumulation modes of fine particles; VMD (volume median diameter) and σ_g (geometric standard deviation); and formation and growth mechanisms	9-7
9-2	Submicron number size distribution observed in a boreal forest in Finland showing the tri-modal structure of fine particles	9-8
9-3	Geometric mean infiltration factor (indoor/outdoor ratio) for hourly nighttime, nonsource data for two seasons	9-12
9-4	Deposition fraction as a function of particle size for nasal and oral breathing during rest and exercise: (a) extrathoracic (ET), (b) tracheobronchial (TB), and (c) alveolar (A) regions	9-14
9-5	Excess risk estimates for total non-accidental, cardiovascular, and respiratory mortality in single-pollutant models for U.S. and Canadian studies	9-21
9-6	Excess risk estimates for hospital admissions and emergency department visits for cardiovascular and respiratory diseases in single-pollutant models from U.S. and Canadian studies	9-22
9-7	Inhalation rates on a per body-weight basis for males (■) and females (▲) by age (Layton, 1993)	9-91

9. INTEGRATIVE SYNTHESIS

9.1 INTRODUCTION

This chapter integrates key information drawn from the preceding detailed chapters, in order to provide coherent frameworks for assessment of human health and welfare risks posed by ambient PM in the United States. This chapter focuses on integrating newly available scientific information with the information available in the last review to address a series of overarching questions central to EPA's assessment of scientific information upon which the PM NAAQS review is to be based.

As such, this chapter is not intended merely as an Executive Summary of the information presented in the earlier chapters; nor is its goal to simply resummairize key information from those chapters. Rather, the goal of this chapter, in particular, is to provide an integrative exposition of the scientific basis for the Agency's review of the PM NAAQS. More specifically, this chapter attempts to provide an updated syntheses of scientific information in a manner so as to facilitate consideration of the key policy-related NAAQS issues to be addressed in the PM Staff Paper, prepared by EPA's Office of Air Quality Planning and Standards (OAQPS) staff. These policy-related issues include consideration of information that will facilitate selection of appropriate indicators, averaging times, forms, and levels for primary and secondary PM NAAQS in the United States. EPA's consideration of these issues will be informed not only by the scientific information and integrative assessment presented here and throughout this document, but also by additional policy evaluations of scientific and technical information to be included in the PM Staff Paper so as to “bridge the gap” between the scientific review and the judgments required of the EPA Administrator in deciding whether to retain or revise the existing PM NAAQS.

While this synthesis focuses on what has been learned from the new information that has become available since the last PM NAAQS review, it also highlights important uncertainties that remain and recognizes the value of continuing research in a number of key areas. Although the delineation of detailed research recommendations in these areas is beyond the scope of this document, such recommendations are to be discussed in later PM research needs documents and/or research plans to be prepared by EPA.

In considering the PM-related health effects information, Section 9.2 builds specifically upon the integrative synthesis presented in Chapter 13 of the 1996 PM AQCD (U.S.

1 Environmental Protection Agency, 1996). The synthesis of PM-related health effects
2 information in Section 9.2 is organized around five key questions dealing with the following
3 central issues in assessing the available scientific information: (1) consideration of fine and
4 coarse thoracic particles as separate subclasses of PM pollution, taking into account atmospheric
5 science, exposure-related and dosimetric information; (2) the strengths and limitations of the
6 epidemiological evidence of associations between health effects and fine and coarse thoracic PM
7 within the mix of ambient air pollutants; (3) the extent to which effects observed in
8 epidemiologic studies can plausibly be attributed to various indicators or constituents of ambient
9 PM, acting alone and/or in combination with other pollutants, based on consideration of
10 dosimetric, toxicologic, and other types of information; (4) characterization of susceptible
11 subpopulations potentially at increased risk for PM-related health effects and factors enhancing
12 such risk; and (5) potential public health impacts of human exposures to ambient PM in the
13 United States.

14 With regard to considering PM-related welfare effects information important for decisions
15 related to secondary standards, Section 9.3 then addresses each of the major types of welfare
16 effects, first building upon information presented in the 1996 PM AQCD where possible. This
17 includes allusion to key findings and conclusions on visibility and climate effects from Chapter 8
18 and on damage to manmade materials from Chapter 9 of the 1996 document and consideration of
19 new findings discussed in Chapter 4 of this document. However, PM-related effects on
20 vegetation and ecosystems were not addressed in the 1996 PM AQCD; and, so, the discussion of
21 information concerning such effects is based entirely on pertinent findings characterized in
22 Chapter 4 of this current document. Each subsection within Section 9.3 is organized around
23 specific questions that serve to synthesize the available scientific information relevant to each of
24 these classes of PM-related welfare effects.

25 26 27 **9.2 SYNTHESIS OF AVAILABLE INFORMATION ON PM-RELATED** 28 **HEALTH EFFECTS**

29 The integrative synthesis of the latest available information on PM-related health effects
30 poses especially large challenges in view of:
31

- 1 (1) The unprecedented amount of new information generated since the 1996 PM AQCD
adds greatly to the complexity of any integrative assessment;
- 2 (2) Extensive new information available from epidemiologic studies, while reflecting much
progress in addressing many research recommendations from the last review, also raises
new issues or resurfaces issues earlier thought to have been adequately addressed but
which remain important in interpreting the body of epidemiologic evidence and the
characterization of its strengths and limitations;
- 3 (3) Much new information from dosimetric and toxicologic studies, making notable
progress toward identifying and exploring potential mechanisms of action and
characteristics of PM that may underlie health effects observed in experimental studies,
but still leaving open many issues to be more fully addressed in the future.

4 Thus, despite substantial progress, uncertainties remain in integrating these different types of
5 evidence into a coherent synthesis.

6 The present Section 9.2 is organized so as to first address the question of whether there is
7 continued support for considering fine and coarse thoracic PM as separate subclasses of PM
8 based on atmospheric science, air quality, exposure, and dosimetric information. Next, the
9 strengths and limitations of epidemiologic evidence are evaluated, taking into account various
10 factors, such as consideration of the magnitude, statistical significance, precision, and robustness
11 of reported associations; assessment of the consistency or general concordance of study results
12 and consideration of potential reasons for observed differences; and information from so-called
13 intervention studies of “natural” or “found” experiments. Looking beyond just epidemiologic
14 evidence, consideration is then given to toxicological and other information bearing on the
15 biological plausibility of the PM-effects associations observed in the epidemiologic studies and
16 the coherence of the effects associations to reach conclusions as to the extent to which observed
17 effects can be attributed to ambient fine and coarse thoracic PM, acting alone and in combination
18 with other pollutants. This is then followed by discussion of evidence regarding various risk
19 factors (e.g., pre-existing disease and age-related factors) to reach conclusions as to which
20 susceptible subpopulations may be potentially at special risk for health effects related to fine and
21 coarse thoracic PM. Finally, information on the magnitude of population subgroups having
22 health conditions thought to put them at increased risk for PM effects is discussed, to provide
23 inputs to evaluation of potential public health impacts of exposures to ambient fine and coarse
24 thoracic PM in the United States.

25

1 **9.2.1 Does the Newly Available Information Continue to Support**
2 **Consideration of Fine and Coarse Particles as Separate Subclasses**
3 **of PM Pollution?**

4 This question is addressed below by drawing upon the information and assessments found
5 primarily in Chapters 2, 3, 5, and 6 of this document, concerning: the physics and chemistry of
6 particle pollution, the measurement of airborne particles, relationships between ambient PM
7 concentrations and population exposure, and PM dosimetry. The focus here is on whether the
8 newly available science in these areas continues to support consideration of fine and coarse
9 thoracic PM separately in the context of the Agency's periodic review of the PM NAAQS.
10 The scientific bases for selecting certain key features of such standards, including health effects
11 information and assessments presented in Chapters 7 and 8, will then be considered in
12 addressing the subsequent questions posed and addressed beyond this Section.
13

14 **9.2.1.1 Key Points from Previous PM NAAQS Reviews**

15 The primary focus in the last review was on thoracic particles, defined for regulatory
16 purposes as being indexed by an indicator of PM₁₀, and whether fine and coarse thoracic
17 particles should be addressed by separate standards with different indicators. The 1996 PM
18 AQCD noted that the PM₁₀ indicator was established as a result of the 1987 PM NAAQS review,
19 which concluded that the indicator for primary standards should represent those particles small
20 enough to penetrate to the thoracic region (including the tracheobronchial and pulmonary
21 regions) of the lower respiratory tract and generally exclude particles that deposit only in the
22 extrathoracic region (the latter being particles previously included in the original TSP indicator).

23 As discussed in the 1996 PM AQCD, the natural division of ambient PM into fine particles
24 and coarse particles has been understood since it was enunciated by Whitby (1978):

25 “The distinction between “fine particles” and “coarse particles” is a fundamental one.
26 There is now an overwhelming amount of evidence that not only are two modes in the
27 mass or volume distribution usually observed, but that these fine and coarse modes are
28 usually chemically quite different. The physical separation of the fine and coarse modes
29 originates because condensation produces fine particles while mechanical processes
30 produce mostly coarse particles . . . the dynamics of fine particle growth ordinarily
31 operate to prevent the fine particles from growing larger than about 1 μm. Thus, the
32 fine and coarse modes originate separately, are transformed separately, are removed
33 separately, and are usually chemically different . . . practically all of the sulfur found in
34 atmospheric aerosol is found in the fine particle fraction. Thus, the distinction between
35 fine and coarse fractions is of fundamental importance to any discussion of aerosol
36 physics, chemistry, measurement, or aerosol air quality standards.”
37

1 Consistent with this view, the 1996 PM AQCD stated that the evidence indicates that
2 “it would be appropriate to consider fine and coarse mode particles as separate subclasses”
3 of PM pollution. This conclusion was based on various considerations:

- 4 (1) Differences in formation processes and sources of fine- and coarse-mode thoracic
particles, as well as differences in chemical and physical properties, atmospheric
residence times and distances transported in the atmosphere;
- 5 (2) Resulting differences in patterns of ambient population exposures to fine- and
coarse-mode thoracic particles;
- 6 (3) Evidence from dosimetric studies showing differences in the fractions inhaled,
deposited, and/or retained in various regions of the respiratory tract for fine- versus
coarse-mode thoracic particles; and
- 7 (4) Evidence from health studies leading to conclusions that fine particles are more
strongly associated with more serious health effects and that chemical components
likely to have higher relative toxicity primarily occur in the fine fraction.

8 The EPA’s selection of 2.5 μm as the appropriate cut-point between fine and coarse
9 thoracic particles for use in defining an indicator for fine particle standards (i.e., $\text{PM}_{2.5}$) was
10 based primarily on the following considerations:

- 11 (1) Recognition that overlap between fine and coarse thoracic particles occurs generally
12 between 1 and 3 μm ; within this range, no one cut-point would clearly separate fine-
13 and coarse-mode particles in all areas. That is, although fine particles are generally
14 below 1 μm in size, under high humidity conditions, constituent accumulation-mode
15 particles may grow above 1 μm . Also, in clouds or fogs, accumulation-mode particles
16 may grow to above 2.5 μm , and reactions of gases that dissolve in clouds or fog
17 droplets may lead to formation of fine particles larger than 1 μm . As for coarse-mode
18 particles, in dry dusty areas, resuspended coarse-mode soil particles may extend down
19 to about 1 μm ; and, in cities near oceans, coarse-mode sea salt particles may be found
20 in the 1 to 2.5 μm range.
21
22
- 23 (2) The 2.5 μm cut-point was selected mainly to reflect the regulatory importance that
24 was placed on defining an indicator for fine particle standards that would more
25 completely capture fine-mode particles under all conditions likely to be encountered
26 across the U.S., while recognizing that some small coarse-mode particles would also
27 be captured by $\text{PM}_{2.5}$ monitoring.
28
- 29 (3) The decision to retain PM_{10} as an indicator for standards to address coarse-mode
30 particles, rather than an indicator that would not also encompass fine particles
31 (e.g., $\text{PM}_{10-2.5}$), was based in large part on the very limited epidemiologic studies and
32 air quality data specifically available for coarse-mode thoracic particles beyond that

1 which could be inferred or derived from PM₁₀ studies and data in areas dominated by
2 coarse-mode particles.¹
3

4 **9.2.1.2 Integration of New Information**

5 The ensuing discussion builds upon the information base provided by the most salient key
6 findings from the previous PM NAAQS review(s), while updating and integrating key findings
7 and conclusions from the newly available studies assessed in earlier chapters of this document.

8 As a consequence of the decisions made by EPA in the last PM NAAQS review on the
9 separate indicators for fine and coarse thoracic PM, a national PM_{2.5} monitoring network was
10 established that has provided extensive air quality data on PM_{2.5} and, by difference between co-
11 located PM₁₀ and PM_{2.5} monitors, more limited data on PM_{10-2.5}. The availability of such air
12 quality data has prompted the increased use of PM_{2.5} and, to a far lesser degree, PM_{10-2.5} as
13 indicators in new epidemiologic studies, as well as increasing focus on these PM size fractions in
14 other types of studies (exposure, dosimetry, toxicology, etc.).
15

16 **9.2.1.2.1 Physics and chemistry considerations**

17 Since the last PM NAAQS review, the physical and chemical properties of fine and coarse
18 particles have become better understood. Nonetheless, the fundamental concept of the natural
19 division of thoracic particles into somewhat overlapping ranges of fine and coarse particles, as
20 illustrated in Figure 9-1, remains unchanged. Improved measurement techniques have provided
21 additional information that refines the general characterization of particles below 0.1 μm
22 diameter (i.e., ultrafine particles) from a single mode to a bi-modal structure. Thus, as shown in
23 Figure 9-2, fine particles are now divided into three modes: a nucleation mode (< 0.01 μm);
24 an Aitken mode (~0.01 to ~0.1 μm); and an accumulation mode (~0.1 to ~1.0 μm). The
25 nucleation mode is transient and rapidly grows into the Aitken mode. The Aitken mode grows
26 more slowly into the accumulation mode. Under normal atmospheric conditions, the
27 accumulation mode does not grow into the coarse mode. However, as was previously
28 recognized, an overlap between accumulation-mode fine particles and coarse particles can occur
29 at times between 1 and 3 μm. For example, under high humidity conditions, accumulation-mode

¹ As discussed in Chapter 1, subsequent litigation resulted in the court finding the use of PM₁₀ as an indicator for coarse-mode particles (in conjunction with PM_{2.5} standards) to be arbitrary, since PM₁₀ includes all fine particles; the court remanded this aspect of EPA's 1997 decision to the Agency for further consideration.

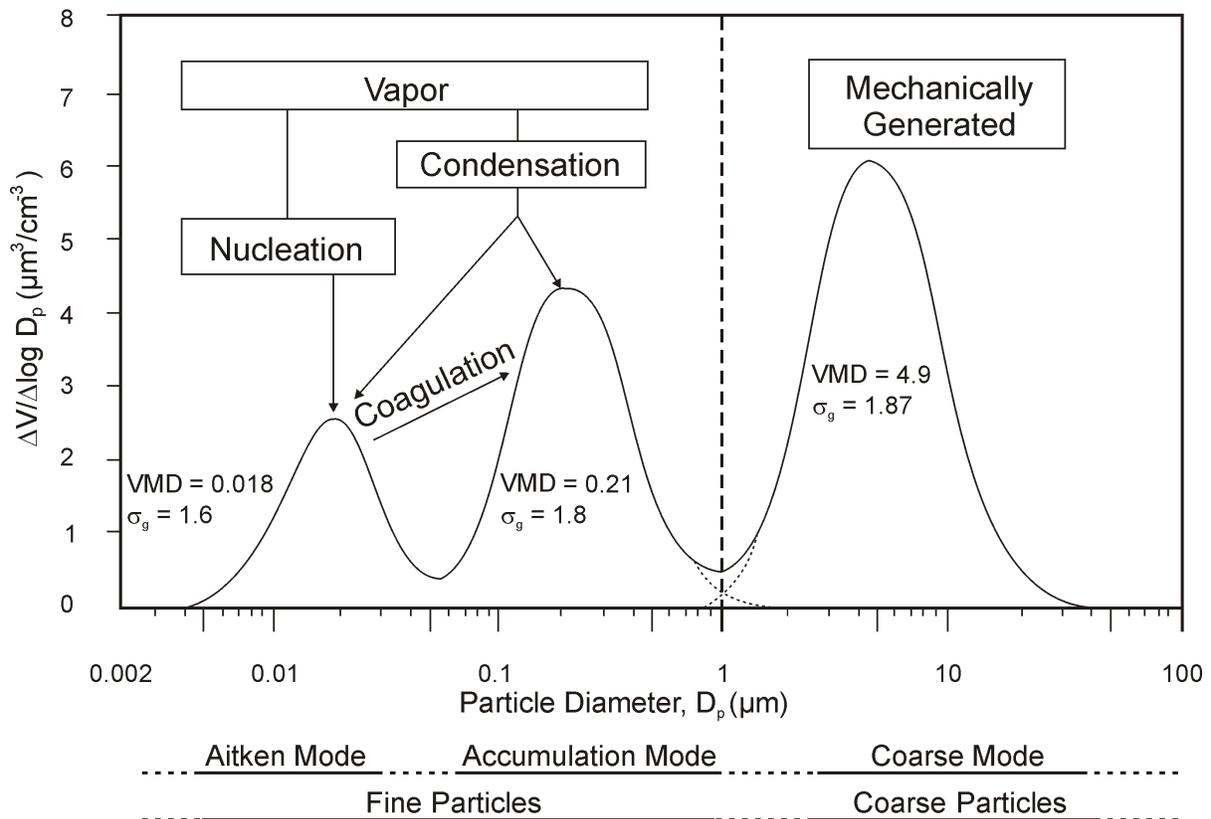


Figure 9-1. Volume size distribution, measured in traffic, showing fine and coarse particles and the Aitken and accumulation modes of fine particles; VMD (volume median diameter) and σ_g (geometric standard deviation); and formation and growth mechanisms.

Source: Adapted from Wilson and Suh (1997).

1 particles may grow above 1 μm , and in dry dusty areas, coarse PM (e.g., resuspended soil) may
 2 have a tail reaching to 1 μm or below. Sampling fine and coarse particles using a cut point of
 3 2.5 μm , as was specified as a regulatory choice in the last PM NAAQS review, thus helps to
 4 ensure that high concentrations of accumulation-mode particles will be collected even under
 5 high humidity conditions. However, it is recognized that a 1 μm cut point, using monitors that
 6 dehumidify the airstream before collecting particles, might give a better separation of fine- and
 7 coarse-mode particles, especially in dry, dusty areas.

8 It takes increasing amounts of energy to break non-biological materials into smaller and
 9 smaller particles. As a result, natural processes, such as suspension of soil dust by wind,

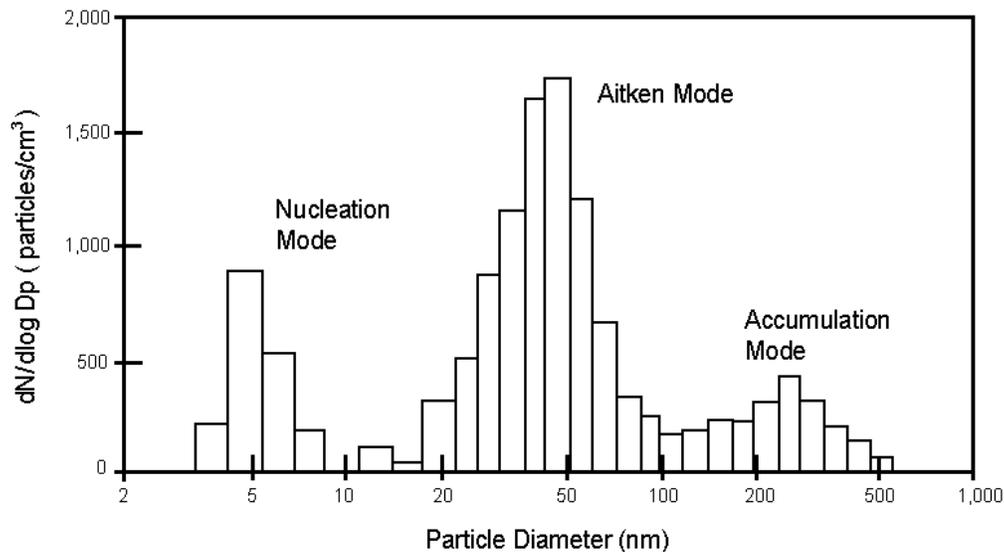


Figure 9-2. Submicron number size distribution observed in a boreal forest in Finland showing the tri-modal structure of fine particles.

Source: Mäkelä et al. (1997).

1 produce few particles below 1 μm in diameter. However, in the years since the 1996 PM
 2 AQCD, it has been discovered that biological material, although originally in the coarse mode,
 3 may deteriorate or fragment and produce particles in the fine-particle size range. Thus,
 4 fragments of pollen, endotoxins, and other biological material may be found in the fine-particle
 5 size range. Progress has also been made in understanding the semivolatile components of PM
 6 (particle-bound water, ammonium nitrate, and semivolatile organic compounds) and new
 7 techniques have been developed to measure the semivolatile components of mass, either
 8 separately or included with the nonvolatile component. Much progress has also been made in
 9 understanding the many organic compounds formed in the atmospheric reactions of biogenic and
 10 anthropogenic hydrocarbons, including condensible species that form organic particles. Progress
 11 also has been made in measurement of carbonaceous particles from diesel engines.

12 Progress of the above types has helped to enhance our understanding of ambient aerosol
 13 components and interrelationships between them that may contribute to ambient PM-related
 14 effects. Of much importance, for example, is emerging new evidence related to the role of
 15 particle-bound water and associated submicron PM constituents serving as vectors by which

1 water soluble gases (e.g., SO₂), short-lived reactive species (e.g., peroxides), and organic species
2 (e.g., formaldehyde) present in atmospheric aerosol mixes can be delivered in enhanced
3 proportions to lower regions of the respiratory tract (as discussed in Section 9.2.3). The
4 importance of nonbiological ambient PM components serving as carriers or vectors enhancing
5 deposition of bioaerosols (e.g., allergen-laden pollen fragments and endotoxins) in the lower
6 respiratory tract is also noted in Section 9.2.3.

7 The 1996 PM AQCD listed properties of fine and coarse particles. Because of the
8 increasing interest in ultrafine particles and additional information on their properties, this
9 current document provides new information on the chemical and physical properties of ultrafine
10 and accumulation-mode fine particles and coarse particles, as shown in Table 9-1. As shown,
11 ultrafine and accumulation-mode particles share similar formation processes and mechanisms,
12 sources, and compositions. However, their fate and transport are quite dissimilar. In breathing
13 and infiltration into homes, ultrafine particles are removed rather efficiently by diffusion to
14 surfaces. In the atmosphere, ultrafine particles are removed largely by coagulation with other
15 ultrafine particles (or accumulation-mode particles) and grow into the accumulation mode.
16 Coarse particles, however, are removed from the atmosphere rather rapidly by gravitational
17 settling. With regard to the volume or mass of ambient PM, accumulation-mode and coarse-
18 mode particles both contribute appreciably in most areas, with very little contribution from
19 ultrafine particles. With regard to particle surface area, however, ultrafine and accumulation-
20 mode particles both contribute appreciably, with very little contribution from coarse-mode
21 particles.

22 Ultrafine, accumulation mode, and coarse particles also behave differently with regard to
23 exposure and dosimetric considerations, as discussed below, as well as in toxicologic and
24 epidemiologic studies, as discussed in subsequent sections of this chapter.

25 26 **9.2.1.2.2 Exposure-related considerations**

27 The critical relationship to be considered is that between ambient PM *concentrations* and
28 *personal exposures* to ambient PM. (Ambient PM means that PM measured at a community
29 monitoring site, or the average over several such sites.) It is convenient to consider two aspects
30 of this relationship. One important aspect is the relationship between the ambient concentration
31 measured at one or more monitoring sites, and the distribution of outdoor concentrations across

**TABLE 9-1. COMPARISON OF AMBIENT PARTICLES,
FINE PARTICLES (Ultrafine Plus Accumulation-Mode) AND COARSE PARTICLES**

	Fine		
	Ultrafine	Accumulation	Coarse
Formation Processes:	Combustion, high-temperature processes, and atmospheric reactions		Break-up of large solids/droplets
Formed by:	Nucleation Condensation Coagulation	Condensation Coagulation Reactions of gases in or on particles Evaporation of fog and cloud droplets in which gases have dissolved and reacted	Mechanical disruption (crushing, grinding, abrasion of surfaces) Evaporation of sprays Suspension of dusts Reactions of gases in or on particles
Composed of:	Sulfate Elemental carbon Metal compounds Organic compounds with very low saturation vapor pressure at ambient temperature	Sulfate, nitrate, ammonium, and hydrogen ions Elemental carbon Large variety of organic compounds Metals: compounds of Pb, Cd, V, Ni, Cu, Zn, Mn, Fe, etc. Particle-bound water	Suspended soil or street dust Fly ash from uncontrolled combustion of coal, oil, and wood Nitrates/chlorides/sulfates from HNO ₃ /HCl/SO ₂ reactions with coarse particles. Oxides of crustal elements (Si, Al, Ti, Fe) CaCO ₃ , CaSO ₄ , NaCl, sea salt Pollen, mold, fungal spores Plant and animal fragments Tire, brake pad, and road wear debris
Solubility:	Probably less soluble than accumulation mode	Largely soluble, hygroscopic, and deliquescent	Largely insoluble and nonhygroscopic
Sources:	Combustion Atmospheric transformation of SO ₂ and some organic compounds High temperature processes	Combustion of coal, oil, gasoline, diesel fuel, wood Atmospheric transformation products of NO _x , SO ₂ , and organic compounds, including biogenic organic species (e.g., terpenes) High-temperature processes, smelters, steel mills, etc.	Resuspension of industrial dust and soil tracked onto roads and streets Suspension from disturbed soil (e.g., farming, mining, unpaved roads) Construction and demolition Uncontrolled coal and oil combustion Ocean spray Biological sources
Atmospheric half-life:	Minutes to hours	Days to weeks	Minutes to hours
Removal Processes:	Grows into accumulation mode Diffuses to raindrops	Forms cloud droplets and rains out Dry deposition	Dry deposition by fallout Scavenging by falling rain drops
Travel distance:	< 1 to 10s of km	100s to 1000s of km	< 1 to 10s of km (100s to 1000s in dust storms)

Source: Adapted from Wilson and Suh (1997).

1 an area (e.g., outside homes and other microenvironments). This relationship will depend in part
2 on the uniformity with which the PM indicator of interest is distributed across the community.
3 For time-series epidemiologic analyses of associations between 24-h concentrations of ambient
4 PM and health endpoints, another important parameter in this relationship is the day-to-day
5 correlation of 24-hour concentration values at various monitoring sites in the community. For
6 long-term epidemiologic analyses, the variation in the seasonal or yearly average at various sites
7 in the community is the important parameter. Much new information on the uniformity of PM_{2.5}
8 and PM_{10-2.5} concentrations across cities is available from the new monitoring networks and is
9 presented in detail in Chapter 3. The data show that, in general, PM_{2.5} is more evenly distributed
10 than PM_{10-2.5} in terms of both daily/seasonal/yearly averages and day-to-day correlations,
11 although there are significant differences among cities. Little is known about the spatial
12 distribution of ultrafine particle concentrations. However, because of their rapid growth into the
13 accumulation mode, they probably tend to be concentrated most heavily near sources such as
14 traffic and, thus, likely have a far more heterogeneous distribution across a community.

15 The second aspect is the relationship between the concentration of PM outdoors and the
16 concentration of that outdoor PM which has infiltrated into the home or other microenvironment.
17 The relationship between the concentrations of ambient particles outdoors, C , and the indoor
18 concentrations of those ambient particles that have infiltrated indoors, C_{ai} , is given by

$$F_{INF} = C_{ai} / C = Pa / (a + k), \quad 9-1$$

20 where P is the penetration factor; a is the air exchange rate; and k is the particle deposition rate.
21 P and k vary with particle size, so that infiltration factor, F_{INF} , also varies with particle size.
22 As shown in Figure 9-3, the infiltration factor is high for accumulation-mode particles and
23 decreases to low levels with decreasing size within the ultrafine range and with increasing size
24 within the coarse-mode range. Exposure-related relationships for the three particle size classes
25 are summarized in Table 9-2.

27 In most community time-series studies and long-term cohort studies, the ambient
28 concentration is used as a surrogate for personal exposure to ambient PM (ambient exposure).
29 For the ambient concentration to be a satisfactory surrogate, there must be a reasonable
30 correlation between ambient concentration and ambient exposure. Because of the lower and

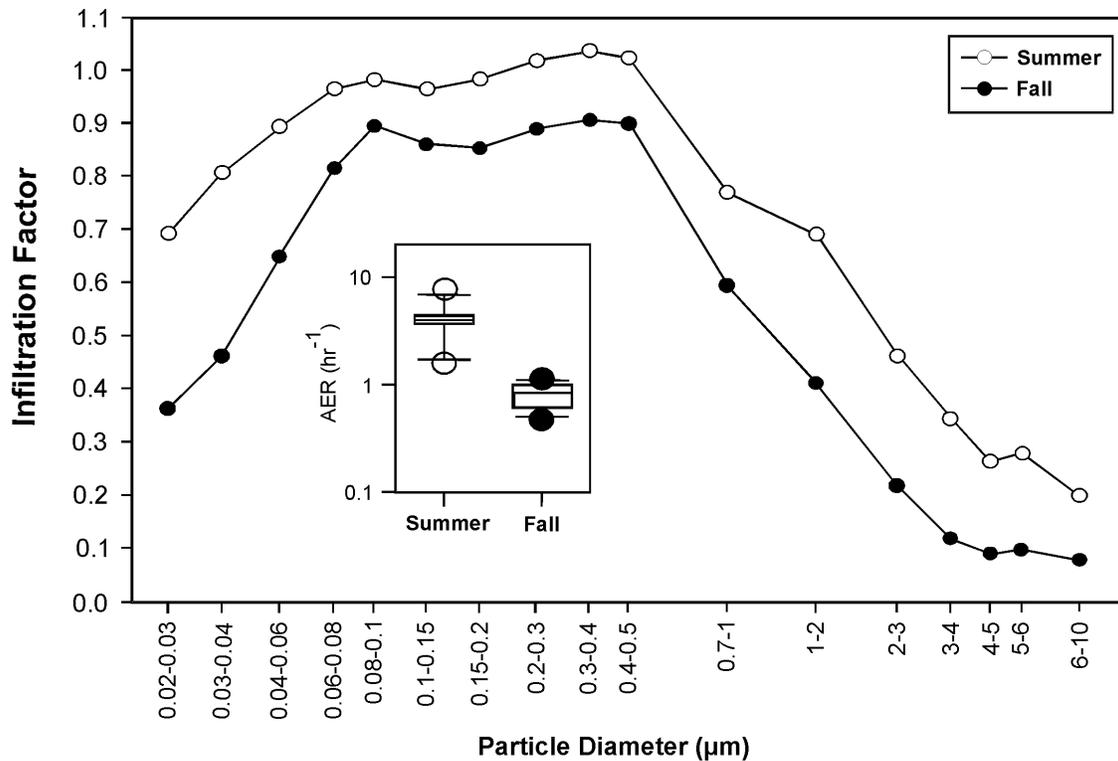


Figure 9-3. Geometric mean infiltration factor (indoor/outdoor ratio) for hourly nighttime, nonsource data for two seasons. Box plots of air exchange rates are shown as inserts for each plot (Boston, 1998).

Source: Long et al. (2001).

TABLE 9-2. EXPOSURE-RELATED RELATIONSHIPS FOR PARTICLE SIZE FRACTIONS

	Ultrafine	Accumulation-Mode	Thoracic-Coarse
Even distribution across city	probably not	frequently	seldom
Site-to-site correlation	probably low	frequently high	frequently low
Infiltration factor	generally low	high	generally low

1 more variable infiltration factors for ultrafine and coarse particles and their less even distribution
2 and lower site-to-site correlations across the community, it is likely that their ambient
3 concentrations will be a poorer surrogate for their ambient exposures than is the case for PM_{2.5}.
4 Nonambient PM may also be responsible for health effects. However, since the ambient and
5 nonambient components of personal exposure are independent, the health effects due to
6 nonambient PM exposures generally will not bias the risk calculated for ambient PM exposures.

7 8 **9.2.1.2.3 Dosimetric considerations**

9 The fraction of inhaled particles that are deposited in the various regions of the lung
10 depends on the particle size, the breathing route (nasal or oral), the breathing frequency (breaths
11 per minute), and the volume of air inhaled (tidal volume). The fractional depositions in the
12 extrathoracic (ET), tracheobronchial (TB), and gas exchange or alveolar (A) regions of the
13 respiratory tract are shown as a function of particle size in Figure 9-4 for nasal and oral breathing
14 at two levels of activity (resting and light exercise). Particles in the accumulation-mode size
15 range generally have very low deposition fractions, especially in the ET and TB regions, that
16 are relatively insensitive to breathing pattern or exercise. However, for nose breathing the
17 deposition of larger accumulation-mode particles in the ET region does increase with exertion.
18 Thus, most accumulation mode particles that enter the lungs are exhaled rather than deposited.

19 Ultrafine particles generally have much higher fractional depositions than accumulation
20 mode particles. However, the smaller nucleation-mode (< 0.01 μm) ultrafine particles behave
21 differently from the larger Aitken mode (0.01 to 0.1 μm) ultrafine particles. With decreasing
22 particle size below 0.1 μm, the total deposition of particles increases, and the pattern of
23 deposition within the respiratory tract slowly moves proximally, i.e., toward the ET region. This
24 shift in the pattern of deposition is quite obvious for decreases in particle size below 0.01 μm
25 where A deposition fractions rapidly decline and the ET deposition fractions correspondingly
26 increase. The TB deposition fraction increases to a maximum near 3 nm. For the Aitken mode
27 particles, the deposition fraction for the A region increases with exertion whereas in the TB
28 region it decreases. Deposition fractions in the A region for particles less than 1 μm are
29 relatively insensitive to route of breathing.

30 The fractional deposition for coarse particles is even more complex. For both the A and
31 TB regions, the deposition fraction increases with particle diameter above 1 μm, reaches a peak

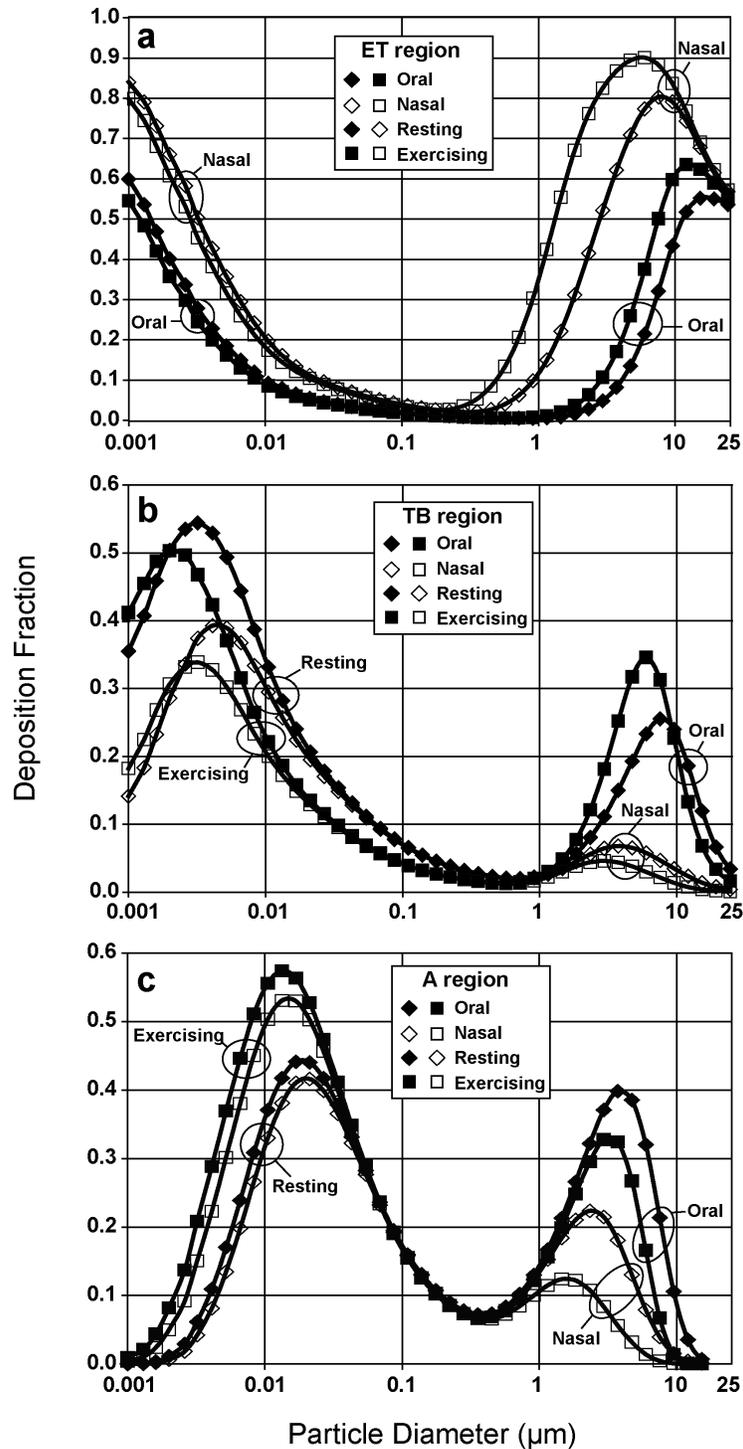


Figure 9-4. Deposition fraction as a function of particle size for nasal and oral breathing during rest and exercise: (a) extrathoracic (ET), (b) tracheobronchial (TB), and (c) alveolar (A) regions. Data shown here were calculated with the ICRP model and were also shown in Figures 6-16 and 6-17 along with similar results from the MPPD model. The data below 0.01 μm are uncertain but are shown to indicate trends. Note the different scale for the ET region.

1 before the diameter reaches 10 μm , and then declines. The deposition fractions for the A and TB
2 regions are lower during nasal breathing because a large fraction of the coarse particles deposit
3 within the nose. For mouth breathing, the A and TB deposition fractions are higher than during
4 nasal breathing but not as high as those for the ultrafine mode during mouth breathing. For
5 mouth breathing, the deposition fractions for both the A and the TB regions are greater for
6 coarse particles than for accumulation-mode particles. Even for nose breathing, some coarse
7 particles, of a specific size, will have higher deposition fractions than accumulation mode
8 particles.

9 In general, given these complex deposition patterns, there are no sharp cut points that
10 clearly distinguish between particle size ranges with relatively high versus relatively low
11 fractional deposition rates. For example, in the ET region, particles ranging in size from roughly
12 0.01 μm on up to ~ 1 μm (for nasal breathing) to over 3 μm (for oral breathing) exhibit relatively
13 low fractional deposition rates. For the TB region, relatively low rates are exhibited by particles
14 ranging in size from roughly 0.05 μm up to ~ 2 μm (for oral breathing) to over 10 μm (for nasal
15 breathing). For the A region, relatively low rates are exhibited not only by particles from ~ 0.1 to
16 1 μm , but also for particles in the low end of the ultrafine size range and in the upper end of the
17 coarse-mode range. Thus, while differences in dosimetric properties also generally support the
18 division of ambient particles into fine and coarse fractions, dosimetric considerations now
19 further suggest subdividing fine particles, although appropriate size cut point(s) would depend in
20 part on the relative importance placed on deposition in the different regions of the lung.

21 22 **9.2.1.3 Summary and Conclusions**

23 The fundamental distinctions between fine and coarse ambient particles based on the
24 physics and chemistry of ambient particles that were articulated in the last review, including
25 differences in formation, sources, composition, fate, and transport, remain generally unchanged.
26 However, some important advances have been made in our understanding of such distinctions,
27 especially with regard to characteristics of particles below ~ 0.1 μm in diameter (ultrafine
28 particles). In particular, whereas fine particles were previously characterized in two modes, they
29 are now characterized in terms of three modes, and distinctions among these modes allow for
30 more differentiation in characterizing properties of fine particles. Also, progress has been made
31 in better understanding the size distribution of biological materials. While previously

1 understood mainly to be present in the coarse particle size range, newly available information
2 indicates that such particles (e.g., pollen grains, endotoxins) may fragment or deteriorate into the
3 fine particle size range. This information expands our understanding of the types of particles
4 that can occur in particular within the intermodal size range of ~1 to ~3 μm .

5 Data now available from the new national $\text{PM}_{2.5}$ monitoring network and speciation sites
6 have allowed for better assessments of exposure-related considerations which broaden but do not
7 fundamentally change our understanding of the substantial differences between fine particles in
8 the accumulation mode and coarse particles. Relationships between ambient PM concentrations
9 and personal exposure to ambient PM are now better understood, primarily for fine particles, but
10 also to a more limited degree for coarse particles. For example, new data reinforce our earlier
11 understanding that ambient concentrations of fine particles (measured as $\text{PM}_{2.5}$) are typically
12 more highly correlated and/or are more uniform across community monitors within an urban
13 area than are coarse particles (measured as $\text{PM}_{10-2.5}$), although in some areas the differences are
14 much less pronounced than in others. More limited data and knowledge of the behavior of
15 ultrafine particles suggest that spatial distributions of their concentrations (which decrease
16 quickly from peak levels around major highways) are likely more similar to those for coarse
17 particles (which decrease quickly from peak levels around primary sources) than for other
18 (accumulation-mode) fine particles. Further, new studies reinforce our earlier understanding that
19 fine particles generally infiltrate indoors much better than do either coarse or ultrafine particles.
20 Thus, central site ambient concentration measurements are a better surrogate for population
21 exposure to accumulation-mode fine particles, measured as $\text{PM}_{2.5}$, than for either coarse or
22 ultrafine particles. Such similarities between ultrafine particles and coarse particles are based on
23 far more limited data, highlighting a need for further research on these particle size ranges.
24 However, since ultrafine particles represent only a very small mass fraction of typical ambient
25 fine particles, the most important exposure-related distinctions for particle mass, but not
26 necessarily for surface area, remain between fine particles (measured as $\text{PM}_{2.5}$) and coarse
27 particles (measured as $\text{PM}_{10-2.5}$).

28 Newly available dosimetry information continues to reinforce important distinctions
29 between fine and coarse particles, and submodes within fine particles, with regard to deposition
30 patterns within the respiratory tract. In general, while deposition patterns within the major
31 respiratory tract regions as a function of particle size are complex and dependent in varying

1 degrees on breathing route and ventilation levels, accumulation-mode particles exhibit distinctly
2 lower fractional deposition rates in any of the major respiratory tract regions than do ultrafine or
3 coarse particles on average. With increasing levels of activity, associated increases in breathing
4 rate, and associated increased oral nasal/oral breathing, the fractional deposition tends to
5 increase in the TB and A regions for ultrafine and coarse particles, while decreasing in the ET
6 region. Peak fractional deposition rates occur approximately in the mid-range of ultrafine and
7 coarse particle sizes in the A region, with the peaks tending more to the lower end of the
8 ultrafine size range and the upper end of the coarse particle range in the TB and ET regions.
9 Thus, it is difficult to characterize more specific size fractions within the range of thoracic
10 particles that would clearly delineate ranges of relatively high and relatively low fractional
11 deposition across all respiratory tract regions.

12 Overall, then, the above considerations generally reinforce the recommendation made in
13 the 1996 PM AQCD that fine and coarse atmospheric particles be considered as separate
14 subclasses of PM pollution. Further progress in characterizing these subclasses of pollutants will
15 depend upon obtaining additional ambient concentration/composition, exposure-related, and
16 dosimetric data, especially to supplement the far more limited data on coarse particles. Also,
17 new information suggests that important exposure-related and dosimetric distinctions exist
18 between ultrafine and accumulation-mode particles within the fine fraction, although there is as
19 yet only very limited data available to characterize these distinctions. Thus, it would also be
20 particularly useful to obtain additional ambient measurements of ultrafine particles to facilitate
21 future research to investigate their toxic potential.

22

23 **9.2.2 How Does the Newly Available Information Inform Our Judgments**

24 **about the Strengths and Limitations of the Epidemiologic Evidence for**

25 **Health Effects Related to Ambient Fine and Coarse Thoracic PM,**

26 **Acting Alone and/or in Combination With Other Pollutants?**

27 In assessing the strengths and limitations of the epidemiologic evidence, information is
28 drawn primarily from Chapter 8, as well as from Chapter 5 of this document. More specifically,
29 as discussed in Section 8.1.4, the approach used here to assess the epidemiologic information
30 focuses on a number of salient aspects of associations between mortality and morbidity effects
31 and short- and long-term exposures to ambient thoracic PM and common copollutants. This
32 includes consideration of: (1) the magnitude, statistical significance, precision/power, and

1 robustness of reported health effects associations for various size fractions of ambient PM, as
2 well as for PM from various types of sources; (2) assessment of the consistency or general
3 concordance of study results across the various PM size fractions, and consideration of potential
4 reasons for observed differences; and (3) information from so-called “intervention” studies or
5 “found experiments” as to the extent to which reductions in PM-related air pollution have been
6 observed to be associated with improvements in health measures. In assessing the robustness of
7 the PM-related health effects associations, consideration is given to potential confounding by
8 looking at associations with PM alone and with other common gaseous air pollutants and
9 associations based on different statistical modeling approaches. Exposure-related information is
10 also considered to assess the potential consequences of exposure misclassification. In assessing
11 these issues in light of the newly available information, Section 9.2.2.2 is organized primarily to
12 focus on discussions of the evidence of associations for PM₁₀, PM_{2.5}, PM_{10-2.5}, and source-
13 oriented PM.

14 15 **9.2.2.1 Key Points from 1996 Integrative Synthesis**

16 Based on the then available PM epidemiologic studies, the 1996 PM AQCD, arrived at the
17 following overall conclusions:

18
19 “The evidence for PM-related effects from epidemiologic studies is fairly strong,
20 with most studies showing increases in mortality, hospital admissions, respiratory
21 symptoms, and pulmonary function decrements associated with several PM indices.
22 These epidemiologic findings cannot be wholly attributed to inappropriate or incorrect
23 statistical methods, misspecification of concentration-effect models, biases in study
24 design or implementation, measurement errors in health endpoint, pollution exposure,
25 weather, or other variables, nor confounding of PM effects with effects of other factors.
26 While the results of the epidemiology studies should be interpreted cautiously, they
27 nonetheless provide ample reason to be concerned that there are detectable human
28 health effects attributable to PM at levels below the current NAAQS.” (U.S. EPA, 1996,
29 p. 13-92).

30
31 The 1996 PM AQCD went on to state further that, while the epidemiological studies
32 indicate increased health risks associated with exposure to PM, alone or in combination with
33 other air pollutants, the role of PM as an independent causal factor has not been completely
34 resolved, based on the available studies using multiple air pollutants as predictors of health
35 effects (U.S. EPA, 1996, p. 13-92).

1 **9.2.2.2 Integration of New Information**

2 Many recent epidemiologic studies have built upon what was previously know, showing
3 statistically significant associations of ambient PM with a variety of cardiovascular and
4 respiratory health endpoints, including mortality, hospital admissions, emergency department
5 visits, other medical visits, respiratory illness and symptoms, physiological or biochemical
6 changes related to the cardiovascular system, and physiologic changes in pulmonary function.
7 Associations have been consistently observed between short-term and all of these endpoints; and
8 long-term PM exposure has been associated with increased risk of mortality, development of
9 respiratory disease, and changes in lung function. As summarized in Appendices 8A and 8B,
10 epidemiologic studies have been conducted in areas across the U.S. and Canada, as well as in
11 Central and South America, Europe, Asia and Australia, and various methods have been used to
12 measure ambient PM concentrations. Considering the evidence from the full body of
13 epidemiologic studies using various PM indicators, the available findings demonstrate well that
14 human health effects are associated with ambient PM. Discussions in the following sections will
15 focus primarily on studies conducted in the U.S. and Canada using various mass measurements
16 of thoracic particles (e.g., PM₁₀, PM_{2.5}, PM_{10-2.5}).

17 The etiology of most air pollution-related health outcomes is highly multifactorial, and the
18 impact of ambient air pollution exposure on these outcomes may be small in comparison to that
19 of other etiologic factors (e.g., smoking). In contrast with the marked increase in health effects
20 observed during historic episodes of very high air pollution levels, relatively small effect
21 estimates would generally be expected with current ambient PM concentrations in the U.S.

22 In epidemiologic studies of ambient air pollution, associations with health outcomes have
23 been observed quite consistently, frequently being statistically significant or nearly so. Also,
24 magnitudes and significance levels of observed air pollution-related effects estimates would be
25 expected to vary somewhat from place to place, if the observed epidemiologic associations
26 denote actual effects, because (a) not only would the complex mixture of PM vary from place to
27 place, but also (b) affected populations may differ in characteristics that could affect
28 susceptibility to air pollution health effects, and (c) areas may differ in factors that affect
29 population exposures to ambient pollutants.

1 ***9.2.2.2.1 Strength of epidemiologic evidence on health effects associations with PM***

2 ***Short-term Exposure Studies***

3
4 Many new epidemiologic studies have built upon what was available in the 1996 PM
5 AQCD. These include several multi-city studies that can provide more precise estimates of
6 effects than individual city studies, offer consistency in data handling and modeling, allow for
7 systematic evaluation of geographic patterns in effects, and clearly do not suffer from potential
8 omission of negative findings due to “publication bias.” In addition, there are studies of new
9 health indices (e.g., physician visits) and cardiovascular health outcomes, analyses that provide
10 insight into the sensitivity of PM effects to alternative statistical modeling, new assessments on
11 the potential for confounding by gaseous copollutants, and new evidence from “found
12 experiments” that evaluate improvement in health with reductions in air pollution levels.

13 The results from key U.S. and Canadian studies on short-term PM exposure for several
14 commonly-used health outcomes — mortality, hospitalization and medical visits – are presented
15 in Figures 9-5 and 9-6. Epidemiologic studies of short-term air pollution exposures have also
16 evaluated other health outcomes (e.g., respiratory symptoms, medical visits, cardiovascular
17 health indicators, lung function changes). In addition, Chapter 8 also summarizes the results
18 of numerous studies of health effects linked with long-term exposures (discussed later in
19 Section 9.2.2.2.2). Thus, these figures do not attempt to present the full range of
20 epidemiological study results, but rather to illustrate results for a few major health outcome
21 categories commonly used in time-series epidemiology studies of short-term PM exposure
22 effects. Appendix Tables 9A-1 and 9A-2 present a fuller array of results for short-term (24 h)
23 ambient PM exposure effects on mortality and morbidity, respectively, including information on
24 effect size estimates derived by varying models for different mortality and morbidity endpoints
25 for U.S. and Canadian cities.

26 Figures 9-5 and 9-6 draw from findings presented in separate figures for the various
27 endpoints in Chapter 8 to pull together results of studies for the three major PM mass indicators.
28 As is the case for figures in Chapter 8, the results are drawn from U.S. and Canadian studies that
29 either did not use GAM or were reanalyzed to address GAM-related questions. Single-pollutant
30 (PM only) results are presented for purposes of comparison across studies, and it is noted that
31 multipollutant model results are presented and discussed in Chapter 8 (see especially Section
32 8.4.3). In many studies, the authors identified the model or lag period that appeared to best fit

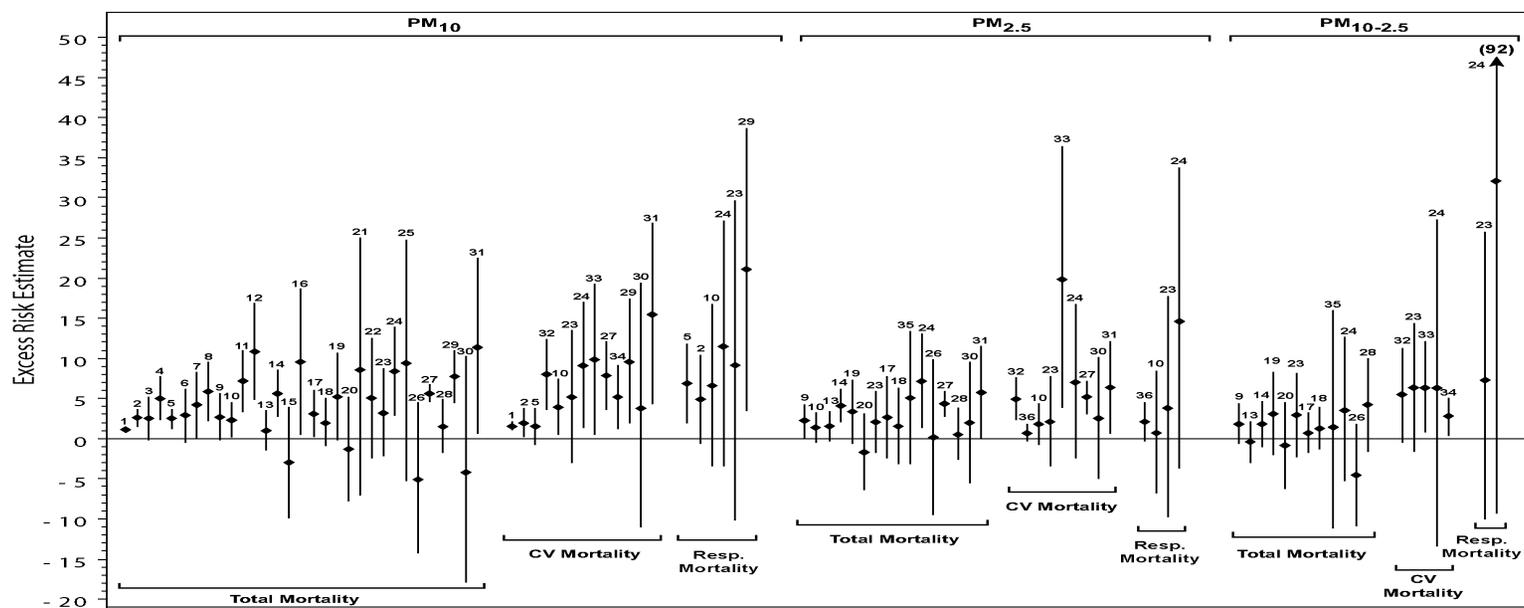


Figure 9-5. Excess risk estimates for total non-accidental, cardiovascular, and respiratory mortality in single-pollutant models for U.S. and Canadian studies. PM increments: $50 \mu\text{g}/\text{m}^3$ for PM_{10} and $25 \mu\text{g}/\text{m}^3$ for $\text{PM}_{2.5}$ and $\text{PM}_{10-2.5}$. Results presented from time-series studies that did not use GAM or were reanalyzed using GLM.

- | | | |
|--|--|--|
| 1. Dominici et al. (2003a), 90 U.S. cities | 13. Klemm and Mason (2003), St. Louis | 25. Schwartz (2003a), Colorado Springs |
| 2. Moolgavkar (2003), Cook County | 14. Klemm and Mason (2003), Boston | 26. Klemm and Mason (2003), Topeka |
| 3. Kinney et al. (1995), Los Angeles | 15. Schwartz (2003a), Birmingham | 27. Tsai et al. (2000), Newark |
| 4. Schwartz (2003a), Chicago | 16. Schwartz (2003a), New Haven | 28. Klemm and Mason (2003), Steubenville |
| 5. Ito and Thurston (1996), Cook County | 17. Chock et al. (2000) Pittsburgh (< 75 y.o.) | 29. Pope et al. (1992), Utah Valley |
| 6. Schwartz (2003a), Pittsburgh | 18. Chock et al. (2000) Pittsburgh (75+ y.o.) | 30. Tsai et al. (2000), Elizabeth |
| 7. Styer et al. (1995), Cook County | 19. Klemm and Mason (2003),
Kingston-Harriman | 31. Tsai et al. (2000), Camden |
| 8. Schwartz (2003a), Detroit | 20. Klemm and Mason (2003), Portage | 32. Lipfert et al. (2000a), Philadelphia |
| 9. Burnett and Goldberg (2003),
8 Canadian cities | 21. Schwartz (2003a), Canton | 33. Mar et al. (2003), Phoenix |
| 10. Moolgavkar (2003), Los Angeles | 22. Schwartz (2003a), Spokane | 34. Ostro et al. (2003), Coachella Valley |
| 11. Schwartz (2003a), Seattle | 23. Ito (2003), Detroit | 35. Klemm and Mason (2000), Atlanta |
| 12. Schwartz (2003a), Minneapolis | 24. Fairley (2003), Santa Clara County | 36. Ostro et al. (1995), Southern California |

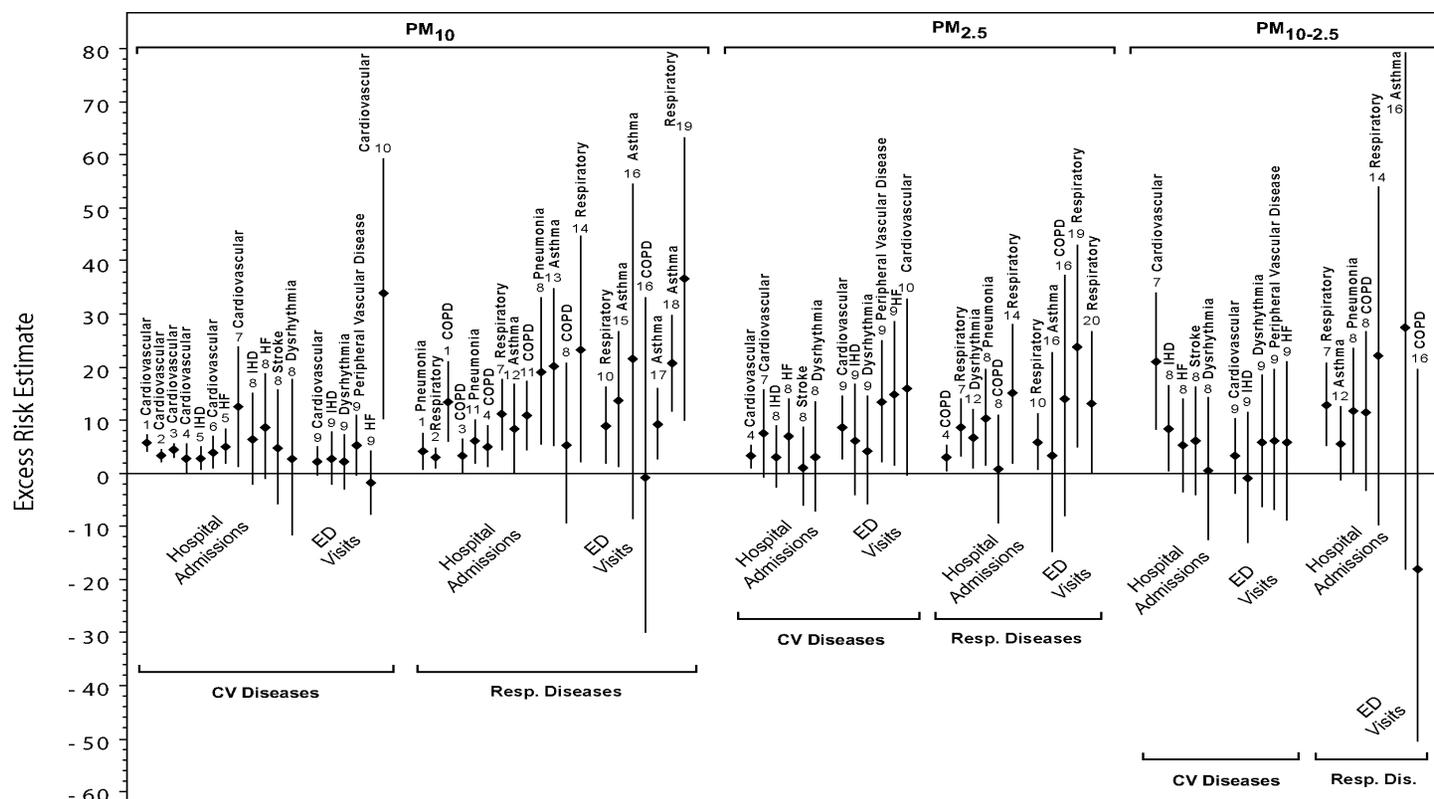


Figure 9-6. Excess risk estimates for hospital admissions and emergency department visits for cardiovascular and respiratory diseases in single-pollutant models from U.S. and Canadian studies. PM increments: 50 $\mu\text{g}/\text{m}^3$ for PM_{10} and 25 $\mu\text{g}/\text{m}^3$ for $\text{PM}_{2.5}$ and $\text{PM}_{10-2.5}$. Results presented from time-series studies that did not use GAM or were reanalyzed using GLM.

- | | | |
|--|---|---|
| 1. Zanobetti and Schwartz (2003)
U.S. 14 cities | 8. Ito (2003), Detroit | 14. Thurston et al. (1994), Toronto |
| 2. Linn et al. (2000), Los Angeles | 9. Metzger et al. (2004), Atlanta | 15. Tolbert et al. (2000b), Atlanta |
| 3. Moolgavkar (2003), Cook County | 10. Stieb et al. (2000), St. John | 16. Tolbert et al. (2000a), Atlanta |
| 4. Moolgavkar (2003), Los Angeles | 11. Schwartz (1994), Detroit | 17. Lipsett et al. (1997), Santa Clara County |
| 5. Schwartz and Morris (1995), Detroit | 12. Sheppard (2003), Seattle | 18. Choudhury et al. (1997), Montreal |
| 6. Morris and Naumova (1998), Chicago | 13. Nauenberg and Basu (1999),
Los Angeles | 19. Delfino et al. (1997), Montreal |
| 7. Burnett et al. (1997), Toronto | | 20. Delfino et al. (1998), Montreal |

1 the data, and those results are presented here. For other studies in which results for several lag
2 periods are presented, results from the model with the most precise results (e.g., largest
3 t-statistic) are shown. The results of models using different lag periods from time-series
4 epidemiologic studies are also presented and discussed in Chapter 8 (see Section 8.4.4). Finally,
5 for each health outcome, the results are presented in the figures in order (from left to right) of
6 decreasing study power, using as an indicator the product of the number of study days and
7 number of health events per day.

8 In Figure 9-5, effect estimates for associations between mortality and PM are grouped both
9 by PM indicator (PM_{10} , $PM_{2.5}$, and $PM_{10-2.5}$) and by mortality category (total non-accidental,
10 cardiovascular or cardiorespiratory, and respiratory). Looking across the results with particular
11 focus on the more precise estimates, some general observations can be made:

- 12 (1) Most all of the associations between PM_{10} and total mortality are positive and over
half are statistically significant, including most all of those with more precise
estimates. All associations reported between PM_{10} and cardiovascular and respiratory
mortality are positive; most of the cardiovascular mortality associations are also
statistically significant, where as most of the respiratory associations are less precise
and not statistically significant. In studies where all three types of mortality were
evaluated, the effects estimates for respiratory mortality were generally larger and less
precise. The more precise effect estimates range from ~1% to 8% increased risk of
mortality per $50 \mu\text{g}/\text{m}^3 PM_{10}$.
- 13 (2) A similar pattern can be seen for $PM_{2.5}$, though fewer studies are available, and the
effects estimates are generally somewhat less precise and less frequently statistically
significant. In particular, most all of the $PM_{2.5}$ associations with total mortality are
positive, although less than half are statistically significant. All $PM_{2.5}$ associations
with cardiovascular and respiratory mortality are positive; about half of the
cardiovascular associations, but none of the respiratory associations, are statistically
significant. The more precise effect estimates range from about 2% to 6% increased
risk of mortality per $25 \mu\text{g}/\text{m}^3 PM_{2.5}$.
- 14 (3) Still fewer studies have used $PM_{2.5}$ measurements. Though the effect estimates are
most all positive and similar in magnitude to those reported for $PM_{2.5}$ and PM_{10} , few
reach statistical significance. In this figure, statistically significant associations are
seen only between cardiovascular mortality and $PM_{10-2.5}$, and only in Phoenix (Mar
et al., 2003) and Coachella Valley, CA (Ostro et al., 2003).

15 The results for U.S. and Canadian studies are generally consistent with those presented
16 in Chapter 8 based on all available epidemiologic studies world wide. These results indicate
17 that there is substantial strength in the epidemiological evidence for association between PM_{10}

1 and PM_{2.5} and mortality, especially for total and cardiovascular mortality, but also for respiratory
2 mortality. For PM_{10-2.5}, the evidence for associations with mortality is more limited and clearly
3 not as strong, though it will be important to consider the influence of issues such as exposure
4 error in interpreting these results.

5 In Figure 9-5, the effect estimates presented for associations between morbidity and
6 ambient PM are grouped by PM indicator (PM₁₀, PM_{2.5}, and PM_{10-2.5}), general health outcome
7 category (cardiovascular and respiratory), and more specific outcome measures (hospital
8 admissions and medical visits). Several general observations can be made:

- 9
- 10 (1) All associations between PM₁₀ and hospitalization for cardiovascular and respiratory
diseases are positive and most are statistically significant, including all of the more
precise estimates. Most all PM₁₀ associations with ED visits for cardiovascular and
respiratory diseases are positive, and most respiratory (but not cardiovascular)
associations are statistically significant. The more precise effect estimates range from
about 2% to 6% increased risk per 50 µg/m³ PM₁₀ for cardiovascular diseases, and
2% to 12% increased risk per 50 µg/m³ PM₁₀ for respiratory diseases, with some
effect estimates for respiratory medical visits up to about 30% per 50 µg/m³ PM₁₀.
- 11 (2) For PM_{2.5}, all associations with hospitalization for cardiovascular and respiratory
diseases are positive and many are statistically significant, especially for respiratory
diseases. All PM_{2.5} associations with ED visits for cardiovascular and respiratory
diseases are positive, and about half are statistically significant. The more precise
effect estimates range from about 1% to 10% increased risk per 25 µg/m³ PM_{2.5}
for cardiovascular diseases, and 1% to 12% increased risk per 25 µg/m³ PM_{2.5} for
respiratory diseases.
- 12 (3) Associations between PM_{10-2.5} and hospitalization for cardiovascular and respiratory
diseases are all positive, though the confidence intervals are broader than those for
associations with PM₁₀ or PM_{2.5} and few associations are statistically significant.
Most PM_{10-2.5} associations with ED visits are positive, but none are statistically
significant.
- 13 (4) For all PM indicators, associations with medical visits tend to be less precise than
those for hospital admissions. As was noted in Section 8.3.2.4, many of the
medical/physician visits effect estimates are larger in magnitude than those for
hospital admissions.

14 As was found for mortality, the epidemiologic evidence for associations between
15 hospitalization and medical visits for cardiovascular and respiratory diseases is strong for PM₁₀
16 and PM_{2.5}. The few available PM_{10-2.5} studies also provide some evidence, though not as strong,

1 for associations between hospitalization and medical visits for cardiovascular and respiratory
2 diseases with $PM_{10-2.5}$.

3 For both mortality and morbidity effects, many more epidemiologic studies have
4 used PM_{10} than have used $PM_{2.5}$ and $PM_{10-2.5}$ measurements, since there is a much more
5 extensive set of air quality monitoring data available for PM_{10} . It is not surprising that the
6 strength of epidemiologic evidence for PM_{10} may appear somewhat stronger than the
7 epidemiologic evidence for $PM_{2.5}$ and certainly stronger than the evidence for effects of $PM_{10-2.5}$.
8 It is important to recognize, however, that information summarized in the previous section
9 strongly indicates that $PM_{2.5}$ and $PM_{10-2.5}$ are quite different pollutants and thus associations
10 reported for PM_{10} are representing some combination of effects of $PM_{2.5}$ and $PM_{10-2.5}$. The few
11 studies that have tested multipollutant models that include both $PM_{2.5}$ and $PM_{10-2.5}$ have reported
12 that the two PM size fractions have independent effects (e.g., Lippmann et al., 2000). In areas
13 where PM_{10} is predominantly fine particles, including most areas in the eastern U.S., it is likely
14 that associations with PM_{10} primarily reflect effects of fine particles. Likewise, associations
15 reported in areas where PM_{10} is predominantly coarse fraction particles, including many areas in
16 the western U.S., effect estimates for PM_{10} likely primarily reflect effects of $PM_{10-2.5}$. It should
17 be noted that epidemiological effect estimates have been presented using standardized
18 increments to allow for comparison across studies. As described in Section 8.1.1, based on
19 current air quality data distributions, increments of $50 \mu\text{g}/\text{m}^3$ for PM_{10} and $25 \mu\text{g}/\text{m}^3$ for $PM_{2.5}$
20 and $PM_{10-2.5}$ were selected as representative of a realistic high-to-low range of concentrations for
21 most U.S. communities. If one were to present the effect estimates per $\mu\text{g}/\text{m}^3$ for each PM mass
22 measure, the effect estimates for both $PM_{2.5}$ and $PM_{10-2.5}$ are generally larger than those for PM_{10} .

23 In addition to studies using PM_{10} , $PM_{2.5}$, and $PM_{10-2.5}$, a few new studies have evaluated
24 associations with ultrafine particle concentrations (using generally number of particles per m^3).
25 One mortality study in Erfurt, Germany (Stölzel et al., 2003; Wichmann et al., 2000) reported
26 associations for both $PM_{2.5}$ and ultrafine particles that were positive, but not statistically
27 significant. Four panel studies of subjects with asthma, two in Germany and two in Finland,
28 have included measurements of particle number. Peak expiratory flow and daily medication use
29 in adults, but not children, were more closely associated with particle number than particle mass.
30 One major source of ultrafine particles is motor vehicles, thus ultrafine particles may be used as

1 an indicator of traffic pollution, and, as discussed below, a number of studies have linked
2 particles from motor vehicle emissions with mortality and other adverse health effects.

3 As discussed in more detail in Section 8.2.2.5, various PM components or characteristics
4 have been associated with mortality. In general, evidence for associations have been reported
5 for most components that have been studied. However, many PM components are correlated
6 with each other and also with PM mass, making it is difficult to distinguish effects of the various
7 components. Also, different PM components or characteristics would be expected to be more
8 closely linked with different health outcomes.

9 One new approach used to evaluate effects associated with various PM components is to
10 conduct a source apportionment analysis of the composition data base in conjunction with an
11 epidemiologic analysis using source category factors as the surrogate for exposure. Three such
12 studies were discussed in Section 8.2.2.5.3 (Laden et al., 2000, reanalyzed in Schwartz et al.,
13 2003; Mar et al., 2000, 2003; Tsai et al., 2000). Motor vehicles, or more precisely particles
14 associated with vehicular traffic, stand out clearly as a source category associated with mortality
15 in all three studies. A regional sulfate source category was also identified as being associated
16 with mortality in all three studies; however, particles of crustal origin were not significantly
17 associated with mortality. Laden et al. (2000) and Tsai et al. (2000) reported associations with
18 an oil combustion factor. In addition, a source category related to vegetative burning was also
19 identified as being associated with mortality by Mar et al. (2000, 2003). These studies suggest
20 that many different chemical components of fine particles and a variety of different types of
21 source categories are all associated with, and probably contribute to, mortality, either
22 independently or in combinations.

23 One key research question that has not been addressed in epidemiologic studies is the
24 relationship between sources or composition of coarse fraction particles and health outcomes.
25 The studies described above used source apportionment based on components of fine particles
26 (Laden et al., 2000; Mar et al., 2000) or PM_{15} in an area dominated by fine particles (Tsai et al.,
27 2000). However, some limited information is available from air quality analyses that may help
28 inform the assessment of epidemiological evidence for coarse fraction particles. Mar and
29 colleagues (2000) conducted factor analysis of $PM_{10-2.5}$ data in Phoenix using the limited
30 speciation data that were available, and reported that two major source factors for coarse fraction
31 particles were crustal particles and coarse fraction metals. In addition to the studies described

1 above for fine particles, new studies have shown no increases in mortality on days with high
2 concentrations of wind-blown dust (crustal particles), using PM_{10} concentrations and data on
3 wind speed as indicators of dust-storm days. Thus, it is possible that the association reported
4 between $PM_{10-2.5}$ and cardiovascular mortality in Phoenix is influenced by the metal content of
5 the coarse fraction particles, but perhaps not the crustal component. Clearly, further research is
6 needed in this area.

7 In summary, as discussed in more detail in Chapter 8, there is strong evidence from
8 epidemiologic studies that short-term increases in PM_{10} and $PM_{2.5}$ are linked with increased risk
9 of both mortality and morbidity. The more limited body of studies on $PM_{10-2.5}$ provides evidence
10 that is suggestive of associations with morbidity and, perhaps less so, with mortality, but the
11 evidence is less strong than that for PM_{10} and $PM_{2.5}$. Some new epidemiological evidence also
12 suggests associations between health effects and ultrafine particles and other fine particle
13 components and sources, but the data are as yet too limited to characterize the relative toxicities
14 of these various components or indicators of fine particles.

15 16 *Long-term Exposure Studies*

17 In the 1996 PM AQCD, results of prospective cohort studies linked long-term exposure to
18 fine particles and mortality, and there was limited evidence indicating that long-term PM
19 exposure was linked with chronic respiratory morbidity, such as the development of bronchitis.
20 The more recent long-term exposure studies, summarized in Appendix Table 9A-3, have built
21 upon these findings and provide further evidence for associations with both mortality and
22 respiratory morbidity.

23 A series of analyses using data from the ACS cohort have shown significant associations
24 between total and cardiopulmonary mortality and fine particles or sulfates, and the most recent
25 analyses have also reported significant associations with lung cancer mortality. The Six Cities
26 study found significant associations of $PM_{2.5}$ with total and cardiopulmonary (but not lung
27 cancer) mortality, but not with coarse particle indicators. The results most recently reported for
28 the AHSMOG study reported some significant associations between PM_{10} and total mortality and
29 deaths with contributing respiratory causes. In further investigation of the results found for PM_{10}
30 among males, the associations with $PM_{2.5}$ had larger effect estimates than those for $PM_{10-2.5}$ for

1 males in the AHSMOG cohort, though none reached statistical significance. For the VA study,
2 inconsistent results were reported for associations between PM indicators and mortality.

3 Based on several factors – the larger study population in the ACS study, the larger air
4 quality data set in the Six Cities study, the more generally representative study populations used
5 in the Six Cities and ACS studies, and the fact that these studies have undergone extensive
6 reanalyses – the greatest weight should be placed on the results of the ACS and Six Cities cohort
7 studies in assessing relationships between long-term PM exposure and mortality. The results of
8 these studies, including the reanalyses results for the Six Cities and ACS studies and the results
9 of the ACS study extension, provide substantial evidence for positive associations between long-
10 term ambient PM (especially fine PM) exposure and mortality.

11 For morbidity, results of studies in a cohort of children in Southern California have built
12 upon the limited evidence available in 1996 PM AQCD to indicate that long-term exposure to
13 PM is associated with development of chronic respiratory disease and reduced lung function
14 growth. Long-term exposure to PM was associated with significant decreases in lung function
15 growth among a cohort of Southern California school children, but the earlier cross-sectional
16 analysis for the same cohort found no relationship between respiratory symptoms and annual
17 average PM₁₀ levels. These findings support the results of the cross-sectional study in 24 U.S.
18 and Canadian cities from the 1996 PM AQCD, in which long-term PM exposure was associated
19 with some effects on respiratory function changes and respiratory illness.

20 As was true in the 1996 PM AQCD, it is more difficult to assess strength of evidence for
21 long-term exposure studies, since there are fewer studies available. For mortality, reanalyses
22 and extended analyses of cohort studies provide strong evidence for the link between mortality
23 and long-term exposure to fine particles; however, the available studies have provided no
24 evidence for associations between long-term exposure to coarse fraction particles and mortality.
25 In addition, prospective cohort and cross-sectional analyses have reported associations between
26 respiratory morbidity and PM₁₀, and sometimes also PM_{2.5}, providing fairly strong evidence for
27 effects of long-term fine particle exposures on respiratory morbidity. The morbidity studies
28 have not generally included PM_{10-2.5} data, so no conclusions can be drawn regarding long-term
29 exposure to coarse fraction particles and morbidity.

1 **9.2.2.2.2 Assessment of robustness of associations for epidemiologic studies**

2 Many epidemiologic studies have also included assessment of whether the associations
3 were robust to such factors as model specification and potential confounding by copollutants.
4 Another factor that is relevant to robustness of epidemiologic findings is exposure error.
5 Chapter 8 includes detailed discussions on each of these topics, and the following discussion will
6 focus on the extent to which the current epidemiologic findings can be considered robust.

7 8 **Model Specification**

9 The 1996 PM AQCD included considerable discussion of issues regarding model
10 specification for time-series epidemiologic studies, including results of reanalyses using several
11 data sets, with a special focus on the large data set available from Philadelphia, PA. In this set of
12 reanalyses, results reported with the use of alternative modeling strategies were not substantially
13 different from the original investigators' findings. Also, at the time of completion of the 1996
14 PM AQCD, it appeared that issues related to model specifications used to control for weather
15 effects in daily time-series analyses of ambient PM relationships to mortality/morbidity had
16 largely been resolved. Based on two major studies extensively evaluating a number of different
17 approaches to adjust for weather effects (including evaluations using synoptic weather patterns),
18 it was concluded that significant PM-mortality associations were robust and verifiable via a
19 variety of model specifications controlling for weather.

20 More recently, the influence of using default parameters in a widely used software package
21 for GAM on epidemiologic study results has been investigated, and in this process, the question
22 of appropriate adjustment for weather, temporal trends and other covariates in time- series
23 models was reopened. Numerous study findings were reanalyzed to test the effect of using more
24 stringent convergence criteria in the GAM program, as well as alternative modeling methods
25 such as GLM. The results from the GAM reanalysis studies indicate that PM risk estimates from
26 GAM models were often, but not always, reduced when more stringent convergence criteria
27 were used, although the extent of the reduction was not substantial in most cases. Also, the
28 extent of downward bias in standard errors reported for these data (a few percent to ~15%)
29 appears not to be very substantial, especially when compared to the range of standard errors
30 across studies due to differences in population size and numbers of days available. These
31 GAM-related issues, however, were seen to have less influence on effect estimates than

1 investigator-to-investigator variations in model specifications, including the number of weather
2 terms and extent of smoothing.

3 As observed in the HEI reanalysis report (HEI, 2003), the use of alternative modeling
4 strategies tended to reduce the effect estimate size, but did not change the overall findings and
5 qualitative conclusions of epidemiologic studies showing associations between PM and both
6 mortality and morbidity. While it is clear that there will not be one “correct” model or approach
7 for covariate adjustment, further research can help inform modeling strategies to adjust for
8 temporal trends and weather variables in time-series epidemiology studies.

9 10 *Assessment of Confounding by Copollutants*

11 Airborne particles are found among a complex mixture of atmospheric pollutants, some of
12 which are widely measured (such as gaseous criteria copollutants O₃, CO, NO₂, SO₂) and others
13 which are not routinely measured. Because many of the pollutants are closely correlated due to
14 emissions by common sources and dispersion by common meteorological factors, and some are
15 in the pathway of formation of other pollutants (e.g., NO → NO₂ → NO₃⁻ → PM nitrates), it is
16 generally difficult to disentangle their effects. In addition, as described in Section 8.1.3.2,
17 copollutants could possibly act as effect modifiers; for example, exposure to one pollutant could
18 result in greater sensitivity to another pollutant. Potential effect modification between pollutants
19 has been investigated in some toxicological or controlled human exposure studies (Section 7.9.3)
20 but little evidence is available from epidemiologic studies to characterize any such effects.

21 The potential for co-pollutant confounding in the epidemiologic time-series studies was
22 assessed in some detail in Section 8.4.3. Multipollutant modeling is the most common method
23 used to test for potential confounding in epidemiologic studies; however, interpretation of the
24 results of multipollutant models is complicated by the correlations that often exist between air
25 pollutants. In interpreting the results of any of these studies, it is important to consider factors
26 such as the biological plausibility of associations between the pollutants and health outcomes,
27 and questions related to model specification and exposure error. Some new studies, described in
28 Section 5.3.3.4 have reported that while ambient PM_{2.5} concentrations are well correlated with
29 personal PM_{2.5} exposure measurements, this is not generally the case for O₃, SO₂ and NO₂.
30 However, ambient gaseous pollutant concentrations were generally found to be correlated with
31 personal PM_{2.5} exposures, suggesting that ambient concentrations of O₃, NO₂, SO₂ may act as

1 surrogates, as opposed to confounders of PM_{2.5} in the estimation of PM health effects based on
2 multipollutant models.

3 Multipollutant modeling results for associations between a range of health outcomes and
4 PM with gaseous pollutants in single-city studies are presented in Section 8.4.3 (Figures 8-16
5 through 8-19). For most studies, there was little change in coefficients for PM between single-
6 pollutant and multipollutant models; however, in some cases the PM effect estimate was
7 markedly reduced in size and lost statistical significance in models with one or more gaseous
8 pollutants. Key results are also available from the NMMAPS evaluation of associations across
9 many U.S. cities with varying climates and mixes of pollutants; associations between PM₁₀ and
10 both mortality and morbidity were not changed with adjustment for gaseous pollutant
11 concentrations. Thus, for the most part, effect estimates for PM were not substantially changed
12 when gaseous copollutants were included in the models. PM and the gaseous copollutants were
13 often highly correlated, especially for fine particles and CO, SO₂ and NO₂, and it was generally
14 the case that high correlations existed between pollutants where PM effect estimates were
15 reduced in size with the inclusion of gaseous copollutants.

16 In the prospective cohort and cross-sectional studies, the potential for confounding by
17 copollutants has been assessed in some studies of mortality, but little studied for morbidity. The
18 reanalysis of data from the ACS cohort indicated that the relationships with fine particles and
19 sulfates were reduced in size in co-pollutant models including SO₂, but not the other gaseous
20 pollutants. SO₂ is a precursor for fine particle sulfates, thus complicating the interpretation of
21 multipollutant models including fine particles and SO₂ for this study. The authors concluded
22 that their results suggested that mortality may be associated with more than one component of
23 the ambient air pollutant mix, and that there were robust associations between mortality and fine
24 particles and sulfates.

25 In summary, ambient PM exposure usually is accompanied by exposure to many other
26 pollutants, and PM itself is composed of numerous physical/chemical components. Assessment
27 of the health effects attributable to ambient PM and its constituents within an already subtle total
28 air pollution effect is therefore very challenging, even with well-designed studies. Indeed,
29 statistical partitioning of separate pollutant effects is not likely to characterize fully the effects
30 that actually depend on simultaneous exposure to multiple air pollutants. Overall, the new
31 evidence tends to substantiate that for both long-term and short-term exposures, observed PM

1 effects are at least partly due to ambient PM acting either alone or in conjunction with covarying
2 gaseous pollutants.

4 *Exposure Error*

5 Numerous analyses of the potential influence of measurement error on time-series
6 epidemiologic study results are discussed in Section 8.4.5. One consideration in comparing
7 epidemiologic findings for different pollutants is the relative precision with which the pollutants
8 are measured. If two pollutants have effects, and there is correlation between both the
9 pollutants, the effect estimate of the pollutant that is less precisely measured may be attenuated
10 when the pollutants are considered together. One would expect that PM_{2.5}, CO, and NO₂ would
11 often have a high positive correlation due to common activity patterns, weather, and source
12 emissions. PM_{10-2.5} is generally less precisely measured than PM_{2.5}, but the two are not generally
13 highly correlated. Several recent studies have focused on this question, and reported that for
14 most situations, it is unlikely that differential measurement error would result in shifting
15 apparent effects from one pollutant to another. The most extreme case, complete transfer of
16 apparently causal effects from one pollutant to another, required very high correlation between
17 the covariates, no error in measurement of the “false” covariate and moderate error in
18 measurement of the “true” predictor. Thus, it is unlikely that effects attributed to PM (generally
19 focusing on PM₁₀ or PM_{2.5}) are falsely transferred from other pollutants that are measured with
20 less precision. Another facet of exposure error is the degree to which the measurements made at
21 monitoring sites reflect population exposures to PM.

22 As discussed in Section 5.2, a number of studies have shown that ambient fine particle
23 concentrations are well correlated with temporal changes in personal exposures to ambient fine
24 particles. However, it should be noted that the spatial variability across the city is generally
25 much greater for PM_{10-2.5} than for PM_{2.5}. In addition, the infiltration factors for PM_{10-2.5} and a
26 number of the gases (e.g., ozone, SO₂) are lower and more variable than that of PM_{2.5}, likely
27 leading to a lower correlation between ambient concentration, used as an exposure surrogate in
28 community time-series studies, and personal exposure to the ambient pollutant for PM_{10-2.5} and
29 these gases. Thus, the new exposure studies indicate that fine particles measurements at central
30 monitoring sites are good indicators of personal exposures to PM_{2.5} in time-series studies.
31 Exposure relationships for PM_{10-2.5} have been less well studied, but exposure error and

1 measurement error would be expected to have greater influence for associations with $PM_{10-2.5}$
2 than for $PM_{2.5}$.

3 4 **9.2.2.2.3 Assessment of consistency in epidemiologic study results**

5 One key conclusion in the 1996 PM AQCD was that there was considerable consistency in
6 findings among the PM time-series epidemiologic studies. The much larger set of studies
7 available now includes evidence for somewhat greater heterogeneity among risk estimates from
8 different locations, in studies of both mortality and morbidity effects.

9 This potential heterogeneity in risk estimates is most notable in the reports from multi-city
10 studies conducted in the U.S., Canada and Europe, that have also included quantitative
11 assessments of heterogeneity and potential factors that would influence heterogeneity. The city-
12 specific and regional PM-mortality associations presented in NMMAPS results suggested greater
13 variability in effect estimates than had been observed in the studies available in the 1996 PM
14 AQCD. However, statistical analyses indicated that there was no significant heterogeneity in
15 mortality effect estimates for 90 U.S. cities (Samet et al., 2000a, Dominici et al., 2003a). For
16 eight Canadian cities, no evidence of heterogeneity was reported in the initial analysis, but in
17 reanalysis to address GAM issues, there appeared to be greater heterogeneity in the PM-
18 mortality associations (Burnett and Goldberg, 2003). Finally, initial analyses of mortality
19 associations for 29 European cities indicated differences between eastern and western cities, but
20 these differences were less clear with reanalysis to address GAM questions (Katsouyanni et al.,
21 2003).

22 There are a number of reasons to expect variation in PM-health effect associations for
23 different geographic regions. Regional differences can include differences in PM sources or
24 composition, differences in population exposures, and differences in potentially susceptible
25 groups. In the European multi-city study, APHEA, PM-mortality associations were found to be
26 larger in areas with higher average NO_2 levels (considered an indicator of traffic pollution),
27 warmer climates (possibly due to more open windows resulting in better exposure estimation)
28 and lower overall mortality rates (potentially more susceptible people). In NMMAPS, no
29 apparent associations were found between PM-mortality association and socioeconomic
30 indicators or $PM_{2.5}/PM_{10}$ ratios, but there was also no statistically significant measure of
31 heterogeneity in this study. However, for hospital admissions in the NMMAPS, the PM_{10} -

1 admissions associations were greater in areas with less use of central air conditioning (possibly
2 an indicator of increased exposure to ambient pollutants) and with larger contributions of PM₁₀
3 emissions from vehicle emissions and oil combustion.

4 In general, it has been found that PM concentrations, especially for fine particles, are quite
5 regional in distribution and concentrations measured at different monitors across a city are well
6 correlated. However, as discussed in Section 8.4.5.2, some U.S. cities show more variability in
7 PM measurements across monitoring sites. For larger metropolitan areas, including monitors in
8 outlying areas can also increase variability in PM measurements across the area. From those
9 U.S. cities in which epidemiological studies have been conducted, areas with more uniformity
10 in PM_{2.5} concentrations include Chicago and Detroit; areas with more variability include Seattle
11 and Los Angeles. There are a number of factors that could influence spatial variability of PM
12 concentrations, including topography, location of major PM sources and weather patterns.
13 Greater spatial variability in PM concentrations would be expected to increase exposure error
14 and thus potentially affect epidemiologic study results in those areas.

15 One factor unrelated to geographic location that would likely affect the consistency of
16 results across studies is the amount of data available for analysis. For time-series studies, the
17 number of days with measurements is one important indicator of study size, or statistical power.
18 In Figure 9-5, the PM-mortality associations are plotted in order of decreasing statistical power,
19 using the product of daily death rate and number of PM measurement days as the indicator. For
20 single-city mortality studies, the number of PM measurement days ranges from about 150 (Tsai
21 et al., 2000) to over 2000 days (e.g., Ito and Thurston, 1996). Multi-city studies included ranges
22 of about 500-900 days for eight Canadian cities (Burnett and Goldberg, 2003), about 200-3000
23 days for 90 U.S. cities (Dominici, 2003), and 1500-3000 days for 10 U.S. cities (Schwartz,
24 2003a). For several studies, more data are available for PM₁₀ than for PM_{2.5} or PM_{10-2.5}; for
25 example, Fairley (2003) used a data set with approximately 800 days of PM₁₀ measurements and
26 400 days for PM_{2.5} and PM_{10-2.5}. In the 1996 PM AQCD, studies conducted in the U.S. had about
27 300 to 4000 days of PM measurements, and a clear correlation between t-ratio and number of
28 monitoring days could be seen (Figure 12-17, Table 12-25). Similarly, Figures 9-5 and 9-6 show
29 a tendency for larger studies to have more consistent effect estimates that are more likely
30 statistically significant. A number of the newer studies, however, particularly those using PM_{2.5}
31 and PM_{10-2.5} data, are somewhat smaller in size than those available in the 1996 PM AQCD. This

1 would be expected to result in decreased precision, and more variability in effect estimate size
2 for the smaller studies.

3 In addition, while many single-city epidemiologic studies have used availability of
4 everyday monitoring data as a criterion for selecting study locations, a number of the newer
5 studies have used PM_{2.5} and PM_{10-2.5} data measured every sixth day. Beyond limiting the number
6 of days of data available, the use of 1-in-6 day data may also complicate the time-series
7 analyses. As discussed in Section 8.4.5.2, one analysis of data from Chicago used data from an
8 everyday monitor and created six 1-in-6 day data sets from these data; the resulting
9 PM-mortality associations for these six data sets were quite inconsistent.

10 Focusing on the results of epidemiologic studies with larger study sizes, there remains
11 an overall indication of consistency in effect estimate size for health effect associations
12 with PM₁₀, PM_{2.5} and PM_{10-2.5}. The new multi-city study results provide some evidence for
13 differences in PM-mortality associations across locations. However, there are reasons to
14 expect heterogeneity in study results across cities, based on different topographies, distribution
15 of sources or emissions, mixes of pollutants, and population characteristics.

16 17 **9.2.2.2.4 Lag period between exposure and effect**

18 The lag time between exposure and health effect has been investigated in many time-series
19 epidemiological studies, as described in more detail in Section 8.4.4. In considering the results
20 of models for a series of lag days, it is important to consider the pattern of results that is seen
21 across the series of lag periods. If there is an apparent pattern of results across the different lags,
22 then selecting the single-day lag with the largest effect from a series of positive associations is
23 reasonable. In fact, the single-day lag effect estimate is likely to underestimate the overall effect
24 size since the largest single-lag day results do not fully capture the risk also distributed over
25 adjacent or other days; a distributed lag model should more correctly capture the effect size.
26 In addition, it should be noted that the pattern of results with different lag periods may depend
27 on the pattern of persistence of air pollution (i.e., episodes may persist for a few days), which
28 may vary from city to city and from pollutant to pollutant. If this is the case, fixing the lag
29 across cities or across pollutants may not be ideal, and may tend to obscure important nuances of
30 lag structures that might provide important clues to possible different lags between PM
31 exposures and different cause-specific effects.

1 It would be expected that differing lag periods would be appropriate for different health
2 outcome-pollutant associations. For example, the time-series studies of cardiovascular hospital
3 admissions or emergency department visits suggest that PM effects are stronger with a lag 0 with
4 some carryover to the day 1 lag. In a few studies of cardiac physiological changes, strongest
5 associations were reported for some effects with 1- to 2-hour lag periods. In panel studies of
6 respiratory symptoms and in several studies of asthma hospitalization or emergency department
7 visits, longer moving average lag periods (up to 5- to 7-day moving averages) yielded larger PM
8 effect estimates.

9 Studies have shown different lag days for different source categories and different size
10 fractions. For example, Mar et al. (2000) found positive and significant associations with
11 cardiovascular mortality for regional sulfate for lag day 0, for traffic-related PM on lag day 1,
12 and for wood-burning related PM on lag day 3; for PM_{10-2.5} on lag day 0 and PM_{2.5} on lag
13 day 1. PM₁₀, presumably due to its correlation with both PM_{2.5} and PM_{10-2.5}, was positive
14 and significant on both lag day 1 and 2.

15 Studies of long-term exposure have included less evaluation of temporal relationships
16 between PM exposure and health effect. The prospective cohort studies have used air quality
17 measurements made over a period of years as an indicator of long-term exposure to air pollution,
18 The associations reported in these studies are for relationships with PM across various levels of
19 exposure, not as a measure of latency of effect.

20 However, some new studies have included some assessment of temporal relationships
21 between PM exposure and mortality. In the reanalysis of the Six City Study, the decline in fine
22 particle levels over the monitoring period was included as a time-dependent variable, to assess
23 the effect of changing PM concentrations over time on the association with mortality. The
24 association between total mortality and fine particles was reduced in size, though still
25 statistically significant, as compared with the model not allowing for temporal change in
26 pollution level. This is likely indicative of the effectiveness of control measures in reducing
27 source emissions importantly contributing to the toxicity of ambient particles in cities where PM
28 levels were substantially decreased over time.

29 The VA study analysis tested associations between different subsets of pollution and
30 mortality data. While the associations found between PM and mortality varied and were often
31 negative and generally not statistically significant, it was observed that the associations were

1 larger and more likely to be statistically significant with the air quality data from the earliest
2 time periods, as well as the average across all data. Further study is needed to evaluate the
3 relationship between health effects and long-term PM exposure where PM concentrations are
4 changing over time.

5 In summary, for time-series studies, it is likely that the most appropriate lag period for a
6 study will vary depending on the health outcome under study and air quality patterns in the study
7 area. Where effects are found for a series of lag periods, the effect estimate for any one lag
8 period will likely underestimate the effect size and a distributed lag model will more accurately
9 characterize the effect estimate size. Caution should be used in selecting results for single lag
10 periods if the pattern of results across lag periods is highly variable. For effects of chronic
11 exposure, less is known about the importance of different time windows for exposure and some
12 recent studies indicate that further investigation will be important.

13 14 **9.2.2.2.5 Form of concentration-response function**

15 In the 1996 PM AQCD, the limitations of identifying possible “thresholds” in the
16 concentration-response relationships in observational studies were discussed, including
17 difficulties related to the low data density in the lower PM concentration range, the small
18 number of quantile indicators often used, and the possible influence of measurement error. Few
19 studies had quantitatively assessed the form of PM-effect concentration-response functions and
20 the potential for a threshold level.

21 A threshold for a population, as opposed to a threshold for an individual, has some
22 conceptual issues that should be noted. For example, since individual thresholds vary from
23 person to person due to individual differences in genetic-level susceptibility and pre-existing
24 disease conditions (and even can vary from one time to another for a given person), it is
25 extremely difficult mathematically to demonstrate convincingly that a clear threshold exists in
26 the population studies. This is especially true if the most sensitive members of a population are
27 unusually sensitive even down to very low concentrations. The person-to-person difference in
28 the relationship between personal exposure to PM of ambient origin and the concentration
29 observed at a monitor may also add to the variability in observed exposure-response
30 relationships, possibly obscuring otherwise more evident thresholds. Since one cannot directly

1 measure but can only compute or estimate a population threshold, it would be difficult to
2 interpret an observed population threshold biologically, without pertinent collateral
3 dosimetric/toxicologic information. Despite these issues, several PM-related epidemiologic
4 studies have attempted to address the question of threshold.

5 Analyses using data for 90 U.S. cities showed that, for total and cardiorespiratory
6 mortality, the exposure-response spline curves for mean lag (0- and 1-day) were roughly linear,
7 but less so for current and previous day PM₁₀, making it difficult to discern any evident
8 threshold. For daily total or cardiorespiratory mortality, the likelihood of a threshold occurring
9 above PM₁₀ levels of ~25 µg/m³ seems to be essentially zero (see Figure 8-31); there was
10 increasing probability of a threshold occurring at levels below 25 µg/m³. The hypothesis of
11 linearity was examined, with the results indicating that the linear model was preferred over the
12 spline and the threshold models. In some single-city analyses, there were indications of potential
13 population thresholds for associations between mortality and PM₁₀ in the range of 80 µg/m³ to
14 100 µg/m³, and with PM_{2.5} in the range of 20-25 µg/m³. However, other single-city analyses
15 reported no evidence of a threshold level for PM-mortality associations.

16 In summary, the results from large multi-city studies suggest that there is no strong
17 evidence of a clear threshold for PM mortality effect. Some single-city studies provide some
18 suggestive hints for possible thresholds, but not in a statistically clear manner. More data need
19 to be examined with alternative approaches, but, in the meantime, the use of linear PM effect
20 models appears to be appropriate.

21 22 **9.2.2.2.6 Intervention studies**

23 Although many studies have reported short-term associations between PM indices and
24 mortality, a largely unaddressed question remains as to the extent to which reductions in ambient
25 air PM actually lead to reductions in health effects attributable to PM. This question is not only
26 important in terms of “accountability” from the regulatory point of view, but it is also a scientific
27 question that challenges the predictive validity of statistical models and their underlying
28 assumptions used thus far to estimate excess mortality due to ambient PM.

29 The opportunities to address this question are rare. However, at the time of the 1996 PM
30 AQCD, results were available from epidemiologic studies of a “natural” or “found experiment”
31 in the Utah Valley, where mortality and respiratory hospital admissions were found to decrease

1 during the time a major PM source was closed. Recent toxicologic and controlled human
2 exposure studies using particle extracts from ambient community PM₁₀ sampling filters from the
3 Utah Valley have also shown reduced effects with exposure to particles collected during the time
4 period when the source was not operating. A recent epidemiologic study in Dublin, Ireland also
5 provides evidence for reductions in ambient PM being associated with reductions in mortality
6 rates. Other “found experiments” also provide evidence for decreases in mortality and/or
7 morbidity being associated with notable declines in PM (and/or gases such as SO₂) as the result
8 of interventions aimed at reducing air pollution.

9 By providing evidence for improvement in community health following reduction in air
10 pollutant emissions, these studies add further support to the results of the hundreds of
11 epidemiologic studies linking ambient PM exposure to an array of health effects. The studies
12 available generally show improvements in health with reductions in emissions of both PM and
13 gaseous pollutants and thus do not distinguish effects from the different pollutants. However,
14 they provide strong evidence that reducing emissions of PM and gaseous pollutants has
15 beneficial public health impacts.

16 17 **9.2.2.2.7 Summary and conclusions**

18 There is substantial evidence that both long-term and short-term exposures to PM_{2.5} are
19 associated with both mortality and morbidity. The fewer studies available for PM_{10-2.5} provide
20 less evidence for associations with mortality, but somewhat more support for associations
21 between short-term exposures and morbidity effects; little evidence is available to allow
22 conclusions to be drawn about long-term PM_{10-2.5} exposures and morbidity. There is also
23 extensive and convincing evidence for associations between short-term exposures to PM₁₀ and
24 both mortality and morbidity; however, as discussed above, these PM₁₀ associations likely reflect
25 underlying relationships with either PM_{2.5} or PM_{10-2.5} or with both PM mass fractions. Results of
26 new source apportionment studies and “found experiments” lend support to the results of the
27 other epidemiological studies.

9.2.3 How Does Newly Available Information Inform Assessment of Biological Plausibility and Coherence of Health Effects Attributed to Ambient Fine and Coarse Thoracic PM and/or Their Components?

In more broadly assessing the extent to which the overall body of evidence supports the attribution of observed health effects to exposure to fine and coarse thoracic PM and related chemical constituents, one needs to look beyond just epidemiologic evidence to consider the implications of newly available dosimetric, toxicologic, and other evidence as well. More specifically, the following assessment (a) evaluates information pertaining to the biological plausibility of the types of health effect associations observed in the epidemiologic studies, taking into account toxicologic findings and potential mechanisms of action; and (b) considers information about the coherence of the overall body of evidence relevant to PM-related health effects to reach conclusions regarding attribution of observed effects to ambient fine or coarse thoracic PM and related chemical constituents, acting alone and/or in combination with other pollutants.

9.2.3.1 Key Points from 1996 Integrative Synthesis

The 1996 PM AQCD highlighted several key findings and conclusions concerning attribution of observed health effects to specific ambient PM size fractions or chemical compounds:

- (1) “The likelihood of ambient fine mode particles being significant contributors to PM-related mortality and morbidity among [the] elderly population is bolstered by: (1) the more uniform distribution of fine particles across urban areas. . . ; (2) the penetration of ambient particles to indoor environments. . . ; and (3) the longer residence time of ambient fine particles in indoor air, enhancing the probability of indoor exposure to ambient fine particles more so than for indoor exposure to ambient coarse particles.”
- (2) The PM indices that have been “most consistently associated with health endpoints are fine particles (indexed by BS, COH, and $PM_{2.5}$), inhalable particles (PM_{10} or PM_{15}), and sulfate (SO_4^-),” whereas “[l]ess consistent relationships have been observed for TSP, strong acidity (H^+), and coarse PM ($PM_{10-2.5}$). . . . [and] none of these indices can completely be ruled out as a biologically relevant indicator of PM exposure.”
- (3) “Based on current evidence from epidemiologic, controlled human, human occupational, and laboratory animal studies, no conclusions can be reached regarding the specific chemical components of PM_{10} that may have the strongest biologic activity.” Further, none of the various subclasses of PM [e.g., acid aerosols, bioaerosols, metals (including

transition metals), and insoluble ultrafine particles] that have been considered “can be specifically implicated as the sole or even primary cause of specific morbidity and mortality effects.” (U.S. EPA, 1996, p. 13-93)

1 Hence, although at the time of the 1996 PM AQCD, the epidemiologic evidence was viewed as
2 substantiating well PM₁₀ or PM_{2.5} associations with human mortality and morbidity, uncertainties
3 remained with regard to (a) the relative toxicity of specific PM constituents and (b) the
4 biological plausibility of the reported effects and/or the mechanisms of action underlying them.
5

6 **9.2.3.2 Integration of New Evidence**

7 In the ensuing years since the 1996 PM AQCD evaluations, progress has been made in
8 (1) further substantiating and expanding epidemiologic findings indicative of ambient PM-health
9 effect associations, (2) identifying likely constituents contributing to observed effects, and
10 (3) obtaining evidence bearing on the biological plausibility of observed effects and possible
11 mechanisms of action involved. Efforts to interpret the overall meaning of the epidemiologic
12 finds and to evaluate their biological plausibility and pertinent mechanisms of action are
13 complicated by the fact that ambient PM exists as a component of a complex air pollution
14 mixture that includes other criteria pollutants, as well as many other airborne contaminants that
15 may convey risks to health. This section addresses these complexities by first considering the
16 physical and chemical components and source categories that have been associated with health
17 effects in epidemiologic studies. This is followed by a discussion of the toxicologic links that
18 have been reported between specific PM components identified in epidemiologic studies and
19 health effects and/or related biologic changes in controlled exposure human, animal, and in vitro
20 studies. Potential mechanisms of actions are then summarized, followed by a discussion of
21 inhaled particles as potential carriers of toxic agents. The coherence of findings from
22 epidemiologic and toxicologic studies is then discussed, leading to conclusions with regard to
23 the attribution of health effects to ambient fine and coarse thoracic PM and/or their constituents.
24

25 **9.2.3.2.1 Chemical components and source categories associated with health effects in** 26 **epidemiologic studies**

27 As discussed above, the numerous newly available epidemiologic studies provide growing
28 evidence that substantiates well statistical associations between (a) increased risk of total and
29 cause-specific mortality, as well as various morbidity endpoints (e.g., hospital admissions,

1 doctors visits, etc.), and (b) short- or long-term exposures to ambient PM indexed by mass
2 concentrations of several ambient PM size fractions and/or constituent components measured at
3 community monitoring stations. Probably the most extensive database substantiating such
4 associations is that demonstrating relationships between increased mortality/morbidity and
5 ambient levels of PM₁₀, which subsumes both fine- and coarse-mode fractions of thoracic
6 particles capable of reaching lower (thoracic) TB and A regions of the respiratory tract.
7 Extensive new evidence also substantiates further the judgments made in the 1996 PM AQCD
8 that significant associations exist between increased risk of mortality and morbidity and
9 exposures to ambient fine particles (indexed mainly by PM_{2.5} mass measurements), certain fine
10 particle components (e.g., sulfates), and sources of fine particles (e.g., coal or oil combustion,
11 motor vehicles, etc.). Less extensive, but growing, evidence has also begun to accumulate for
12 the ambient coarse fraction (PM_{10-2.5}) of thoracic particles also being associated, at least under
13 some circumstances, with increased risk of human morbidity and, possibly, mortality. For
14 example, while the epidemiologic findings generally do not implicate crustal materials primarily
15 in coarse fraction, they suggest that soil particles contaminated with metals (originally deposited
16 as fine particle components) or serving as carriers for bioaerosol materials (e.g., pollen grain
17 fragments, fungi spores, endotoxin) may contribute to observed effects.

18 Inherent in the PM research agenda recommended by the NRC (1998) was the recognition
19 of the importance of evaluating the relative toxicity of various components or characteristics of
20 PM so as to concentrate other aspects of future PM-related research (e.g., exposure monitoring)
21 on components that may be relatively more toxic. However, currently a wide array of PM
22 characteristics have been found to be associated with toxicity through epidemiologic studies, as
23 listed in Table 9-3.

24 Epidemiologic studies using either individual chemical species or classes or using source
25 category factors (SCF) derived from factor analysis have been particularly useful in helping to
26 identify a variety of species whose ambient concentrations are statistically associated with either
27 total mortality or more specific mortality groupings. A number of techniques have been
28 developed that apportion PM in ambient samples to its sources (see Section 3.3 of this document
29 and Section 5.5 of the 1996 PM AQCD for descriptions of these techniques). These powerful
30 techniques are limited by their ability to differentiate between PM produced by sources having
31 similar compositional profiles and by the lack of data for the composition (especially the organic

TABLE 9-3. PARTICULATE MATTER CHARACTERISTICS, COMPONENTS, OR SOURCE CATEGORIES SHOWN TO BE ASSOCIATED WITH MORTALITY IN U.S., CANADIAN, OR EUROPEAN EPIDEMIOLOGIC STUDIES^{1,2}

PM Size Fractions	Ions/Elements	Carbon/Organic Fractions	Source Categories (Tracers)
Mass TSP	Sulfate (SO ₄ ⁻)	TC (Total Carbon)	Motor Vehicles (CO, Pb)
Mass PM ₁₀	Nitrate (NO ₃ ⁻)	BC (Black Carbon)	Motor Vehicles plus resuspended road dust (CO, NO ₂ , EC, OC, Mn, Fe, Zn, Pb)
Mass-thoracic coarse PM [PM _{10-2.5} or PM ₁₀₋₁]	Ammonium (NH ₄ ⁻)	EC (Elemental Carbon)	
Mass-fine PM [PM _{2.5} or PM _{1.0}]	Transition metals (e.g., Cd, Cu, Fe, Ni, Mn, Zn)	COH (Coefficient of Haze)	Fuel oil combustion (Ni, V)
Mass-ultrafine PM [PM _{0.1}]	Other toxic metals (e.g., Pb)	OC (Organic Carbon)	Coal burning (Se)
Particle number	Strong Acid (H ⁺)	CX (Cyclohexene-extractable Carbon)	Sulfate or regional sulfate (S)
Particle surface area		Organic PM compounds	Industrial (Zn, Cd)

1. Components measured in PM_{2.5} unless otherwise specified.
2. Organic PM compounds extracted by three techniques.

composition) of emissions from many sources. This limitation may be mitigated in the future by further analytical developments in analyzing the composition of PM samples and broader availability of compositional data from the new PM speciation monitoring network.

Source categories found to be significantly associated with total, cardiovascular, or cardiovascular plus respiratory mortality in one or more cities are shown in Table 9-3, based on results from several studies (Laden et al., 2000; Schwartz, 2003a; Mar et al., 2000, 2003; and Tsai et al., 2000). A source category associated with motor vehicles was found in all three studies, which may include as causal elements one or more of: gaseous copollutants (CO and NO₂); soot particles from cars (indexed by BS, COH, or EC); organic PM from vehicles; Pb or transition metals emitted by vehicles (Mn, Fe, Zn); or other particles generated or

1 resuspended by vehicular traffic. Each of the three studies also identified a sulfate factor. The
2 factor reported by Laden et al. (2000) as “coal burning” contains high loadings of both selenium
3 and sulfur and could have also been called “regional sulfate.” Mar et al. (2000) refer to the
4 factor with high sulfate specifically as “regional sulfate,” distinguishing it from a factor with a
5 high loading of SO₂ (called a “local SO₂” factor). The regression with elemental S (assumed to
6 be sulfate) was not significant, but the regression with the regional sulfate factor was significant,
7 perhaps because the factor analysis tends to remove other more localized sulfate sources
8 (e.g., CaSO₄ and Na₂SO₄), leaving only acid sulfates ([NH₄]₂SO₄, NH₄HSO₄, and H₂SO₄) for
9 a regional sulfate factor. (In Phoenix, there was also a modest loading of S in the soil factor.)
10 Thus, all three sulfate factors should be considered as regional sulfate. These studies of specific
11 chemical components and source categories are also important because they indicate the
12 association of human health effects with three major components of PM_{2.5} mass: sulfate, nitrate,
13 and organic PM. Examination of the lag structure from the Phoenix study reveals that neither
14 the regional sulfate factor nor the vegetative burning factor was confounded by NO₂, CO, SO₂,
15 or O₃. Also, in another study, examination of PM_{2.5} and nitrate effects, alone and in multiple
16 regressions, indicated that PM_{2.5} and nitrate were not confounded by NO₂, CO or O₃ in
17 Santa Clara, CA (Fairley, 1999).

18 Also of much importance, all of the above studies that investigated multiple source
19 categories found a soil or crustal source that was negatively associated with mortality. This
20 suggests that the components of natural soil may have minimal toxicity unless contaminated by
21 toxic agents from anthropogenic sources, e.g., transition metals or polyaromatic hydrocarbons
22 (PAHs).

23 Although results such as those presented above are illuminating, it should be noted that
24 there can be ambiguity regarding the identification of source categories, as the marker elements
25 used in many of the methods used (e.g., specific rotation factor analysis) can have more than one
26 source. As an example, before Pb was phased down in gasoline, it was (and still is) both
27 produced by smelters and other industries as well as used in gasoline (see Appendix 3D of
28 Chapter 3). Also, there can be substantial spatial variability in source contributions across an
29 urban area, increasing potential exposure characterization error. Still, these new epidemiologic
30 findings suggest that various specific chemical components of PM and a variety of different

1 types of source categories are associated with and probably contribute to mortality, either acting
2 alone or in combination with other agents in ambient aerosol mixes.

3 Based on the overall available epidemiologic information, then, one or more size fractions
4 and/or constituent components of ambient PM have been most clearly shown to be associated
5 with increased risk of mortality and/or morbidity manifested in terms of (1) cardiovascular/
6 systemic, (2) respiratory, and (3) lung cancer effects. Acute, short-term (\leq 24-hr) exposures to
7 ambient PM appear to exert cardiovascular/systemic effects rather quickly, with peak lags of 0-1
8 days being generally seen, and one study reporting myocardial infarction increases even as early
9 as 2 h post exposure. Respiratory effects typically exhibit somewhat longer and more extended
10 lag periods, from 1 to 2 days on out to a week or so. Both cardiovascular and respiratory risks
11 have also been shown to be elevated in relation to long-term (years, decades) exposures to
12 ambient PM (especially the fine fraction), as has lung cancer mortality and morbidity.

13 14 **9.2.3.2.2 *Approaches to experimental evaluation of PM health effects***

15 As discussed in Chapter 7, various experimental approaches have been used to evaluate
16 PM health effects, including: studies of human volunteers exposed to PM under controlled
17 conditions; in vivo studies of laboratory animals including nonhuman primates, dogs, and rodent
18 species; and in vitro studies of tissue, cellular, genetic, and biochemical systems. A variety of
19 exposure conditions have been employed, including: whole body, mouth-only, and nose-only
20 inhalation exposures to concentrated ambient particles (CAPs) or laboratory-generated particles;
21 intratracheal, intrapulmonary, and intranasal instillation; and in vitro exposures to test materials
22 in solution or suspension. These approaches have been used mainly to test hypotheses regarding
23 the role of PM in producing the types of health effects identified by PM-related epidemiologic
24 studies. Thus, most new toxicological studies have thus far been designed to address the
25 question of biologic plausibility of epidemiologically-demonstrated effects and mechanisms of
26 action, rather than dose-response relationships.

27 Reflecting this, most of the toxicology studies have generally used exposure concentrations
28 or doses that are relatively high compared to concentrations commonly observed in ambient air.
29 One consideration underlying the use of such experimental exposure concentrations is the fact
30 that healthy animals have most typically been used in many controlled-exposure toxicology
31 studies, whereas epidemiologic findings often reflect ambient pollutant effects on compromised

1 humans (e.g., those with one or another chronic disease) or other susceptible groups at increased
2 risk due to other factors. Implicit in using relatively high concentrations in experimental studies
3 of healthy subjects is the assumption that increasing the dose makes up for compromised
4 tissue/organ functions that may contribute to observed ambient PM effects. However, this may
5 not be the case. Recognizing this, there has been growing attention to development and use of
6 compromised animal models that are thought to mimic important characteristics contributing to
7 increased human susceptibility to ambient PM effects.

8 One example is the use of monocrotaline (MCT)-treated rats, in which the MCT-induced
9 pulmonary vasculitis/hypertension is thought to render them at possible increased risk for PM
10 effects. Another example is a compromised animal model of chronic bronchitis (induced by
11 repeated prolonged exposure to SO₂, before exposure to PM). Partial coronary artery occlusion
12 is yet another example of a compromised animal model, evaluated for increased cardiovascular
13 risk. Possible PM exacerbation of respiratory infections has also been evaluated in animals
14 intratracheally exposed to various bacteria.

15 Given the relatively high concentrations used, much caution is needed in attempting to
16 interpret and extrapolate effects seen in these studies to provide insight into the biological
17 plausibility and mechanisms of action underlying effects seen in humans under “real world”
18 exposure conditions. Some reported responses may only be seen at the higher concentrations
19 (more typical of occupational exposures) and not necessarily at (usually much lower) ambient
20 particle exposure levels. On the other hand, differences between humans and rodents with
21 regard to the inhalability, deposition, clearance, and retention profiles for PM (see Chapter 6 for
22 details) could conceivably make doses to some specific respiratory tract tissues from
23 experimental exposures relatively similar to doses from human ambient exposures.

24 Since the 1996 PM AQCD, the effects of controlled exposures to ambient PM have been
25 evaluated by use of urban air particles (UAP) collected from ambient samplers (e.g., impactors,
26 diffusion denuders, etc.) and, more recently, by the use of aerosol concentrators. In the first type
27 of study, particles from ambient air samplers are collected on filters or other media, then stored
28 for varying time periods (hours to years or even decades) before later being resuspended in an
29 aqueous medium and used in inhalation, instillation, or in vitro studies. Depending on the
30 storage conditions for the filters (e.g., whether or not kept refrigerated or in the dark) varying

1 amounts of some originally collected materials (including highly biologically active semivolatile
2 compounds) may be lost and their possible effects missed in UAP studies.

3 Particle concentrators allow exposure under controlled conditions of animals or humans by
4 inhalation to concentrated “real-world” ambient particles (CAPs) at levels higher than typical
5 ambient PM concentrations. However, CAPs studies cannot control closely the mass
6 concentration and day-to-day variability in ambient particle composition, and they often lack
7 detailed characterization of variations in chemical composition from one CAPs exposure to
8 another. Because the composition of CAPs vary across both time and location, thorough
9 physical-chemical characterization is needed (but rarely done or reported) in order to facilitate
10 comparison of results between studies or even among exposures within studies, so as to better
11 link specific particle composition to effects. Another limitation is the fact that concentrators
12 used in many of the studies assessed here do not efficiently concentrate ambient particles
13 $\leq 0.1 \mu\text{m}$. Thus, it is likely that a large portion of potentially important combustion-generated
14 particles (e.g, from diesel, gasoline vehicle, wood smoke, coal smoke, etc.) were present only at
15 ambient (not higher concentrated) levels in most or all of the CAPs studies assessed here; and
16 many other potentially toxic co-components (e.g., SO_2 , O_3 , peroxides, etc.) of the ambient
17 aerosols may be excluded from the CAPs exposure mix as well. Thus, even “real-world” CAPs
18 exposures do not fully reflect important interactive effects of the overall aerosol mix (see also
19 Section 9.2.3.2.4).

20 Controlled human and laboratory animal exposures to particulate material obtained from
21 combustion-source bag house filters or other combustion-source collection devices have also
22 been used to evaluate the in vitro and in vivo respiratory toxicity of complex combustion-related
23 PM. Residual oil fly ash (ROFA) collected from large industrial sources (e.g., oil-fired power
24 plants) has been extensively used, and, less often, domestic oil furnace ash (DOFA) or coal fly
25 ash (CFA). The major disadvantage associated with the use of such materials derives from
26 questions about the potential relevance of results obtained in understanding ambient PM
27 exposure effects. Before extensive implementation of air pollution controls, ambient U.S. air
28 contained mixtures of PM species (at higher than current concentrations) analogous to those in
29 many of the source samples used in toxicologic studies during the past decade or so. However, it
30 is unlikely that high concentrations of certain materials that typify such samples would be found
31 or approached in ambient air PM samples from community monitoring sites in U.S., Canada, and

1 much of western Europe that generated the aerometric data (collected during the past 20 to 30
2 years) that were used to estimate PM exposures in most PM epidemiology studies assessed here.
3 Very high concentrations of metals (especially Ni and V, for example) typify most ROFA
4 samples, and experimental exposures to such materials have generally resulted in exposures and
5 doses that are orders of magnitude (100s of times) higher than for usual concentrations of such
6 metals in ambient PM measured routinely since the 1970s at community monitoring sites across
7 the United States. Thus, significant issues arise concerning the extent to which the effects of
8 high concentrations of ROFA or other combustion-source particle mixes can be extrapolated to
9 help interpret ambient air PM effects.

10 Analogous issues arise with evaluation of the toxicity of PM emitted from mobile source
11 combustion devices, e.g., diesel and gasoline vehicle engines. Complex combustion-related
12 mixtures in such mobile source emissions include many different types of particles and gaseous
13 compounds in high concentrations that are not necessarily representative of ambient PM derived
14 from such sources after passage through particle traps, catalytic converters, exhaust pipes, etc.
15 For example, ultrafine particles emitted from gasoline and diesel engines are reduced in numbers
16 and concentrations as they agglomerate to form larger, accumulation-mode particles as they cool
17 in passing through exhaust systems and/or as they undergo further physical and chemical
18 transformation as they “age” in ambient air. Further complicating evaluation of the toxicity of
19 mobile source emission components is: (1) the difficulty in separating out toxic effects
20 attributable to particles versus those of gaseous components in automotive exhausts; and (2) the
21 changing nature of those exhaust mixes as a function of variations in engine operating mode
22 (e.g., cold start versus warm start or “light” versus “heavy” load operation, etc.) and changes in
23 engine technology (e.g., “old diesels” versus “new diesels”).

24 The in vivo and in vitro PM exposure studies have almost exclusively used PM₁₀ or PM_{2.5}
25 as particle size cutoffs for studying the effects of ambient PM. Collection and study of particles
26 in these size fractions has been made easier by widespread availability of ambient sampling
27 equipment for PM₁₀ and PM_{2.5}. However, other important size fractions, such as the coarse
28 fraction (PM_{10-2.5}) and PM_{1.0}, largely have been ignored; and only limited toxicology data are
29 now available to assess effects of these particle sizes. Similarly, relatively little research has
30 addressed mechanisms by which organic compounds may contribute to ambient PM-related

1 effects. Both UAP extracts and CAPs have been used to evaluate effects in healthy and
2 compromised laboratory animals and humans.

3 4 **9.2.3.2.3 Interspecies comparisons of experimental results**

5 Much of the new toxicologic data assessed in Chapter 7 and discussed here was derived
6 from either: (a) in vivo exposures of human subjects or laboratory animals via inhalation
7 exposures or instillation of PM materials; or (b) in vitro exposures of various (mostly respiratory
8 tract) cells or tissues to diverse types of PM. As already noted, the experimental exposure
9 conditions used in these studies are typically different from those experienced through inhalation
10 of airborne PM by human populations in ambient environments. Thus, comparisons between
11 experimental tissue doses leading to observed PM effects and exposure/doses associated with
12 effects observed with human ambient PM exposures is useful, especially if any quantitative
13 extrapolation of experimental results across species or to ambient conditions is to be attempted.

14 To help place the toxicologically relevant concentrations/doses into context in relation to
15 ambient conditions, EPA carried out illustrative dosimetric/extrapolation modeling analyses to
16 provide comparisons between the high doses typically used in toxicological studies and doses
17 typical of human exposures under ambient conditions. Building upon advances in dosimetric
18 modeling discussed in Chapter 6, the EPA analyses compare PM doses delivered to human or rat
19 lung tissue from experimental exposures and PM doses to the human lung from exposures during
20 normal activities. These analyses and interpretation of their results (see Appendix 7-A) provide
21 context for exposure concentrations used and toxicological results assessed here.

22 23 ***Dosimetric Considerations in Comparing Dosages for Inhalation, Instillation, and*** 24 ***Exposure of Cultured Cells***

25 From among the three common experimental approaches for studying biological effects of
26 PM, inhalation studies are the most realistic physiologically and, thus, the most applicable to risk
27 assessment. However, because they are expensive, time consuming and require specialized
28 equipment and personnel, they are often supplemented by other techniques (instillation and in
29 vitro studies). Instillation studies, in which particles suspended in a carrier such as physiological
30 saline are applied to the airways, have certain advantages over in vitro studies. The exposed
31 cells have normal attachments to basement membranes and adjacent cells, circulatory support,
32 surrounding cells and normal endocrine, exocrine and neuronal relationships. Although the TB

1 region is most heavily dosed in such studies, alveolar regions can also be exposed via instillation
2 techniques. In vitro studies using live cells are cost-effective, allow for precise dose delivery,
3 and provide a useful avenue by which to conduct rapid PM mechanistic and comparative toxicity
4 studies. Often, the initial information on likely mechanisms of action of particles is obtained
5 through in vitro techniques. For in vitro studies, dose selection is important because it is easy to
6 overwhelm normal defense mechanisms.

7 It is difficult to compare particle deposition and clearance among different inhalation and
8 instillation studies because of differences in experimental methods and in quantification of
9 particle deposition and clearance. Key points from a discussion by Driscoll et al. (2000) of the
10 role of instillation in respiratory tract dosimetry and toxicology studies are informative. In brief,
11 inhalation may result in deposition within the ET region, the extent of which depends on the size
12 of the particles used; but intratracheal instillation bypasses this portion of the respiratory tract
13 and delivers particles directly into the TB tree. Although some studies indicate that short (0 to
14 2 days) and long (100 to 300 days postexposure) phases of clearance of insoluble particles
15 delivered either by inhalation or intratracheal instillation are similar, others indicate that the
16 percent retention of instilled particles is greater than for inhalation, at least up to 30 days
17 postexposure. Also, inhalation generally results in a fairly homogeneous distribution of particles
18 throughout the lungs, but instillation is typified by heterogeneous distribution (especially in the
19 A region) and high focal levels of particles. Most instilled material penetrates beyond the major
20 tracheobronchial airways, but the lung periphery is often virtually devoid of particles. This
21 difference is reflected in particle burdens within macrophages, those from animals inhaling
22 particles being burdened more homogeneously and those from animals with instilled particles
23 showing some populations of cells with heavy burdens and others with no particles, and is likely
24 to impact clearance pathways, dose to cells and tissues, and systemic absorption. Exposure
25 method, thus, clearly influences dose distribution; thus arguing for much caution in interpreting
26 results from instillation studies.

27 Dosimetric calculations must be performed to relate TB cell exposures from instillation in
28 terms of particle concentrations (on a number of particles per unit surface area basis) to those
29 occurring in human environmental exposures. Such calculations require selecting characteristics
30 associated with the particles, the exposed subject and the environmental exposure scenario.
31 Hence each study can present a unique dosimetric analysis. In most cases, it will be useful to

1 know the relationship between the surface doses in instillation studies and realistic local surface
2 doses that could occur in vivo in human subpopulations receiving the maximum potential dose.
3 Some characteristics of individuals serve to enhance the local surface deposition doses to
4 respiratory tract cells. These characteristics include: exercise and mouth breathing; non-uniform
5 inhaled air distribution (such as occurs in chronic bronchitis and other COPD conditions),
6 impaired particle clearance as occurs in some disease states; and location near pollutant sources.
7 In addition, even normal subjects exposed by inhalation are expected to have numerous sites of
8 locally high (“hot spots”) particle deposition (specifically at airway bifurcations) within the TB
9 tree.

10 In many studies, both toxicologic and epidemiologic, health endpoints are presented and
11 analyzed as a function of exposure concentration. However, it is generally accepted that the
12 dose to target cells or tissues, rather than exposure concentration per se, is responsible for
13 adverse responses. Appendix 7A provides analyses of relationships between rat and human lung
14 doses predicted for various exposure scenarios ranging from ambient PM exposures to PM
15 instillations into the lung. As noted in Appendix 7A, establishing firm linkages between
16 exposure and dose requires consideration of particle characteristics and biological normalizing
17 factors. Optimally, the dose metrics and normalizing factors should be based on the biological
18 mechanisms mediating an effect. For some effects, the mass of soluble PM depositing in a
19 region of the lung may be an appropriate dose metric. For example, an appropriate normalizing
20 factor for soluble PM could be the surface area of the airways for irritants, whereas body mass
21 might be more suitable when considering systemic effects.

22 First, experimental exposure concentrations can be estimated that should result in the same
23 tissue dose in a rat as received by a human exposed to various levels of ambient PM as a
24 function of dose metric, normalizing factor, and level of human exertion. As no single dose
25 metric nor normalizing factor appears to be appropriate for all situations, numerous scenarios
26 were considered in Appendix 7A. The parameters chosen can dramatically affect the rat
27 exposure concentration estimated to be required to provide a normalized dose equivalent to that
28 occurring in a human, as illustrated in Appendix 7A, (Tables 7A-7a through 7A-9b).

29 Second, the dose to the lung can be estimated for both animal and human inhalation
30 studies. These analyses make it possible to compare biological responses as a function of dose
31 rather than just exposure. Equal lung doses should not be assumed in comparing studies, even if

1 PM mass concentrations, animal species, and exposure times are identical. Differences in the
2 aerosol size distributions to which animals are exposed also affect dose delivered or retained.
3 For example, in an Appendix 7A comparison of several CAPs studies, one study was estimated
4 to have 1.7 times the alveolar dose of another study despite a 10% lower exposure concentration
5 in the first study. Thus, to make accurate estimates of dose, it is essential to have accurate and
6 complete information regarding exposure conditions, i.e., not only concentration and duration of
7 exposure, but also the aerosol size distribution and the level of exertion (and hence breathing
8 rates) for exposed subjects.

9 It was obviously not feasible, given the complexity involved, to attempt extrapolation
10 modeling for more than a few illustrative health endpoints from among those evaluated in the
11 vast array of studies assessed in Chapter 7 here. However, providing some illustrative modeling
12 results here that estimate comparative exposure concentrations/doses shown experimentally in
13 animal or human studies to be effective in producing a few important types of health endpoints
14 should be of value in helping to provide a context by which to gauge the potential relevance of
15 experimental results for ambient human exposure conditions.

16 ***Dosimetric Intercomparison for PMN Influx as a Marker for Lung Inflammation***

17 Various types of particulate materials (both ambient PM and combustion source particles)
18 have been shown to cause inflammation of the lung by migration of PMNs (predominantly
19 neutrophils) into the airways as discussed in Chapter 7 and summarized below. These cells are
20 initially produced by bone marrow and, along with alveolar macrophages (AM), constitute an
21 important defense mechanism triggered by invasion of PM, bacteria, or some other foreign
22 matter. The PMNs, once in the lung, ingest PM and then degranulate, forming hydrogen
23 peroxide and superoxide anions. Excessive quantities of PM in the lung can cause the lysosomal
24 enzymes in PMNs to enter the extracellular fluid, creating further inflammatory responses.
25 Additionally, PMN produce thromboxanes, prostaglandins, and leukotrienes.

26 Three new studies discussed in Chapter 7 and Appendix 7A provide data on PMN
27 increases following CAPs exposure that allow comparison of rat to human responses. (Clarke
28 et al., 1999; Kodavanti et al., 2000a; Ghio et al., 2000a). Chapter 7 dosimetric intercomparison
29 analysis of polymorphic neutrophil (PMN) data generated from exposures of rats and humans to
30 CAPs in these studies demonstrated that healthy humans are more susceptible to the
31

1 inflammatory effects of CAPs than are rats. By assessing increases in PMN numbers in both
 2 species, a retained alveolar dose of 28 to 47 $\mu\text{g}/\text{m}^2$ causes a 60 to 500% increase in PMNs in rats,
 3 whereas it was estimated that a retained dose of 0.7 $\mu\text{g}/\text{m}^2$ causes a 267% increase in PMNs in
 4 humans. The full array of modeling results is presented in Table 9-4.

TABLE 9-4. CAPS: RAT AND HUMAN INHALATION STUDY COMPARISONS

Study	Species	Particle	Exposure Conc. ($\mu\text{g}/\text{m}^3$)	MADD (σ_g)	Exposure duration	Analysis PE	Change in PMN	Estimated alveolar dose per surface area
Kodavanti et al. (2000a)	SD rat SO ₂ -SD	RTP CAPs	740	0.98 (1.41)	6 h/day for 2-3 days	< 3 h	255% \uparrow PMN in 2 of 4 exp (bronchitic rats only) no change in PMN	ND
						18 h		28 $\mu\text{g}/\text{m}^2$ retained
Clarke et al. (1999)	SD rat SO ₂ -SD	Boston CAPs	515	0.18 (2.9)	5 h/day for 3 days	24 h	500% \uparrow PMN 367% \uparrow PMN	47 $\mu\text{g}/\text{m}^2$ retained
Ghio et al. (2000a)	humans	Chapel Hill CAPs	120	0.65 (2.35)	2 h	18 h	267% \uparrow PMN	0.7 $\mu\text{g}/\text{m}^2$ retained

1 ***Inhibition of Phagocytosis by PM Exposure***

2 Phagocytosis is a form of endocytosis wherein bacteria, dead tissue, or other foreign
 3 material (e.g., inhaled ambient particles) are engulfed by cells such as AM, MO, or PMN as part
 4 of normal lung defense mechanisms. Hence, increased numbers of AM, MO, or PMN cells in
 5 lung tissue are an indicator of mobilization of lung defenses in response to infection or
 6 deposition of inhaled particles. Once ingested by AM, lysosomes act to digest engulfed
 7 materials. Inhibition of the phagocytosis by AM would signal interference with lung defense
 8 mechanisms by which inhaled bacteria and viruses are killed or other foreign particles are
 9 detoxified and/or cleared from the lung. Also, if an AM is overwhelmed by the amount or
 10 toxicity of ingested material, that material may be released along with the AM's digestive
 11 enzymes onto the alveolar surface and numbers of AM or their phagocytic activities may
 12 decrease.

1 Several experimental (especially in vitro) studies discussed in Chapter 7 have
2 demonstrated, that in some instances, one or another type of PM has caused an inhibition of
3 phagocytosis. As with other endpoints affected by PM, this inhibitory effect is determined by
4 the size and composition of the specific particle mixes tested.

5 Analysis in Chapter 7 of in vitro exposure data evaluating inhibition of phagocytosis in
6 rodent and humans showed some important species differences. Human AMs demonstrated
7 inhibition of phagocytosis at 0.2 to 0.5 ng/cell (UAP and ROFA) and 0.05 ng/cell (Utah Valley
8 PM). Hamster AMs showed no inhibition of phagocytosis at doses up to 0.04 ng/cell CAPs and
9 0.4 ROFA. A mouse AM cell line showed inhibition of phagocytosis at concentrations of
10 0.013 to 0.025 ng/cell. Differences in inhibition may be attributed to interspecies variability in
11 the capacity of AM, wherein rodent AMs are smaller, have less capacity for phagocytosis, and
12 are inhibited at a lower burden of PM per cell.

13 14 **9.2.3.2.4 General overview of toxicologic findings**

15 Dose-response relationships and extrapolation of experimental PM effects on both
16 cardiovascular and respiratory endpoints were discussed in Chapter 7. Some of the more salient
17 new toxicological findings that have emerged for the three general categories of effects
18 implicated by the epidemiology studies are summarized below.

19 20 **Cardiovascular/Systemic Effects**

21 Controlled human exposure studies have yielded some limited but interesting evidence for
22 ambient PM effects on cardiac physiological function (as indexed by ECG readings) or systemic
23 endpoints (as indexed by vasopressor control, blood coagulation control, etc.) linked to more
24 serious cardiovascular events. Cardiovascular and systemic effects of inhaled PM were observed
25 with CAPs, UAP, and ROFA. Probably of most note, the controlled human exposure CAPs
26 study by Ghio et al. (2000a) and another by Petrovic et al. (2000) did find evidence indicating
27 that ambient levels (~25 to ~125 to 300 $\mu\text{g}/\text{m}^3$) of inhaled $\text{PM}_{2.5}$ can produce some biochemical
28 changes (increased fibrinogen) in blood suggestive of PM-related increased risk for
29 prothrombotic effects. Blood fibrinogen levels increased in humans with exposures of 125 to
30 330 $\mu\text{g}/\text{m}^3$ CAPs and in both normal and compromised dogs at 69 to 828 $\mu\text{g}/\text{m}^3$; and Ulrich et al.
31 (2002) found a 20% increase in plasma fibrinogen in rats 2 days after instillation exposure to 6.7

1 or 22.2 mg/kg of Ottawa EHC93 UAP extract. Also, decreased Factor VII levels were observed
2 by Gong et al. (2003) in humans (with 2-h CAPs exposure at $\sim 174 \mu\text{g}/\text{m}^3$) and by Reed et al.
3 (2004) in rats (with DE exposure 6h/day, 7 day/wk, for 1 wk at 300 and 1000 $\mu\text{g}/\text{m}^3$), perhaps
4 reflecting that enzyme being consumed in an ongoing coagulation process. Also, strain
5 differences were found for effects on plasma fibrinogen levels, blood cell counts, and cardiac
6 lesions in rats at doses of 10 to 15 mg/m^3 with exposures to ROFA via inhalation. On the other
7 hand, the same and many other human and animal studies did not find significant changes in
8 other factors (e.g., increased platelets or their aggregation) related to blood coagulation control.
9 Additional other studies have shown no cardiovascular effects in rats and dogs with CAPs
10 exposures of 3-360 $\mu\text{g}/\text{m}^3$. Inhalation of ROFA exposures demonstrated effects such as
11 arrhythmias, ECG abnormalities, and decreased heart rate variability in rodents and dogs at
12 3 to 15 mg/m^3 .

13 Instilled UAP and ROFA have been found to have cardiovascular and systemic effects in
14 laboratory animals. Ottawa UAP instilled intratracheally at 7 mg/kg induced hypothermia and
15 bradycardia in rats and at 1.6 to 2 mg/kg caused increases in circulating PMN band cell numbers
16 and atherosclerotic lesions in rabbits. ROFA exposures of 0.7 mg/kg (compromised rats) and of
17 3 to 7 mg/kg (normal rat) have been shown to induce arrhythmias. The hypothermic response
18 was also seen in a similar concentration range in rats. Increased fibrinogen has been observed in
19 rats with exposures as low as 5 mg/kg. In many cases, compromised animals which model
20 human cardiovascular disease show effects at lower doses than their normal counterparts. More
21 rigorous characterizations of dose-response relationships with environmentally relevant levels
22 and species of PM are necessary to evaluate more fully the risk posed by ambient exposures.

23 Among the most salient hypotheses proposed to account for cardiovascular/systemic
24 effects of PM are: alterations in coagulability (Seaton et al., 1995; Sjögren, 1997); cytokine
25 effects on heart tissue (Killingsworth et al., 1997); perturbations in both conductive and
26 hypoxemic arrhythmogenic mechanisms (Watkinson et al., 1998; Campen et al., 2000); altered
27 endothelin levels (Vincent et al., 2001); and activation of neural reflexes (Veronesi and
28 Oortgiesen, 2001). Only limited progress has been made in obtaining evidence bearing on such
29 hypotheses, as discussed later; and much future research using controlled exposures to PM of
30 laboratory animals and human subjects will be needed to test further such mechanistic

1 hypotheses so as to more fully understand pathways by which low concentrations of inhaled
2 ambient PM may be able to produce life-threatening cardiovascular/systemic changes.

3 Overall, then, some available laboratory studies provide limited evidence suggesting that
4 relatively high concentrations/doses of inhaled or instilled particles can exert cardiovascular-
5 related systemic effects. However, many of the studies provide conflicting evidence, especially
6 with regard to heart rate, heart rate variability, or other ECG markers of cardiac function. Thus,
7 although some of the reported changes have been used as clinical “markers” for cardiovascular
8 diseases, the causal relationship between such PM-related changes and potential life-threatening
9 alterations in cardiovascular function remains to be better established.

11 *Respiratory Effects*

12 The respiratory effects of PM having varying physical and chemical characteristics have
13 been extensively studied for more than 30 years using a wide range of techniques and with
14 exposure durations ranging from brief periods to months. The most extensively studied
15 materials have been sulfates and acid aerosols formed as secondary pollutants in the atmosphere.
16 Fly ash from coal-fired power plants or other coal-combustion sources has been less extensively
17 studied. The toxicological data available today provide little basis for concluding that these
18 specific PM constituents have substantial respiratory effects at current ambient levels of
19 exposure. Recently, ROFA, a very specific kind of PM, has been studied extensively and found
20 to produce a range of respiratory effects, especially lung inflammation.

21 Probably of more direct relevance for present purposes, other recent studies evaluating
22 controlled human exposures to concentrated ambient particles (CAPs) from diverse locations
23 (e.g., Boston, New York City, Los Angeles, Toronto, and Chapel Hill, NC) have found little or
24 no effects on pulmonary function or respiratory symptoms in healthy human adults acutely
25 exposed (for 2 h) by inhalation to CAPs concentrations that ranged from about 25 up to about
26 300 $\mu\text{g}/\text{m}^3$. Some indications of mild lung inflammation were reported with such exposures in
27 some of the studies, but not others. Analogous controlled exposures to CAPs of rats, hamsters,
28 and dogs at concentrations varying across a range of ~ 100 to $1000 \mu\text{g}/\text{m}^3$ for 1-6 h/day for 1 to
29 3 days yielded similar minimal effects on respiratory functions, but did find some signs of mild
30 inflammation in normal healthy animals and somewhat enhanced indications of lung
31 inflammation in at least one compromised animal model of chronic bronchitis. More

1 specifically, inhalation CAPs exposures of ~100 to 1055 $\mu\text{g}/\text{m}^3$ caused decreased respiratory
2 rates and increased in BAL neutrophils in dogs, 200 to 700 $\mu\text{g}/\text{m}^3$ caused functional changes in
3 rats, and 650 $\mu\text{g}/\text{m}^3$ caused increased BAL protein and neutrophil in bronchitic rats. ROFA at
4 concentrations of 10 to 15 mg/m^3 caused increases in PMN, AM, BAL protein, LDH, and airway
5 hyperreactivity. Followup evaluations have produced new evidence for the transition metal
6 components of ambient PM from diverse locations and of ROFA having a mediating role in
7 producing inflammatory responses. Another inhalation study found indications of some
8 impairment of lung immune defense functions and exacerbation of bacterial infection with an
9 acute (3 h) exposure of rats to New York City CAPs (at 100-350 $\mu\text{g}/\text{m}^3$).

10 Instillation studies have also shown respiratory effects of PM on a variety of endpoints.
11 Exposures of humans to Utah Valley dust at concentrations of 0.007 mg/kg caused increases in
12 cytokines, fibronectin, fibrinogen, PMN, BAL protein, and tissue factor. Exposures of rats to
13 3 mg/kg caused similar changes, and 8 mg/kg caused lung lesions and airway reactivity.
14 Respiratory effects from ROFA instillations were seen in rodents in dose ranges of ~1 to
15 10 mg/kg .

16 Also, CAPs, UAPs, and ROFA, have all been used in in vitro experiments to demonstrate
17 effects and explore mechanisms whereby PM causes effects. Approximately 0.02 to 0.2 ng
18 PM/cell is the concentration range where in vitro effects (e.g., cytokine production, inhibition of
19 phagocytosis, and oxidant formation) were observed.

20 There still remains, however, a critical need for the systematic conduct of studies of the
21 potential respiratory effects of major components of PM from different regions of the U.S., in
22 recognition that PM of different composition and from different sources can vary markedly in its
23 potency for producing different respiratory effects. Of particular importance are studies that
24 more systematically evaluate mixtures of ambient constituents found in various airsheds,
25 including short-lived species, e.g., peroxides.

26 ***Mutagenic/Genotoxic Effects of PM***

27 As discussed in Chapter 8 and in Section 9.2.2, the Pope et al. (2002) extension of analyses
28 evaluating long-term ambient PM exposure effects on total (non-accidental) and cause-specific
29 mortality (using longer term follow-up data from the American Cancer Society or “ACS”
30

1 database) provides additional strong evidence for chronic ambient PM exposure being associated
2 with increased risks for lung cancer.

3 Several recent *in vivo* and *in vitro* toxicological studies have suggested that ambient urban
4 PM is mutagenic. Research evaluating the mutagenicity of ambient PM from the Los Angeles
5 area has pointed to ubiquitous emission sources as being responsible for mutagenic activity
6 observed *in vitro* (Hannigan et al., 1997, 1998). Fractionation of those ambient samples and
7 subsequent mutagenicity assessments have indicated that six unsubstituted polyaromatic
8 compounds and two semi-polar compounds are the likely mutagens. The former include pyrene
9 compounds (mainly from non-catalyst equipped gasoline engines), and the latter fluoranthene
10 compounds commonly found in vehicle exhaust or emitted by natural gas combustion.
11 Mutagenicity of urban air from heavily industrialized or traffic urban areas of Germany has also
12 been demonstrated (Hornberg et al., 1996, 1998; Seemayer and Hornberg, 1998) with evidence
13 showing that ambient PM_{2.5} exerted much stronger effects than PM₁₀. Additionally, ambient PM
14 from high traffic areas in The Netherlands has also been shown to induce genotoxic activity.

15 Emissions from wood/biomass burning have been shown to be mutagenic. Studies of
16 human exposures in The Netherlands (Heussen et al., 1994) and China (Vinitketkumnuen et al.,
17 2002), examining both chronic seasonal and acute exposures, have demonstrated increased
18 mutagenicity with environmental exposures. Characterization of wood smoke fractions to assign
19 mutagenicity have shown that the gaseous fraction is more mutagenic than the PM component
20 and that the condensate is not mutagenic (Putnam et al., 1999). Wood smoke emissions can
21 cause both frameshift and base pair mutations but have not yet demonstrated the production of
22 DNA adducts.

23 Coal combustion emissions have been shown to be mutagenic, especially the polar and
24 aromatic fractions. Research in China examining populations with high lung cancer rates have
25 shown that emission samples from homes burning smoky coal are mutagenic in the Ames assay,
26 and implicate PAHs as contributors to the mutagenicity (Mumford et al., 1987, 1999; Lan et al.,
27 2002). More recent work (Granville et al., 2003) characterizing the mechanism of genotoxicity
28 has examined the mutation spectra of coal smoke emissions from these Chinese homes.
29 Sequencing the revertants has demonstrated that the mutations in *Salmonella* exposed to coal
30 smoke extract are similar to mutations seen in lung tumors of women exposed environmentally
31 to the coal smoke.

1 Extensive past diesel exhaust (DE) studies have demonstrated mutagenic activity in both
2 particulate and gaseous fractions of DE. By sequential fractionation of DE, apportionment of the
3 mutagenicity is possible, which has implicated nitrated polynuclear aromatic compounds as
4 being responsible for a substantial portion of the mutagenicity. Other mutagenically active
5 compounds include ethylene, benzene, 1,3-butadiene, acrolein, and several PAHs in the gas
6 phase. In addition to Ames assay studies, the induction of gene mutations has been reported in
7 several in vitro mammalian cell lines after exposure to extracts of DPM. Structural chromosome
8 aberrations and SCE in mammalian cells have been induced by DE particles and extracts.

9 Older studies comparing the mutagenicity of gasoline and diesel exhaust showed that the
10 PM component of the exhaust is more mutagenic than the condensate fraction, and that overall,
11 diesel exhaust is more mutagenic than gasoline exhaust. More mutagenicity is also observed in
12 exhaust from cold starts than from exhausts at room temperature. Examining the fractional
13 mutagenicity of gasoline and diesel exhausts, it was shown that, as with coal smoke, the polar
14 component has the most mutagenicity, and further, that nitro-PAH is present in this fraction.
15 A comprehensive study (Seagrave et al., 2002) comparing gasoline and diesel exhaust
16 genotoxicity, using both the PM and SVOC fractions, demonstrated that both exhausts are
17 mutagenic, but, in general, diesel exhaust is more mutagenic. Further, the study implicates PAH
18 and nitroarenes in the genotoxicity. Another current study (Pohjola et al., 2003) corroborates
19 these findings, and includes data suggesting that DNA adduct formation is a component of the
20 mutagenicity.

21 Thus, there is qualitative evidence for the mutagenic/ genotoxic potential of both ambient
22 PM and some fuel combustion products. Many of the published in vitro studies failed to provide
23 details about the dose of PM extract delivered to the cells in vitro. In general, equal volumes of
24 air or amounts of time were generally sampled and reported, but little characterization of the
25 amount of PM mass or size was done or reported. Thus, any quantitative extrapolation of the
26 reported findings would be quite difficult. Still, they collectively provide extensive credible
27 evidence substantiating the biologic plausibility of, and/or elucidating potential mechanisms
28 underlying reported epidemiologic associations between lung cancer and long-term human
29 exposure to ambient fine particles.

1 **9.2.3.2.5 *Links between specific PM components/characteristics and health effects***

2 The epidemiology evidence reviewed in the 1996 PM AQCD and updated in this document
3 clearly shows positive associations between ambient PM pollution and mortality/morbidity.

4 Approaches to assessing likely “causation” and “biological plausibility” have attempted to
5 integrate the wealth of epidemiologic data with the growing body of toxicology information in
6 order to reveal coherence among the findings that support newly emerging sound hypotheses.
7 Thus, while it is often difficult to separate the physicochemical attributes of PM that may be of
8 health significance from the mechanisms by which individual factor(s) may function in the
9 response, hypotheses have been proposed that focus on various PM characteristics as potentially
10 significant contributors to the observed health effects (reviewed by Dreher, 2000). Each of the
11 attribute-based hypotheses has a sufficient data base to merit consideration and further
12 investigation.

13 To date, toxicologic studies on PM have provided important, albeit still limited, evidence
14 for specific PM attributes being important factors involved in the induction of cardiopulmonary
15 effects linked to ambient PM. In most cases, however, exposure concentrations in laboratory
16 studies have been inordinately high compared to the exposures at which epidemiologic studies
17 have found effects. Reasons for this dosimetric discrepancy include the typically limited use of
18 very young, elderly, unhealthy, or otherwise at-high-risk animals or humans, especially in light
19 of poorly understood risk factors. However, sufficient coherence in the epidemiologic and
20 toxicological data has added a level of “plausibility” to the observational studies and has thus
21 opened new avenues for investigation to link PM properties and constituents to specific sources
22 and to health outcomes.

23 The plausibility of epidemiologically-demonstrated associations between ambient PM and
24 increases in morbidity and mortality has been questioned because cardiovascular and pulmonary
25 effects have been observed among human populations at very low ambient PM concentrations.
26 To date, experimental toxicology studies have provided some intriguing, but limited, evidence
27 for ambient PM mixes or specific PM components potentially being responsible for reported
28 health effects of ambient PM. Overall, the new studies suggest that some types of particles are
29 more toxic than others. New findings substantiating the occurrence of health effects in response
30 to controlled exposures to (a) ambient PM mixes and/or (b) their constituent substances are

1 useful in demonstrating or clarifying potential contributions of physical/chemical factors of
2 constituent particles.

3 4 *Physical Properties*

5 **Ultrafine Particles (Size, Surface Area, Number).** The physical attributes of PM - size,
6 surface area and number - are intimately interrelated. These properties influence lung
7 deposition, penetrance and persistence in lung tissues, and systemic transport, and, in several
8 studies, apparently the inherent toxicity of the particle itself. While a few epidemiologic studies
9 (Wichmann et al., 2000) show correlations between health outcomes and ultrafine (< 100 nm)
10 ambient PM, the bulk of the information regarding its toxic potential, and the role of surface
11 area, has derived from studies of surrogate insoluble particles, such as mineral oxides
12 (e.g., TiO₂) and carbon black. Studies of various types of ultrafine particles have demonstrated
13 a significantly greater inflammatory response than that seen with fine particles of the same
14 chemical composition at similar mass doses (Oberdörster et al., 1992; Li et al., 1996, 1997,
15 1999). Instillation of 125 µg of ultrafine carbon black (20 nm) caused substantially more
16 inflammation per unit mass than did the same dose of fine particles of carbon black (200 to
17 250 nm), suggesting that ultrafine particles may cause more inflammation per unit mass than
18 larger particles (Li et al., 1997). However, the chemical constituents of the two sizes of carbon
19 black used in this study were not analyzed, and it cannot be assumed that the chemical
20 composition was the same. Further, when the particle surface area is used as a dosimetric, the
21 inflammatory response to both fine and ultrafine particles may be basically the same
22 (Oberdörster, 1996; Oberdörster et al., 2000; Li et al., 1996). In other more limited studies,
23 ultrafines also have generated greater oxidative stress in experimental animals. Inhalation
24 exposure of normal rats to ultrafine carbon particles generated by electric arc discharge
25 (100 µg/m³ for 6 h) caused minimal lung inflammation per unit mass (Elder et al., 2000a,b),
26 compared to ultrafine PTFE or metal particles.

27 These studies have shown that on an equivalent mass exposure-dose metric, ultrafine PM
28 can induce more acute lung injury than fine PM. Similarly, surrogate PM with high surface
29 areas induced more toxicity than those of like composition, but having smaller surface areas
30 (Lison et al., 1997). On the other hand, studies have shown that composition also matters; for

1 example MgO ultrafines produce less injury than ZnO (Kuschner et al., 1997), as did sparked
2 carbon versus similarly generated metal oxides (Elder et al., 2000a,b).

3 With regard to acid aerosols, studies of low concentrations of ultrafine sulfuric acid and
4 metal oxide particles have demonstrated effects in the lung.

5 Studies of ultrafine particles have focused largely on effects in the lung, but inhaled
6 ultrafine particles may also have the potential to be distributed systemically and have effects that
7 are independent of lung effects. Recent epidemiologic studies evaluating blood viscosity as a
8 biologic correlate of ultrafine exposures, have reported slight increases that raise the prospect of
9 potential cardiovascular implications (Wichmann et al., 2000). Thus, there is still insufficient
10 toxicological evidence to elucidate clearly the extent to which ambient concentrations or high
11 number counts of ultrafine particles may differentially contribute to increased health effects risks
12 associated with ambient PM air pollution.

13
14 **Fine and Thoracic Coarse Particles.** In contrast to ultrafine particles, the respective roles
15 of PM_{2.5} (indicator for fine PM) and PM_{10-2.5} (indicator for thoracic coarse PM) in defining health
16 outcomes have garnered considerable research attention because they are the most frequently
17 measured size-fractions of ambient PM and for which most health effects data exist. The fine
18 fraction comprises most of the combustion-related constituents discussed below under chemical
19 properties. The fine fraction has greater surface area than the thoracic coarse fraction, but less
20 surface area and much larger particle number than the ultrafine fraction. To the extent that
21 inhaled PM may carry chemicals or reactive species on their surfaces, these smaller size
22 fractions may have an additional dimension to their toxicity (in terms of surface chemical
23 bioavailabililty) that is not found with coarse PM. For example, acute exposure to sulfate-coated
24 carbon black was found to impair alveolar macrophage phagocytosis and intrapulmonary
25 bactericidal activity in mice (Jakab et al., 1996; Clarke et al., 2000). On the other hand, coarse
26 PM usually is of mineral (earthen) or biologic (discussed below) origin and, thus, has a less
27 complex bioavailable chemical matrix than the finer PM mode. The relative toxicity of most
28 earthen-derived PM has been observed to be less than that of the finer combustion-derived or
29 surrogate ultrafine particles. However, because ambient coarse PM would tend to impact on the
30 airways of humans, it is thought this fraction may be adverse to those with airways sensitivities
31 or disease (e.g., asthma).

1 *Chemical Properties*

2 **Acid Aerosols.** Controlled exposure studies assessed in the 1996 PM AQCD showed that
3 aqueous acid aerosols had little effect on pulmonary function or respiratory symptoms in healthy
4 young adults with inhatim exposure at concentrations as high as 1000 $\mu\text{g}/\text{m}^3$. On the other hand,
5 lung function effects were observed in adolescent asthmatics at concentrations as low as
6 68 $\mu\text{g}/\text{m}^3$; and modest bronchoconstriction was seen in adult asthmatics exposed to
7 concentrations $\geq 400 \mu\text{g}/\text{m}^3$, but the available data were not consistent. However, at
8 concentrations as low as 100 $\mu\text{g}/\text{m}^3$, acid aerosols can alter mucociliary clearance. That is, brief
9 exposures (≤ 1 h) to low concentrations ($\sim 100 \mu\text{g}/\text{m}^3$ may accelerate clearance while longer
10 (multihour) exposures to higher concentrations ($> 100 \mu\text{g}/\text{m}^3$) can depress clearance.

11 There is relatively little new information on the effects of acid aerosols, and the basic
12 conclusions of the 1996 PM AQCD largely remain unchanged. As noted, It was previously
13 concluded that acid aerosols cause little or no change in pulmonary function in healthy subjects
14 at levels well exceeding ambient concentrations, but asthmatics may experience small
15 decrements in pulmonary function at distinctly lower (but supra-ambient) levels. Long-term
16 exposures of animals to acid aerosols, on the other hand, have been shown to alter airway
17 morphology with epithelial cell desquamation and an increase in secretory cells, but these
18 changes have been considered relatively minor. The conclusions about acute health effects,
19 however, appear to be supported by a newer study by Linn and colleagues (1997), in which
20 healthy children (and children with allergy or asthma) were exposed to sulfuric acid aerosol
21 ($100 \mu\text{g}/\text{m}^3$) for 4 hours. While there were no significant effects on symptoms or pulmonary
22 function when the entire group was analyzed, the allergy group did have significant acid-related
23 increases in symptoms, although the acid concentrations were distinctly higher than typical
24 ambient concentrations. Also, Leduc et al. (1995) found no increased bronchoconstriction or
25 bronchial hyperresponsiveness in asthmatic adults exposed via a facemask to $500 \mu\text{g}/\text{m}^3$ of acid
26 fog containing H_2SO_4 or ammonium sulfate aerosol.

27 Several other laboratory animal studies found little effect of sulfate (ammonium bisulfate,
28 ammonium ferrosulfate) or ammonium nitrate aerosols on lung inflammation markers or
29 indicators of AM function in normal, sensitized, or MCT-compromised mice with inhalation
30 exposures (4 h/day for 3 days) at concentrations varying from 70 to $\sim 970 \mu\text{g}/\text{m}^3$ (Cassee et al.,
31 1998a,b,c). In another study (Zelikoff et al., 1997), of 3 h exposures to $1000 \mu\text{g}/\text{m}^3$ of H_2SO_4 ,

1 for both rabbits and humans, superoxide production by macrophages was somewhat depressed in
2 both species, and macrophage phagocytosis and antimicrobial activity was reduced in the
3 rabbits.

4 Although pulmonary effects of acid aerosols have been the subject of extensive research,
5 the cardiovascular effects of acid aerosols have received little attention. One example which
6 raises the issue is a study of acetic acid fumes where reflex mediated increases in blood pressure
7 were found in normal and spontaneously hypertensive rats (Zhang et al., 1997). Similarly, acidic
8 residual oil fly ash (ROFA), which also contains a considerable amount of metal sulfates, was
9 found to alter ecocardiogram (ECG) patterns in the same strain of rats at high air concentrations
10 (Kodavanti et al., 2000b). Thus, acidic components should not be entirely dismissed as possible
11 mediators of ambient PM health effects, since so little is known about potential cardiovascular
12 impacts or impacts in compromised subjects.

13
14 **Transition Metals.** The 1996 PM AQCD (U.S. Environmental Protection Agency, 1996a)
15 mainly relied on data related to occupational exposures to evaluate the potential toxicity of
16 metals in contributing to health effects associated with ambient PM exposures. Since that time,
17 numerous newly published in vivo and in vitro studies using exposures to ambient PM extracts,
18 ROFA, other combustion source emission materials (e.g., CFA, etc.), or specific soluble
19 transition metals have contributed substantial further information on the health effects of
20 particle-associated soluble metals. Although there are some uncertainties about differential
21 effects of one transition metal versus another, some water soluble metals (e.g., Ni, V, Zn, Fe)
22 leached from ambient filter extracts or ROFA have been shown consistently (albeit at high
23 concentrations) to cause cell injury and inflammatory changes in vitro and in vivo.

24
25 **Other Inorganic Constituents.** The inorganic constituents of ambient PM comprise a
26 number of compounds and elements that derive from either natural or combustion sources. The
27 earthen or natural constituents of PM are typically silicates that contain surface and matrix
28 bound metals such as calcium, magnesium, aluminum, and iron. As noted above, most of these
29 silicates do not appear to contribute much toxicity to ambient PM, as considered in this
30 document. Sulfate and nitrate anions derived from combustion or photochemical processes
31 usually complex with other constituents in PM - often more water-soluble ammonium ions or

1 organic acids, as well as elemental cations, such as metals. The intrinsic, independent toxicities
2 of sulfates (as per above) and nitrates appear to be rather low, but they may influence the toxicity
3 or bioavailability of other PM components. Of the cations, metals represent a potential class of
4 causal constituents for PM-associated health effects that have received considerable attention
5 (discussed further below). Sulfate, nitrate, ammonium, and metals make up a substantial part of
6 the mass of ambient PM, often with a silicate or carbonaceous (see below) core, layering, or
7 matrix. The majority of PM-associated metals in fine PM are derived from stationary or mobile
8 combustion sources whereas particle sulfate, nitrate and ammonium originate from secondary
9 atmospheric transformation reactions of involving SO₂, NO_x and biomass ammonia emissions.
10 Organic PM has both primary and secondary sources.

11
12 **Organic Constituents.** Published research on the acute effects of PM-associated organic
13 carbon constituents is conspicuous by its relative absence, except for diesel exhaust particles
14 (DPM). Like metals, organics are common constituents of combustion-generated PM and are
15 found in ambient PM samples over a wide geographical range. Organic carbon constituents
16 comprise a substantial portion of the mass of ambient PM (10 to 60% of the total dry mass
17 [Turpin, 1999]). Although the organic fraction of PM is a poorly characterized heterogeneous
18 mixture of a widely varying number of different compounds, strategies have been proposed for
19 examining the health effects of potentially important organic constituents (Turpin, 1999).
20 In contrast, the mutagenic effects of ambient PM and evidence of DNA-adducts have had more
21 extensive study and have been linked to specific organic fractions (Binková et al., 1999; Chorąży
22 et al., 1994; Izzotti et al., 1996). The extent to which organic constituents of ambient PM
23 contribute to adverse health effects identified by current epidemiology studies is not known.
24 Nevertheless, organic constituents remain of concern regarding PM health effects due in large
25 part to the contribution of DE particles to the fine PM fraction and the health effects associated
26 with exposure to these particles.

27
28 **Biological Constituents.** Recent studies do not fully support the strong conclusion of the
29 1996 PM AQCD that bioaerosols (e.g., fungal spores, plant and insect fragments, airborne
30 bacteria, etc.), at concentrations present in the ambient environment, are unlikely to contribute to
31 health effects of ambient PM. On the one hand, dose-response inhalation studies in healthy

1 volunteers exposed to 0.55 and 50 μg endotoxin showed the threshold for pulmonary and
2 systemic effects for endotoxin to be between 0.5 and 5.0 μg (Michel et al., 1997). Urban
3 ambient air PM contains variable amounts of endotoxin, but the levels typically are several
4 orders of magnitude less. In vitro toxicological studies have also shown endotoxin associated
5 with ambient PM to be pro-inflammatory, inducing cytokine expression in human and rat
6 alveolar macrophages, which appears to depend heavily on the endotoxin dose to cell ratio
7 (Becker et al., 1996; Dong et al., 1996). However, endotoxin content does appear to vary by PM
8 size-mode. Monn and Becker (1999) demonstrated cytokine induction by human monocytes,
9 characteristic of endotoxin activity, in the coarse size fraction of outdoor PM, but not in the fine
10 fraction. Interestingly, while studies in animal models also require more endotoxin than
11 typically found in ambient PM to induce inflammation, some findings suggest that endotoxin
12 may have a priming effect on PM-induced inflammatory processes (Imrich et al., 1999). Thus,
13 the role of biogenic material like endotoxin may have a subtle role that is poorly understood.

14 On the other hand, clearer new insights have been gained with regard to the fact that
15 allergen-laden cytoplasmic fragments of pollen grains are produced that range to very small
16 (0.1 - 0.4 μm) fine-mode size upon rupture of pollen granules, which is highly moisture
17 dependent and thought to be the main cause of increased incidence of “thunderstorm” asthma
18 characterized by dramatic post-storm increases in asthma attacks and medical visits/treatments.
19 Important roles of pollen spores or grains as carriers of other biologic agents (e.g., endotoxins,
20 fungi fragments) are discussed below in Section 9.2.3.2.4, as are the roles of non-biological
21 particles in serving as carriers of allergen-laden pollen fragments or other toxic bioaerosol
22 agents.

23 24 ***Mixtures: Ambient and Source PM***

25 Ambient PM comprises a complex mix of constituents derived from many sources, both
26 natural and anthropogenic. Hence, the physicochemical composition of PM generally reflects
27 the major contributing sources locally and regionally. Within this framework of source or
28 origin, PM composition also varies significantly by the size-mode within which it is classified
29 (ultrafine, accumulation, or coarse). It should be clear that any given particle can differ
30 appreciably from another individual particle of similar size, but that the region of origin with all
31 of its contributing sources determines the general composition of the generic PM in that

1 classification mode. By its nature then, exposure to airborne ambient PM constitutes an
2 exposure to what is very clearly a mixture of different particles of differing composition and to
3 other gaseous copollutants that coexist in that air-shed. Particle concentrators are in use that
4 separate particles between about 0.1 μm and 2.5 μm and concentrate them for use in inhalation
5 exposure studies without removing the particles from the atmospheric gases in which they occur.
6 Studies have also used diluted diesel without removing the particles from the diesel exhaust.
7 Other studies make use of particles collected by impactors, filters, electrostatic precipitators, or
8 bag houses and then resuspend them for inhalation studies or make extracts or suspensions for
9 instillation or *in vitro* studies.

10
11 **Concentrated Ambient Particles (CAPs).** Studies using CAPS are probably most useful
12 in helping to substantiate that particles present in “real-world” ambient air mixes are indeed
13 capable of inducing notable pathophysiological effects under controlled exposure conditions and
14 to clarify further factors affecting increased susceptibility of “at risk” groups for PM effects.
15 CAPs studies, on the other hand, have thus far been somewhat less helpful than other toxicologic
16 approaches in helping to delineate the specific characteristics of PM producing toxicity and
17 pertinent underlying mechanisms. Some, but not all, studies with inhaled CAPs have found
18 cardiopulmonary changes in rodents and dogs at high concentrations of fine PM. However, no
19 comparative studies to examine the effects of ultrafine and coarse ambient PM have been done.

20 Importantly, it has become evident that, although the concentrated ambient PM (CAPs)
21 studies have provided important exposure-response information for some PM size fractions
22 (especially $\text{PM}_{2.5}$), they have not, to date, been very helpful in identifying specific toxic
23 components in urban PM. Insufficient attention has been accorded to characterization of day-to-
24 day variations in specific PM constituents in order to relate such variations to observed variable
25 health responses to CAPs exposures. New particle concentrator systems now coming on-line at
26 the U.S. EPA and elsewhere that permit selective concentration of ultrafine, fine, and thoracic
27 coarse PM hold promise for enhancing our future understanding of PM characteristics producing
28 toxicity. Future CAPs studies also hold promise for helping to identify susceptibility factors in
29 animal models and to permit examination of mechanisms related to PM toxicity.

1 **Collected Urban Air Particles.** Studies using extracts of collected urban air PM (UAP)
2 for intratracheal administration to healthy and compromised animals have also produced
3 interesting new information. Despite the difficulties associated with extrapolating from the
4 bolus delivery used in such studies, they have provided evidence indicating that the chemical
5 composition of ambient particles can have a major influence on toxicity. Instillation of rats with
6 filter extracts of ambient air particles collected from Ottawa CN air (Watkinson, et al., 2002a,b)
7 at 2.5 mg, for example, induced pronounced biphasic hypothermia, severe drop in heart rate, and
8 increased arrhythmias; this was in contrast to no cardiac effects seen with comparable instilled
9 dose of Mt. St. Helens volcanic ash (shown by many studies to be relatively inert
10 toxicologically). Similarly, dose-dependent increases in polymorphonuclear leukocytes (PMNs),
11 other markers of lung inflammation, and decreases in alveolar macrophages (AMs) were seen
12 with intratracheal exposures of hamsters to urban ambient particles from St. Louis or to Kuwaiti
13 oil fire particles (Brain et al., 1998).

14 Perhaps most notable in this argument are the Utah Valley studies that have linked the
15 toxicology (*in vitro* cell culture as well as human and rodent instillation) with published
16 epidemiological findings. In these studies, filter extracts of Utah Valley PM collected from the
17 State/Federal sampling sites yielding aerometric data used to ascribe the impact of PM on
18 hospital admissions and population mortality rates showed remarkable qualitative coherence
19 with toxicological and clinical endpoints (BAL fluid markers, lung dysfunction) among the
20 human and rodent test subjects. Moreover, the data were themselves consistent with the
21 hypothesized underlying mode of action (oxidant generation, inflammation) for metal-associated
22 PM cardiorespiratory effects (Frampton et al., 1999; Dye et al., 2001; Ghio and Devlin, 2001;
23 Soukup et al., 2000; Wu et al., 2001; Pagan et al., 2003). Studies comparing human (Ghio and
24 Devlin, 2001) and rat (Dye et al., 2001) exposures to both high and low metal content PM
25 collected near a steel plant, suggest that the metal content of the PM was an important
26 contributor to the toxicity of the PM. Both species showed similar inflammatory responses to
27 exposures from PM with high metal content (collected while the steel mill was operating).
28 Hence, these findings provide an important linkages across study disciplines used in the human
29 and animal toxicology as well as in the *in vitro* studies.

30 Since the Utah studies were completed, an analogous study has addressed differential
31 exacerbation of allergic asthma-related responses by PM from two German cities (Hettstedt and

1 Zerst) of contrasting industrial activity. An allergic mouse model (representing an allergic
2 asthma population) was intratracheally instilled with filter extracts from each city and the
3 appropriate allergen to activate the model. The respective responses of the model corresponded
4 to the prevalence of allergy and respiratory disease in the cities and appeared to be influenced by
5 the ambient PM metal content in the respective cities. Hence, the data base is growing for
6 studies linking animal and human responses. Some of these linkages are in the laboratory while
7 others are with epidemiology. Why these collective data show coherence despite exposure/dose
8 discrepancies, not to mention species and other differences, is unclear, but the data and findings
9 stand on their merits and attest to the legitimacy of the approach and the value of the animal data
10 in establishing biologic plausibility and insight into potential mechanisms.

11 Even though it is clear that combustion particles that have a high content of soluble metals
12 can cause lung injury in compromised animals and correlate well with epidemiological findings
13 in some cases (e.g., the Utah Valley Studies), it has not been fully established that the small
14 quantities of metals (typically ≤ 0.5 to $1.0 \mu\text{g}/\text{m}^3$) associated with current U.S. ambient PM mass
15 concentrations exhibit greater toxicity than other PM components typically present in ambient
16 air. In studies in which various ambient and combustion-source particulates were instilled into
17 rats, the soluble metal content did appear to be one important determinant of lung injury (Costa
18 and Dreher, 1997). However, one published study (Kodavanti et al., 2000b) has compared the
19 effects of inhaled ROFA (at $1 \text{ mg}/\text{m}^3$) to concentrated ambient PM (four experiments, at mean
20 concentrations of 475 to $900 \mu\text{g}/\text{m}^3$) in normal and SO_2 -induced bronchitic rats. A statistically
21 significant increase in at least one lung injury marker was seen in the bronchitic rats with one out
22 of four of the CAPs exposures; whereas the inhaled ROFA had no effect, even though the
23 content of soluble iron, vanadium, and nickel was much higher in the ROFA sample than in the
24 concentrated ambient PM. This suggests that substances present in some ambient air (but
25 perhaps not in ROFA) besides soluble metals may contribute to a stronger potency of ambient
26 air, at times, than seen with some oil combustion source-related materials.

27 Other particularly interesting new findings do point toward ambient PM exacerbation of
28 allergic airway hyperresponsiveness and/or antigen-induced immune responses. Both metal and
29 diesel particles have been implicated, with an expanding array of new studies showing DPM in
30 particular as being effective in exacerbating allergic asthmatic responses, as noted below.

31

1 **Diesel Exhaust Particles.** As described in Section 7.5.3, there is growing toxicological
2 evidence that, analogously to several other types of PM (silica, carbon black, road dust, etc.),
3 diesel PM may exacerbate allergic responses to inhaled antigens. The organic fraction of diesel
4 exhaust has been linked to eosinophil degranulation and induction of cytokine production,
5 suggesting that the organic constituents of diesel PM are the responsible part for the immune
6 effects. It is known that the adjuvant-like activity of DEP is not unique, and that certain metals
7 have analogous adjuvant effects (Lambert et al., 2000). It is important to compare the immune
8 effects of other source-specific emissions, as well as concentrated ambient PM, to diesel PM to
9 determine the extent to which exposure to diesel exhaust PM may contribute to the incidence and
10 severity of allergic rhinitis and asthma. It is also notable that rather direct evidence has been
11 obtained which demonstrates adherence of allergen-laden pollen cytoplasm fragments to diesel
12 particles, providing a likely mechanism by which diesel PM acts to concentrate bioaerosol
13 materials and to increase their focal accumulation in lower regions of the respiratory tract. Other
14 evidence substantiates mutagenic/genotoxic effects of diesel emission particles (e.g., PAHs),
15 consistent with qualitative findings in several studies of increased lung cancer effects being
16 associated with long-term, occupational exposure to diesel emissions.

17 18 ***Summary***

19 Toxicological studies have provided considerable supportive evidence that certain
20 physicochemical particle attributes can provide elements of “causality” to observed health
21 effects of ambient PM. There is probably no single primary causative attribute, but rather many
22 attributes may contribute to complex mechanisms driven by the nature of a given type of PM and
23 its contributing sources. The multiple interactions that may occur in eliciting a response in a
24 host may make the identification of any single causal component difficult and may account for
25 the fact that mass, as the most basic PM metric, shows the relationships to health outcomes that
26 it does.

27 28 **9.2.3.2.6 *Mechanisms of action***

29 As discussed in Chapter 7, the body of evidence supporting various hypotheses regarding
30 induction of PM effects has grown substantially since the last review. Various toxicologic
31 studies using PM having diverse physicochemical characteristics have shown that such

1 characteristics have a great impact on the specific response that is observed. Thus, there appear
2 to be multiple biological mechanisms that may be responsible for observed morbidity/mortality
3 due to exposure to ambient PM, and these mechanisms appear to be highly dependent on the
4 type and dose of particle in the exposure atmosphere. It also appears that many biological
5 responses are produced by PM whether it is composed of a single component or a complex
6 mixture. The potential mechanisms for action on the cardiovascular and pulmonary systems that
7 were discussed in more detail in Section 7.9.1 are summarized briefly below.

8
9 ***Direct Pulmonary Effects:***

10 Lung Injury and Inflammation. Exacerbation of respiratory disease by ambient PM may be
11 caused in part by lung injury and inflammation. Recent studies have reported effects including
12 increased levels of neutrophils, protein, and inflammatory cytokines, and increase in airway
13 hyperresponsiveness. Some recent studies have implicated metal components of PM as
14 contributing to inflammation and injury in the lung. In contrast, controlled exposures of animals
15 to sulfuric acid aerosols, acid-coated carbon, and sulfate salts cause little lung injury or
16 inflammation, even at high concentrations. Some (but not all) studies have shown CAPs
17 exposure to cause mild pulmonary injury and inflammation. There are also new data indicating
18 a potential neurogenic basis for the effects of particulate matter, with particles potentially
19 activating certain receptors found on human airway epithelial cells and sensory terminals; this
20 activation, could initiate and sustain inflammatory events in the pathophysiology of neurogenic
21 inflammation. Particle surface charge is important, with negatively charged particles
22 surrounded by proton cloud being effective in activating the receptor.

23
24 Increased Airway Reactivity and Exacerbation of Asthma. The strongest evidence supporting
25 this hypothesis is derived from studies on diesel particulate matter (DPM), which has been
26 shown to increase production of antigen-specific IgE in mice and humans (as summarized in
27 Chapter 7). Biological agents (e.g., pollen fragments) also contribute to asthma effects.

28
29 Increased Susceptibility to Respiratory Infections. A few newly published studies have provided
30 some evidence for ambient PM potentially affecting lung defense mechanisms and increasing
31 susceptibility to infection. Several new studies have suggested that particles can increase

1 numbers of alveolar macrophages and increase bacterial burden, slow clearance of the bacteria,
2 reduce AM NO production, and decrease phagocytic activity of alveolar macrophages (AM).
3 These data are suggestive of possible impairment of an important lung defense mechanism even
4 in the absence of lung injury.

6 *Cardiovascular and Other Systemic Effects Secondary to Lung Injury*

7 **Decreased Pulmonary Function and Oxygenation Adversely Affect the Heart Secondary to**
8 **Lung Injury**. Results from new toxicology studies in which animals (normal and compromised)
9 were exposed to CAPs (at concentrations many times higher than would be encountered in the
10 United States) indicate that ambient PM is unlikely to cause severe disturbances in oxygenation
11 or pulmonary function. However, even a modest decrease in oxygenation can have serious
12 consequences in individuals with ischemic heart disease. Evidence from some new studies
13 indicate that it is plausible that instilled ROFA may cause severe hypoxemia leading to death;
14 more information is needed, however, on the effects of PM on arterial blood gases and
15 pulmonary function to fully address the above hypothesis.

17 **Systemic Hemodynamic Effects Secondary to Lung Inflammation and Increased**

18 **Cytokine Production**. It has been suggested that systemic effects of ambient PM may result
19 from activation of cytokine production in the lung, with cytokine releases in the lung resulting in
20 systemic changes such as arrhythmias or changes in heart rhythm. While some new studies have
21 provided suggestive evidence, further information is needed on the effects of mild pulmonary
22 injury on cardiovascular function to more fully evaluate this hypothesis.

24 **Increased Blood Coagulability Secondary to Lung Inflammation**. Several new studies have
25 investigated possible effects of ambient PM or surrogate particles on blood chemistry
26 constituents that would be indicative of increased blood coagulability. There is abundant
27 evidence linking small prothrombotic changes in the blood coagulation system to increased long-
28 term risk of heart attacks and strokes. Some studies have shown both indicators of lung
29 inflammation and increased coagulability; however, the published toxicological evidence
30 bearing on whether moderate lung inflammation causes increased blood coagulability is very
31 mixed and inconsistent. The coagulation system is multifaceted and complex; and there are

1 many other sensitive and clinically significant parameters that should, in addition to fibrinogen,
2 show more extensive and consistent patterns of change reflective of PM effects on blood
3 coagulation.

4
5 **Hematopoiesis Effects Secondary to PM Interactions With the Lung.** Some new studies
6 have shown increased release of immune cells, such as PMNs, from bone marrow; however,
7 consistent evidence that PM ambient concentrations can affect hematopoiesis remains to be
8 demonstrated.

9
10 *Direct Effects on the Heart*

11 **Cardiac Effects Secondary to Uptake of Particles and/or Release of Soluble Substances into**
12 **the Circulation.** Particles or particle components could conceivably be rapidly transported to
13 the heart, where they might exert effects directly on cardiac vasculature (e.g., exacerbation of
14 atherosclerosis) or heart muscle itself. Alternatively, they could also exert very rapid effects on
15 cardiac function through stimulation of nerve ending receptors in lung tissue, resulting in
16 secretion of inflammatory messenger substances and/or activation of neurally-mediated
17 autonomic reflexes.

18
19 **Inhaled PM Effects on Autonomic Control of the Heart and Cardiovascular System.**

20 Changes in sympathetic and parasympathetic input to the cardiovascular system can result in
21 changes in heart rhythm; these changes may be mediated by neural reflexes. Though not always
22 consistent and somewhat difficult to interpret, changes in heart rate variability and conductance
23 system function associated with ambient PM exposure have been reported in some animal
24 studies and in several epidemiologic studies.

25
26 *9.2.3.2.7 Inhaled particles as potential carriers of toxic agents*

27
28 *Particle-Bound Water*

29 In Chapter 2, it was noted that, although water vapor is not considered a pollutant per se
30 and particle-bound water is not measured as part of the ambient PM mass typically monitored for
31 regulatory purposes, particle bound water may serve as a carrier for other pollutants. Wilson
32 (1995) proposed that water-soluble gases that are usually largely removed by deposition to wet

1 surfaces in the upper (ET) portion of the respiratory tract could be dissolved in particle bound
2 water and, thereby, be carried into the lower regions of the respiratory tract. Such water-soluble
3 gases commonly found in polluted air masses include: oxidant species (e.g., O₃, H₂O₂, and
4 organic peroxides); acid gases (e.g., SO₂, HCl, HNO₃, HONO, and formic acid); and polar
5 organic species (e.g., formaldehyde). Thus, water may be a vector by which these gases may be
6 delivered in enhanced proportions to the TB and A regions of the deep lung.

7 Kao and Friedlander (1995) noted further that, in evaluating health effects of ambient
8 aerosol components, it is “important to realize that the chemical analyses of routinely collected
9 particulate samples are not necessarily an accurate representation of the atmosphere”. They
10 further noted that many short-lived chemical species in the gas or particle phase, such as free
11 radicals, may not be present in the sampled materials when analyzed hours to weeks (or longer)
12 after being collected on filters and stored. Also, the unmeasured metastable species may be
13 much more biochemically active than “dead” components collected or remaining on analyzed
14 filters. They concluded that, “since inhalation toxicology studies using both human and animal
15 subjects often do not include the potential for metastable species and reactive intermediaries to
16 be present, they could greatly underestimate the effects seen in field or epidemiologic studies.”

17 Friedlander and Yeh (1998) elaborated further on the fact that the aqueous component of
18 the atmospheric submicron aerosol contains short-lived reactive chemical species. That is, they
19 explained that submicron atmospheric aerosols contain several types of components:

20 (1) Primary components that include elemental (black) carbon; high molecular weight organic
21 compounds emitted in aerosol form directly into the atmosphere; metallic compounds from
22 smelting, welding, and other high temperature processes; and some small particles from soil dust
23 or, in marine aerosols near coastal sites, sea salts; (2) Secondary components resulting from
24 atmospheric reactions that yield inorganic ionic species (NH₄⁺, SO₄⁼ and NO₃⁻ being most
25 important per mass basis) and, also, polar condensible products from atmospheric reactions
26 involving organic vapors; (3) Water, the presence of which in the ambient submicron aerosol
27 depends heavily on the relative humidity and the concentration of which can range from ~10 to
28 50 µg/m³ in urban aerosols; and (4) Very short-lived reaction intermediates, such as hydrogen
29 peroxides, aldehydes, and organic acids found in cloud and rain water.

30 Friedlander and Yeh (1998) further noted (1) that particle phase concentrations of
31 hydrogen peroxide fall in a range for which significant biochemical effects were elicited with

1 treatment of respiratory tract epithelial cells; (2) that this may help to explain epidemiologic
2 study results showing significant health effects to be associated with fine-mode aerosols or
3 sulfate (the submicron sulfate-containing aerosol often being the product of atmospheric
4 reactions involving hydrogen peroxide), and (3) that such aerosols may be serving as a surrogate
5 or indicator for the hydrogen peroxide or other reactive species.

6 Wexler and Sarangapani (1998) used a physical model of “gas-particle-mucus heat and
7 mass transport in the human airways” to investigate the transport by particles of soluble vapors
8 to the tracheobronchial and air exchange regions of the lung. When the atmospheric aerosol is
9 inhaled, water soluble gases will begin to dissolve in the mucus on the surface of the airways.
10 However, hygroscopic particles will add particle-bound water in the high relative humidity of
11 the respiratory tract and more soluble gas can dissolve in the particle. The amount of soluble gas
12 in the particle will depend on the solubility of the gas (expressed as the Henry’s Law coefficient),
13 the size of the particle, and the position of the particle in the respiratory tract. In the presence of
14 particles, the pattern of deposition of soluble gases may be moved deeper into the respiratory
15 tract. Very soluble gases, such as H_2O_2 and formaldehyde will still be almost completely
16 removed from the gas phase to the mucus on the airways. However, soluble gases dissolved in
17 particles may be carried into the air exchange region. If equilibrium is reached rapidly, such
18 highly soluble gases will evaporate from particles smaller than $0.1 \mu m$ diameter before the
19 particles reach the air exchange region. However, particles larger than $\sim 0.3 \mu m$ diameter can
20 efficiently carry such gases into the air exchange region.

21 Wexler and Sarangapani (1998) point out that due to the small volume of particle-bound
22 water, even in the case of highly soluble gases, only on the order of 1% of the soluble gas will be
23 found in the particles. However, particles will change the pattern of vapor deposition and
24 particles will carry dissolved gases deeper into the respiratory tract where the particles can
25 deposit on air exchange surfaces not protected by mucus. Furthermore, the Wexler and
26 Sarangapani (1998) analysis was based on considerations of physical solubility only. If adducts
27 or complexes form, such peroxohydrates from H_2O_2 (Friedlander and Yeh, 1998; Elvers et al.,
28 1991), or if the gas reacts chemically with water, as SO_2 does to form $SO_2(aq)$, $H_2SO_3(aq)$,
29 and $HSO_3^-(aq)$ (Schwartz, 1984), the solubility of the gas may be increased greatly and the time
30 to reach equilibrium may be increased. Both factors would enable particles to transfer greater
31 quantities of dissolved gases to the air exchange region.

1 Morio et al. (2001) evaluated whether hygroscopic components of PM may transport H₂O₂
2 into the lower respiratory tract and induce tissue injury. Rats were exposed via inhalation
3 to (NH₄)₂ SO₄ (0.3 to 0.4 μm MMD) at 215 or 249 μg/m³ or H₂O₂ at 10, 20, or 100 ppb alone or
4 in combination for 2 h. No major effect was observed on BAL cell number or viability or on
5 protein content or LDH levels immediately or 24 h post exposure. However, rats treated with the
6 combination of sulfate and peroxide showed increased tumor necrosis factor (TNF - ∞)
7 produced by alveolar macrophages and increased numbers of neutrophils in pulmonary
8 capillaries (as seen via EM). These results and other effects on NO levels were interpreted by
9 the authors as showing that biological effects of inhaled PM are augmented by coexposure to
10 sulfate and peroxide, including altered production of cytokine mediators by AM.

11 Also of note, Hung and Wang (2001) observed high reactive oxygen species (ROS) activity
12 (reflective of hydrogen peroxide levels) in atmospheric aerosols collected roadside in China,
13 with higher ROS activity among fresh fine particles (~0.18 μm) than among ultrafine (< 0.1) or
14 coarse (3.2 to 10 μm) particles. They noted that ambient temperature and water vapor content
15 may affect ROS content of ambient particles.

16 The information summarized above has substantial implications for interpreting and
17 understanding the vast array of epidemiological and toxicologic results discussed in preceding
18 sections of this chapter and earlier chapters of this document. Their full significance becomes
19 more evident when considered in light of dosimetric information discussed in Chapter 6. It is
20 worth restating a few basic points here from Chapter 6 and expanding on them further with
21 regard to the importance of dosimetric considerations in relation to particles as carriers of other
22 toxic agents.

23 First, particle size is one of the most basic parameters governing particle behavior and
24 deposition in the respiratory tract. Particles between 0.3 and 0.7 μm in diameter have minimal
25 deposition in the respiratory tract. Above and below this range of minimum deposition, the
26 efficiency of deposition increases. The pattern of deposition within the respiratory tract also
27 slowly shifts from the alveolar region to the TB and ET regions with increasing particle size over
28 1 to 2 μm and with decreasing particle size below 0.1 μm.

29 Hygroscopicity, the propensity of a material for taking up and retaining moisture, is an
30 important property of some ambient particle species and affects respiratory tract deposition.
31 Such particles can increase in size in humid air in the atmosphere or in the respiratory tract and,

1 when inhaled, deposit according to their hydrated size rather than their initial size. Compared to
2 nonhygroscopic particles of the same initial size, deposition of hygroscopic aerosols in different
3 regions varies, depending on initial size: hygroscopicity generally increases total deposition for
4 particles with initial sizes larger than $\sim 0.5 \mu\text{m}$, but decreases deposition for particles between
5 ~ 0.01 and 0.5 and again increases deposition for particles $< 0.01 \mu\text{m}$. Thus, under high humidity
6 conditions, there is increased deposition of smaller (nucleation-mode; $< 0.01 \mu\text{m}$) ultrafine
7 particles and of larger accumulation-mode ($\geq 0.5 \mu\text{m}$) particles, the latter of which can grow to
8 sizes exceeding $1.0 \mu\text{m}$ and both of which would contain enhanced amounts of particle bound
9 water and other toxic agents (e.g., O_3 , SO_2 , peroxide, formaldehyde) dissolved therein.

10 Enhanced particle retention occurs on carinal ridges in the trachea and segmental bronchi;
11 and deposition “hot spots” occur at airway bifurcations or branching points. Peak deposition
12 sites shift from distal to proximal sites as a function of particle size, with greater surface dose in
13 conducting airways than in the A region for all particle sizes. To some extent then, the growth
14 of ultrafine and accumulation mode particles under humid conditions would also likely increase
15 “hot spot” deposition at airway branching points and thereby increase PM doses to tissues at
16 those points.

17 Ventilation rate, gender, age, and respiratory disease status all affect total and regional
18 respiratory tract particle deposition. Of likely most concern from among all these factors
19 affecting respiratory particle deposit patterns are altered PM deposition patterns due to
20 respiratory disease status that may put certain groups of adults (including some elderly) and
21 children at greater risk for PM effects. Importantly, COPD contributes to more heterogenous
22 deposition patterns and differences in regional deposition. One study indicates that people with
23 COPD tend to breathe faster and deeper than those with normal lungs (i.e., about 50% higher
24 resting ventilation) and have $\sim 50\%$ greater deposition than age-matched healthy adults under
25 typical breathing conditions, with average deposition rates 2.5 times higher under elevated
26 ventilation rates. Enhanced deposition appears to be associated more with the chronic bronchitic
27 than the emphysematous component of COPD. In this and other new studies, fine-particle
28 deposition increased markedly with increased degree of airway obstruction. With increasing
29 airway obstruction and uneven airflow because of irregular obstruction patterns, particles tend to
30 penetrate more into remaining better ventilated lung areas, leading to enhanced focal deposition

1 at airway bifurcations and alveoli in those A region areas. In contrast, TB deposition increases
2 with increasingly more severe bronchoconstrictive states, as occur with asthmatic conditions.

3 Disease states can also alter clearance rates for removal of deposited particles from the
4 lung. Bronchial mucus transport is slowed by asthma, chronic bronchitis, bronchial carcinoma,
5 and various acute respiratory infections - all being disease conditions expected to increase
6 retention of deposited particle material and, thereby, increase the probability of toxic effects
7 from inhaled ambient PM components reaching the TB region. Also, spontaneous coughing, an
8 important TB region clearance mechanism, does not appear to fully compensate for impaired
9 mucociliary clearance in small airways and may become depressed with worsening airway
10 disease, as seen in COPD patients. Clearance of particles from the A region by alveolar
11 macrophages and their mucociliary transport is usually rapid (< 24 h), but alveolar region
12 clearance rates are decreased in human COPD sufferers and slowed by acute respiratory
13 infections; and the viability and functioning of alveolar macrophages are reduced in human
14 asthmatics and in animals with viral lung infections.

15 All this suggests that persons with asthma, chronic bronchitis, or acute lung infections are
16 likely to experience increased deposition and retention of inhaled particles and to be at risk for
17 ambient PM exposure effects. Such individuals can reasonably be expected to be put at even
18 greater risk when inhaling ambient PM under high humidity conditions (with increased delivery
19 of peroxides, O₃, SO₂, and other noxious agents into the deep lung in particle-bound water and
20 enhanced “hot spot” deposition of hygroscopic aerosols at branching points in bronchial
21 airways).

22 23 ***Bioaerosols as Contributors to Ambient PM Effects***

24 Bioaerosols, from sources such as plants, fungi, and microorganisms, range in size from
25 0.01 µm to > 20 µm. They comprise a small fraction of ambient PM, but likely contribute to the
26 some types of ambient PM-related health effects exposure.

27 Intact pollen grains from flowering plants, trees and grasses are by far most abundant in
28 warm/humid spring and summer months and can deposit in upper airways to induce allergic
29 rhinitis. Allergen-laden cytoplasmic fragments (~0.1 to 0.4 µm in size) of pollen grains (which
30 rupture under high moisture conditions) can enter the deep lung, where they can exacerbate
31 asthma. Binding of allergen-laden pollen cytoplasmic fragments to ambient fine particles (e.g.,

1 DPM) has also been observed; and synergistic interactions between pollen debris and other
2 ambient PM (e.g., the polycyclic hydrocarbon component of DE) are thought to be a mechanism
3 that may explain the increased incidence of asthma morbidity and mortality. Pollen granules can
4 also act as vectors for binding of other bioaerosols (e.g., endotoxins, fungi or fragments, glucans)
5 and thereby enhance their inhalation and deposition in the respiratory tract.

6 Fungal spores and fungi fragments are among the largest and most consistently present
7 bioaerosols found outdoors (levels being higher during warm/humid months). They cause
8 allergic rhinitis and asthma, which is highly dependent on seasonal variations in concentration.
9 Exposures have been linked to asthma hospitalization and death.

10 Bacteria and viruses are significant bioaerosols. Much of the toxicity of bacteria is due to
11 the endotoxins present in the outer cell membrane, which trigger production of cytokines and a
12 cascade of inflammation. Ambient airborne concentrations of endotoxins vary with seasons
13 (being higher in warm/humid periods and low in colder months) and tend to be higher in samples
14 of coarse-mode than in fine-mode ambient PM. Another cell wall component of bacteria and
15 fungi, (1→3)-β-D-glucan, has also been shown to cause respiratory inflammation.

16 Animals and insects produce bioaerosols capable of producing hypersensitivity diseases.
17 Most notably, exposure to dust mite and cockroach material has been linked to sensitization in
18 children. However, indoor exposures to such materials probably are of most importance with
19 regard to human exposures to such materials.

20 It thus appears that certain ambient bioaerosols (e.g., pollen, fungi, endotoxins, glucans)
21 that become abundant during warm/humid weather may contribute to seasonal increases in PM-
22 associated risk during spring/summer months, but not during colder winter months. The
23 copresence of non-biological particles, serving as vectors concentrating such bioaerosols and
24 enhancing their delivery into the deep lung, appears to likely be important.

25 26 ***Summary and Conclusions***

27 It has been proposed that particles also may act as carriers to transport toxic gases into the
28 deep lung. Water-soluble gases, which would be removed by deposition to wet surfaces in the
29 upper respiratory system during inhalation, could dissolve in particle-bound water and be carried
30 with the particles into the deep lung. Equilibrium calculations indicate that particles do not
31 increase vapor deposition in human airways. However, these calculations do show that soluble

1 gases are carried to higher generation airways (i.e., deeper into the lung) in the presence of
2 particles than in the absence of particles. In addition, species such as SO₂ and formaldehyde
3 react in water, reducing the concentration of the dissolved gas-phase species and providing a
4 kinetic resistance to evaporation of the dissolved gas. Thus, the concentration of the dissolved
5 species may be greater than that predicted by the equilibrium calculations. Of much concern,
6 particle-bound water appears to be a means by which dissolved hydrogen peroxide and other
7 short-lived reactive oxygen species can be carried into lower respiratory tract regions and
8 contribute to the induction of inflammatory responses. Also, certain other toxic species (e.g.,
9 nitric oxide [NO], nitrogen dioxide [NO₂], benzene, polycyclic aromatic hydrocarbons [PAH],
10 nitro-PAH, a variety of allergens) may be absorbed onto solid particles and carried into the
11 lungs. Thus, ambient particles may play important roles not only in inducing direct health
12 impacts of their constituent components but also in facilitating delivery of toxic gaseous
13 pollutants or bioagents into the lung and may, thereby, serve as key mediators of health effects
14 caused by the overall air pollutant mix.

15 16 **9.2.3.2.8 Coherence of evidence**

17 One of the key factors for evaluating the associations between exposure and outcome
18 variables derived from epidemiologic studies is the coherence in the evidence. As described in
19 Section 13.4.2.5 of the 1996 PM AQCD, an assessment of the coherence across a body of
20 evidence considers the logical and systematic relationships among various health outcomes that
21 may be related to exposure. In assessing coherence, one should compare outcomes with similar
22 time frames, for example, looking across various respiratory-related health outcomes linked with
23 short-term (e.g., daily) ambient PM concentrations. An assessment of coherence is primarily
24 qualitative in nature, not quantitative, since it involves consideration of evidence from across
25 disciplines and varying study methodologies. For example, Bates (1992) suggested evaluating
26 coherence not only within epidemiologic data, but also between epidemiologic and animal
27 toxicologic data and among epidemiologic, controlled human exposure and animal data.

28 Looking first within the epidemiologic literature, considerable coherence can be seen to
29 exist across the now extensive body of available epidemiologic study findings. In the 1996 PM
30 AQCD, consideration was given to the coherence of evidence of various effects within the same
31 geographic area. In particular, epidemiologic evidence from studies conducted in four U.S.

1 locations using varying indicators such as PM₁₀ and TSP – Detroit, Birmingham, Philadelphia
2 and Utah Valley – generally showed coherence across cardiovascular and respiratory health
3 outcomes within each area. The health outcomes included mortality from cardiovascular or
4 respiratory diseases, hospital admissions for respiratory causes and for cardiovascular causes in
5 the elderly, and respiratory symptoms.

6 The expanded body of epidemiological evidence available in this review provides further
7 support for those findings. Effect estimates for associations between short-term exposure
8 to PM₁₀ and various effects ranging from mortality to respiratory symptoms or cardiovascular
9 health indicators are available from multiple studies in a number of urban areas, including
10 Chicago, Los Angeles, Detroit, Seattle and Pittsburgh. As shown in Section 8.4.4, results for
11 associations between PM₁₀ and various health outcomes are summarized in a series of figures,
12 using single-pollutant model results from the available studies in each location. While in
13 Detroit, Los Angeles, Seattle and Pittsburgh, some studies also reported associations with PM_{2.5}
14 and PM_{10-2.5} (included in the presentation of results for mortality and hospitalization and medical
15 visits studies in Figure 9-5 and 9-6), the more numerous results for PM₁₀ better allow for an
16 assessment of coherence within these areas.

17 These results for PM₁₀ include an array of health outcome measures, summarized below,
18 that expand upon the findings in the 1996 PM AQCD with much more extensive evidence on
19 cardiovascular and respiratory morbidity outcomes. As discussed in Section 9.2.2, coherence
20 can be observed in considering the pattern of findings across the studies within each area,
21 especially focusing on those study results with greater precision, in that almost all studies report
22 positive associations (at least for some of the lag periods examined in those studies that reported
23 results for multiple lag periods), many of which are statistically significant.

- 24 • Chicago (Figure 8-24) - total, cardiovascular and respiratory mortality, hospital
admissions for respiratory and cardiovascular diseases; especially for lag periods of day 0
and/or day 1.
- 25 • Los Angeles (Figure 8-25) - total, cardiovascular and respiratory mortality, hospital
admissions for respiratory and cardiovascular diseases, hospital admissions for asthma,
COPD, myocardial infarction, congestive heart failure, cardiac arrhythmia,
cerebrovascular and occlusive stroke, and respiratory symptoms in asthmatic children.
Positive, statistically significant results are more generally reported for lag periods of day
0 and/or day1, although less consistency is observed across different lag periods in some
studies, perhaps due to inherent limitations in the 1-in-6-day ambient PM data used in
some of these studies.

- 1 • Pittsburgh (Figure 8-26) - total mortality, hospital admissions for cardiovascular diseases, COPD and pneumonia.
- 2 • Detroit (Figure 8-27) - total, cardiovascular and respiratory mortality, and hospital admissions for pneumonia, COPD, ischemic heart disease, dysrhythmia, heart failure and stroke.
- 3 • Seattle (Figure 8-28) - total mortality, hospital admissions for cardiovascular diseases, asthma, COPD and pneumonia, asthma symptoms; a notable exception is the negative risk estimate reported for sudden cardiac arrest, although the result is highly imprecise.

4 In addition to the evidence of associations with measures such as mortality or
5 hospitalization, new epidemiological studies have reported associations between PM,
6 primarily PM₁₀ or PM_{2.5}, and health outcome measures related to cardiovascular and respiratory
7 disease such as physician visits for respiratory diseases, incidence of myocardial infarctions, and
8 physiological or biochemical indicators of cardiovascular health. Epidemiologic panel studies
9 have reported changes in blood characteristics (e.g., increased fibrinogen or C-reactive protein
10 levels) related to increased risk of ischemic heart disease as also being associated with ambient
11 PM exposures. New studies have also reported associations between PM and changes related to
12 heart rhythm, including cardiac arrhythmia or changes in heart rate variability that may be linked
13 with more serious cardiac effects. In addition, new evidence exists for ambient PM associations
14 with reductions in pulmonary function and/or increased respiratory symptoms, especially of note
15 in relation to asthmatic or other chronic lung disease individuals. In considering the evidence for
16 different PM size fractions or components, it can be observed that most of these studies have
17 data on fine particles or fine particle constituents. All these cardiovascular and respiratory
18 morbidity effects add to the coherence of the overall evidence substantiating effects of short-
19 term exposure to PM, especially fine PM, in susceptible population groups.

20 Beyond epidemiologic studies, there are now many more studies from other disciplines to
21 use in considering the results of epidemiologic, toxicologic and controlled human exposure
22 studies together than were available for the 1996 PM AQCD. For example, epidemiologic
23 studies conducted in Boston have linked fine particle concentrations with increased risk of
24 myocardial infarction, cardiac arrhythmia and changes in heart rate variability (discussed in
25 Section 8.3.1.3.4) and toxicologic studies using Boston CAPs (fine particles) have also shown
26 some evidence for changes in blood parameters or heart rhythm (summarized in Table 7-1).
27 Section 9.2.3.2 summarizes the results of new toxicologic or controlled human exposure studies

1 that have shed light on potential mechanisms underlying cardiovascular and respiratory effects
2 observed in epidemiologic studies, and thus provide support for the coherence of PM-related
3 effects.

4 Compelling evidence of coherence is offered by a body of epidemiologic, toxicologic and
5 controlled human exposure studies on effects of particles from the Utah Valley area. As
6 discussed above, a series of epidemiologic studies from Utah Valley reported associations
7 between PM and health outcomes ranging from increased respiratory symptoms to mortality. A
8 special feature of these studies was the closure of a steel mill, a major source of PM emissions in
9 the area, for a 13-month period. As discussed in Section 8.2.3.4, respiratory hospital admissions
10 for children were reduced during the period the source was not operating, and assessment of
11 mortality risk also indicated that mortality rates were 3.2% greater when the steel mill was
12 operating. New toxicologic and human studies have used extracts of ambient particles collected
13 on filters from ambient monitors operating during the time periods before, during and after steel
14 mill closure. Intratracheal instillation of particle extracts in both human volunteers and animals
15 resulted in greater lung inflammatory responses for materials obtained before and after the plant
16 closure period (further discussed in section 7.3.1.2). The health responses were indicative of
17 inflammatory changes in the lung, including increased levels of neutrophils, protein and
18 inflammatory cytokines. As discussed previously, consideration of dosimetry information
19 indicates that the doses of particles used in these experiments are equivalent to higher-level
20 exposure concentrations that the community could experience during typical winter inversions in
21 the Utah Valley. *In vitro* studies using human airway epithelial cell lines also showed evidence
22 for inflammatory responses, such as increases in cytokine levels, indicators of oxidative response
23 in alveolar macrophages and some evidence of cytotoxicity (see Section 7.4.2). This body of
24 evidence links results of community epidemiologic studies reporting increases in respiratory
25 hospitalization with toxicologic studies showing evidence of respiratory inflammation in humans
26 and animals.

27 Further evidence was obtained from the toxicologic studies to suggest that metals were an
28 important particle component for the inflammatory changes described above. The Utah Valley
29 studies used particles collected from the PM₁₀ monitoring network, but further analysis was done
30 on the extracted particle material. Notably larger proportions of metals (e.g., Cu, Zn, Fe, Pb, As,
31 Mn, Ni), as well as sulfate and cationic salts (e.g., calcium, potassium, magnesium), were found

1 in the particles collected while the source was operating. The addition of a chelating agent to the
2 particle extract was found to attenuate responses found in *in vitro* studies, providing further
3 evidence that metals are an important particle component for this group of health responses.

4 More limited evidence is also available on the effects of long-term exposure to particles
5 from all health disciplines. The epidemiologic studies indicate associations with mortality from
6 cardiopulmonary diseases and lung cancer, and with potential development of chronic
7 respiratory diseases or reduction in lung function. For lung cancer, there is substantial evidence
8 for coherence in the results of recent epidemiologic studies and toxicologic studies on
9 mutagenicity or genotoxicity, as described in Section 9.2.3.2.2. There have not been many
10 toxicologic studies using chronic particle exposures to evaluate responses relevant to respiratory
11 or cardiovascular health outcomes. In addition, epidemiologic studies have not as yet addressed
12 potential links between long-term PM exposure and indicators for the development of
13 cardiovascular disease that would provide coherence with findings of cardiovascular mortality
14 risk. Thus, the evidence with regard to coherence of effects related to long-term particle
15 exposures is somewhat limited.

16 The overall body of controlled human and/or laboratory animal exposure studies discussed
17 earlier also add coherence to the evidence for ambient PM-related health impacts. A number of
18 studies provide evidence that supports one or another hypothesis with regard to (a) PM
19 components (by size, chemical composition, source) and/or (b) mechanisms likely contributing
20 to PM effects on various cardiovascular or respiratory endpoints. For example, the results of
21 instillation studies, using filter extracts from community monitoring stations in the Utah Valley
22 before, during, and after temporary shut down of a steel mill there are particularly compelling on
23 two accounts: (1) the evidence of greater lung inflammation from instilled extracts from periods
24 of mill operation parallel epidemiologic findings of increased cardiorespiratory hospitalizations
25 during such periods; and (2) dosimetric calculations indicate that concentrations of particulate
26 extract materials likely delivered to affected lung tissue with the instillation would probably be
27 reasonably comparable to those likely experienced in connection with ambient inhalation
28 exposures over several weeks to PM₁₀ concentrations in the Utah Valley PM mixture.

29 Overall, new evidence from epidemiologic, toxicologic, and controlled human exposure
30 studies has built a strong foundation for coherence for fine particle related effects. The evidence
31 for coherence of effects related to PM_{10-2.5}, however, is far more limited, with some evidence for

1 related cardiopulmonary effects from epidemiologic studies but little supporting mechanistic
2 evidence from toxicologic studies.

3 4 **9.2.3.2.9 Summary and conclusions**

5 Consideration of the plausibility and coherence of PM-related effects involves looking
6 across evidence from dosimetric, toxicologic, and epidemiologic studies. In comparison with the
7 1996 PM AQCD, there is much more such evidence available in recent studies on fine particles
8 or fine particle constituents.

9 Toxicological studies, largely studies of fine particles, contribute support for biological
10 plausibility for the effects on the cardiovascular and respiratory systems observed in
11 epidemiologic studies. While often high exposures/doses are used in toxicologic studies, the
12 tissue doses achieved are often not necessarily that far removed from doses derived from
13 exposures of humans at higher ambient levels, as indicated by the quantitative assessment of
14 doses used in toxicologic and controlled human exposure studies, described in Appendix A to
15 Chapter 7, along with the assessment of dose-response functions in toxicologic studies. The
16 recent studies have linked components of fine particles with various health outcomes. There is
17 probably no single primary causative attribute of fine particles, but rather many attributes may
18 contribute to complex mechanisms for the different health outcomes. Overall, the toxicological
19 evidence provides considerable evidence for biological plausibility for effects on the respiratory
20 and cardiovascular systems, including new evidence for lung cancer. There is as yet little
21 toxicological evidence available on coarse fraction particles.

22 Within the body of epidemiological evidence, there is good evidence of coherence across
23 respiratory and cardiovascular health outcomes, especially for effects of short-term exposures.
24 New toxicologic and controlled human exposure studies offer new insights into coherence for
25 effects on the cardiovascular and respiratory systems; compelling new evidence is available, for
26 example, from toxicological and controlled human exposure studies conducted in Utah Valley
27 using particles collected in the same time period as the published epidemiological studies. The
28 results of new studies build a strong foundation of coherence for fine particle effects; however,
29 the evidence available on coherence of coarse fraction particles is far more limited.

30 There is also important new information highlighting potentially crucial roles that particle
31 bound water plays in serving as a carrier or vector by which other toxic agents (e.g., O₃, SO₂,

1 peroxides, formaldehyde) can be accumulate within inhalable PM and delivered in enhanced
2 quantities into the deep lung. The increased availability of certain bioaerosol materials (e.g.,
3 pollen fragments) in small (0.1 - 0.4 micrometers) fine particle sizes that deposit in TB and
4 A regions of the lung (where they can exacerbate asthma effects) is also now recognized.
5

6 **9.2.4 How Does Newly Available Information Inform Our Understanding of** 7 **Subpopulations Potentially Susceptible to PM-Related Health Effects?**

8 **9.2.4.1 Key Points from 1996 Integrative Synthesis**

9 The 1996 PM AQCD included only a relatively limited discussion of susceptible
10 population groups potentially at increased risk for ambient PM effects, noting:
11

12 “There is considerable agreement among different studies that the elderly are particularly
13 susceptible to effects from both short-term and long-term exposures to PM, especially if
14 they have underlying respiratory or cardiac disease. . . Children, especially those with
15 respiratory diseases, may also be susceptible to pulmonary function decrements
16 associated with exposure to PM or acid aerosols.” (U.S. EPA, 1996, p. 13-92)
17

18 The term susceptibility generally encompasses innate or acquired factors that make
19 individuals more likely to experience effects with exposure to pollutants. Genetic or
20 developmental factors can lead to innate susceptibility, while acquired susceptibility may result
21 from age, from disease, personal risk factors such as smoking or exercise, or socioeconomic
22 factors such as reduced access to health care. Other factors can also increase an individual’s
23 vulnerability to adverse effects related to pollution exposure, such as having increased pollutant
24 exposure due to characteristics of the home or due to residence near a specific pollution source.

25 The 1996 PM AQCD identified several population groups potentially as being at increased
26 risk for experiencing health impacts of ambient PM exposure. Elderly individuals (> 65 years)
27 were most clearly identified, along with people having preexisting cardiovascular or respiratory
28 disease conditions. Individuals with asthma, especially children, also were identified as a
29 potential susceptible population group.

30 New studies appearing since the 1996 PM AQCD provide additional evidence that
31 substantiates the above named groups as likely being at increased risk for ambient PM-related
32 morbidity or mortality effects. The newly available studies continue to indicate that the elderly
33 and children are likely more susceptible to PM-related effects. There are also numerous new

1 studies which substantiate the finding that preexisting disease conditions represent an important
2 risk factor for ambient PM health effects. Cardiovascular and respiratory diseases continue to
3 appear to be of greatest concern in relation to increasing the risk for PM mortality and morbidity
4 effects. Indeed, the fact that these disease “entities” often involve both organ systems, albeit to
5 varying degrees, might argue for their compilation under a broader combined classification of
6 “cardiopulmonary” disease.

7 8 **9.2.4.2 Integration of Newly Available Information**

9 ***9.2.4.2.1 Preexisting disease as a risk factor for particulate matter health effects***

10 A number of epidemiologic studies have reported increased risk in study subsets of
11 individuals with preexisting heart or lung diseases. For example, Sunyer et al. (2000) reported
12 large relative risk estimates for total mortality in people with preexisting COPD. Also, Goldberg
13 et al. (2000) originally reported larger effect sizes for total mortality in persons with cancer,
14 diabetes, lower respiratory disease, cardiovascular disease, coronary artery disease, and
15 congestive heart failure; however, upon reanalysis (Goldberg and Burnett, 2003) the pattern of
16 results remained the same, but all lost statistical significance for new analyses using more
17 stringent GAM or GLM. Both Linn et al. (2000) and Zanobetti and Schwartz (2001, 2002, not
18 reanalyzed) reported increased risk of hospitalization for cardiovascular diseases in subgroups
19 with diabetes. In addition, Boezen et al. (1998, Europe) reported significant effects in the subset
20 of adults who had bronchial hyperreactivity or increased peak flow variability; and Vedal et al.
21 (1998) reported greater effects in a subset of children who had asthma.

22 Toxicologists have used several animal models of cardiopulmonary disease to evaluate PM
23 susceptibility aspects. Such animal models include rats with monocrotaline-induced pulmonary
24 vasculitis/hypertension, SO₂-induced chronic bronchitis, spontaneously hypertensive rats, and
25 animals infected with various viral or bacterial agents. As summarized in Section 7.5.1,
26 increased magnitude or frequency of effects have been reported with PM exposure for these
27 groups of animals relative to healthy animals. In addition, toxicologists have also studied effects
28 of particles, including diesel exhaust particles, in animals with heightened allergic sensitivity and
29 via *in vitro* studies (summarized in Section 7.5.2). Overall, the results from newly available
30 toxicological studies provide evidence suggestive of enhanced susceptibility to inhaled PM in
31 “compromised” hosts.

1 The underlying biology of lung diseases might also lead to heightened sensitivity to PM,
2 but this attribute of disease remains hypothetical in the context of PM. The functional linkages
3 with the cardiac system for maintenance of adequate gas exchange and fluid balance
4 notwithstanding, the role of inflammation in the diseased respiratory tract (airways and alveoli)
5 could play a key role. Studies in animals genetically or exogenously altered to induce
6 inflammation are sometimes intrinsically more responsive to concentrated ambient PM, to
7 specific combustion-source-generated PM, or to other laboratory-generated particles. While a
8 PM-induced response may on the one hand be cumulative with the underlying injury or
9 condition, the responses may, on the other hand, be magnified by any number of mechanisms
10 that are poorly understood. There is sufficient basic biological data to hypothesize that the
11 exudated fluids in the airspaces may either interact differently with deposited PM (e.g., to
12 generate oxidants - Costa and Dreher, 1999; Ghio et al., 2001), to augment injury, or to
13 predispose the lung (e.g., sensitize receptors - Undem and Carr, 2002) to enhance the response to
14 a stereotypic PM stimulus through otherwise normal pathways. Less appreciated is the loss of
15 reserve (functional or biochemical), wherein the susceptible individual may be incapable of
16 sufficient compensation (e.g., antioxidant responses - Kodavanti et al., 2000b). Any of these or
17 related mechanisms may contribute to increased “susceptibility” and may indeed be a common
18 factor possibly attributable to other susceptible groups. Understanding these and other
19 mechanisms will ultimately aid in better assessing any increased risk of susceptible groups to
20 PM.

21 Studies with humans that might reveal more specific data have been limited both ethically,
22 as well as by the absence of or limitations associated with biomarkers of response (such as
23 interpretation of ECG indicators of cardiac function and disease). Measures of blood-gas
24 saturation and lung function appear not to be sufficiently revealing or sensitive to mild
25 physiologic changes in those with moderate disease conditions who might be amenable to
26 participation in laboratory studies. In the field, assessing the degree of underlying disease and
27 how that relates to responsiveness of these biomarkers is unclear. However, subjects with COPD
28 and asthma have been studied under controlled conditions with inert aerosols for the purpose of
29 assessing distribution of PM within the lung, and it is now quite clear that airways disease leads
30 to very heterogeneous distribution of PM deposited within the lung. Studies have shown up to
31 10-fold higher than normal deposition at airway bifurcations, thus creating “hot-spots” that may

1 well have biologic implications, especially if the individual already has diminished function or
2 other debilitations due to the underlying disease, even cardiovascular disease (CVD). Thus, the
3 dosimetry of PM within the lung must be considered an important element of the susceptibility
4 paradigm with most any cardiopulmonary disease condition.
5

6 ***9.2.4.2.2 Age-related at-risk population groups: the elderly and children***

7 The very young and the very old apparently constitute two other groups thought to be
8 especially at risk for ambient PM air pollution health effects. Numerous epidemiological studies
9 have reported health responses to PM and other pollutants for one or another specific age group.

10 These studies, as summarized in Section 8.4.9 of Chapter 8, tend to support previous
11 findings that, depending on the effect under study, older adults and children may be more
12 susceptible to certain PM-related effects. More specifically, older adults (aged 65+ years)
13 appear to be most clearly at somewhat higher risk for PM exacerbation of cardiovascular-related
14 disease effects and, perhaps, tend to experience higher PM-related total (non-accidental)
15 mortality risk, as well. On the other hand, more limited evidence points to children possibly
16 being at somewhat higher risk for respiratory-related (especially asthma) PM effects than adults.

17 A major factor in increased susceptibility to air pollution is the presence of a preexisting
18 illness and susceptibility related to age group may well be closely linked with the potential for
19 preexisting cardiopulmonary diseases. Cardiopulmonary diseases more common to the elderly
20 play into the risk within older age groups, but some panel studies of morbidity focusing on
21 generally healthy people in retirement homes or elderly volunteers exposed to concentrated
22 ambient PM in chambers show subtle alterations of autonomic control of cardiac function (e.g.,
23 slight depression of heart rate variability) and blood factors concordant with a putative response
24 to ambient PM levels. However, given the overall patterns of results observed in these (and in
25 controlled exposure) studies, there currently exists a conflicting array of evidence which makes
26 it difficult to ascribe any clinical significance to the generally small changes observed, even
27 though similar at times to changes indicative of increased risk based on studies of risk in cardiac
28 patients and general population studies of cardiac disease progression. Over the long term,
29 innate differences in metabolism or other mechanisms may impact the likelihood of chronic
30 outcomes, e.g., COPD or lung cancer. To what extent progression occurs with repeated PM

1 exposures and how much disease or other risk factors add to or complicate the magnitude of
2 response remains uncertain.

3 Although infection as a risk factor for PM has already been noted, it is important to
4 emphasize that there are clear age differences in both the incidence and type of infections across
5 age groups. Young children have the highest rates of respiratory illnesses related to infection
6 (notably respiratory syncytial virus), while adults are affected by other infectious agents such as
7 influenza that may also lend increased susceptibility to PM effects. Data to address fully the
8 importance of these differences is incomplete, but some of the newly available toxicological
9 studies (e.g., Zelikoff et al., 2003) provide evidence for ambient PM exposures affecting lung
10 defense mechanisms so as to exacerbate preexisting respiratory infections.

11 In addition to their higher incidences of preexisting respiratory conditions, several other
12 factors may render children and infants more susceptible to PM exposures, including more time
13 spent outdoors, greater activity levels and ventilation, higher doses per body weight and lung
14 surface area, and the potential for irreversible effects on the developing lung. For example, PM
15 doses on a per kilogram body weight basis are much higher for children than for adults, as is
16 displayed graphically in Figure 9-7. The amount of air inhaled per kilogram body weight
17 decreases dramatically with increasing age, due in part to ventilation differences (in cubic meters
18 per kilogram a day) of a 10-year-old being roughly twice that of a 30-year-old person, even
19 without the consideration of activity level. Child-adult dosage disparities are even greater when
20 viewed on a per lung surface-area basis.

21 As to potential lung developmental impacts of PM, there exist both experimental and
22 epidemiologic data, which although limited, suggest that the early post-neonatal period of lung
23 development is a time of high susceptibility for lung damage by environmental toxicants. In
24 experimental animals, for example, elevated neonatal susceptibility to lung-targeted toxicants
25 has been reported at doses “well below the no-effects level for adults” (Plopper and Fanucchi,
26 2000); and acute injury to the lung during early postnatal development may impair normal repair
27 processes, such as down-regulation of cellular proliferation (Smiley-Jewel et al., 2000, Fanucchi
28 et al., 2000). These results in animals appear to be concordant with recent findings for young
29 children growing up in the Los Angeles area, where both oxidants and high PM prevail
30 (Gauderman et al., 2000).

31

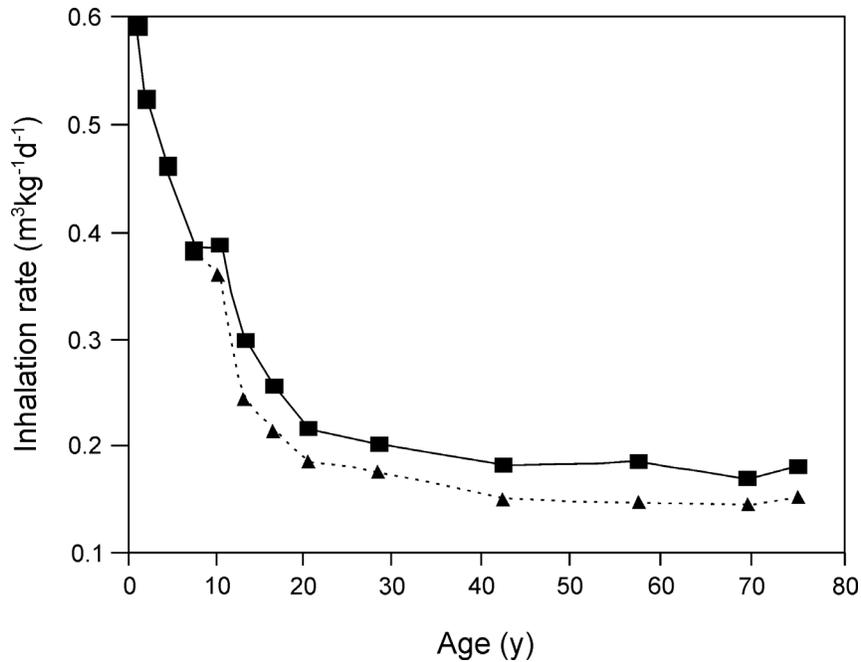


Figure 9-7. Inhalation rates on a per body-weight basis for males (■) and females (▲) by age (Layton, 1993).

1 **9.2.4.2.3 Genetic susceptibility**

2 A key issue in understanding adverse health effects of inhaled ambient PM is the
 3 identification of which classes of individuals are susceptible to PM. Although factors such as
 4 age and health status have been studied in both epidemiology and toxicology studies, some
 5 investigators have begun to examine the importance of genetic susceptibility in the response to
 6 inhaled particles because of evidence that genetic factors play a role in the response to inhaled
 7 pollutant gases. To accomplish this goal, toxicologists typically have sought to detect
 8 inter-strain differences in responses to particles in rodents; little evidence is available from
 9 epidemiological studies at this time. The small group of newly-available toxicological studies
 10 have begun to demonstrate that genetic susceptibility can play a role in the response to inhaled
 11 particles (Section 7.5.2); for example, Kodavanti et al. (1996, 1997a) found a genetic based
 12 difference in susceptibility to lung injury induced by instilled ROFA, using several strains of rats
 13 with varying genetic characteristics.

14

1 **9.2.4.2.4 Gender**

2 There are significant gender differences in the homogeneity of deposition as well as the
3 deposition rate of particles. These differences derive from differences between males and
4 females in body size, conductive airway size, and ventilatory parameters. Females have a
5 somewhat greater deposition of coarse mode particles in the ET and TB regions, but lower
6 deposition in the A region. This gender effect appears to be particle-size dependent, showing a
7 greater fractional deposition in females for very small ultrafine and large coarse thoracic
8 particles. Total fractional lung deposition for 0.04 and 0.06 μm particles also appears to be
9 somewhat greater in females than males but only negligibly so for particles in the size range
10 0.8 to 1.0 μm . As the particle size increases (3 to 5 μm), total fractional deposition increases in
11 females. While deposition appears to be more localized in females than males, deposition rate
12 appears to be greater in males.

13 Little evidence is available from toxicology studies regarding gender differences in
14 susceptibility to pollution effects. In the epidemiology studies that have included stratified
15 analyses based on gender, there is no clear pattern of increased vulnerability for either males or
16 females. A number of studies using long-term and short-term PM exposures report no clear
17 pattern of differences in effects across genders (e.g., Linn et al., 2000; Ostro et al., 2001;
18 Dockery et al., 1996; Raizenne et al., 1996; Krewski et al., 2000). Where differences in effects
19 between males and females were reported, they were generally not significantly different, and
20 the findings are not consistent. For example, from PM_{10} -mortality studies conducted in Chicago,
21 Styer et al. (1995) report larger effect estimates for men, but Ito and Thurston (1996) report
22 larger effect estimates for women.

23 Thus, insufficient evidence exists overall to allow for any clear conclusions to be drawn as
24 to potential gender differences with regard to PM health effects. More systematic research on
25 the subject is needed.

26 **9.2.4.2.5 Socioeconomic status**

27 Epidemiological studies of long-term PM exposures have suggested that there is effect
28 modification of PM-mortality associations due to socioeconomic factors. In the ACS and Six
29 Cities cohort analyses on mortality risk with long-term exposure to $\text{PM}_{2.5}$, there was clear
30 evidence of effect modification (though not confounding) by education level, with greater effects
31

1 being reported in the cohort subgroups with lower education levels (Krewski et al., 2000; Pope
2 et al., 2002).

3 Among the studies of short-term PM exposure, the evidence is more mixed regarding
4 potential influence of socioeconomic status on PM-related health risks. Schwartz (2000a) found
5 no evidence of effect modification for PM₁₀-mortality associations in 10 U.S. cities using four
6 measures of social or economic status: greater percent of population living in poverty status;
7 higher unemployment rate; greater percent of population with college degrees; or greater percent
8 of the population being nonwhite. Zanobetti and Schwartz (2003; reanalyzed Zanobetti et al.,
9 2000) conducted similar analyses with data for hospital admissions in 10 U.S. cities, and none of
10 the four measures of social or economic status mentioned above significantly modified the
11 relationship between PM₁₀ and hospitalization for COPD or pneumonia. However, for CVD
12 admissions, PM₁₀ effect estimates were greater in communities with greater percentages of the
13 population being unemployed, nonwhite, or living in poverty. The authors postulate that this
14 effect would be a result of increased exposure, increased prevalence of predisposing diseases or
15 other factors. Tolbert et al. (2000b) found race (black vs. white) and insurance Medicaid vs.
16 non-Medicaid) to be effect modifiers for emergency department admissions for asthma in
17 children (< 17 years) in Atlanta, but no associations with interaction terms for these factors and
18 PM₁₀ or ozone. Also, Norris et al. (1999) reported no effect estimate differences for asthma
19 hospitalization in children (< 18 years) when comparing the inner city area with the rest of
20 Seattle.

21 However, some studies have reported evidence for socioeconomic factors increasing risk
22 of hospitalization or emergency department visits with PM and other pollutants. Gwynn and
23 Thurston (2001, not reanalyzed) reported generally greater effect estimates for respiratory
24 hospitalization for nonwhite persons, as compared to the white persons, and for the subgroup
25 with no health insurance, compared with those who had insurance or Medicaid coverage;
26 differences in effect estimates were more notable for ozone than PM₁₀. The authors suggest that
27 a large portion of the apparent difference in pollutant risk estimates between racial subgroups
28 can be explained by socioeconomic factors such as insurance status and poverty. Nauenberg and
29 Basu (1998) analyzed associations between hospitalization for asthma with PM₁₀ and ozone in
30 Los Angeles for subsets of patients who were uninsured, insured by MediCal, or had other
31 insurance. Significant associations with PM₁₀ were reported only for the subset of patients using

1 MediCal, not for the privately insured or uninsured; the authors speculate that the small sample
2 size for uninsured patients may have precluded detection of an effect.

3 In summary, evidence from cohort studies of long-term PM exposure effects indicates that
4 PM-mortality risk may be greater for those with lower socioeconomic status. The time-series
5 epidemiologic studies provide less evidence of effect modification for short-term exposure
6 effects by socioeconomic status, though there is some limited evidence suggesting possible
7 greater effects on respiratory hospitalization with lower socioeconomic status.

8 9 ***9.2.4.2.6 Enhanced vulnerability due to heightened exposure levels***

10 Exercise may increase the potential health risks of inhaled particles because exercise
11 increases the rate of oxygen consumption and changes ventilatory parameters affecting airflow
12 rate and breathing patterns. The switch from nose breathing to mouth breathing, which occurs as
13 exercise intensity increases, leads to an increase in fractional deposition of ultrafine and coarse
14 thoracic particles in the tracheobronchial and alveolar regions. The higher breathing rate and
15 larger tidal volume lead to a greater amount of deposition. Total lung deposition rate may be
16 3 to 4 times greater during exercise. The more rapid breathing of children also leads to a greater
17 amount of deposition.

18 In several reports from the Southern California children's study, larger effect estimates for
19 reduced lung function or increased respiratory illness with long-term exposure to PM and other
20 pollutants were reported for the subset of children spending a larger amount of time outdoors
21 (Peters et al., 1999a,b, Gauderman et al., 2000, 2002). Also, using data from 14 U.S. cities,
22 Janssen et al. (2000; 2002; Zanobetti and Schwartz, 2003) reported that effect estimates between
23 PM₁₀ and hospitalization for CVD and COPD increased with less air conditioning use in homes
24 (such use being an indicator of decreased exposure due to less penetration of particles into the
25 home). PM₁₀-hospital admission effect estimates were also found to increase with increasing
26 population density, which was strongly correlated with estimates of vehicle miles traveled in
27 these cities and, thus, is likely an indicator for increased exposure to vehicle-related pollution.
28 Increased vulnerability to the effects of pollution may come from living near a source of PM and
29 other pollutants, such as a major roadway. Numerous recent studies have linked adverse health
30 effects with indicators of traffic-related pollution. For example, Hoek et al. (2002) reported

1 statistically significant associations for mortality with long-term ambient PM exposure
2 (measured as black smoke, BS) and also with residence near a major road.

4 **9.2.4.3 Summary and Conclusions**

5 In summary, host variability may come to be one of the most important factors in
6 determining the response profile of any population exposed to PM. Studies to date suggest that
7 certain subpopulations are indeed more acutely responsive to PM, perhaps due to differences in
8 lung deposition (either in terms of dose and/or intrapulmonary distribution) or other biologic
9 aspects of the cardiopulmonary system or disease thereof. The role of innate attributes of risk
10 grounded in one's genetic code is largely unknown, but of potentially great importance. Animal
11 models have been used to show clear differences in response to PM and other pollutants, and the
12 critical involvement of varied genes in the induction of asthma, emphysema, and many other
13 ailments is widely accepted, but poorly understood. Long-term epidemiologic studies indicate
14 an increase in risk associated with various indications of lower socioeconomic status.

16 **9.2.5 What Does the Newly Available Information Imply With Regard to 17 Potential Public Health Impacts of Human Exposures to Ambient 18 PM in the United States?**

19 **9.2.5.1 Key Points from 1996 Integrative Synthesis**

20 The 1996 PM AQCD highlighted the then considerable uncertainty related to estimating
21 public health impact of ambient PM exposure, stating:

22
23 “Efforts to quantify the number of deaths attributable to, and the years of life lost to,
24 ambient PM exposure are currently subject to much uncertainty.” (U.S. EPA, 1996,
25 p. 13-87). Nonetheless, while “PM-related increases in individual health risks are small,”
26 they are “likely significant from an overall public health perspective because of the large
27 numbers of individuals in susceptible risk groups that are exposed to ambient PM.”
28 (U.S. EPA, 1996, p. 1-21)

30 **9.2.5.2 Integration of New Information**

31 **9.2.5.2.1 Magnitude of susceptible groups**

32 As summarized in Section 9.2.4, numerous U.S. population groups may be identified as
33 having increased susceptibility or vulnerability to adverse health effects from PM. Considering
34 together the subpopulations of persons with preexisting cardiopulmonary disease, older adults,

1 children, people of lower socioeconomic status and those with higher potential exposure levels
2 as potentially susceptible or vulnerable, it is clear that the impact of PM on public health could
3 be very extensive.

4 Table 9-5 summarizes information on the prevalence of chronic respiratory and circulatory
5 conditions and diabetes in the U.S. population in 2000. It can be seen that people with
6 preexisting cardiopulmonary disease constitute a fairly large proportion of the population, with
7 tens of millions of people included in each disease category. For circulatory conditions,
8 approximately 22 million people, or 11% of the U.S. adult population, have received a diagnosis
9 of heart disease. Approximately 20% of the U.S. adult population has hypertension, with 6%
10 reporting diagnoses of coronary heart disease. For respiratory conditions, approximately 9% of
11 U.S. adults (and 11% of children) have been diagnosed with asthma, and 6% of adults diagnosed
12 with conditions included in COPD. Table 9-6 provides further information on the number of
13 various specific respiratory conditions per 100 persons by age among the U.S. population during
14 the mid-1990s. In addition, approximately 6% of the U.S. adult population has diabetes. Both
15 cardiovascular conditions and diabetes are more common among older age groups, while asthma
16 prevalence is higher in children.

17 Potentially susceptible subpopulations based on age group or socioeconomic status would
18 also comprise substantial segments of the population. Based on U.S. census data from 2000,
19 about 26% of people in the U.S. are under 18 years of age, and 12% are 65 years of age or older.
20 From among commonly-used indicators of socioeconomic status, about 12% of individuals and
21 9% of families are below the poverty level, and 20% of the U.S. population does not have a high
22 school or higher level of education. Clearly, large proportions of the U.S. population are
23 included in groups that are thought likely to be at increased risk (i.e., susceptible) to the effects
24 of PM. Thus, even a small percentage reduction in PM-associated admissions or deaths from
25 cardiopulmonary disease would predict a large number of avoided cases.

27 ***9.2.5.2.2 Evidence of new endpoints and potentially susceptible groups***

28 The expanded body of epidemiologic studies have identified a number of health outcomes
29 that are linked with exposure to ambient PM, in addition to data that were available in the 1996
30 PM AQCD on cardiopulmonary mortality, hospitalization for respiratory disease, respiratory
31 symptoms and changes in lung function.

TABLE 9-5. PREVALENCE OF SELECTED CARDIORESPIRATORY DISORDERS BY AGE GROUP AND BY GEOGRAPHIC REGION, 2000 (reported as percent or numbers of cases in millions)

Chronic Condition/Disease	Adults (18+)*		Age				Regional			
	Number (× 10 ⁶)	%	18-44	45-64	65-74	75+	NE	MW	S	W
			%	%	%	%	%	%	%	%
Respiratory conditions										
Asthma	18.7	9.3	9.8	8.7	8.7	8.1	8.9	9.3	9	10.3
<i>Asthma (<18 years)*</i>	<i>8.92*</i>	<i>12.4*</i>								
COPD:										
Chronic bronchitis	9.36	4.6	3.6	5.5	6.4	6.6	3.9	4.6	5.4	4.1
Emphysema	3.13	1.6	0.2	1.9	4.7	5.9	1	1.7	2	1.2
Circulatory conditions										
All heart disease	21.99	10.9	4.2	12.5	26.4	35	10.4	11.5	11.5	9.5
Coronary heart disease	11.23	5.6	0.7	6.6	17.3	22.7	5.1	5.3	6.3	5
Hypertension	39.21	19.5	6.4	27.3	46.3	51.5	17.9	18.8	21.6	18.1
Stroke	4.36	2.2	0.3	2.1	6.5	10.5	1.6	2.1	2.6	2.1
Diabetes	11.86	5.9	1.9	8.4	15.9	13.4	5.5	5.6	6.4	5.9

Source: Pleis et al. (2003).

*All data are for adults except asthma prevalence data for children under 18 years of age, responding to “ever told had asthma”; source for data on children is Blackwell et al. (2003).

**TABLE 9-6. NUMBER OF ACUTE RESPIRATORY CONDITIONS PER
100 PERSONS PER YEAR, BY AGE: UNITED STATES, 1996**

Type of Acute Condition	All Ages	Under 5 Years	5-17 Years	18-24 Years	25-44 Years	45 Years and Over		
						Total	45-64 Years	65 Years and Over
Respiratory Conditions	78.9	129.4	101.5	86	76.9	53.3	55.9	49
Common Cold	23.6	48.6	33.8	23.8	18.7	16.1	16.4	15.7
Other Acute Upper Respiratory Infections	11.3	13.1	15	16.1	11.6	7	7.5	6.1
Influenza	36	53.7	44.3	40.5	38.1	23.3	26.1	18.6
Acute Bronchitis	4.6	*7.2	4.3	*3.9	5.1	3.8	3.5	*4.4
Pneumonia	1.8	*3.9	*1.7	*1.4	*1.3	*2.0	*0.9	*3.8
Other Respiratory Conditions	1.7	*2.9	*2.4	*0.4	*2.0	*1.1	*1.5	*0.5

Source: Adams et al. (1999).

*All data are for adults except asthma prevalence data for children under 18 years of age, responding to “ever told had asthma”; source for data on children is Blackwell et al. (2003).

1 New information from prospective cohort studies has suggested that long-term exposure to
2 PM is linked not only with mortality due to cardiovascular diseases, but also with lung cancer
3 mortality. This new data adds to what had been reported in earlier cross-sectional studies and
4 provides strong supporting evidence for the link between ambient pollution and lung cancer
5 mortality.

6 The recent appreciation for underlying cardiovascular dysfunction as a risk factor for PM
7 health effects derives from a growing and diverse body of literature. A number of epidemiologic
8 studies had reported associations with cardiopulmonary mortality, and limited evidence was
9 available on hospitalization for cardiovascular diseases in the 1996 PM AQCD. Numerous new
10 epidemiologic studies have built upon those findings, and new studies summarized in Section
11 8.3.1.3 have reported associations between PM and risk of myocardial infarction, measures of
12 heart beat rhythm, and changes in electrocardiographic (ECG) markers of cardiac function, e.g.,
13 altered heart rate variability (HRV), shown in other studies to be indicators of increased risk for
14 serious cardiovascular outcomes (e.g., heart attacks), though it is noted that interpretation of
15 changes in heart rate variability is complicated. Other studies point toward changes in blood
16 characteristics (e.g., alterations in C-reactive protein levels, fibrinogen levels, blood viscosity,

1 etc.) related to increased risk of ischemic heart disease also being associated with ambient PM
2 exposures. These results provide suggestive evidence indicative of potential pathophysiologic
3 alterations contributing to serious PM-related cardiovascular effects (e.g. myocardial infarction,
4 stroke, death). Collectively, these new epidemiologic and toxicologic studies provide important
5 new insights into potential cardiac responses to PM.

6 As noted above, studies using data on visits to physicians' offices or outpatient clinics
7 provide new evidence on respiratory morbidity. That is, comparing the number of admissions in
8 London from an earlier study (Anderson et al., 1996) with those for GP visits in the 1999 study,
9 Hajat et al. (1999) observed about 24 asthma GP visits for every asthma hospital admission in
10 that city. This suggests that looking only at numbers of hospital admissions and emergency
11 hospital visits may markedly underestimate the overall numbers of respiratory morbidity events
12 due to acute ambient PM exposure.

13 PM-related health effects in infants and children are emerging as an area of more concern
14 than in the 1996 PM AQCD; and ultimately, such health effects could have very substantial
15 implications for life expectancy calculations. However, only very limited evidence currently
16 exists about potential ambient PM relationships with some of the more serious pertinent health
17 endpoints (low birth weight, preterm birth, neonatal and infant mortality, emergency hospital
18 admissions, and mortality in older children). Also, little is yet known about involvement of PM
19 exposure in the progression from less serious childhood conditions, such as asthma and
20 respiratory symptoms, to more serious disease endpoints later in life. This is an important health
21 issue, because childhood illness or death may cost a very large number of productive life-years.

22 Small relative risk estimates for health effects have generally been observed for ambient air
23 pollutants, as would be expected on biological and epidemiologic grounds. In contrast to effect
24 estimates for mortality derived for the 1952 London smog episode, i.e., relative risk (RR)
25 exceeding 4.0 (i.e., 400% increase over baseline) for extremely high (≥ 2 mg/m³) ambient PM
26 levels, effects estimates in most current epidemiology studies at distinctly lower PM
27 concentrations (often ≤ 100 μ g/m³) are relatively small.

28 It is important to recognize that even a small percentage reduction in PM health impacts on
29 respiratory-related diseases would reflect a large number of avoided cases. As described earlier,
30 the potentially susceptible population subgroups can include a large portion of the U.S.
31 population. Data available from national surveys can provide some useful information on U.S.

1 annual health outcome statistics, and also provide background information on what is known as
2 the “pyramid” of effects. At the top of the pyramid, there are approximately 2.5 millions deaths
3 per year in the U.S. population, with about 900,000 deaths due to cardiovascular diseases, and
4 100,000 from chronic lower respiratory diseases (Arias et al., 2003). For measures of
5 cardiovascular disease morbidity, there are approximately 6 million hospital discharges per year
6 (Hall and DeFrances, 2003), nearly 5 million emergency department visits (McCaig et al., 2004),
7 to over 70 million ambulatory care visits for circulatory system disorders (Cherry et al., 2003).
8 For chronic respiratory health diseases, there are over 3 million hospital discharges for
9 respiratory diseases (Hall and DeFrances, 2003), nearly 13 million emergency department visits
10 (McCaig et al., 2004), over 200 million ambulatory care visits per year for respiratory conditions
11 (Cherry et al., 2003) and an estimated 700 million restricted activity days per year due to
12 respiratory conditions (Adams et al., 1999). Combining small risk estimates with relatively large
13 baseline estimates of health outcomes can result in quite large public health impacts.

14 15 **9.2.5.2.3 *Impact on life-expectancy***

16 Conceptually, ambient PM exposures may be associated with both the long-term
17 development of underlying health problems (“frailty”) and with the short-term variations in
18 timing of mortality among a susceptible population with some underlying health condition
19 (Künzli et al. 2001). New evidence from toxicological studies have provided insights into
20 potential mechanisms for PM-related health effects, but this evidence is not sufficient to allow
21 direct conclusions to be drawn regarding specific effects linked with short-term or long-term PM
22 exposures. Epidemiologic studies of the mortality effects of short-term exposure to particulate
23 matter using time-series studies can only capture PM's association with short-term variations in
24 mortality and, therefore, must systematically underestimate the proportion of total mortality
25 attributable to PM. The relative risk estimates for mortality from the prospective cohort studies
26 have converged in the range of 7 to 13 percent increase in the non-external mortality rate
27 associated with a 10 $\mu\text{g}/\text{m}^3$ increment in a long-term average of $\text{PM}_{2.5}$. Risk estimates from
28 short-term exposure studies are considerably smaller in magnitude, on the order of 2 to 6%
29 increased risk of mortality per 25 $\mu\text{g}/\text{m}^3$ change in 24-hour average $\text{PM}_{2.5}$. A recent time-series
30 study that examined the contribution of daily PM levels over an extended lag period (42 days)
31 could only partially bridge the gap between the effects of short-term and long-term exposures to

1 particulate matter (Zanobetti et al., 2002). The PM effect size estimates for total mortality from
2 these studies also indicate that a substantial portion of these deaths reflect cumulative PM effects
3 above and beyond those exerted by short-term exposure events.

4 Recent investigations of the public health implications of effect estimates for long-term
5 PM exposures also were reviewed in Chapter 8. Life table calculations by Brunekreef (1997)
6 found that relatively small differences in long-term exposure to ambient airborne PM can have
7 substantial effects on life expectancy. For example, a calculation for the 1969 to 1971 life table
8 for U.S. white males indicated that a chronic exposure increase of 10 $\mu\text{g}/\text{m}^3$ PM was associated
9 with a reduction of ~ 1.3 years for the entire population's life expectancy at age 25. The new
10 evidence noted above of infant mortality associations with PM exposure suggests that life
11 shortening in the entire population from long-term PM exposure could well be significantly
12 larger than estimated by Brunekreef (1997).

13 It is also useful to highlight the newer results of the extension of the ACS study analyses
14 (that include more years of participant follow-up and address previous criticisms of the earlier
15 ACS analyses), which provide the strongest evidence to date that long-term ambient PM
16 exposures are associated with increased risk of lung cancer. That increased risk appears to be in
17 about the same range as that seen for a non-smoker residing with a smoker and, therefore,
18 passively exposed chronically to tobacco smoke, with any consequent life-shortening impacts
19 due to lung cancer.

21 **9.2.5.3 Summary and Conclusions**

22 Clearly, the public health impact of exposures to ambient PM can be quite large. The
23 population groups that are likely more susceptible to the effects of ambient particles, including
24 those with heart or lung diseases, children and older adults, comprise substantial portions of the
25 U.S. population. Even relatively small increases in PM-related risks for serious health effects
26 (e.g., premature mortality, hospital admissions) in such large population groups result in
27 substantial public health impacts. In addition, somewhat larger increases in PM-related risks for
28 less serious health effects (e.g., medical visits, respiratory symptoms) can add substantially to
29 this overall public health burden. Looking beyond the question of how many PM-related
30 premature deaths are likely to occur, it is also important to address the question of the extent of
31 life lost due to PM-related premature mortality. Findings from recent studies indicate that loss

1 of population life expectancy may be substantial, on the order of a year or so, with long-term
2 exposure to PM; however, further research is needed on this question. Further research is also
3 needed to build upon currently only very limited evidence about potential PM-related health
4 endpoints in infants and children, which is emerging as an area of more concern than in the
5 previous review.

6 7 8 **9.3 SYNTHESIS OF AVAILABLE INFORMATION ON PM-RELATED** 9 **WELFARE EFFECTS**

10 The synthesis of available information on PM-related welfare effects presented in this
11 section focuses on four types of effects, i.e., PM-related effects on: visibility, vegetation and
12 ecosystems, man-made materials, and climate change processes. The resulting synthesis of
13 information and conclusions are intended to provide the scientific bases for options to be
14 considered by the EPA Administrator as to whether currently available scientific information
15 supports retention or revision of existing secondary PM NAAQS.

16 17 **9.3.1 Airborne Particle Effects on Visibility**

18 The following discussion of the effects of airborne particles on visibility is drawn primarily
19 from information in Chapter 4 of this document, which itself is supplementary to several other
20 significant reviews of the science of visibility. These reviews include reports of the National
21 Acid Precipitation Assessment Program (1991, 1998), the National Research Council's report on
22 Protecting Visibility in National Parks and Wilderness Areas (1993), and U.S. EPA's Interim
23 Findings on the Status of Visibility Research (1995). The focus here is on characterizing:
24 (a) how ambient PM (in particular ambient fine PM) affects visibility, and (b) how the public
25 values improvements in visibility, especially in urban areas.

26 27 **9.3.1.1 How Does Newly Available Information Inform Our Understanding of How** 28 **Ambient PM and Its Major Constituents Affect Visibility?**

29 The role of ambient PM in impairing visibility has long been well understood, as was
30 recognized in the 1996 PM AQCD as follows:

1 “The relationships between air quality and visibility are well understood. Ambient fine
2 particles are the major cause of visibility impairment. Significant scientific evidence
3 exists showing that reducing fine particle concentrations will improve visibility.”
4 (U.S. EPA, 1996, p. 1-18).
5

6 More specifically, the efficiency with which airborne particles cause visibility impairment
7 depends on not just the mass of fine particles, but also on particle composition, particle size, and
8 relative humidity. Airborne particles degrade visibility due to their optical properties of light
9 scattering and absorption, which can be well characterized in terms of a light extinction
10 coefficient. The contribution of airborne particles to total light extinction can be derived from
11 well-established relationships for the major fine particle components, with relative-humidity
12 adjustment factors to account for the hygroscopic behavior of the sulfate and nitrate components;
13 coarse mode particles generally play a much smaller role. Sulfates, nitrates, and organic carbon
14 are the primary light-scattering components of fine particles, with each component being
15 relatively more important to visibility impairment in different parts of the U.S. (e.g., sulfates
16 being the most important contributor in the eastern U.S., organic carbon in the western U.S., and
17 nitrates in southern California). Elemental carbon and, to a much smaller degree, crustal
18 materials are the primary light-absorbing components of fine particles. Some minerals in coarse-
19 mode crustal particles also absorb light and, during events such as dust storms, can be a
20 significant factor in visibility impairment.

21 Particle-related light scattering efficiency depends on particle size, with peak efficiency
22 resulting from particles that are about 0.5 to 0.8 μm in diameter, falling off rapidly for particles
23 below 0.3 or above 1.0 μm in diameter. Therefore, fine particles within the accumulation mode
24 are most effective in scattering light and are more important in visibility degradation than either
25 ultrafine (nuclei-mode) or coarse-mode particles.

26 The overall effect of increasing humidity on light scattering by particles was quantified
27 nearly 20 years ago, but current research is greatly increasing the detailed understanding of the
28 response of aerosol particles to changing humidities and the relationship of this response to the
29 chemical composition of the particles. Humidity effects generally become important at relative
30 humidities between 60 and 70%, and increase particle-related light scattering by a factor of 2 at
31 approximately 85% relative humidity. Light scattering by particles increases rapidly with
32 relative humidity when the humidity exceeds 90%.

1 As discussed in Chapter 4, a number of studies available since the last review have resulted
2 in refinements both (a) in the algorithms and related parameters used to calculate light extinction
3 based on particle properties and (b) in related measurements methods and monitoring
4 instrumentation. For example, a few studies have focused on better characterizing the
5 hygroscopic properties of particles, with a particular focus on organic compounds and mixtures
6 associated with different sources (e.g., Cocker et al., 2001; Chughtai et al., 1999; Hemming and
7 Senfield, 2001). More broadly, Malm (2000) used data from a special study at the Great Smoky
8 Mountain National Park to compare the performance of a number of models for calculating light
9 extinction and found that significant model improvement could be obtained by including the
10 degree of sulfate ammoniation in the model. These studies have served primarily to reinforce
11 and refine our understanding of how airborne particles affect visibility.

12 Our understanding of how ambient PM affects visibility has historically focused on
13 visibility impairment in rural areas, particularly in national parks and wilderness areas (i.e.,
14 Federal Class I areas). Visibility in such areas varies substantially between eastern and western
15 sites in the U.S., with the haziest days in the West typically being roughly equivalent to the
16 clearest days in the East. The largest monitoring network that measures both visibility and
17 aerosol conditions is the Interagency Monitoring of Protected Visual Environments (IMPROVE)
18 network, formed in 1987 as a collaborative effort between Federal, regional, and state entities
19 responsible for visibility protection in such areas. This network has been used in visibility-
20 related research, including the advancement of visibility monitoring instrumentation and analysis
21 techniques and source attribution field studies. This network and related research have provided
22 substantial support to regulatory programs established to protect Federal Class I areas from local
23 and regional sources of visibility impairment.

24 Particle-related visibility impairment also occurs in urban areas, although historically the
25 relationship between ambient PM and visibility has been less well studied in such areas. More
26 recent attention has been given to such efforts, however, drawing upon data now available from
27 the new national monitoring networks designed to assess PM_{2.5} concentrations and composition
28 in urban areas across the country that have been deployed in conjunction with establishment in
29 1997 of the PM_{2.5} NAAQS. In addition, higher resolution visibility data are now becoming
30 available from the Automated Surface Observing System (ASOS) monitoring network in
31 operation at airports across the U.S. These and other sources of visibility and ambient fine

1 particle data provide important information that helps to facilitate the characterization of
2 relationships between ambient PM and visibility especially in urban areas.

3 In addition to empirically derived relationships between ambient PM and visibility
4 measurements, photographic modeling techniques that have been refined in recent years are
5 useful in portraying changes in visibility specifically due to changes in ambient PM levels.
6 For example, the WinHaze system developed by Molenaar et al. (1994) has been used to simulate
7 changes in visibility as a function of changes in air quality for both rural and urban areas. This
8 modeling system can produce a simulated photograph that accurately depicts a cloud-free scene
9 as it would appear to a human observer. Such photographic representations have facilitated the
10 evaluation of how the public values improvements in visibility in a number of urban areas, as
11 discussed below.

13 **9.3.1.2 How Does Newly Available Information Inform Our Understanding of How the** 14 **Public Values Improvements in Visibility, Especially in Urban Areas?**

15 Information about how the public values improvements in visibility comes from both
16 economic studies and from local and/or state initiatives in a number of areas to adopt local
17 visibility goals and standards. There is an extensive scientific literature on the theory and
18 application of economic valuation methods, although, as summarized in Chapter 4, study results
19 vary substantially across different valuation methods and concerns remain about the use of this
20 general approach for quantitative purposes. Initiatives over the past few years in the Denver, CO
21 and Phoenix, AZ areas provide important evidence of public interest in addressing visibility
22 impairment in these urban areas, although uncertainty would be involved in extending the public
23 values implied by these examples to other areas.

24 More specifically, the initiative in Denver began with a series of visibility-related studies
25 in the 1970's through the 1980's, leading to the adoption of a visibility standard for the city of
26 Denver in 1990. This standard is based on a light extinction level of 0.076 km^{-1} , averaged over
27 four daylight hours, reflecting the short-term nature of the perception of changes in visibility
28 conditions. This standard is equivalent to a visual range of approximately 50 km and reflects
29 citizen judgments about acceptable and unacceptable levels of visual air quality. In Phoenix,
30 a study conducted between 1988 and 1990 led to establishment of a Blue Sky Index, which
31 focuses on days in which the visual range, averaged over six daylight hours, is 40 km or more.
32 This target is based on a method very similar to that used in Denver for obtaining citizen's

1 judgments as to acceptable levels of visual air quality. While in practice these standard target
2 values are exceeded many times per year in these areas, they reflect a reasonable degree of
3 consistency in the outcome of the approach used to characterize the value that citizens in these
4 two urban areas place on visual air quality.

6 **9.3.2 Effects of Ambient PM on Vegetation and Ecosystems**

7 **9.3.2.1 What Are the Direct and Indirect Effects of Ambient PM?**

8 The direct and indirect effects of deposited ambient PM can span the full range, scale and
9 properties of biological organization listed under Biotic condition (Chapter 4) and can vary
10 widely depending on the (1) sensitivity of each ecosystem and/or its component biota (biotic
11 receptors) to a given concentration and chemical composition (acid/base, trace metal or
12 nutrients, e.g., nitrates or sulfates) of PM components; (2) the pre-existing buffering capacity of
13 the soils and/or waters (freshwater streams, rivers, ponds, and lakes; estuaries and ocean); (3) the
14 magnitude (rate, deposition velocity), mode, and meteorology of the deposition; and (4) other
15 site specific features (e.g., terrain, hydrology, climate, land use, etc.). The ability of an
16 ecosystem to maintain integrity in the presence of the different stressors in PM deposition is a
17 direct function of the sensitivity level of the ecosystem to the different PM constituents and to
18 the ability of the ecosystem components to ameliorate the effects that can result. Changes in
19 structural patterns and the functioning of ecological processes must be scaled in both time and
20 space and propagated to the more complex levels of community interaction to produce
21 observable ecosystem changes.

22 Direct effects result when PM is deposited onto sensitive receptors. Such effects can be
23 either chemical and/or physical; and they have been observed largely downwind of point sources
24 as the result of dust from limestone quarries and cement kilns or heavy metals from iron and lead
25 smelting factories (Chapter 4). Because these effects tend to be very limited in scope, they do
26 not warrant the level of attention given the more widespread indirect, ecosystem-level, effects
27 discussed below.

28 The indirect effects of major concern are mediated via the soil or aquatic environment and
29 have the potential of degrading ecosystem functioning by altering species diversity, structure,
30 and sustainability of ecosystems to the detriment of animals and plant life, so that ecosystems
31 provide fewer benefits and services for humans (Moomaw, 2002).

1 Ecosystems within the U.S. span the range from remote to urban. Most of the ecosystem
2 impacts of PM that have been reported occurred at non-urban sites and, as such, non-urban
3 ecosystems are the primary focus of the discussion that follows in subsequent subsections.
4 In briefly considering urban ecosystems here, it is recognized that despite a large body of
5 knowledge on concentrations and chemical reactions of air pollutants in cities, there has been
6 little work on the rates of atmospheric deposition to urban ecosystems. However, urban
7 ecosystems are likely to be subjected to large rates of deposition of anthropogenic pollutants
8 (Lovett et al.2000). Decades of research on urban air quality indicate that cities are often
9 sources of nitrogen oxides, sulfur oxides, and dust, among many other pollutants. Some of these
10 air pollutants are major plant nutrients (e.g., nitrogen) and may be affecting nutrient cycles in
11 plant-dominated areas in and around cities. Studying the deposition rates of atmospheric
12 pollutants in urban areas can provide a quantitative estimate of the amounts of gaseous and
13 particulate air pollutants that are removed by urban vegetation. Though these effects of PM as
14 such appear not to have been measured at this time, the deployment of new PM_{2.5} speciated
15 urban monitors and concern over urban visibility impairment could lead to additional
16 information being developed that would be relevant to assessing PM effects on urban
17 ecosystems.

18 19 **9.3.2.2 What are the Components in Ambient PM that are Major Ecosystem Stressors?**

20 In order for any component of ambient PM to impact ecosystems, it must first be removed
21 from the atmosphere through deposition. Deposition can occur in three modes: wet, dry, or
22 occult. The factors that influence the magnitude and mode of particle deposition are numerous
23 and complex and depend in part on particle size, shape, chemistry, atmospheric conditions (e.g.,
24 relative humidity, wind speed) and ecosystem surface features (e.g., elevation, complexity of
25 terrain, land over type, etc.). National deposition monitoring networks routinely measure total
26 wet or dry deposition of certain compounds. Data from these networks demonstrate that
27 nitrogen and sulfur compounds are being deposited onto soils and aquatic ecosystems in
28 sufficient amounts to impact ecosystems at local, regional and national scales. Though the
29 ambient PM contribution to total wet or dry deposition has rarely been characterized and the
30 percentages of nitrogen and sulfur containing compounds in PM vary spatially and temporally,
31 nitrates and sulfates make up a substantial portion of the chemical composition of PM.

1 Therefore, the components of PM that are considered of greatest environmental significance are
2 nitrates, sulfates and the associated hydrogen (H^+) ion (Chapter 4).

3 4 **9.3.2.2.1 Nitrogen**

5 Nitrogen is required by all organisms as it is a major constituent of the nucleic acids that
6 determine the genetic character of all living things and the enzyme proteins that drive the
7 metabolic machinery of every living cell (Galloway, 1998; Galloway and Cowling, 2002). It has
8 long been recognized as the nutrient most important for plant metabolism and, to a large extent it
9 governs the utilization of phosphorus, potassium, and other nutrients. Typically, the availability
10 of nitrogen via the nitrogen cycle controls net primary productivity, and possibly, the
11 decomposition rate of plant litter. Plants usually obtain nitrogen directly from the soil by
12 absorbing NH_4^+ or NO_3^- through their roots, or it is formed in their roots by symbiotic organisms
13 (bacteria, blue-green algae). However, nitrogen (N), unlike other essential nutrients, is not
14 readily available and usually is in short supply.

15 Nitrogen in nature can be divided into two groups: nonreactive (N_2) and reactive (Nr).
16 Molecular nitrogen (N_2), though the most abundant element in the Earth's atmosphere, is not
17 available to more than 99% of living organisms unless converted into reactive forms
18 (Galloway et al., 2003). Reactive Nr includes the inorganic reduced forms of nitrogen (e.g.,
19 ammonia [NH_3] and ammonium [NH_4^+]), inorganic oxidized forms (e.g., nitrogen oxide [NO_x],
20 nitric acid [HNO_3], nitrous oxide [N_2O], and nitrate [NO_3^-]), and the organic compounds (e.g.,
21 urea, amine, proteins, and nucleic acids)] (Galloway and Cowling, 2002).

22 Anthropogenic Nr creation now exceeds the rate of natural terrestrial Nr creation and its
23 conversion back to N_2 by denitrification (Galloway and Cowling, 2002). Thus, increase in
24 global Nr is the result of three main causes: (1) widespread cultivation of legumes, rice and
25 other crops that promote conversion of N_2 to organic nitrogen through biological nitrogen
26 fixation (BNF); (2) combustion of fossil fuels which converts both atmospheric N_2 and fossil N
27 to reactive NO_x ; and (3) the Haber-Bosch process, developed in 1913, which converts
28 nonreactive N_2 to reactive NH_3 to sustain food production and some industrial activities
29 (Galloway and Cowling, 2002; Galloway et al., 2003). As a result, Nr is now accumulating in
30 the atmosphere and terrestrial and aquatic ecosystems on all spatial scales – local, regional and
31 global (Galloway and Cowling, 2002; Galloway et al., 2003).

1 Nitrogen oxides is the only ambient air criteria pollutant that has not decreased since the
2 passage of the Clean Air Act. Despite decreases in emissions from fossil fuel burning industries,
3 emissions from automobiles have increased approximately 10% since 1970 due to greater total
4 miles driven (Howarth et al., 2002). Nitrogen oxides emissions from fuel burning increased
5 exponentially from 1940 until the 1970s, leveled off after the passage in of the Clean Air Act in
6 1970, and stabilized at approximately 7 Tg NO_x /yr in the late 1990s. Contemporary emissions
7 of NO_x in the U.S. from fossil fuel burning are nearly two-thirds the rate of Nr releases from the
8 use of inorganic fertilizers and comprise 30% of the global emissions of NO_x from fossil fuel
9 combustion.

11 *Environmental Effects of Nr*

12 The term “nitrogen cascade” refers to the sequential transfers and transformations of Nr
13 molecules as they move from one environmental system or reservoir (atmosphere, biosphere,
14 hydrosphere) to another and the multiple linkages that develop among the different ecological
15 components. Because of these linkages, the addition of anthropogenic Nr alters a wide range of
16 biogeochemical processes and exchanges as it moves among the different environmental
17 reservoirs, with the consequences becoming magnified through time (Figure 4-15; Galloway and
18 Cowling, 2002; Galloway et al., 2003). These changes in the nitrogen cycle are contributing to
19 both beneficial and detrimental effects to the health and welfare of humans and ecosystems
20 (Rabalais, 2002; van Egmond et al., 2002; Galloway, 1998).

21 Some of the detrimental effects resulting from increased inputs of atmospheric Nr include:
22 (1) increases in productivity of Nr-limited forests and grasslands followed by decreases
23 wherever increase in atmospheric deposition of Nr significantly exceeds critical thresholds; Nr
24 additions have also been shown to decrease biodiversity in many natural habitats (Aber et al.,
25 1995); (2) formation of O₃ and ozone-induced injury to crops, forests, and natural ecosystems
26 and the resulting predisposition to attack by pathogens and insects; (3) nitrogen saturation of
27 soils in forests and other natural ecosystems, leading to shifts in community composition and
28 leaching of Nr into streams, lakes and rivers; (4) eutrophication, hypoxia, loss of biodiversity,
29 and habitat degradation in coastal ecosystems, now considered the biggest pollution problem in
30 coastal waters (Rabalais, 2002); (5) acidification and loss of biodiversity in lakes and streams in
31 many regions of the world when associated with sulfur (Vitousek et al., 1997); and (6) alteration

1 of ecosystem processes through changes in the functioning of beneficial soil organisms
2 (Galloway and Cowling 2002).

3 Indirect effects of Nr on societal values include: (1) increases in fine PM resulting in
4 regional hazes that decrease visibility at scenic rural and urban vistas and airports; (2) depletion
5 of stratospheric ozone by N₂O emissions which can in turn affect ecosystems and human health;
6 (3) global climate change induced by emissions of N₂O; and (4) formation of acidic deposition
7 when in association with sulfate (Galloway et al., 2002).

8 Large uncertainties, however, still exist concerning the rates of Nr accumulation in the
9 various environmental reservoirs which limits our ability to determine the temporal and spatial
10 distribution of environmental effects for a given input of Nr. These uncertainties are of great
11 significance because of the sequential nature of Nr effects on environmental processes. Reactive
12 nitrogen does not cascade at the same rate through all environmental systems. The only way to
13 eliminate Nr accumulation and stop the cascade is to convert Nr back to nonreactive N₂
14 (Galloway et al., 2003).

16 ***Nitrogen Saturation and Ecosystem Response***

17 A major environmental concern is nitrogen saturation of soils. Nitrogen saturation occurs
18 when chronic additions of nitrogen (including nitrate deposition from ambient PM) to soil
19 background levels (nitrogen loading) exceeds the capacity of plants and soil microorganisms to
20 utilize and retain nitrogen (Aber et al., 1989, 1998; Garner 1994; U.S. Environmental Protection
21 Agency, 1993). Nitrogen saturation implies that some resource other than nitrogen is now
22 limiting biotic functions. The appearance of nitrogen in soil solution (leaching) is an early
23 symptom of excess nitrogen.

24 Nitrogen saturation does not occur at a specific point in time, but is a set of gradually
25 developing critical changes in ecosystem processes which represent the integrated response of a
26 system to increased nitrogen availability over time (Aber, 1992). The chronic additions and
27 accumulation of nitrogen alter normal nitrogen cycling and many of the soil and plant processes
28 involving nitrogen that affect an ecosystems' nutrient balance (Waring, 1987; Figure 4-16,
29 Chapter 4).

30 Not all vegetation or ecosystems react in the same manner to nitrogen deposition.
31 Responses vary depending on numerous factors, including soil composition and the length of

1 time nitrate deposition has been occurring. For example, ecosystems comprised of older, mature
2 forests with high stores of soil nitrogen and low C:N ratios receiving high nitrogen deposition
3 are prone to nitrogen saturation (Fenn et al., 1998).

4 Variations in the response of forest ecosystems in the eastern and the western U.S. to
5 differing amounts of nitrate deposition illustrate this point (Chapter 4, Table 4-14). Although
6 soils of most North American forest ecosystems are nitrogen limited, some exhibit severe
7 symptoms of nitrogen saturation (See Figure 4-17; Chapter 4 (Aber et al., 1989). In the east,
8 these include the Great Smoky Mountains National Park (3.1 to 26.6 kg N ha⁻¹ yr) (Johnson and
9 Lindberg, 1992); the Fernow Experimental Forest, WV (15 to 20 kg N ha⁻¹ yr) (Gilliam et al.,
10 1996); Whitetop Mountain, VA (32 kg N ha⁻¹ yr); the Catskill Mountains in southeastern NY
11 (10.2 kg N ha⁻¹ yr); and the Adirondack Mountains of northeastern NY (9.3 kg N ha⁻¹ yr) (see
12 Table 4-14).

13 In the west, wildland ecosystems within the South Coast Air Basin of California receive
14 the highest nitrogen deposition in the United States (Fenn et al., 1998; 2003). The areas
15 receiving the greatest deposition are the south-facing slopes of the San Gabriel Mountains and
16 the western and southern edges of the San Bernardino Mountains where deposition ranges from
17 23.3 to 30 kg N ha⁻¹ per yr. Deposition in the low- and mid-elevation chaparral and mixed
18 conifer forests ranges from 20 to 45 kg N ha⁻¹ per yr in the most exposed areas. However, when
19 fog occurs in late summer with unusually high NO₃⁻ and NH₄⁺ concentrations, deposition values
20 can be higher than 90 kg N ha⁻¹ yr (Fenn et al., 2003). The forests in the southwestern Sierra
21 Nevada of Central California receive 6-11 kg N ha⁻¹ yr as throughfall (Fenn et al; 1998).
22 Nitrogen deposition since the 1980s has resulted in saturation in the high-elevation Front Range
23 in northern Colorado where deposition values currently range from 8 to 10 kg N ha⁻¹ yr
24 (Bowman and Steltzer, 1998; Bowman, 2000; Baron et al., 2000) (Chapter 4, Table 4-14.)

25 On the other hand, the Harvard Forest hardwood stand in Massachusetts has absorbed over
26 900 kg N ha⁻¹ without significant nitrate leaching during a nitrogen amendment study of 8 years.
27 However, leaching losses were high in Harvard pine sites suggesting that deciduous forests may
28 have a greater capacity for nitrogen retention (Fenn et al., 1998). Magill et al. (2000) suggest
29 that the sharp contrasts that exist between hardwood and pine forests indicate that the mosaic of
30 community types across the landscape must be considered when determining regional scale
31 response to nitrogen deposition.

1 Increases in soil nitrogen can also play a selective role in ecosystems, by affecting
2 competition among species, resulting in changes in biodiversity, i.e., community composition.
3 In general, plants adapted to living in an environment of low nitrogen availability will be
4 replaced by nitrophilic plants capable of using increased nitrogen because they have a
5 competitive advantage when nitrogen becomes more readily available (Fenn et al., 1998).
6 Several long-term fertilization studies have observed these effects. For example, fertilization
7 and nitrogen gradient experiments at Mount Ascutney, VT suggest that nitrogen saturation may
8 lead to the slow-growing, slow nitrogen-cycling spruce-fir forest stands being replaced by fast-
9 growing deciduous forests that cycle nitrogen rapidly. Similarly, experimental studies of the
10 effects of nitrogen deposition over a 12-year period on Minnesota grasslands dominated by
11 native warm-season grasses observed the shift to low-diversity mixtures dominated by cool-
12 season grasses at all but the lowest rates of nitrogen addition (Wedin and Tilman, 1996). The
13 shift to low-diversity mixtures was associated with the decrease in biomass carbon to nitrogen
14 (C:N) ratios, increased nitrogen mineralization, increased soil nitrate, high nitrogen losses, and
15 low carbon storage (Wedin and Tilman, 1996).

16 The mutualistic relationship between plant roots, fungi, and microbes is critical for the
17 growth of the organisms involved. The rhizosphere, the soil that surrounds and is influenced by
18 plant roots is an important region of nutrient dynamics. Bacteria are essential components of the
19 nitrogen and sulfur cycles while fungi in association with plant roots form mycorrhizae that are
20 essential in the uptake of mineral nutrients. The action of bacteria make N, S, Ca, P, Mg, K
21 available for plant growth while mycorrhizae are of special importance in the uptake of N and P
22 (Section 4.3.3; Wall and Moore, 1999; Rovira and Davy, 1974). Changes in soil nitrogen
23 influence the mycorrhizal-plant relationship. Mycorrhizal fungal diversity is associated with
24 above-ground plant biodiversity, ecosystem variability, and productivity (Wall and Moore,
25 1999). During nitrogen saturation, soil microbial communities change from being fungal, and
26 dominated by mycorrhizae, to being dominated by bacteria. The decline in the coastal sage
27 scrub species can be directly linked to the decline of the arbuscular mycorrhizal community
28 (Edgerton-Warburton and Allen, 2000; Allen et al., 1998; Padgett et al., 1999).

29

Nitrate Effects on Aquatic Habitats

Aquatic ecosystems (streams, rivers, lakes, estuaries or oceans) receive increased nitrogen inputs either from direct atmospheric deposition (including nitrogen-containing particles), surface runoff, or leaching from saturated soils into ground or surface waters. The primary pathways of nitrogen loss from forest ecosystems are hydrological transport beyond the rooting zone into groundwater or stream water, or surface flows of organic nitrogen as nitrate and nitrogen loss associated with soil erosion (Fenn et al., 1998). Based on data from a number of hydrologic, edaphic, and plant indicators, the mixed conifer forest and chaparral watershed with high smog exposure in the Los Angeles Air Basin exhibited the highest stream water NO_3^- concentrations in wilderness areas of North America (Bytnerowicz and Fenn, 1996; Fenn et al., 1998). High nitrate concentrations have also been observed in streams draining watersheds in the Great Smoky Mountains National Park in Tennessee and North Carolina (Fenn et al., 1998).

Estuaries are among the most intensely fertilized systems on Earth (Fenn et al., 1998). They receive far greater nutrient inputs than other systems. For example, atmospheric nitrogen deposition into soils in watershed areas feeding into estuarine sound complexes (e.g., Chesapeake Bay, the Pamlico Sound of North Carolina) contribute to excess nitrogen flows that also include runoff from agricultural practices or other uses (e.g., fertilization of lawns or gardens). Especially during and after heavy rainfall events such as hurricanes, massive influxes of nitrogen into watersheds and sounds can lead to dramatic decreases of oxygen in water and increases in algae blooms that can cause extensive fish kills and damage to commercial fish and sea food harvesting (Paerl et al., 2001).

9.3.2.2.2 Acidification from PM deposition

Acidic deposition is composed of ions, gases, and particles derived from the precursor gaseous emissions of sulfur dioxide (SO_2), nitrogen oxides (NO_x), ammonia (NH_3) and particulate emissions of acidifying and neutralizing compounds. It connects air pollution to diverse terrestrial and aquatic ecosystems and alters the interactions of the (H^+) and many elements (e.g., S, N, Ca, Mg, Al, and Hg) (Driscoll et al., 2001). Linked also to the nitrogen cascade (see Figure 4-15), acidic precipitation is a critical environmental stress that affects forest landscapes and aquatic ecosystems in North America, Europe, and Asia (Driscoll et al., 2001).

1 Acidic deposition and acidification of soils can lead to high Al-to-nutrient ratios that limit
2 plant uptake of essential nutrients, such as Ca and Mg. Calcium is essential in the formation of
3 wood and the maintenance of the primary plant tissues necessary for tree growth (Shortle and
4 Smith, 1988), and tree species can be adversely affected if altered Ca/Al ratios impair calcium or
5 Mg uptake. A region-wide increase in Ca above expected levels followed by decreasing changes
6 in wood Ca suggests that Ca mobilization began possibly 30 to 40 years ago and has been
7 followed by reduced accumulation in wood, presumably associated with decreasing Ca
8 availability in soil (Chapter 4; Bondietti and McLaughlin, 1992).

9 10 **9.3.2.3 How can Exposures of Concern for Ecosystem Stressor Components of PM** 11 **be Characterized?**

12 The critical loads concept has been used in Europe for estimating the amounts of pollutants
13 that sensitive ecosystems can absorb on a sustained basis without experiencing measurable
14 degradation (Lokke et al., 1996). The estimation of ecosystem critical loads requires an
15 understanding of how an ecosystem will respond to different loading rates in the long term and
16 can be of special value for ecosystems receiving chronic deposition of Nr and sulfur
17 independently and as acidic deposition when in combination. Time scales must be considered
18 when selecting and evaluating ecosystems response(s) to changes in atmospheric deposition.
19 Indicators of ecosystems at risk of nitrogen saturation should include those that can be identified
20 when nitrogen availability exceeds biotic demand. The cardinal indicator of nitrogen saturation
21 in all ecosystem types is increased and prolonged NO_3^- loss below the main rooting zone in
22 stream water (Fenn and Poth, 1998). A paucity of baseline data makes it difficult to determine
23 the time scale for critical loading of most U.S. ecosystems because nitrogen deposition began so
24 many years ago. Though atmospheric sources of nitrogen, including ambient PM, are clearly
25 contributing to the overall excess nitrogen load/burden entering ecosystems annually, there is
26 still insufficient data available at this time to quantify the contribution of ambient PM to total
27 nitrogen or acidic deposition its role varies both temporally and spatially along with a number of
28 other factors.

1 **9.3.2.4 Summary and Conclusions**

2 A number of ecosystem-level conditions (e.g., nitrogen saturation, terrestrial and aquatic
3 acidification, coastal eutrophication) that can lead to negative impacts on human health and
4 welfare have been associated with chronic, long-term exposure of ecosystems to elevated inputs
5 of compounds containing Nr, sulfur and/or associated hydrogen ions. Some percentage of total
6 ecosystem inputs of these chemicals is contributed by deposition of atmospheric particles,
7 although the percentage greatly varies temporally and geographically and has not generally been
8 well quantified. Unfortunately, our ability to relate ambient concentrations of PM to ecosystem
9 response is hampered by a number of significant data gaps and uncertainties.

10 First, U.S. monitoring networks have only recently begun to measure speciated PM.
11 Historically, measurements were focused only on a particular size fraction such as PM₁₀ and,
12 more recently, PM_{2.5}. An exception to this is the IMPROVE network, which collects speciated
13 measurements. Additionally, except for the IMPROVE and some CASTNet sites, much of the
14 PM monitoring effort has focused on urban or near urban exposures, rather than on those in
15 sensitive ecosystems. Thus, the lack of a long-term, historic database of annual speciated PM
16 deposition rates precludes establishing relationships between PM deposition (exposure) and
17 ecosystem response at this time.

18 A second source of uncertainty lies in predicting deposition velocities based on ambient
19 concentrations of PM. There are a multitude of factors that influence the amounts of PM that get
20 deposited from the air onto sensitive receptors, including the mode of deposition (wet, dry,
21 occult), windspeed, surface roughness/stickiness, elevation, particle characteristics (e.g., size,
22 shape, chemical composition, etc.) relative humidity, etc. Therefore, modeled deposition rates,
23 used in the absence of monitored data, can be highly uncertain.

24 Third, each ecosystem has developed within a context framed by the topography,
25 underlying bedrock, soils, climate, meteorology, hydrologic regime, natural and land use history,
26 species associations that co-occur at that location (i.e., soil organisms, plants, etc.), and
27 successional stage, making it unique from all others. Because of this variety, and insufficient
28 baseline data on each of these features for most ecosystems, it is currently impossible to
29 extrapolate with much confidence any effect from one ecosystem to another, or to predict an
30 appropriate “critical load.” Thus, for example, a given PM deposition rate or load of nitrates in

1 one ecosystem may produce entirely different responses than the same deposition rate at another
2 location.

3 Finally, related in part to the complexity and unique set of characteristics belonging to each
4 ecosystem as discussed above, there remain large uncertainties associated with the length of
5 residence time of Nr in a particular ecosystem component or reservoir, and thus, its impact on
6 the ecosystem as it moves through the various levels of the N cascade. As additional PM
7 speciated air quality and deposition monitoring data become available, there is much room for
8 fruitful research into the areas of uncertainty identified above.

9 10 **9.3.3 What Does the Available Information Indicate About the Relationships** 11 **Between Atmospheric PM and Climate Change Processes?**

12 **9.3.3.1 Key Points from 1996 PM AQCD**

13 With regard to the role of ambient PM in affecting climate change-related processes, the
14 1996 PM AQCD stated:

15
16 “Particles [primarily fine particles] suspended in the atmosphere affect the earth's energy
17 budget and thus exert an impact on climate: (a) directly by increasing the reflection of
18 solar radiation by cloud-free portions of the atmosphere, and (b) indirectly by affecting
19 cloud microphysical properties in ways that increase the brightness and stability of
20 clouds.” Since aerosol lifetimes are much shorter than the time required for global
21 mixing, “aerosol radiative effects are most likely to exert their influence on a regional
22 rather than on a global basis.” (U.S. EPA, 1996, p. 1-19, 1-21)
23

24 **9.3.3.2 Integration of New Information**

25 The same physical processes (i.e., light scattering and absorption) responsible for visibility
26 degradation are also responsible for airborne particle effects on transmission of solar visible and
27 ultraviolet radiation. Scattering of solar radiation back to space and absorption of solar radiation
28 determine the effects of an aerosol layer on solar radiation. Atmospheric particles greatly
29 complicate projections of future trends in global warming processes because of emissions of
30 greenhouse gases; consequent increases in global mean temperature; resulting changes in
31 regional and local weather patterns; and mainly deleterious (but some beneficial) location-
32 specific human health and environmental effects. Available evidence, ranging from satellite to
33 in situ measurements of aerosol effects on radiation receipts and cloud properties, is strongly
34 indicative of an important role in climate for aerosols, but this role is poorly quantified. No

1 significant advances have been made since the 1996 PM AQCD in reducing the uncertainties
2 assigned to forcing estimates for aerosol-related forcing, especially for black carbon-containing
3 aerosol. The IPCC characterizes the scientific understanding of greenhouse gas-related forcing
4 as “high” in contrast to that for aerosol, which it describes as “low” to “very low.”

5 In addition to direct climate effects through the scattering and absorption of solar radiation,
6 particles also exert indirect effects on climate by serving as cloud condensation nuclei, thus
7 affecting the abundance and vertical distribution of clouds. The direct and indirect effects of
8 particles appear to have significantly offset global warming effects caused by the buildup of
9 greenhouse gases on a globally averaged basis. However, because the lifetime of particles is
10 much shorter than that required for complete mixing within the Northern Hemisphere, the
11 climate effects of particles generally are felt much less homogeneously than are the effects of
12 long-lived greenhouse gases.

13 Quantification of the effect of anthropogenic aerosol on hydrological cycles requires more
14 information than is presently available regarding ecosystems responses to reduced solar radiation
15 and other changes occurring in the climate system. However, several global-scale studies
16 indicate that aerosol cooling alone can slow down the hydrological cycle, while cooling plus the
17 nucleation of additional cloud droplets can dramatically reduce precipitation rates.

18 Any effort to model the impacts of local alterations in particle concentrations on projected
19 global climate change or consequent local and regional weather patterns would be subject to
20 considerable uncertainty.

21 Atmospheric particles also complicate estimation of potential future impacts on human
22 health and the environment projected as possible to occur because of increased transmission of
23 solar ultraviolet radiation (UV-B) through the Earth’s atmosphere, secondary to stratospheric
24 ozone depletion due to anthropogenic emissions of chlorofluorocarbons (CFCs), halons, and
25 certain other gases. The transmission of solar UV-B radiation is strongly affected by
26 atmospheric particles. Measured attenuations of UV-B under hazy conditions range up to 37%
27 of the incoming solar radiation. Measurements relating variations in PM mass directly to UV-B
28 transmission are lacking. Particles also can affect the rates of photochemical reactions occurring
29 in the atmosphere, e.g., those involved in catalyzing tropospheric ozone formation. Depending
30 on the amount of absorbing substances in the particles, photolysis rates either can be increased or
31 decreased. Thus, atmospheric particle effects on UV-B radiation, which vary depending on size

1 and composition of particles, can differ substantially over different geographic areas and from
2 season to season over the same area. Any projection of effects of location-specific airborne PM
3 alterations on increased atmospheric transmission of solar UV radiation (and associated potential
4 human health or environmental effects) due to stratospheric ozone-depletion would, therefore,
5 also be subject to considerable uncertainty.

7 **9.3.4 What Does the Available Information Indicate About the Effects on** 8 **Man-Made Materials Associated With Ambient PM and its Major** 9 **Constituents?**

10 **9.3.4.1 Key Points from 1996 PM AQCD**

11 The 1996 PM AQCD arrived at the following key findings and conclusions related to PM
12 effects on man-made materials:

14 “Particle exposure results in the soiling of painted surfaces and other building materials,
15 increasing the cleaning frequency for exposed surfaces and possibly reducing their useful
16 lifetimes.” (U.S. EPA, 1996, p. 1-19) Damage to materials can result from the deposition
17 of acid aerosols and the dissolution of acid forming gases on metal surfaces, increasing
18 the corrosion of metals; “exposure to acid forming gases may also limit the life
19 expectancy of paints and may damage various building stones and cement products
20 beyond that resulting from natural weathering processes.” (U.S. EPA, 1996, p. 1-20).

22 **9.3.4.2 Integration of New Information**

23 As noted in the 1996 PM AQCD and restated in Chapter 4 (Section 4.4), building materials
24 (metals, stones, cements, and paints) undergo natural weathering processes from exposure to
25 environmental elements (wind, moisture, temperature fluctuations, sun light, etc.). Metals form
26 a protective film of oxidized metal (e.g., rust) that slows environmentally induced corrosion. On
27 the other hand, the natural process of metal corrosion from exposure to natural environmental
28 elements is enhanced by exposure to anthropogenic pollutants, in particular SO₂ or other acidic
29 substances, that render the protective film less effective. For example, dry deposition of SO₂
30 enhances the effects of environmental elements on calcereous stones (limestone, marble, and
31 cement) by converting calcium carbonate (calcite) to calcium sulfate dihydrate (gypsum). The
32 rate of deterioration is determined by the SO₂ concentration, the deposition rate, and the stone’s
33 permeability and moisture content; however, the extent of the damage to stones produced by the
34 pollutant species above and beyond that from the natural weathering processes is uncertain.

1 Sulfur dioxide also has been found to limit the life expectancy of paints by causing discoloration
2 and loss of gloss and thickness of the paint film layer.

3 As also highlighted in the 1996 PM AQCD, the soiling of painted surfaces and other
4 building materials is a significant detrimental effect of particle pollution. Soiling changes the
5 reflectance of a material from opaque and reduces the transmission of light through transparent
6 materials; it is also a degradation process that requires remediation by cleaning or washing and,
7 depending on the soiled surface, repainting. Available data indicate that airborne particles can
8 result in increased cleaning frequency of exposed surfaces and may reduce the usefulness of
9 soiled materials. Attempts have been made to quantify the pollutant exposure levels at which
10 materials damage and soiling have been perceived; but, to date, insufficient data are available to
11 advance our knowledge regarding perception thresholds with respect to pollutant concentration,
12 particle size, and chemical composition.

13

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APPENDIX 9A

Key Quantitative Estimates of Relative Risk for Particulate Matter-Related Health Effects Based on Epidemiologic Studies of U.S. and Canadian Cities

TABLE 9A-1. ESTIMATED TOTAL, CARDIOVASCULAR, AND RESPIRATORY MORTALITY EFFECT SIZES PER INCREMENTS IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5} FROM U.S. AND CANADIAN STUDIES

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
MORTALITY: Total (nonaccidental) Mortality					
Ito and Thurston (1996) Chicago, IL	GAM not used	2.47 (1.26, 3.69)	—	—	PM ₁₀ 38 (max 128)
Styer et al. (1995) Chicago, IL	GAM not used	4.08 (0.08, 8.24)	—	—	PM ₁₀ 37 (4, 365)
Kinney et al. (1995) Los Angeles, CA	GAM not used	2.47 (-0.17, 5.18)	—	—	PM ₁₀ 58 (15, 177)
Pope et al. (1992) Utah Valley, UT	GAM not used	7.63 (4.41, 10.95)	—	—	PM ₁₀ 47 (11, 297)
Schwartz (1993) Birmingham, AL	GAM not used	5.36 (1.16, 9.73)	—	—	PM ₁₀ 48 (21, 80)
Schwartz et al. (1996) Schwartz (2003a) Boston, MA	GAM Strict GLM NS GLM BS GLM PS	—	5.3 (3.5, 7.1) 5.7 (3.7, 7.6) 5.0 (3.1, 7.0) 4.5 (2.5, 6.5)	0.7 (-1.9, 3.4)	PM ₁₀ 24.5 (SD 12.8) PM _{2.5} 15.7 (SD 9.2) PM _{10-2.5} 8.8 (SD 7.0)
Schwartz et al. (1996) Schwartz (2003a) Kingston-Harriman, TN	GAM Strict GLM NS GLM BS GLM PS	—	3.1 (0.0, 6.2) 3.0 (-0.3, 6.6) 2.8 (-0.5, 6.3) 2.6 (-0.8, 6.1)	1.7 (-2.7, 6.3)	PM ₁₀ 32.0 (SD 14.5) PM _{2.5} 20.8 (SD 9.6) PM _{10-2.5} 11.2 (SD 7.4)
Schwartz et al. (1996) Schwartz (2003a) St. Louis, MO	GAM Strict GLM NS GLM BS GLM PS	—	2.6 (0.9, 4.3) 2.4 (0.6, 4.1) 2.6 (0.9, 4.4) 2.3 (0.6, 4.1)	0.3 (-2.1, 2.7)	PM ₁₀ 30.6 (SD 16.2) PM _{2.5} 18.7 (SD 10.5) PM _{10-2.5} 11.9 (SD 8.5)
Schwartz et al. (1996) Schwartz (2003a) Steubenville, OH	GAM Strict GLM NS GLM BS GLM PS	—	2.4 (-0.4, 5.3) 1.7 (-1.3, 4.8) 1.5 (-1.5, 4.6) 1.8 (-1.2, 4.9)	5.2 (0.0, 10.7)	PM ₁₀ 45.6 (SD 32.3) PM _{2.5} 29.6 (SD 21.9) PM _{10-2.5} 16.1 (SD 13.0)

**TABLE 9A-1 (cont'd). ESTIMATED TOTAL, CARDIOVASCULAR, AND RESPIRATORY MORTALITY
EFFECT SIZES PER INCREMENTS IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
FROM U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
MORTALITY: Total (nonaccidental) Mortality (cont'd)					
Schwartz et al. (1996)	GAM Strict	—	1.6 (-5.3, 9.0)		PM ₁₀ 26.7 (SD 16.1)
Schwartz (2003a)	GLM NS		2.7 (-5.0, 10.9)		PM _{2.5} 12.2 (SD 7.4)
Topeka, KS	GLM BS		1.3 (-6.2, 9.3)		PM _{10-2.5} 14.5 (SD 12.2)
	GLM PS		1.4 (-6.3, 9.6)	-3.0 (-8.1, 2.3)	
Schwartz et al. (1996)	GAM Strict	—	3.5 (2.5, 4.5)	—	PM ₁₀ means 17.8-45.6
Schwartz (2003a)	GLM NS		3.3 (2.2, 4.3)		PM _{2.5} means 11.2-29.6
6 Cities, Overall	GLM BS		3.0 (2.0, 4.0)		PM _{10-2.5} means 6.6-16.1
	GLM PS		2.9 (1.8, 4.0)		
Klemm et al. (2000)	GAM Strict	2.0 (0.0, 4.1)	2.0 (0.5, 3.5)	0.0 (-2.2, 2.3)	PM ₁₀ 30.6 (SD 16.2)
Klemm and Mason (2003)	GLM NS	1.0 (-1.5, 3.6)	1.3 (-0.5, 3.0)	-0.5 (-3.0, 2.0)	PM _{2.5} 18.7 (SD 10.5)
Six City reanalysis-St. Louis					PM _{10-2.5} 11.9 (SD 8.5)
Klemm et al. (2000)	GAM Strict	2.5 (-1.7, 7.0)	1.5 (-1.6, 4.7)	4.6 (-0.7, 10.1)	PM ₁₀ 45.6 (SD 32.3)
Klemm and Mason (2003)	GLM NS	1.5 (-1.7, 4.9)	0.5 (-2.7, 3.8)	4.0 (-1.6, 10.0)	PM _{2.5} 29.6 (SD 21.9)
Six City reanalysis- Steubenville					PM _{10-2.5} 16.1 (SD 13.0)
Klemm et al. (2000)	GAM Strict	-3.5 (-11.6, 5.4)	1.5 (-6.5, 10.2)	-3.7 (-9.2, 2.1)	PM ₁₀ 26.7 (SD 16.1)
Klemm and Mason (2003)	GLM NS	-5.4 (-14.3, 4.4)	-0.5 (-9.5, 9.4)	-4.7 (-10.8, 1.8)	PM _{2.5} 12.2 (SD 7.4)
Six City reanalysis-Topeka					PM _{10-2.5} 14.5 (SD 12.2)
Klemm et al. (2000)	GAM Strict	6.1 (1.5, 11.0)	4.3 (0.9, 7.8)	3.5 (-1.0, 8.2)	PM ₁₀ 32.0 (SD 14.5)
Klemm and Mason (2003)	GLM NS	5.1 (-0.2, 10.7)	3.8 (-0.1, 7.8)	3.0 (-1.9, 8.2)	PM _{2.5} 20.8 (SD 9.6)
Six City reanalysis - Kingston-Harriman					PM _{10-2.5} 11.2 (SD 7.4)
Klemm et al. (2000)	GAM Strict	6.1 (3.6, 8.8)	5.1 (3.3, 6.9)	1.3 (-1.1, 3.7)	PM ₁₀ 24.5 (SD 12.8)
Klemm and Mason (2003)	GLM NS	5.6 (2.8, 8.5)	4.0 (1.9, 6.2)	1.8 (-1.0, 4.6)	PM _{2.5} 15.7 (SD 9.2)
Six City reanalysis - Boston					PM _{10-2.5} 8.8 (SD 7.0)
Klemm et al. (2000)	GAM Strict	1.0 (-4.6, 7.0)	1.5 (-2.7, 5.9)	0.0 (-4.8, 5.0)	PM ₁₀ 17.8 (SD 11.7)
Klemm and Mason (2003)	GLM NS	-1.5 (-7.7, 5.1)	-1.2 (-5.7, 3.5)	-1.0 (-6.2, 4.5)	PM _{2.5} 11.2 (SD 7.8)
Six City reanalysis - Portage					PM _{10-2.5} 6.6 (SD 6.8)

**TABLE 9A-1 (cont'd). ESTIMATED TOTAL, CARDIOVASCULAR, AND RESPIRATORY MORTALITY
EFFECT SIZES PER INCREMENTS IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
FROM U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
MORTALITY: Total (nonaccidental) Mortality (cont'd)					
Klemm et al. (2000)	GAM Strict	3.5 (2.0, 5.1)	3.0 (2.0, 4.1)	0.8 (-0.6, 2.1)	PM ₁₀ means 17.8-45.6
Klemm and Mason (2003)	GLM NS	2.5 (0.8, 4.3)	2.0 (0.9, 3.2)	0.5(-1.0, 2.0)	PM _{2.5} means 11.2-29.6
Six City reanalysis - overall					PM _{10-2.5} means 6.6-16.1
Samet et al. (2000a,b)	GAM strict	1.4 (0.9, 1.9)	—	—	PM ₁₀ mean range 15.3-52.0
Dominici et al. (2002, 2003a)	GLM NS	1.1 (0.5, 1.7)			
90 Largest U.S. Cities					
Schwartz (2000a)	GAM Strict	3.4 (2.6, 4.1)	—	—	PM ₁₀ mean range
Schwartz (2003b)	GLM NS	2.8 (2.0, 3.6)			27.1-40.6
10 U.S. cities					
Schwartz (2000a)	Strict GAM	5.41 (2.36, 8.56)	—	—	PM ₁₀ mean 36.5
Chicago, IL	(dist. lag)				
Schwartz (2003a)					
Schwartz (2000a)	Strict GAM	3.14 (0.25, 6.11)	—	—	PM ₁₀ mean 36.4
Pittsburgh, PA	(dist. lag)				
Schwartz (2003)	GLM PS	2.83 (-0.44, 6.21)			
Schwartz (2000a)	Strict GAM	6.83 (3.73, 10.02)	—	—	PM ₁₀ mean 36.9
Detroit, MN	(dist. lag)				
Schwartz (2003b)	GLM PS	5.83 (2.26, 9.52)			
Schwartz (2000a)	Strict GAM	7.46 (3.94, 11.10)	—	—	PM ₁₀ mean 32.5
Seattle, Wa	(dist. lag)				
Schwartz (2003b)	GLM PS	7.04 (3.33, 10.88)			
Schwartz (2000a)	Strict GAM	10.25 (4.67, 16.12)	—	—	PM ₁₀ mean 27.5
Minneapolis, MN	(dist. lag)				
Schwartz (2003a)	GLM PS	10.68 (4.87, 16.81)			
Schwartz (2000a)	Strict GAM	1.71 (-3.44, 7.13)	—	—	PM ₁₀ mean 34.8
Birmingham, AL	(dist. lag)				
Schwartz (2003a)	GLM PS	-3.21 (-9.80, 3.87)			

TABLE 9A-1 (cont'd). ESTIMATED TOTAL, CARDIOVASCULAR, AND RESPIRATORY MORTALITY EFFECT SIZES PER INCREMENTS IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5} FROM U.S. AND CANADIAN STUDIES

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
MORTALITY: Total (nonaccidental) Mortality (cont'd)					
Schwartz (2000a) New Haven, CT	Strict GAM (dist. lag)	9.17 (1.04, 17.96)	—	—	PM ₁₀ mean 28.6
Schwartz (2003a)	GLM PS	9.22 (0.49, 18.71)			
Schwartz (2000a) Canton, OH	Strict GAM (dist. lag)	8.79 (-4.69, 24.18)	—	—	PM ₁₀ mean 29.31
Schwartz (2003)	GLM PS	7.78 (-7.04, 24.97)			
Schwartz (2000a) Spokane, WA	Strict GAM (dist. lag)	5.62 (-0.31, 11.91)	—	—	PM ₁₀ mean 40.6
Schwartz (2003a)	GLM PS	4.79 (-2.35, 12.45)			
Schwartz (2000a) Colorado Springs, CO	Strict GAM (dist. lag)	8.58 (-3.94, 22.73)	—	—	PM ₁₀ mean 27.1
Schwartz (2003a)	GLM PS	8.69 (-5.25, 24.67)			
Burnett et al. (2000) Burnett and Goldberg (2003) 8 Canadian Cities	GAM Strict GLM NS (6 knots/yr)	3.2 (1.1, 5.5) 2.7 (-0.1, 5.5)	2.8 (1.2, 4.4) 2.1 (0.1, 4.2)	1.9 (-0.1, 3.9) 1.8 (-0.6, 4.4)	PM ₁₀ 25.9 (max 121) PM _{2.5} 13.3 (max 86) PM _{10-2.5} 12.9 (max 99)
Chock et al. (2000) Pittsburgh, PA	GAM not used	< 75 years 3.1 (0.2, 6.1) > 75 years 2.0 (-0.9, 5.0)	< 75 years 2.6 (-2.0, 7.3) > 75 years 1.5 (-3.0, 6.3)	< 75 years 0.7 (-1.7, 3.7) > 75 years 1.3 (-1.3, 3.8)	NR
Clyde et al. (2000) Phoenix, AZ	GAM not used	6 (> 0, 11)	—	—	PM ₁₀ mean 45.4
Fairley (1999) Fairley (2003) Santa Clara County, CA	GAM Strict GLM NS	7.8 (2.8, 13.1) 8.3 (2.9, 13.9)	8.1 (1.6, 15.0) 7.0 (1.4, 13.0)	4.5 (-7.6, 18.1) 3.3 (-5.3, 12.6)	PM ₁₀ 34 (6, 165) PM _{2.5} 13 (2, 105) PM _{10-2.5} 11 (0, 45)
Gamble (1998) Dallas, TX	GAM not used	-3.56 (-12.73, 6.58)	—	—	PM ₁₀ 24.5 (11, 86)
Goldberg et al. (2000) Goldberg and Burnett (2003) Montreal, CAN	GAM Strict GLM NS	—	4.2 (p < 0.05) 1.5 (p > 0.05)	—	PM _{2.5} 17.6 (4.6, 71.7)

TABLE 9A-1 (cont'd). ESTIMATED TOTAL, CARDIOVASCULAR, AND RESPIRATORY MORTALITY EFFECT SIZES PER INCREMENTS IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5} FROM U.S. AND CANADIAN STUDIES

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
MORTALITY: Total (nonaccidental) Mortality (cont'd)					
Klemm and Mason (2000) Atlanta, GA	GAM not used	8.7 (-5.2, 24.7)	4.8 (-3.2, 13.4)	1.4 (-11.3, 15.9)	PM _{2.5} 19.9 (1.0, 54.8) PM _{10-2.5} 10.1 (0.2, 39.5)
Levy (1998) King Co., WA	GAM not used	7.2 (-6.3, 22.8)	1.76 (-3.53, 7.34)	—	PM ₁₀ 29.8 (6.0, 123.0) PM ₁ 28.7 (16.3, 92.2)
Lipfert et al. (2000a) Philadelphia, PA	GAM not used	5.99 (p > 0.055)	4.21 (p < 0.055)	5.07 (p > 0.055)	PM ₁₀ 32.20 (7.0, 95.0) PM _{2.5} 17.28 (-0.6, 72.6) PM _{10-2.5} 6.80 (-20.0, 28.3)
Lippmann et al. (2000) Ito (2003) Detroit, MI	GAM Strict GLM NS	3.3 (-2.0, 8.9) 3.1 (-2.2, 8.7)	1.9 (-1.8, 5.7) 2.0 (-1.7, 5.8)	3.2 (-1.9, 8.6) 2.8 (-2.2, 8.1)	PM ₁₀ 31 (12, 105) PM _{2.5} 18 (6, 86) PM _{10-2.5} 13 (4, 50) mean (5%, 95%)
Moolgavkar (2000a) Moolgavkar (2003) Los Angeles, CA	GAM Strict 30df GLM NS 30df	2.4 (0.5, 4.2) 2.3 (0.5, 4.1)	1.5 (0, 3.0) 1.4 (-0.4, 3.2)	—	PM ₁₀ median 44 (7, 166) PM _{2.5} 22 (4, 86)
Moolgavkar (2000a) Moolgavkar (2003) Cook Co., IL	GAM Strict 100df GLM NS 100df	2.4 (1.4, 3.5) 2.6 (1.6, 3.6)	—	—	PM ₁₀ median 35 (3, 365)
Ostro (1995) San Bernadino and Riverside Counties, CA	GAM not used	—	0.28 (-0.61, 1.17)	—	PM _{2.5} 32.5 (9.3, 190.1) (estimated from visibility)
Schwartz (2000b) Schwartz (2003a) Boston, MA	GLM NS	—	5.8 (4.5, 73) (15-day) 9.7 (8.2, 11.2) (60-day)	—	PM _{2.5} 15.6 (±9.2)
Laden et al. (2000) Schwartz (2003a) Six City source-oriented analysis	GLM PS	—	-5.1 (-13.9, 4.6) crustal 9.3 (4.0, 14.9) traffic 2.0 (-0.3, 4.4) coal	—	PM _{2.5} same as Six City

**TABLE 9A-1 (cont'd). ESTIMATED TOTAL, CARDIOVASCULAR, AND RESPIRATORY MORTALITY
EFFECT SIZES PER INCREMENTS IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
FROM U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
MORTALITY: Total (nonaccidental) Mortality (cont'd)					
Tsai et al. (2000) Newark, NJ	GAM not used	5.65 (4.62, 6.70)	4.34 (2.82, 5.89)	—	PM ₁₅ 55 (SD 6.5) PM _{2.5} 42.1 (SD 22.0)
Tsai et al. (2000) Camden, NJ	GAM not used	11.07 (0.70, 22.51)	5.65 (0.11, 11.51)	—	PM ₁₅ 47.0 (SD 20.9) PM _{2.5} 39.9 (SD 18.0)
Tsai et al. (2000) Elizabeth, NJ	GAM not used	-4.88 (-17.88, 10.19)	1.77 (-5.44, 9.53)	—	PM ₁₅ 47.5 (SD 18.8) PM _{2.5} 37.1 (SD 19.8)
Cardiorespiratory Mortality:					
Samet et al. (2000a, b) 90 U.S. Cities Domenici, 2002	GLM NS	1.6 (1.1, 2.0)	—	—	PM ₁₀ mean range: 15.3 - 52.0
Tsai et al. (2000) Newark, NJ	GAM not used	7.79 (3.65, 12.10)	5.13 (3.09, 7.21)	—	PM ₁₅ 55 (SD 6.5) PM _{2.5} 42.1 (SD 22.0)
Tsai et al. (2000) Camden, NJ	GAM not used	15.03 (4.29, 26.87)	6.18 (0.61, 12.06)	—	PM ₁₅ 47.0 (SD 20.9) PM _{2.5} 39.9 (SD 18.0)
Tsai et al. (2000) Elizabeth, NJ	GAM not used	3.05 (-11.04, 19.36)	2.28 (-4.97, 10.07)	—	PM ₁₅ 47.5 (SD 18.8) PM _{2.5} 37.1 (SD 19.8)
Total Cardiovascular Mortality					
Schwartz (2003b) 10 U.S. Cities	GAM Strict	4.1 (2.5, 5.6)	—	—	PM ₁₀ mean range 27.1- 40.6:
Ito and Thurston (1996) Chicago, IL	GAM not used	1.49 (-0.72, 3.74)	—	—	PM ₁₀ 38 (max 128)
Pope et al. (1992) Utah Valley, UT	GAM not used	9.36 (1.91, 17.36)	—	—	PM ₁₀ 47 (11, 297)

**TABLE 9A-1 (cont'd). ESTIMATED TOTAL, CARDIOVASCULAR, AND RESPIRATORY MORTALITY
EFFECT SIZES PER INCREMENTS IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
FROM U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Total Cardiovascular Mortality (cont'd)					
Fairley (1999)	GAM Strict	8.5 (0.6, 17.0)	6.3 (-4.1, 17.9)	5.0 (-13.3, 27.3)	PM ₁₀ 34 (6, 165)
Fairley (2003)	GLM NS	8.9 (1.3, 17.0)	6.7 (-2.5, 16.7)		PM _{2.5} 13 (2, 105)
Santa Clara County, CA					PM _{10-2.5} 11 (0, 45)
Goldberg et al. (2000)	GAM Strict	—	3.48 (-0.16, 7.26)	—	PM _{2.5} 17.6 (4.6, 71.7)
Goldberg and Burnett (2003)	GLM NS				
Montreal, CAN					
Lipfert et al. (2000a)	GAM not used	8.0 (3.7, 12.3)	5.0 (2.4, 7.5)	5.4 (-0.4, 11.2)	PM ₁₀ 32.20 (7.0, 95.0)
Philadelphia, PA					PM _{2.5} 17.28 (-0.6, 72.6)
(7-county area)					PM _{10-2.5} 6.80 (-20.0, 28.3)
Lippmann et al. (2000)	GAM Strict	5.4 (-2.6, 14.0)	2.2 (-3.2, 7.9)	6.7 (-1.0, 15.0)	PM ₁₀ 31 (12, 105)
Ito (2003)	GLM NS	4.9 (-3.0, 13.5)	2.0 (-3.4, 7.7)	6.0 (-1.6, 14.3)	PM _{2.5} 18 (6, 86)
Detroit, MI					PM _{10-2.5} 13 (4, 50) mean (10%, 90%)
Mar et al. (2000)	GAM Strict	9.7 (1.7, 18.3)	18.0 (4.9, 32.6)	6.4 (1.3, 11.7)	PM ₁₀ 46.5 (5, 213)
Mar et al. (2003)	GLM NS	9.5 (0.6, 19.3)	19.1 (3.9, 36.4)	6.2 (0.8, 12.0)	PM _{2.5} 13.0 (0, 42)
Phoenix, AZ					PM _{10-2.5} 33.5 (5, 187)
Moolgavkar (2000a)	GAM Strict 30df	4.5 (1.6, 7.5)	2.6 (0.4, 4.9)	—	PM ₁₀ median 44 (7, 166)
Moolgavkar (2003)	GLM NS 100df	3.9 (0.6, 7.4)	1.7 (-0.8, 4.3)		PM _{2.5} median 22 (4, 86)
Los Angeles, CA					
Moolgavkar (2000a)	GAM Strict 100df	2.0 (0.3, 4.1)	—	—	PM ₁₀ median 35 (3, 365)
Moolgavkar (2003)	GLM NS 100df	2.0 (0.2, 3.7)			
Cook Co., IL					
Ostro et al. (2000)	GAM Strict	5.5 (1.6, 9.5)	9.8 (-5.7, 27.9)	2.9 (0.7, 5.2)	PM ₁₀ 47.4 (3, 417)
Ostro et al. (2003)	GLM NS	5.1 (1.2, 9.1)	10.2 (-5.3, 28.3)	2.7 (0.4, 5.1)	PM _{2.5} 16.8 (5, 48)
Coachella Valley, CA					PM _{10-2.5} 17.9 (0, 149)
Ostro (1995)	GAM not used	—	0.69 (-0.34, 1.74)	—	PM _{2.5} 32.5 (9.3, 190.1) (estimated from visibility)
San Bernadino and Riverside Counties, CA					

**TABLE 9A-1 (cont'd). ESTIMATED TOTAL, CARDIOVASCULAR, AND RESPIRATORY MORTALITY
EFFECT SIZES PER INCREMENTS IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
FROM U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Total Respiratory Mortality:					
Ito and Thurston (1996) Chicago, IL	GAM not used	6.77 (1.97, 11.79)	—	—	PM ₁₀ 38 (max 128)
Pope et al. (1992) Utah Valley, UT	GAM not used	19.78 (3.51, 38.61)	—	—	PM ₁₀ 47 (11, 297)
Fairley (1999) Fairley (2003) Santa Clara County, CA	GAM Strict GLM NS	10.7 (-3.7, 27.2) 10.8 (-3.4, 27.1)	11.7 (-9.8, 38.3) 13.5 (-3.6, 33.7)	(GAM strict) 32.1 (-9.1, 92.2)	PM ₁₀ 34 (6, 165) PM _{2.5} 13 (2, 105) PM _{10-2.5} 11 (0, 45)
Goldberg et al. (2000) Goldberg and Burnett (2003) Montreal, CAN	GAM Strict GLM NS	—	21.6 (13.0, 31.0)	—	PM _{2.5} 17.6 (4.6, 71.7)
Lippmann et al. (2000) Ito (2003) Detroit, MI	GAM Strict GLM NS	7.5 (-10.5, 29.2) 7.9 (-10.2, 29.7)	2.3 (-10.4, 16.7) 3.1 (-9.7, 17.7)	7.0 (-9.5, 26.5) 6.4 (-10.0, 25.7)	PM ₁₀ 31 (12, 105) PM _{2.5} 18 (6, 86) PM _{10-2.5} 13 (4, 50) mean (10%, 90%)
Ostro (1995) San Bernadino and Riverside Counties, CA	GAM not used	—	2.08 (-0.35, 4.51)	—	PM _{2.5} 32.5 (9.3, 190.1) (estimated from visibility)

**TABLE 9A-1 (cont'd). ESTIMATED TOTAL, CARDIOVASCULAR, AND RESPIRATORY MORTALITY
EFFECT SIZES PER INCREMENTS IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
FROM U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
COPD Mortality:					
Schwartz (2003b) 10 U.S. Cities	GAM Strict	7.7 (4.1, 11.5)	—	—	PM ₁₀ mean range 27.1 - 40.6:
Moolgavkar (2000a) Moolgavkar (2003) Cook Co., IL	GAM Strict 30df GLM NS 100df	4.0 (-0.2, 10.1) 4.8 (-0.6, 10.4)	—	—	PM ₁₀ median 35 (3, 365)
Moolgavkar (2000a) Mookgavkar (2003) Los Angeles, CA	GAM Strict 30 df GLM NS 100df	4.4 (-3.2, 12.6) 6.2 (-3.4, 16.7)	1.0 (-5.1, 7.4) 0.5 (-6.8, 8.4)	—	PM ₁₀ median 44 (7, 166) PM _{2.5} 22 (4, 86)

* Both original published studies and recent reanalyses reported in HEI (2003) Special Report for many cited here. Original studies published before 1996 and Schwartz et al. (1996) were assessed in 1996 PM AQCD.

** Where GAM not used in original analysis cited, original results are reported here. Otherwise reanalyses results are reported here if GAM (default) was used in original analysis. GAM strict = GAM with stringent criteria; GLM = general linear model; NS = natural splines; BS = B splines; PS = penalized splines.

*** Mean (minimum, maximum) 24-h PM level in parentheses unless otherwise noted.

TABLE 9A-2. CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5} IN U.S. AND CANADIAN STUDIES

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
CARDIOVASCULAR MORBIDITY					
Total Cardiovascular Hospital Admissions or Medical Visits:					
Samet et al. (2000a,b) 14 U.S. Cities (> 65 years)	strict GAM GLM NS GLM PS	4.95 (3.95-5.95) 4.8 (3.55-6.0) 5.0 (4.0-5.95)	—	—	PM ₁₀ means 24.4-45.3
Zanobetti and Schwartz (2003)	GLM PS	5.7 (4.2-7.30)			
Linn et al. (2000) Los Angeles, CA (> 29 years)	GAM not used	3.25 (2.04, 4.47)	—	—	PM ₁₀ 45.5 (5, 132)
Metzger et al. (2004) Atlanta, GA Period 1 (1993 - 2000) Period 2 (1998 - 2000)	GAM not used	5.1 (-7.9, 19.9) 2.3 (-0.4, 5.0)	8.2 (2.6, 14.7)	3.0 (-3.7, 10.3)	PM ₁₀ median 26.3 PM _{2.5} median 17.8 PM _{10-2.5} median 9.1
Moolgavkar (2000b) Moolgavkar (2003) Cook Co., IL (> 65 years)	strict GAM 100df GLM NS100df	4.05 (2.9-5.2) 4.25 (3.0-5.5)	—	—	PM ₁₀ median 35 (3, 365)
Moolgavkar (2000b) Moolgavkar (2003) Los Angeles, CA (> 65 years)	GAM30df GAM100df GLM NS100df	3.35 (1.2-5.5) 2.7 (0.6-4.8) 2.75 (0.1-5.4)	3.95 (2.2-5.7) 2.9 (1.2-4.6) 3.15 (1.1-5.2)	—	PM ₁₀ median 44 (7, 166) PM _{2.5} median 22 (4, 86)
Morris and Naumova (1998) Chicago, IL (> 65 years)	GAM not used	3.92 (1.02, 6.90)	—	—	PM ₁₀ 41 (6, 117)
Stieb et al. (2000) St. John, CAN (all ages)	GAM not used	32.5 (10.2, 59.3)	15.11 (-0.25, 32.8)	—	Summer 93 PM ₁₀ 14.0 (max 70.3) PM _{2.5} 8.5 (max 53.2)

**TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT
SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
IN U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Total Cardiovascular Hospital Admissions: (cont'd)					
Burnett et al. (1997) Toronto, CAN (all ages)	GAM not used	12.07 (1.43, 23.81)	7.18 (-0.61, 15.60)	20.46 (8.24, 34.06)	PM ₁₀ 28.4 (4, 102) PM _{2.5} 16.8 (1, 66) PM _{10-2.5} 11.6 (1, 56)
Ischemic Heart Disease Hospital Admissions:					
Schwartz and Morris (1995) Detroit (> 65 years)	GAM not used	2.8 (0.7, 5.0)	—	—	PM ₁₀ (98 (22, 82) (10%, 90%)
Lippmann et al. (2000) Detroit, MI (> 65 years) Ito 2003	Strict GAM GLM NS	8.0 (-0.3-17.1) 6.2 (-2.0-15.0)	3.65% (-2.05-9.7) 3.0% (-2.7-9.0)	10.2% (2.4-18.6) 8.1% (0.4-16.4)	PM ₁₀ 31 (max 105) PM _{2.5} 18 (6, 86) PM _{10-2.5} 13 (4, 50)
Metzger et al. (2004) Atlanta, GA Period 1 (1993 - 2000) Period 2 (1998 - 2000)	GAM not used	2.8 (-1.9, 7.7)	5.8 (-4.1, 16.9)	-1.5 (-13.0, 11.6)	PM ₁₀ median 26.3 PM _{2.5} median 17.8 PM _{10-2.5} median 9.1
Dysrhythmias Hospital Admissions:					
Lippmann et al. (2000) Detroit, MI (> 65 years) Ito (2003)	Strict GAM GLM NS	2.8 (-10.9-18.7) 2.0 (-11.7-17.7)	3.2 (-6.6-14.0) 2.6 (-7.1-13.3)	0.1% (-12.4-14.4) 0.0% (-12.5-14.3)	PM ₁₀ 31 (max 105) PM _{2.5} 18 (6, 86) PM _{10-2.5} 13 (4, 50)
Metzger et al. (2004) Atlanta, GA Period 1 (1993 - 2000) Period 2 (1998 - 2000)	GAM not used	2.0 (-2.9, 7.2)	3.8 (-5.8, 14.4)	5.3 (-6.3, 18.5)	PM ₁₀ median 26.3 PM _{2.5} median 17.8 PM _{10-2.5} median 9.1
Heart Failure Hospital Admissions:					
Linn et al. (2000) Los Angeles, CA (> 29 years)	GAM not used	2.02 (-0.94, 5.06)	—	—	PM ₁₀ 45.5 (5, 132)

**TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT
SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
IN U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Heart Failure Hospital Admissions: (cont'd)					
Lippmann et al. (2000)	Strict GAM	9.2 (-0.3-19.6)	8.0 (1.4-15.0)	4.4% (-4.0-13.5)	PM ₁₀ 31 (max 105)
Ito (2003) Detroit, MI (> 65 years)	GLM NS	8.4 (-1.0-18.7)	6.8 (0.3-13.8)	4.9% (-3.55-14.1)	PM _{2.5} 18 (6, 86) PM _{10-2.5} 13 (4, 50)
Metzger et al. (2004) Atlanta, GA Period 1 (1993 - 2000) Period 2 (1998 - 2000)	GAM not used	2.0 (7.7, 4.1)	14.3 (1.7, 28.6)	5.1 (-8.7, 21.0)	PM ₁₀ median 26.3 PM _{2.5} median 17.8 PM _{10-2.5} median 9.1
Schwartz and Morris (1995) Detroit (> 65 years)	GAM not used	5.0 (1.9, 8.3)			
Myocardial Infarction Hospital Admissions:					
Linn et al. (2000) Los Angeles, CA (> 29 years)	GAM not used	3.04 (0.06, 6.12)	—	—	PM ₁₀ 45.5 (5, 132)
Cardiac arrhythmia Hospital Admissions:					
Linn et al. (2000) Los Angeles, CA (> 29 years)	GAM not used	1.01 (-1.93, 4.02)	—	—	PM ₁₀ 45.5 (5, 132)
Cerebrovascular Hospital Admissions:					
Linn et al. (2000) Los Angeles, CA (> 29 years)	GAM not used	0.30 (-2.13, 2.79)	—	—	PM ₁₀ 45.5 (5, 132)
Metzger et al. (2004) Atlanta, GA	GAM not used	5.1 (-0.4, 10.9)	13.0 (2.1, 25.0)	5.6 (-6.8, 19.6)	PM ₁₀ median 26.3 PM _{2.5} median 17.8 PM _{10-2.5} median 9.1

**TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT
SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
IN U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
RESPIRATORY MORBIDITY					
Stroke Hospital Admissions:					
Linn et al. (2000) Los Angeles, CA (> 29 years)	GAM not used	6.72 (3.64, 9.90)	—	—	PM ₁₀ 45.5 (5, 132)
Lippmann et al. (2000) Ito (2003) Detroit, MI (> 65 years)	GLM NS	4.4 (-5.8, 15.7)	1.0 (-6.1, 8.5)	5.6 (-4.0, 16.2)	PM ₁₀ 31 (max 105) PM _{2.5} 18 (6, 86) PM _{10-2.5} 13 (4, 50)
Total Respiratory Hospital Admissions or Medical Visits:					
Thurston et al. (1994) Toronto, Canada	GAM not used	23.26 (2.03, 44.49)	15.00 (1.97, 28.03)	22.25 (-9.53, 54.03)	PM ₁₀ 29.5-38.8 (max 96.0) PM _{2.5} 15.8-22.3 (max 66.0) PM _{10-2.5} 12.7-16.5 (max 33.0)
Linn et al. (2000) Los Angeles, CA (> 29 years)	GAM not used	2.89 (1.09, 4.72)	—	—	PM ₁₀ 45.5 (5, 132)
Schwartz et al. (1996) Cleveland, OH (> 65 years)	GAM not used	5.8 (0.5, 11.4)	—	—	PM ₁₀ 43
Lumley and Heagerty (1999) King County, WA (all ages)	GAM not used	—	5.91 (1.10, 10.97)	—	PM ₁ NR
Burnett et al. (1997) Toronto, CAN (all ages)	GAM not used	10.93 (4.53, 17.72)	8.61 (3.39, 14.08)	12.71 (5.33, 20.74)	PM ₁₀ 28.1 (4, 102) PM _{2.5} 16.8 (1, 66) PM _{10-2.5} 11.6 (1, 56)
Delfino et al. (1997) Montreal, CAN (> 64 years)	GAM not used	36.62 (10.02, 63.21)	23.88 (4.94, 42.83)	—	summer 93 PM ₁₀ 21.7 (max 51) PM _{2.5} 12.2 (max 31)

**TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT
SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
IN U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Total Respiratory Hospital Admissions: (cont'd)					
Delfino et al. (1998) Montreal, CAN (> 64 years)	GAM not used	—	13.17 (-0.22, 26.57)	—	PM _{2.5} 18.6 (SD 9.3)
Stieb et al. (2000) St. John, CAN (all ages)	GAM not used	8.8 (1.8, 16.4)	5.69 (0.61, 11.03)	—	summer 93 PM ₁₀ 14.0 (max 70.3) PM _{2.5} 8.5 (max 53.2)
Pneumonia Hospital Admissions:					
Schwartz, 1995 Detroit (> 65 years)	GAM not used	5.9 (1.9, 10.0)	—	—	PM ₁₀ 48 (22, 82) mean (10%, 90%)
Samet et al. (2000a,b) 14 U.S. Cities (> 65 years)	Strict GAM GLM NS GLM PS	8.8 (5.9, 11.8) 2.9 (0.2, 5.6) 6.3 (2.5, 10.3)	—	—	PM ₁₀ means 24.4-45.3
Zanobetti and Schwartz (2003)	(dist. lag)	4.1 (0.7, 7.5)			
Lippmann et al. (2000) Detroit, MI (> 65 years) Ito (2003)	Strict GAM GLM NS	18.1 (5.3, 32.5) 18.6 (5.6, 33.1)	10.5 (1.8, 19.8) 10.1 (1.5, 19.5)	9.9 (-0.1, 22.0) 11.2 (-0.02, 23.6)	PM ₁₀ 31 (max 105) PM _{2.5} 18 (6, 86) PM _{10-2.5} 13 (4, 50)
COPD Hospital Admissions:					
Schwartz, 1995 Detroit (> 65 years)	GAM not used	10.6 (4.4, 17.2)	—	—	PM ₁₀ 48 (22, 82) mean (10, 90)
Samet et al. (2000a,b) 14 U.S. Cities (> 65 years)	Strict GAM GLM NS GLM PS	8.8 (4.8, 13.0) 6.8 (2.8, 10.8) 8.0 (4.3, 11.9)	—	—	PM ₁₀ means 24.4-45.3
Zanobetti and Schwartz (2003)	(dist. lag)	13.4 (6.2, 21.0)			
Linn et al. (2000) Los Angeles, CA (> 29 years)	GAM not used	1.5 (-0.5, 3.5)	—	—	PM ₁₀ 45.5 (5, 132)

**TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT
SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
IN U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
COPD Hospital Admissions: (cont'd)					
Tolbert et al. (2000a) Atlanta, GA (all ages)	GAM not used	-3.5 (33.0, -29.9)	12.44 (-7.89, 37.24)	-23.03 (-50.69, 20.15)	PM ₁₀ 29.1 (SD 12.0) PM _{2.5} 19.4 (SD 9.35) PM _{10-2.5} 9.39 (SD 4.52)
Lippmann et al. (2000) Detroit, MI (> 65 years)	Strict GAM GLM NS	6.5 (-7.8, 23.0) 4.6 (-9.4, 20.8)	3.0(-6.9, 13.9) 0.3(-9.3, 10.9)	8.7 (-4.8, 24.0) 10.8 (-3.1, 26.5)	PM ₁₀ 31 (max 105) PM _{2.5} 18 (6, 86) PM _{10-2.5} 13 (4, 50)
Ito (2003)					
Moolgavkar (2000c) Cook Co., IL (> 65 years) Moolgavkar 2003	Strict GAM 100df	3.24 (.031, 6.24)	—	—	PM ₁₀ median 35 (3, 365)
Moolgavkar (2000c) Los Angeles, CA (> 65 years) Moolgavkar 2003	Strict GAM: 100df GLM NS: 100df	5.52 (2.53-8.59) 5.00 (1.22, 8.91)	2.87 (0.53, 5.27) 2.59 (-0.29, 5.56)		PM ₁₀ median 44 (7, 166) PM _{2.5} median 22 (4.86)
Asthma Hospital Admissions or Medical Visits:					
Choudbury et al. (1997) Anchorage, AK Medical Visits (all ages)	GAM not used	20.9 (11.8, 30.8)	—	—	PM ₁₀ 42.5 (1, 565)
Jacobs et al. (1997) Butte County, CA (all ages)	GAM not used	6.11 (p > 0.05)	—	—	PM ₁₀ 34.3 (6.6, 636)
Linn et al. (2000) Los Angeles, CA (> 29 years)	GAM not used	1.5 (-2.4, 5.6)	—	—	PM ₁₀ 45.5 (5, 132)
Lipsett et al. (1997) Santa Clara Co., CA (all ages)	GAM not used	34.7 (16, 56.5) (min. temp. 20° F) 9.1 (2.7, 15.9) (min. temp. 40° F)	—	—	PM ₁₀ 61.2 (9, 165)

TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5} IN U.S. AND CANADIAN STUDIES

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Asthma Hospital Admissions or Medical Visits: (cont'd)					
Nauenberg and Basu (1999) Los Angeles, CA (all ages)	GAM not used	20.0 (5.3, 35)	—	—	44.8 (SE 17.23)
Tolbert et al. (2000b) Atlanta, GA (< 17 years)	GAM not used	13.2 (1.2, 26.7)	—	—	PM ₁₀ 38.9 (9, 105)
Tolbert et al. (2000a) Atlanta, GA (all ages)	GAM not used	18.8 (-8.7, 54.4)	2.3 (-14.8, 22.7)	21.1 (-18.2, 79.3)	PM ₁₀ 29.1 (SD 12.0) PM _{2.5} 19.4 (SD 9.35) PM _{10-2.5} 9.39 (SD 4.52)
Sheppard et al. (1999) Seattle, WA (< 65 years)	Strict GAM GLM NS	10.9 (2.8, 19.6) 8.1 (0.1, 16.7)	8.7 (3.2, 14.4) 6.5 (1.1, 12.0)	5.5 (0, 14.0) 5.5 (-2.7, 11.1)	PM ₁₀ 31.5 (90% 55) PM _{2.5} 16.7 (90% 32) PM _{10-2.5} 16.2 (90% 29)
Respiratory Symptoms		Odds Ratio (95% CI) for 50 µg/m ³ increase in PM ₁₀	Odds Ratio (95% CI) for 25 µg/m ³ increase in PM _{2.5}	Odds Ratio (95% CI) for 25 µg/m ³ increase in PM _{10-2.5}	PM _{10-2.5} Mean (Range) Levels Reported**
Schwartz et al. (1994) 6 U.S. cities (children, cough)	GAM not used	1.39 (1.05, 1.85)	1.24 (1.00, 1.54)	—	PM ₁₀ median 30.0 (max 117) PM _{2.5} median 18.0 (max 86)
Schwartz et al. (1994) 6 U.S. cities (children, lower respiratory symptoms)	GAM not used	2.03 (1.36, 3.04)	1.58 (1.18, 2.10)	—	PM ₁₀ median 30.0 (max 117) PM _{2.5} median 18.0 (max 86)
Neas et al. (1995) Uniontown, PA (children, cough)	GAM not used	—	2.45 (1.29, 4.64)	—	PM _{2.5} 24.5 (max 88.1)
Ostro et al. (1991) Denver, CO (adults, cough)	GAM not used	1.09 (0.57, 2.10)	—	—	PM ₁₀ 22 (0.5, 73)

**TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT
SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
IN U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Respiratory Symptoms (cont'd)					
Pope et al. (1991) Utah Valley, UT (lower respiratory symptoms, schoolchildren)	GAM not used	1.28 (1.06, 1.56)	—	—	PM ₁₀ 44 (11, 195)
Pope et al. (1991) Utah Valley, UT (lower respiratory symptoms, asthmatic patients)	GAM not used	1.01 (0.81, 1.27)	—	—	PM ₁₀ 44 (11, 195)
Neas et al. (1996) State College, PA (children, cough)	GAM not used	NR	1.48 (1.17, 1.88) (1-d)	—	PM ₁₀ 31.9 (max 82.7) PM _{2.5} 23.5 (max 85.8)
Neas et al. (1996) State College, PA (children, wheeze)	GAM not used	NR	1.59 (0.93, 2.70) (1-d)	—	PM ₁₀ 31.9 (max 82.7) PM _{2.5} 23.5 (max 85.8)
Neas et al. (1996) State College, PA (children, cold)	GAM not used	NR	1.61 (1.21, 2.17) (0-d)	—	PM ₁₀ 31.9 (max 82.7) PM _{2.5} 23.5 (max 85.8)
Ostro et al. (1995) Los Angeles, CA (children, asthma episode)	GAM not used	1.05 (0.64, 1.73)	—	—	PM ₁₀ 55.87 (19.63, 101.42)
Ostro et al. (1995) Los Angeles, CA (children, shortness of breath)	GAM not used	1.51 (1.04, 2.17)	—	—	PM ₁₀ 55.87 (19.63, 101.42)
Schwartz and Neas (2000) Six Cities reanalysis (children, cough)	GAM not used	—	1.28 (0.98, 1.67)	1.77 (1.23, 2.54)	PM _{2.5} (same as Six Cities) PM _{10-2.5} NR

**TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT
SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
IN U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Respiratory Symptoms (cont'd)					
Schwartz and Neas (2000) Six Cities reanalysis (children, lower respiratory symptoms)	GAM not used	—	1.61 (1.20, 2.16)	1.51 (0.66, 3.43)	PM _{2.5} (same as Six Cities) PM _{10-2.5} NR
Vedal et al. (1998) Port Alberni, CAN (children, cough)	GAM not used	1.40 (1.14, 1.73)	—	—	PM ₁₀ median 22.1 (0.2, 159.0) (north site)
Vedal et al. (1998) Port Alberni, CAN (children, phlegm)	GAM not used	1.40 (1.03, 1.90)	—	—	PM ₁₀ median 22.1 (0.2, 159.0) (north site)
Vedal et al. (1998) Port Alberni, CAN (children, nose symptoms)	GAM not used	1.22 (1.00, 1.47)	—	—	PM ₁₀ median 22.1 (0.2, 159.0) (north site)
Vedal et al. (1998) Port Alberni, CAN (children, sore throat)	GAM not used	1.34 (1.06, 1.69)	—	—	PM ₁₀ median 22.1 (0.2, 159.0) (north site)
Vedal et al. (1998) Port Alberni, CAN (children, wheeze)	GAM not used	1.16 (0.82, 1.63)	—	—	PM ₁₀ median 22.1 (0.2, 159.0) (north site)
Vedal et al. (1998) Port Alberni, CAN (children, chest tightness)	GAM not used	1.34 (0.86, 2.09)	—	—	PM ₁₀ median 22.1 (0.2, 159.0) (north site)
Vedal et al. (1998) Port Alberni, CAN (children, dyspnea)	GAM not used	1.05 (0.74, 1.49)	—	—	PM ₁₀ median 22.1 (0.2, 159.0) (north site)

**TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT
SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
IN U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Respiratory Symptoms (cont'd)					
Vedal et al. (1998) Port Alberni, CAN (children, any symptom)	GAM not used	1.16 (1.00, 1.34)	—	—	PM ₁₀ median 22.1 (0.2, 159.0) (north site)
Lung Function Changes					
		Lung Function change (L/min) (95% CI) for 50 µg/m ³ increase in PM ₁₀	Lung Function change (L/min) (95% CI) for 25 µg/m ³ increase in PM _{2.5}	Lung Function change (L/min) (95% CI) for 25 µg/m ³ increase in PM _{10-2.5}	PM _{10-2.5} Mean (Range) Levels Reported**
Neas et al. (1995) Uniontown, PA (children)	GAM not used	—	-2.58 (-5.33, +0.35)	—	PM _{2.5} 24.5 (max 88.1)
Thurston et al. (1997) Connecticut summer camp (children)	GAM not used	—	PEFR -5.4 (-12.3, 1.5) (15 µg/m ³ SO ₄ ⁼)	—	SO ₄ ⁼ 7.0 (1.1, 26.7)
Naeher et al. (1999) Southwest VA (adult women)	GAM not used	am PEFR -3.65 (-6.79, -0.51) pm PEFR -1.8 (-5.03, 1.43)	am PEFR -1.83 (-3.44, -0.21) pm PEFR -1.05 (-2.77, 0.67)	am PEFR -6.33 (-12.50, -0.15) pm PEFR -2.4 (-8.48, 3.68)	PM ₁₀ 27.07 (4.89, 69.07) PM _{2.5} 21.62 (3.48, 59.65) PM _{10-2.5} 5.72 (0.00, 19.78)
Neas et al. (1996) State College, PA (children)	GAM not used	—	pm PEFR -0.64 (-1.73, 0.44)	—	PM _{2.5} 23.5 (max 85.8)
Neas et al. (1999) Philadelphia, PA (children)	GAM not used	am PEFR -8.17 (-14.81, -1.56) pm PEFR -1.44 (-7.33, 4.44)	am PEFR -3.29 (-6.64, 0.07) pm PEFR -0.91 (-4.04, 2.21)	am PEFR -4.31 (-11.44, 2.75) pm PEFR 1.88 (-4.75, 8.44)	PM _{2.5} 22.2 (IQR 16.2) PM _{10-2.5} 9.5 (IQR 5.1)
Schwartz and Neas (2000) Uniontown, PA (reanalysis) (children)	GAM not used	—	pm PEFR -1.52, (-2.80, -0.24)	pm PEFR +1.73 (-2.2, 5.67)	PM _{2.5} 24.5 (max 88.1) PM _{10-2.5} NR

**TABLE 9A-2 (cont'd). CARDIOVASCULAR AND RESPIRATORY-RELATED MORBIDITY EFFECT
SIZE ESTIMATES PER INCREMENT IN 24-h CONCENTRATIONS OF PM₁₀, PM_{2.5}, AND PM_{10-2.5}
IN U.S. AND CANADIAN STUDIES**

Original study* Reanalysis study Study location	Analysis Comments**	% increase (95% CI) per 50 µg/m ³ PM ₁₀ Increase	% increase (95% CI) per 25 µg/m ³ PM _{2.5} Increase	% increase (95% CI) per 25 µg/m ³ PM _{10-2.5} Increase	PM ₁₀ , PM _{2.5} and PM _{10-2.5} Mean (Range) Levels Reported***
Lung Function Changes (cont'd)		Lung Function change (L/min) (95% CI) for 50 ug/m ³ increase in PM ₁₀	Lung Function change (L/min) (95% CI) for 25 ug/m ³ increase in PM _{2.5}	Lung Function change (L/min) (95% CI) for 25 ug/m ³ increase in PM _{10-2.5}	PM _{10-2.5} Mean (Range) Levels Reported**
Schwartz and Neas (2000) State College PA (reanalysis) (children)	GAM not used	—	pm PEFr -0.93 (-1.88, 0.01)	pm PEFr -0.28 (-3.45, 2.87)	PM _{2.5} 23.5 (max 85.8) PM _{10-2.5} NR
Vedal et al. (1998) Port Alberni, CAN (children)	GAM not used	PEF -1.35 (-2.7, -0.05)	—	—	PM ₁₀ median 22.1 (0.2, 159.0) (north site)

*Both original published studies and recent reanalyses reported in HEI (2003) Special Report for many cited here. Original studies published before 1996 and Schwartz et al. (1996) were assessed in 1996 PM AQCD.

**Where GAM not used in original analysis cited, original results reported here. Otherwise reanalyses results reported here if GAM (default) used in original analysis. GAM strict = GAM with stringent criteria. GLM = general linear model; NS = natural splines; BS = B splines; PS = penalized splines.

***Mean (minimum, maximum) 24-h PM level in parentheses unless otherwise noted.

TABLE 9-A3. EFFECT ESTIMATES PER INCREMENTS* IN LONG-TERM MEAN LEVELS OF FINE AND COARSE FRACTION PARTICLE INDICATORS FROM U.S. AND CANADIAN STUDIES

Type of Health Effect Study and Location	Indicator (PM Increment)	Change in Health Indicator per Increment in PM	Range of City PM Levels ** Means ($\mu\text{g}/\text{m}^3$)
Increased Total Mortality in Adults			
Relative Risk (95% CI)			
Six City ^A	PM _{15/10} (20 $\mu\text{g}/\text{m}^3$)	1.18 (1.06-1.32)	18-47
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.13 (1.04-1.23)	11-30
	SO ₄ ⁻ (15 $\mu\text{g}/\text{m}^3$)	1.46 (1.16-2.16)	38119
Six City Reanalysis ^C	PM _{15/10} (20 $\mu\text{g}/\text{m}^3$)	1.19 (1.06-1.34)	18.2-46.5
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.14 (1.05-1.23)	11.0-29.6
ACS Study ^B (151 U.S. SMSA)	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.07 (1.04-1.10)	9-34
	SO ₄ ⁻ (15 $\mu\text{g}/\text{m}^3$)	1.10 (1.06-1.16)	4-24
ACS Study Reanalysis ^C	PM _{15/10} (20 $\mu\text{g}/\text{m}^3$) (dichot)	1.04 (1.01, 1.07)	58.7 (34-101)
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.07 (1.04-1.10)	9.0-33.4
	PM _{15-2.5} (10 $\mu\text{g}/\text{m}^3$)	1.00 (0.99, 1.02)	9-42
ACS Study Extended Analyses ^D	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.04 (1.01-1.08)	21.1 (SD=4.6)
Southern California ^E	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	1.091 (0.985-1.212; males)	51 (\pm 17)
	PM ₁₀ (cutoff = 30 days/year > 100 $\mu\text{g}/\text{m}^3$)	1.082 (1.008-1.162; males)	
	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	0.950 (0.873-1.033; females)	51 (\pm 17)
	PM ₁₀ (cutoff = 30 days/year > 100 $\mu\text{g}/\text{m}^3$)	0.958 (0.899-1.021; females)	
Southern California ^F	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.09 (0.98, 1.21) (males)	32 (17, 45)
	PM _{10-2.5} (10 $\mu\text{g}/\text{m}^3$)	1.05 (0.92, 1.21) (males)	27 (4, 44)
Veterans Cohort ^G	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$) (mortality period 1976-96)	1.003 (NS) ****	5.6-42.3
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$) (mortality period 1982-88)	0.90 (SS) ****	5.6-42.3
	PM _{15-2.5} (10 $\mu\text{g}/\text{m}^3$) (mortality period 1976-96)	1.007 (NS)****	3.6-64.2
	PM _{15-2.5} (10 $\mu\text{g}/\text{m}^3$) (mortality period 1982-88)	0.98 (NS)****	3.6-64.3
	PM ₁₅ (10 $\mu\text{g}/\text{m}^3$) (mortality period 1976-96)	1.007 (NS)****	9.74-101.7
	PM ₁₅ (10 $\mu\text{g}/\text{m}^3$) (mortality period 1982-88)	0.92 (NS)****	9.74-101.7

TABLE 9A-3 (cont'd). EFFECT ESTIMATES PER INCREMENTS* IN LONG-TERM MEAN LEVELS OF FINE AND COARSE FRACTION PARTICLE INDICATORS FROM U.S. AND CANADIAN STUDIES

Type of Health Effect Study and Location	Indicator (PM Increment)	Change in Health Indicator per Increment in PM	Range of City PM Levels ** Means ($\mu\text{g}/\text{m}^3$)
Increased Cardiopulmonary Mortality in Adults		Relative Risk (95% CI)	
Six City ^A	PM _{15/10} (20 $\mu\text{g}/\text{m}^3$)	***	18-47
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.18 (1.06, 1.32)	37954
Six City Reanalysis ^C	PM _{15/10} (20 $\mu\text{g}/\text{m}^3$)	1.20 (1.29, 1.41)	18.2-46.5
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.19 (1.07, 1.33)	11.0-29.6
ACS Study ^B (151 U.S. SMSA)	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.12 (1.07-1.17)	9-34
ACS Study Reanalysis ^C	PM _{15/10} (20 $\mu\text{g}/\text{m}^3$) (dichot)	1.07 (1.03, 1.12)	58.7 (34-101)
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.12 (1.07-1.17)	9.0-33.4
	PM _{15-2.5} (10 $\mu\text{g}/\text{m}^3$)	1.00 (0.98, 1.03)	9-42
	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	1.01 (0.92, 1.10)	51 (\pm 17)
ACS Study Extended Analyses ^D	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$) (1979-83)	1.06 (1.02, 1.10)	21 (10, 30)***
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$) (1999-00)	1.08 (1.02, 1.14)	14 (5, 20)(***
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$) (average)	1.09 (1.03, 1.16)	18 (\pm 4)
Southern California ^E	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	1.01 (0.92, 1.10)	51 (0, 84)
Southern California ^F	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.23 (0.97, 1.55) (males)	32 (17, 45)
	PM _{10-2.5} (10 $\mu\text{g}/\text{m}^3$)	1.20 (0.87, 1.64) (males)	27 (4, 44)
Increased Lung Cancer Mortality in Adults			
Six City ^A	PM _{15/10} (20 $\mu\text{g}/\text{m}^3$)	****	NR (18, 47)
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.18 (0.89, 1.57)	NR (11, 30)
Six City Reanalysis ^C	PM _{15/10} (20 $\mu\text{g}/\text{m}^3$)	1.14 (0.75, 1.74)	NR (18, 47)
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.21 (0.92, 1.60)	NR (11, 30)
ACS Study ^B	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.01 (0.91, 1.12)	18** (9, 34)
ACS Study Reanalysis ^C	PM _{15/10} (10 $\mu\text{g}/\text{m}^3$) (dichot)	1.01 (0.91, 1.11)	59 (34, 101)
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.01 (0.91, 1.11)	20 (10, 38)
	PM _{15-2.5} (10 $\mu\text{g}/\text{m}^3$)	0.99 (0.93, 1.05)	7.1 (9, 42)
ACS Study Extended Analyses ^D	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$) (1979-83)	1.08 (1.01, 1.16)	21 (10, 30)***
	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$) (1999-00)	1.13 (1.04, 1.22)	14 (5, 20)***
	PM _{2.5} (average)	1.14 (1.05, 1.24)	18 (\pm 4)
Southern California ^E	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	1.81 (1.14, 2.86) (males)	51 (0, 84)
Southern California ^F	PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.39 (0.79, 2.50) (males)	32 (17, 45)
		1.26 (0.62, 2.55) (males)	27 (4, 44)

TABLE 9A-3 (cont'd). EFFECT ESTIMATES PER INCREMENTS* IN LONG-TERM MEAN LEVELS OF FINE AND COARSE FRACTION PARTICLE INDICATORS FROM U.S. AND CANADIAN STUDIES

Type of Health Effect Study and Location	Indicator (PM Increment)	Change in Health Indicator per Increment in PM	Range of City PM Levels ** Means ($\mu\text{g}/\text{m}^3$)
Increased Bronchitis in Children		Odds Ratio (95% CI)	
Six City ^H	PM _{15/10} (50 $\mu\text{g}/\text{m}^3$)	3.26 (1.13, 10.28)	20-59
24 City ^I	SO ₄ ⁻ (15 $\mu\text{g}/\text{m}^3$)	3.02 (1.28, 7.03)	18.1-67.3
24 City ^I	PM _{2.1} (10 $\mu\text{g}/\text{m}^3$)	1.31 (0.94, 1.84)	9.1-17.3
24 City ^I	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	1.60 (0.92, 2.78)	22.0-28.6
Southern California ^J	SO ₄ ⁻ (15 $\mu\text{g}/\text{m}^3$)	1.39 (0.99, 1.92)	—
12 Southern California communities ^K (all children)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	0.95 (0.79, 1.15)	28.0-84.9
12 Southern California communities ^L (children with asthma)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$) PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.4 (1.1, 1.8) 1.3 (0.9, 1.7)	13.0-70.7 6.7-31.5
Increased Cough in Children		Odds Ratio (95% CI)	
12 Southern California communities ^K (all children)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	1.05 (0.94, 1.16)	28.0-84.9
12 Southern California communities ^L (children with asthma)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$) PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	1.1 (0.7, 1.8) 1.2 (0.8, 1.8)	13.0-70.7 6.7-31.5
Increased Airway Obstruction in Adults		Odds Ratio (95% CI)	
Southern California ^M	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	1.19 (0.84, 1.68)	NR
Decreased Lung Function in Children		Odds Ratio (95% CI)	
Six City ^H	PM _{15/10} (50 $\mu\text{g}/\text{m}^3$)	NS Changes	20-59
24 City ^N	PM _{2.1} (10 $\mu\text{g}/\text{m}^3$)	-2.15% (-3.34, -0.95) FVC	18.1-67.3
24 City ^N	SO ₄ ⁻ (7 $\mu\text{g}/\text{m}^3$)	-3.06% (-4.50, -1.60) FVC	9.1-17.3
24 City ^N	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	-2.80% (-4.97, -0.59) FVC	22.0-28.6
12 Southern California communities ^O (all children)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	-19.9 (-37.8, -2.6) FVC	28.0-84.9

TABLE 9A-3 (cont'd). EFFECT ESTIMATES PER INCREMENTS* IN LONG-TERM MEAN LEVELS OF FINE AND COARSE FRACTION PARTICLE INDICATORS FROM U.S. AND CANADIAN STUDIES

Type of Health Effect Study and Location	Indicator (PM Increment)	Change in Health Indicator per Increment in PM	Range of City PM Levels ** Means ($\mu\text{g}/\text{m}^3$)
Decreased Lung Function in Children (cont'd)			
Odds Ratio (95% CI)			
12 Southern California communities ^O (all children)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	-25.6 (-47.1, -5.1) MMEF	28.0-84.9
12 Southern California communities ^P (4 th grade cohort)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$) PM _{2.5} (10 $\mu\text{g}/\text{m}^3$) PM _{10-2.5} (10 $\mu\text{g}/\text{m}^3$)	-0.23 (-0.44, -0.01) FVC % growth -0.18 (-0.36, 0.0) FVC % growth -0.22 (-0.47, 0.02) FVC % growth	NR
12 Southern California communities ^P (4 th grade cohort)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$) PM _{2.5} (10 $\mu\text{g}/\text{m}^3$) PM _{10-2.5} (10 $\mu\text{g}/\text{m}^3$)	-0.51 (-0.94, -0.08) MMEF % growth -0.4 (-0.75, -0.04) MMEF % growth -0.54 (-1.0, -0.06) MMEF % growth	NR
12 Southern California communities ^Q (second 4 th grade cohort)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$) PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	-0.23 (-0.46, -0.0) FVC % growth -0.19 (-0.39, 0.01) FVC % growth	NR
12 Southern California communities ^Q (second 4 th grade cohort)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$) PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	-0.55 (-1.0, -0.08) MMEF % growth -0.42 (-0.85, 0.01) MMEF % growth	NR
12 Southern California communities ^Q (second 4 th grade cohort)	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$) PM _{2.5} (10 $\mu\text{g}/\text{m}^3$)	-0.49 (-0.84, -0.14) PEFR % growth -0.37 (-0.70, -0.04) PEFR % growth	NR
Southern California ^R	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	-3.6 (-18, 11) FVC growth	15.0-66.2
Southern California ^R	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	-33 (-64, -2.2) MMEF growth	15.0-66.2
Southern California ^R	PM ₁₀ (20 $\mu\text{g}/\text{m}^3$)	-70 (-120, -20) PEFR growth	15.0-66.2

TABLE 9A-3 (cont'd). EFFECT ESTIMATES PER INCREMENTS* IN LONG-TERM MEAN LEVELS OF FINE AND COARSE FRACTION PARTICLE INDICATORS FROM U.S. AND CANADIAN STUDIES

Type of Health Effect Study and Location	Indicator (PM Increment)	Change in Health Indicator per Increment in PM	Range of City PM Levels ** Means ($\mu\text{g}/\text{m}^3$)
Lung Function Changes in Adults			
Southern California ^S (% predicted FEV ₁ , females)	PM ₁₀ (cutoff of 54.2 days/year > 100 $\mu\text{g}/\text{m}^3$)	+0.9 % (-0.8, 2.5) FEV ₁	52.7 (21.3, 80.6)
Southern California ^S (% predicted FEV ₁ , males)	PM ₁₀ (cutoff of 54.2 days/year > 100 $\mu\text{g}/\text{m}^3$)	+0.3 % (-2.2, 2.8) FEV ₁	54.1 (20.0, 80.6)
Southern California ^S (% predicted FEV ₁ , males whose parents had asthma, bronchitis, emphysema)	PM ₁₀ (cutoff of 54.2 days/year > 100 $\mu\text{g}/\text{m}^3$)	-7.2 % (-11.5, -2.7) FEV ₁	54.1 (20.0, 80.6)
Southern California ^S (% predicted FEV ₁ , males)	SO ₄ ⁼ (1.6 $\mu\text{g}/\text{m}^3$)	-1.5 % (-2.9, -0.1) FEV ₁	7.3 (2.0, 10.1)

* Results calculated using PM increment between the high and low levels in cities, or other PM increments given in parentheses; NS Changes = No significant changes.

** Range of mean PM levels given unless, as indicated, studies reported overall study mean (min, max), or mean (\pm SD); NR=not reported.

*** Results only for smoking category subgroups.

**** NS = not significant. SS = statistically significant; as reported by the author.

References:

^A Dockery et al. (1993)	^K Peters et al. (1999b)
^B Pope et al. (1995)	^L McConnell et al. (1999)
^C Krewski et al. (2000)	^M Berglund et al. (1999)
^D Pope et al. (2002)	^N Raizenne et al. (1996)
^E Abbey et al. (1999)	^O Peters et al. (1999a)
^F McDonnell et al. (2000)	^P Gauderman et al. (2000)
^G Lipfert et al. (2000b)	^Q Gauderman et al. (2002)
^H Dockery et al. (1989)	^R Avol et al. (2001)
^I Dockery et al. (1996)	^S Abbey et al. (1998)
^J Abbey et al. (1995a,b,c)	