1	Emissions from Forest and Grass Burns:
2	Comparison of Aerial and Ground Field
3	Measurements with Laboratory Burns
4	Johanna Aurell ^{a,1} , Brian K. Gullett ^{b,2} , Dennis Tabor ^b
5	^a National Research Council Post Doctoral Fellow to the U.S. Environmental Protection Agency,
6	Office of Research and Development, National Risk Management Research Laboratory,
7	Research Triangle Park, NC 27711, USA
8	^b U.S. Environmental Protection Agency, Office of Research and Development, National Risk
9	Management Research Laboratory, Research Triangle Park, NC 27711, USA
10	¹ Present address: University of Dayton Research Institute, 300 College Park, Dayton, OH
11	45469-0141, USA
12	² Corresponding Author: gullett.brian@epa.gov, 919/541-1534
13	
14	ABSTRACT
15	Emissions from prescribed burns of forest and grass stands in western Florida were measured by
16	simultaneous aerial and ground sampling. Results were compared with biomass gathered from the
17	same stands and tested in an open burn laboratory test facility. Measurements included polycyclic
18	aromatic hydrocarbons (PAHs), volatile organic compounds (VOCs), particulate matter (PM2.5),
19	elemental carbon (EC), organic carbon (OC), black carbon (BC), brown carbon (BrC),
20	polychlorinated dibenzo-p-dioxins (PCDDs), and polychlorinated dibenzofurans (PCDFs). The

flaming phase (high modified combustion efficiency) was characterized by high levels of BC and 21 BrC yet low levels of VOCs. In general, ground based measurements of PM_{2.5}, BC, and BrC 22 reported marginally higher emission factors than those measured by aerostat-lofted instruments in 23 24 the plume. The optically-determined BC emission factor was approximately ten times higher than 25 many previously reported results. Simultaneous BC and EC measurements showed that EC values were, on average, 42% lower than the BC values, lending uncertainty to the common use of EC 26 27 measurements as a BC surrogate. PAH emission factors were indistinguishable across the sampling scenarios, while PCDDs/PCDFs saw a significant decline in the laboratory testing. 28 29

30 **KEYWORDS:** Black carbon, PAH, Brown Carbon, PCDD, PCDF, VOCs, PM2.5

31 1. INTRODUCTION

Wildfires and prescribed forest burns generate a variety of emissions that can cause adverse health effects for humans, contribute to climate change, and give rise to decreased visibility. Prescribed forest burns are carried out during the cooler, wetter months to reduce fuel buildup, improve the habitat for animal and plant species, minimize spread of disease, and reduce the risk of wildfires during the warmer months.

Measuring air emissions and deriving emission factors from prescribed forest burns are 37 important because it will provide data to better understand effects on air quality, as the emission 38 39 factors are used in national emission inventory calculations, air climate change models, and risk assessments. Forest burns are one of the largest sources of particulate matter in the U.S.A (U.S. 40 EPA, 2012). Light-absorbing carbonaceous aerosols known as black carbon (BC) are considered 41 to be one of the major contributors to the global climate change (Ramanathan and Carmichael, 42 2008). Elemental carbon (EC), sometimes used interchangeably with BC despite measurement by 43 different means (thermal-optical versus optical/aethalometer), is an indicator of combustion extent. 44 The light-absorbing organic matter aerosols referred to as brown carbon (BrC) is a possible global 45 warming agent (Alexander et al., 2008; Andreae and Gelencser, 2006; Kirchstetter et al., 2004) 46 47 that is starting to get more attention. Volatile organic compounds (VOCs) from combustion such as benzene, and semi-volatile organic compounds such as polycyclic aromatic hydrocarbons 48 (PAHs), polychlorinated dibenzo-p-dioxins (PCDDs) and polychlorinated dibenzofurans 49 50 (PCDFs), include pollutants with properties that are toxic, carcinogenic, bioaccumulative, and persistent in the environment. 51

Measuring air emissions from actual field burns is challenging due to variation in plume height,
changes in wind direction, and safety for sampling equipment and personnel, all of which may

54 limit the time the sampling equipment can spend in the plume. Emissions can be measured in the 55 field by aerial or ground based sampling platforms or, alternatively, in laboratory test facilities. 56 Aerial measurements are important when plumes are lofted and when ground-based sampling may 57 create a bias toward the safer, smoldering phase of the burn. Laboratory measurements are utilized 58 when field measurements are impractical or costly and when species-, phase-, and condition-59 specific burn data are desired.

Emission sampling is typically separated into flaming (good combustion) and smoldering 60 (poor combustion) phases, as each has reasonably distinctive emission factors. Samples further 61 62 afield from large burns may be a mix of both flaming and smoldering combustion phases as the flame front and post-flame smoldering emissions are mixed and entrained into the plume. 63 Measurements at heights above the fire are more likely to generate a composite sample of flaming 64 65 and smoldering emissions than would be produced from point source measurements on the ground. Collection of ground based samples may be biased toward the smoldering phase because they are 66 more amenable to safe sampling and equipment survival than intense flaming phases. During 67 ground-based sampling, mixing of the plume is limited, the pollutants have less aging, heavier 68 particles are less likely entrained, and pollutants are less likely to have been affected by 69 atmospheric chemistry such as formation of secondary organic aerosols (Cubison et al., 2011). 70 Evidence for differences between emissions sampled from airborne and ground-based platforms 71 was observed for hydrocarbons and oxygenated organic species by Burling et al. (2011). Methods 72 73 for aerial and ground measurements were demonstrated during three southeastern U.S. prescribed burns (Aurell and Gullett, 2013). Initial reporting of concurrent application of these aerial- and 74 ground-based measurements (Strand et al., 2014) showed little difference in a variety of pollutants, 75 76 but only single aerial samples were available for comparison.

77 Due to their small size and controlled combustion conditions, laboratory tests can have the advantage of allowing both species-specific studies and separate measurements of different 78 combustion phases. Comparability between laboratory and field emission data has been 79 demonstrated for African fires (Christian et al., 2003) and extensively covered by Yokelson et al. 80 (2012). Differences observed in laboratory versus field emission factors have been attributed to 81 higher modified combustion efficiency (MCE) values in laboratory burns (Christian et al., 2003) 82 of African savanna fuels. When they account for MCE, laboratory-based emission data predict 83 emission factors within 15% of the field values. 84

Work reported here, part of a comprehensive study to develop a fire and fuels dataset (Ottmar et al., 2013), presents multipollutant emission factors derived through concurrent aerial and ground sampling and accompanied by laboratory combustion tests on the same biomass.

88

89 2. EXPERIMENTAL SECTION

90 2.1 Sampling Method and Instrumentation

A light-weight instrument package termed the "Flyer" was attached to a helium filled, tethered 91 aerostat (4.3 m in diameter) mounted to an all terrain vehicle (ATV). The use of the aerostat and 92 Flyer instrumentation with the ATV has been described in detail elsewhere (Aurell and Gullett, 93 2013; Aurell et al., 2011). In summary, the instrumentation on the Flyer consisted of: non-94 dispersive infrared for continuous measurement of carbon dioxide (CO₂) (LICOR 820, LICOR 95 96 Biosciences U.S.A), light-scattering photometers for continuous PM_{2.5} (DustTrak 8520, DustTrak DRX, TSI Inc. U.S.A), micro aethlometers for BC and BrC (AE51, AE52, Aethlabs, U.S.A), 97 SUMMA canisters for VOCs, CO₂, and carbon monoxide (CO) (Columbia Analytical Services -98 99 CAS, U.S.A.), batch sampling of $PM_{2.5}$ onto a 47 mm in diameter Teflon filter (pore size 2 μ m),

batch sampling onto quartz filters for EC, OC, and TC measurements, and batch sampling of
PCDDs/PCDFs onto a quartz filter and polyurethane foam sorbent. The Flyer's position and
altitude were tracked using a global positioning system (MTi-G, Xsens, Netherlands). The ground
based ATV was also equipped with a smaller sampling package for measurements of PAHs using
a quartz filter, polyurethane foam, and XAD-2 sorbent, described in detail in Aurell et al. (2011).

105 2.2 Field Burn Description

Simultaneous aerial and ground sampling was conducted during prescribed burns of one 106 grass/forb field (126.5 ha) and one managed longleaf pine (*Pinus palustris*) forest (151 ha) at Eglin 107 108 Air Force Base, Florida, during two days in November of 2012 (Ottmar et al., 2014; Ottmar et al., 109 2013). This site was selected to complement previous testing done during the Prescribed Fire Combustion and Atmospheric Dynamic Research (RxCADRE) research campaign at Eglin Air 110 111 Force Base (http://www.firelab.org/document/rxcadre-overview). The burns were ignited by drip torch from all terrain vehicles (ATVs) in separate parabola shaped lines, described in detail 112 elsewhere (Ottmar et al., 2014). The aerostat sampler collected emissions downwind of the grass 113 114 field and forest area at an altitude of 50-110 m and 45-85 m, respectively. Forest biomass was gathered on an adjacent, unburned area and transported for subsequent laboratory burn testing four 115 days later. 116

117 **2.3 Open burn test facility**

The U.S. EPA's 70 m³ open burn test facility (OBTF) was used for laboratory simulation of prescribed burns of forest litter (no grass burns were conducted), (Aurell and Gullett, 2013). The OBTF (described in detail elsewhere (Grandesso et al., 2011)) is an enclosed facility ventilated with a high-volume blower that pulls in ambient air and smaller, interior fans that together ensure complete mixing and oxygen concentration close to ambient conditions. Twelve biomass burns 123 were conducted atop an aluminum foil-covered (replaced between each burn) sand layer supported 124 by a steel plate (0.93 m \times 0.93 m). The forest biomass was divided by a quartering procedure (Aurell and Gullett, 2013) and placed on the foil at a charge size (1.4 kg) closely mimicking the 125 field area density $(1.1 \text{ kg/m}^2 \text{ (Ottmar et al., 2014)})$. For consistency, the OBTF sampling used the 126 same instrumentation (the Flyer) as in the field tests with the addition of a bench top CO monitor 127 (California Analytical Instruments Model 200, 0-1,000 ppm, U.S.A.). Ambient air background 128 samples were collected after the burns by sampling inside the OBTF for approximately 9 hours. 129 An ultimate analysis and PCDD/PCDF analysis was conducted on the raw forest biomass. For the 130 131 latter, three raw biomass samples (48 ± 2 g) were Soxhlet-extracted using U.S. EPA Method 3540C (1996) and analyzed for PCDDs/PCDFs by high resolution gas chromatography/high resolution 132 mass spectrometry (HRGC/HRMS). 133

134 **2.4 Data and Sample Analyses**

Emission factors were derived using the carbon mass balance approach (Laursen et al., 1992), 135 which assumes that all combusted carbon in the biomass is emitted to the atmosphere as CO_2 , CO_3 , 136 137 CH₄, and total hydrocarbons (THCs). Due to instrumentation size and weight considerations, only CO₂ was measured in the field, while both CO₂ and CO were measured in the OBTF. The derived 138 emission factors (all except VOCs) in the OBTF were 6.7% \pm 2.7 lower using CO₂ and CO as a 139 carbon source rather than only CO₂, which is within the total error of the method and likely the 140 reproducibility of the event. The VOC emission factors were $2.3\% \pm 0.16$ lower using CO₂, CO, 141 and CH₄ than only CO₂ as a carbon source. Emission factors are expressed in terms of pollutant 142 mass per mass of biomass consumed by the burn using a biomass carbon fraction value of 0.5 143 (Strand et al., 2014). 144

PAHs were cleaned up according to U.S. EPA Method TO-13 (1999b) and analyzed using a 145 Hewlett-Packard gas chromatograph 5890 Series II with a DB-5 60 m \times 0.25 mm \times 0.25 µm column 146 (Agilent/J&W Scientific) connected to a Hewlett Packard 5972 mass spectrometer. PAH emission 147 factors were evaluated using toxic equivalent factors (TEFs) relative to the benzo-[a]-pyrene 148 toxicity equivalent (B[a]P-TEQs) (Larsen and Larsen, 1998). PCDD/PCDF samples were cleaned 149 150 up via a modified U.S. EPA Method TO-9A (1999a) and analyzed with a HRGC/HRMS using a Hewlett-Packard gas chromatograph 6890 Series coupled to a Micromass Premier (Waters Inc., 151 UK) with a Dioxin-2 60 m \times 0.25 mm \times 0.25 µm column (Restek). The PCDD/PCDF TEQ 152 153 emission factors were determined by using the 2005 World Health Organization (WHO) TEFs (Van den Berg et al., 2006). Non-detect (ND) congeners were noted for some of the seventeen 154 PCDD/PCDF TEF congeners. The ND congeners were set to zero as well as to the limit of 155 156 detection (SI Table S2).

EC, TC, and OC were analyzed via a modified, thermal-optical analysis (TOA) using NIOSH 157 Method 5040 (1999c) (Khan et al., 2012). Custom correction factors were calculated for the 158 159 DustTrak, as per the manufacturer's recommendations, by dividing the average continuous $PM_{2.5}$ concentration by the PM_{2.5} batch filter concentration collected during the same time period. The 160 average correction factor was 2.4±0.61 for the OBTF measurements. The Modified Combustion 161 Efficiency (MCE), $(\Delta CO_2/(\Delta CO + \Delta CO_2))$, was calculated using the CO and CO₂ concentrations 162 collected by the SUMMA canisters and OBTF test data. The BC and BrC were divided into 163 164 categories of MCE > 0.95 (good combustion), >0.90 and <0.95, and <0.90 (poor combustion) for analysis. 165

166

167 **3. RESULTS AND DISCUSSION**

Simultaneous aerial and ground-based samples were taken with the exception of the forest burn where duplicate ground-based samples were possible due to the burn duration. The forest biomass underwent multiple burns in the OBTF, resulting in three, composite-burn samples of trace pollutants (e.g., PCDD/PCDF) and multiple replicates of filter-based pollutants. Results are presented in Table 1.

173 **3.1 PM_{2.5}, BC, BrC, EC, OC, TC**

174 *3.1.1 Data Comparison*

The PM_{2.5} emission factors from field forest burns, 20-23 g/kg (Table 1), were slightly higher 175 than previously published values derived from the same location (13-14 g/kg) (Aurell and Gullett, 176 2013), as well as from wildfires from Portugal and Mexico 4.5-12 g/kg (Alves et al., 2011; 177 Yokelson et al., 2011). Both the field and OBTF forest burn emission factors (12-23 g/kg) were in 178 the upper range of previously published emission factors from prescribed burning and laboratory 179 burn tests of US biomass species 0.90-24 g/kg (Burling et al., 2011; Lee et al., 2005) and 1.6-34 180 g/kg (Hays et al., 2002; Hosseini et al., 2013), respectively. The PM_{2.5} emission factors from the 181 grass burns (14-18 g/kg) were similar or in the upper range of previously published data of different 182 grass species (savanna, saw grass, wire grass, Kentucky bluegrass) 4.0-15 g/kg (Dhammapala et 183 184 al., 2006; Urbanski et al., 2009; Yokelson et al., 2011). The BrC emission factors in this study (0.68-1.8 g/kg) were in the same range as previously measured (0.9-2.1 g/kg (Aurell and Gullett, 185 2013)) in the southeast U.S. 186

The EC emission factors found in this study (0.39-0.62 g/kg) are in the same range as found in the literature 0.27-1.3 g/kg (Alves et al., 2011; Andreae et al., 1998; Ferek et al., 1998; Hays et al., 2002) (analyzed using thermal measurements) from prescribed burning, wildfires, and laboratory burns of different biomass types. Thermal EC measurements, both current and historical, have 191 often been equated with BC, which is an optically measured property. Few BC emission factors 192 determined by light absorption measurements from biomass burns exist in the literature. Kondo et al. (Kondo et al., 2011) reported an emission factor of 180 ng/m³ BC per ppm CO₂, or 0.11 g 193 194 BC/kg (assuming a biomass carbon fraction of 0.50), which is lower than found in this study (0.85-1.4 g/kg). In a previous study we found BC emission factors of 1.0-2.7 g/kg (Aurell and Gullett, 195 2013) from prescribed burns of biomass from the southeast US, which are similar to those found 196 197 in this study. Seven of nine EC emission factors (0.39-0.62 g/kg) we measured were 42%±13% lower than the simultaneously sampled BC emission factors (0.85-1.4 g/kg) but the concentrations 198 199 for all of the data points correlated well (correlation coefficient of 0.85, SI Figure S1). Importantly, it shows that historical use of EC values to infer BC emission factors may have significantly 200 underestimated BC levels. Moderate to good correlations between BC and EC measurements have 201 previously been found in inter-comparison studies ($r^2 0.44-0.88$) (Ahmed et al., 2009; Jeong et al., 202 2004; Kamboures et al., 2013; Ram et al., 2010). In a recent study using the same measurement 203 methods in this work, BC values using an aethalometer (AE51) were compared to EC 204 205 measurements (NIOSH method) and found to be 39% higher or a BC/EC ratio of 1.64 (Yelverton et al., 2014), in agreement with our findings. 206

The OC emission factors for forest field burns, 11-15 g/kg, were double those from wildfires in Portugal 6.0±2.9 g/kg (Alves et al., 2011) and forest burns in Brazil 6.8 g/kg (estimated Fc 0.5) (Ferek et al., 1998) while the grass burn OC emission factors of 6.5-7.0 g/kg were similar to grass burns in Brazil at 5.1 g/kg (estimated Fc 0.5) (Ferek et al., 1998).

212	I able	1.	Results.

		Grass burn	Grass burn	Forest burn	Forest burn	Forest burn
Compound	Unit	Ground	Aerostat	Ground	Aerostat	$OBTF^{h}$
Filter PM _{2.5}	g/kg biomass	18 ^g	14 ^g	23±1.8 ^{d,g}	20 ^g	12.2±6.9 ^d
						(11.4±6.4)
Continuous PM _{2.5}	g/kg biomass	20 ^g	15 ^g	25 ^g	24 ^g	11±4.6
						$(10\pm4.1^{\rm d})$
Continuous BC ^a	g/kg biomass	1.1 ^g	0.91 ^g	0.89^{g}	1.4 ^g	1.3 ± 0.40^{d}
						(1.2 ± 0.38^{d})
Continuous BrC ^a	g/kg biomass	1.8 ^g	NS^{g}	NS^{g}	0.92^{g}	0.68 ± 0.37^{d}
						$(0.58\pm0.31^{\text{d}})$
Continuous BC ^b	g/kg biomass	1.2	1.0	0.85 ± 0.20^{d}	1.4	1.3 ± 0.31
						(1.2 ± 0.26)
Continuous BrC ^b	g/kg biomass	1.8	NS	NS	1.0	0.66 ± 0.27
				1		(0.58 ± 0.23)
Filter EC	g/kg biomass	0.62 ^g	0.56^{g}	$0.39\pm0.16^{a,g}$	0.46^{g}	0.89±0.29 ^a
F 11 0.0		7 03	< F	1 5 1 0da	119	(0.84 ± 0.28)
Filter OC	g/kg biomass	7.0 ^g	6.5 ^g	15±1.8 ^{d,g}	11 ^g	$6./\pm 6.5^{u}$
Eilter TC	alta hiomogo	7 69	7 19	16 1 0d.g	1.09	(0.3 ± 0.0)
Filler IC	g/kg biomass	/.0*	1.18	10±1.9-%	128	7.0 ± 0.4^{-1}
BC/PMac ^c	mass ratio %	6 8g	7 Og	$3.6+0.67^{d,g}$	7 Og	(7.1 ± 3.9) 128+53 ^d
DC/1 W12.5	mass ratio 70	0.00	7.05	5.0±0.07 *	7.00	12.0±3.3
EC/PM _{2.5}	mass ratio %	3.5 ^g	3.9 ^g	1.6±0.54 ^{d,g}	2.3 ^g	9.6 ± 6.9^{d}
BrC/PM ₂ 5 ^c	mass ratio %	9.8 ^g	NS ^g	NS	5.0 ^g	6.1 ± 2.6^{d}
	ratio	5.2	6.6	19 4 4 4 4	7.0	5 1 1 1
UC/BC	Tatio	5.5	0.0	10.4±4.4	1.9	J.1±4.1
OC/EC	ratio	11.2	11.6	42.5±13.2	24.4	9.6±11.4
PCDD/PCDF	ng WHO TEQ/kg	1.9 [2.2] ^{e,f}	0.80	1.2 [1.5] ^e	0.092±0.10 ^e
	biomass	-	-			(0.086±0.10)
						$[0.18 \pm 0.076^{d}]$
						(0.17±0.069)] ^e
PCDD/PCDF	Total ng/kg biomass	24	47 ^f	65	113	6.9±3.6
						(6.5 ± 3.3)
PCDD/PCDF	mass ratio	7.	.7 ^f	4.0	6.2	1.0 ± 0.17
SUM 16 EPA PAHs	mg/kg biomass	35	NS	23	NS	27 ± 3.7^{d}
		-		-		(25+3.6)

a sampled for a longer time interval than the batch filter. ^b sampled at the same time interval as the batch filter ^c batch filter and BC simultaneously sampled. ^d 1 STDV ^e None detectable congeners = limit of detection within brackets. ^f Combined sample from Aerostat and Ground measurements.
 ^g from Strand et al. 2014. ^h Emission factors calculated using carbon from both CO₂ and CO within parentheses.

216

217 *3.1.2 Aerial versus Ground Sampling*

The $PM_{2.5}$, TC, and OC emission factors were higher on the ground than in the air for both forest and grass burns (Figure 1 and SI Figure S2), although the minimal sample size limits the statistical difference.



Figure 1. Particle and carbon type emission factors. Error bars denote \pm 1 STDV. Ground and aerostat data from this current study, see also (Strand et al., 2014).

225

222

226 *3.1.3 Grass versus Forest Burns*

The PM carbon analysis showed the grass burn had lower emission factors for TC and OC thanthe forest burns for both ground and aerostat-based measurements (Figure 1).

229 3.1.4 Laboratory versus Field Sampling

The OBTF average PM_{2.5}, TC, and OC forest burn emission factors were similar to those 230 derived in the field. A typical concentration and emission factor plot from the OBTF tests is shown 231 in Figures 2A and 2B, respectively. The initial flaming region with high MCE is quickly followed 232 by a rapid spike in BC and BrC concentration that then falls off, replaced by increasing CO and 233 CO₂, followed by PM_{2.5}. BC and BrC emission factors, as well as the BC to PM_{2.5} mass ratio, 234 increase at higher MCE values (Figure 3 and SI Figure S3) suggesting that the OBTF tests may 235 have been conducted under higher average combustion efficiencies than in the field (Table 1). 236 237 The difference in BC and EC concentrations resulted in the BC/PM_{2.5} ratio being 1.3-3.0 times 238 higher than the $EC/PM_{2.5}$ ratio.



Figure 2. Typical OBTF burn of forest biomass. Concentrations (A) and emission factors (B).







Figure 3. BC and BrC versus modified combustion efficiency. Aerostat and Ground measurementsfrom this work; see also Strand et al. (Strand et al., 2014).

251 **3.2 Volatile Organic Compounds**

252 *3.2.1 Data Comparison*

The benzene emission factors for the forest/field burns (277±24 and 283 mg/kg) were in the 253 254 same range as previous burns of understory forest with the same MCE $(276\pm79 \text{ to } 526\pm262 \text{ mg/kg})$ 255 (Aurell and Gullett, 2013) and in both flaming $(211\pm73 \text{ mg/kg})$ and smoldering $(422\pm244 \text{ mg/kg})$ 256 stages of pine dominated forest in Georgia (Lee et al., 2005). The OBTF benzene emission factor 257 (92±20 mg/kg) was lower than previous laboratory forest litter burns (148-284 mg/kg) (Aurell and 258 Gullett, 2013; Hays et al., 2002). The benzene emission factors from the grass burns (192±63 and 259 253 mg/kg) were in the same range as Brazilian grass and slash piles (168 mg/kg) (Yokelson et 260 al., 2007).

261 *3.2.2 Aerial versus Ground Sampling*

262 No correlation between altitude (aerostat-sampled versus ground-sampled) and VOC emission
263 factor levels was found, although more samples would be needed to confirm this finding.

264 *3.2.3 Grass versus Forest Burns*

VOCs for grass burns were lower than those from forest burns. Grass burn measurements, whether by aerostat or ground sampling, had similar MCE values, about 0.92 to 0.94. These values were higher than those from forest burns (0.88 to 0.90). The VOC emission factors decreased with increased MCEs (Table 2, Figure 4), influenced more by the combustion quality than whether the biomass was forest or grass. The higher MCEs in the OBTF consistently led to lower VOC emission factors.

271 3.2.4 Laboratory versus Field Sampling

- 272 Consistent with the grass versus forest VOC/MCE relationship above, the VOC emission
- factors in the OBTF decreased with the higher MCE (Table 2, Figure 4) of the OBTF.
- 274
- Table 2. VOC, CO_2 , CO, and CH_4 emission factors^d.

	Grass burn	Grass burn	Forest burn	Forest burn	Forest burn
	Ground	Aerostat	Ground	Aerostat	OBTF
	mg/kg	mg/kg	mg/kg	mg/kg	mg/kg
Sampling Altitude (m)	2	82	2	48	NA
Modified combustion efficiency	0.933 ± 0.007^{b}	0.933	0.896 ± 0.007^{b}	0.901	0.979±0.006°
CO_2	1711±13	1711	1656	1652	1793±12
СО	78±8.3	78	113	116	24±7.0
Methane (CH ₄)	NA	NA	NA	NA	0.89 ± 0.27
Propene	261±22 ^b	204	329±14 ^b	314	84±31°
Chloromethane ^a	15±2.5 ^b	15	10 ± 1.1^{b}	11	$9.9 \pm 5.4^{\circ}$
1,3-Butadiene ^a	106 ± 6.5^{b}	80	138±27 ^b	114	13±6.8°
Acetonitrile	112±20 ^b	95	156±18 ^b	144	14±6.6°
Acrolein ^a	388±21 ^b	394	537±118 ^b	458	65±26 ^c
Acrylonitrile ^a	39±3.8 ^b	36	37±3.9 ^b	42	6.3±2.4°
Vinyl Acetate ^a	115±94 ^b	168	280±63 ^b	209	30±20°
2-Butanone (MEK)	40±3.1 ^b	34	98±14 ^b	69	12±4.9°
Benzene ^a	192±63 ^b	253	277±24 ^b	283	92±20°
Toluene ^a	89	84	141±26 ^b	111	27±12°
Ethylbenzene ^a	17	15	21±3.8 ^b	18	3.3±2.8°
m,p-Xylene	26	23	47 ± 9.4^{b}	35	5.9±5.7°
Styrene ^a	41	35	50 ± 10^{b}	43	9.2 ± 4.2^{b}
o-Xylene ^a	7.0	7.0	15±3.0 ^b	12	2.2 ± 0.76^{b}
Cumene ^a	1.2	ND	1.8 ± 0.32^{b}	ND	0.49 ± 0.21^{b}
alpha-Pinene	5.4	5.4	34 ± 2.8^{b}	44	6.9 ± 1.78^{b}
1,3,5-Trimethylbenzene	ND	ND	1.9 ± 0.32^{b}	ND	0.37
1,2,4-Trimethylbenzene	4.0	4.6	8.3 ± 1.7^{b}	5.9	1.1 ± 0.40^{b}
d-Limonene	29	23	31 ± 10^{b}	20	6.7±3.3
Naphthalene	40	42	46±7.2 ^b	44	16±3.0 ^b
n-Hexane	1.4 ± 0.62^{b}	2.7	8.7 ± 2.0^{b}	3.7	0.75 ± 0.30^{b}
n-Heptane	0.48	0.44	6.9 ± 2.4^{b}	0.79	0.46 ± 0.087^{b}
Ethanol	21±7.9 ^b	13	22±6.6 ^b	6.5	0.90

^a Included in the EPA list of hazardous air pollutants. ^b Range of data. ^c 1 STDV. ^d Emission factors

277 calculated with CO and CO₂, except for CO, CO₂, and CH₄ for OBTF which included carbon from CH₄.

278 ND = not detected, DNA = does not apply, NA = not analyzed.



283

284 **3.3 PAHs.**

285 3.3.1 Data Comparison

Comparison of PAH emission factors against literature data is cumbersome as different PAHs may be analyzed along with the use of different sampling methods (different configurations of XAD, PUF, and quartz filter). In some cases, a toxicity measure allows inter-comparisons. The B[a]P-TEQ emission factors presented from the literature in this paper have been derived using the reported literature PAH values and the TEF values from Larsen and Larsen (1998) and, in somecases, the carbon fraction of the fuel.

The emission factors for phenanthrene from forest burns in this study (3.1-3.4 mg/kg, Table 3) 292 293 were in the mid-range of literature data obtained from laboratory tests 0.11-11.52 mg/kg (Hays et al., 2002; Hosseini et al., 2013; Jenkins et al., 1996; Singh et al., 2013) and was similar when 294 sampled using the same sampling method (filter and XAD sorbent) 2.6-3.9 mg/kg (Jenkins et al., 295 296 1996). The phenanthrene emission factor from the grass burn, 3.7 mg/kg, was also similar to previous laboratory tests of Kentucky bluegrass, 4.0 mg/kg (Dhammapala et al., 2007). The 297 emission factor from the field grass burn was approximately two times lower than laboratory burns 298 of Kentucky bluegrass reported in the literature: 0.31 mg B[a]P-TEQ/kg and 0.58 mg B[a]P-299 TEQ/kg (Dhammapala et al., 2007), respectively. The forest field burn emission factors were 300 301 similar to laboratory burns of U.S. fir, pine, chaparral and oak 0.12-0.50 mg B[a]P-TEQ/kg (Hosseini et al., 2013; Jenkins et al., 1996) but much lower than laboratory burned fuel woods 302 from India, 3.2 mg B[a]P-TEQ/kg (Singh et al., 2013). The B[a]P-TEQ emission factors from 303 304 prescribed burning of forest litter and grass presented in this paper are low in comparison to conventional outdoor residential wood-fired hydronic heaters 6.1 mg B[a]P-TEQ/kg (Aurell et al., 305 2012a) (white pine) and 2.2 mg B[a]P-TEQ/kg (Aurell et al., 2012a) (red oak), and open burns of 306 municipal waste 2.2 mg B[a]P-TEQ/kg (Aurell et al., 2012b), whereas they are in the same range 307 as a modern three stage combustion hydronic heaters (red oak) 0.31 mg B[a]P-TEQ/kg (Aurell et 308 309 al., 2012a), and higher than a pellet based hydronic heaters 0.063 mg B[a]P-TEQ/kg (Aurell et al., 2012a) 310

311 *3.3.2 Grass versus Forest Burns*

Total PAHs (sum of 16 EPA PAHs) and B[a]P-TEQ emission factors from the forest and grass 312 fuels are presented in Table 3 and SI Table S1, respectively. The most abundant PAHs were 313 naphthalene, acenapthylene, and phenanthrene. Naphthalene contributed to 56% of the total PAH 314 emission factors, on average (SI Figure S4). The total PAH emission factor for the grass burn (35 315 316 mg/kg) was higher than for the forest burn (23 and 27 mg/kg) (Table 3), although this was largely due to the difference in the naphthalene emission factor. When naphthalene was excluded the total 317 PAH emission factors were similar for grass field and forest field at 14 mg/kg and 11 mg/kg, 318 319 respectively.

320 *3.3.3 Laboratory versus Field Sampling*

The total PAH emission factor for forest fuel from the OBTF burns was similar to the field data, at 27±3.7 mg/kg and 23 mg/kg, respectively. The benzo(a)pyrene contributed to 62% (on average) of the B[a]P-TEQ emission factor, and fluroanthene and benzo(b)floranthene contributed 17% and 5%, respectively. The B[a]P-TEQ emission factor from the field forest burn was lower than from the OBTF, 0.14 mg B[a]P-TEQ/kg and 0.33 mg B[a]P-TEQ/kg, respectively.

326 The naphthalene emission factors from the field burns were slightly lower using the quartz 327 filter/XAD method rather than the SUMMA canister method, although some of this difference may be due to different sampling intervals. The SUMMA canisters samples were collected during 328 a short time interval, 1-2 min, representing a segment of the burn process, while the two quartz 329 330 filter/XAD samples were collected for 40 and 160 min representing the entire burn process. The OBTF naphthalene emission factors were similar for both methods even though the MCE was 331 332 lower (0.865) during the quartz filter/XAD method sampling duration than during the SUMMA canister method sampling (0.979). 333

	Grass burn	Forest burn	Forest burn ^a
РАН	Ground	Ground	OBTF
	mg/kg biomass	mg/kg biomass	mg/kg biomass
Naphthalene	22	13	14±2.3 (13±2.3)
Acenaphthylene	4.6	2.9	4.3±0.50 (4.0±0.50)
Acenaphthene	0.34	0.38	0.41±0.087 (0.39±0.081)
Fluorene	1.1	1.4	1.4±0.37 (1.3±0.34)
Phenanthrene	3.7	3.4	3.1±0.57 (2.9±0.52)
Anthracene	0.60	0.63	0.65±0.12 (0.61±0.11)
Fluoranthene	1.1	0.73	0.95±0.13 (0.89±0.12)
Pyrene	0.94	0.62	0.94±0.11 (0.87±0.11)
Benzo(a)anthracene	0.20	0.1	0.29±0.042 (0.28±0.041)
Chrysene	0.27	0.2	0.31±0.042 (0.29±0.039)
Benzo(b)fluoranthene	0.20	0.081	0.16±0.022 (0.15±0.021)
Benzo(k)fluoranthene	0.13	0.052	0.16±0.0046 (0.15±0.0055)
Benzo(a)pyrene	0.20	0.071	0.21±0.022 (0.20±0.021)
Indeno(1,2,3-cd)pyrene	0.13	0.052	0.13±0.011 (0.12±0.011)
Dibenz(a,h)anthracene	ND	0.0048	0.013±0.0033 (0.012±0.0030)
Benzo(g,h,i)perylene	0.13	0.033	0.11±0.0082 (0.10±0.0084)
SUM 16-EPA PAH	35	23	27±3.7 (25±3.6)
SUM PAH B[a]P	3.1E-01	1.4E-01	3.3E-01±3.5E-02

Table 3. Results 16 EPA PAHs.

^a 1 STDV. Emission factors derived using carbon from both CO and CO₂ within parentheses.

336

337

338 **3.4 PCDDs/PCDFs.**

339 *3.4.1 Data Comparison*

The aerial forest emission factor for PCDD/PCDF (1.2 ng TEQ/kg, Table 1) was higher than that previously measured from the same location/area with the same method (0.63 ng TEQ/kg) (Aurell and Gullett, 2013). This may be due to seasonal differences in composition of the biomass as the previous measurements were in February and these are in November. With 17 assigned toxic equivalent factors (TEFs) the four collected PCDD/PCDF samples had 10-17 detected TEF congeners (SI Table S2). The plume emissions exceeded the raw biomass emissions indicating that the emissions are formed rather than evaporated from the raw biomass.

347 *3.4.2 Aerial versus Ground Sampling*

The aerial collected forest burn PCDD/PCDF emission factor was slightly higher than the ground collected sample (1.2 ng TEQ/kg versus 0.80 ng/kg). If this difference is real, it may be due to a higher MCE at the aerial collection point than on the ground, as indicated by the BC/PM_{2.5} ratio discussed above.

352 *3.4.3 Grass versus Forest Burns*

The grass burn had a higher PCDD/PCDF emission factor than the forest burn, 1.9 ng TEQ/kg compared to 1.0 ng TEQ/kg (on average), despite higher MCE values. This is contrary to typical results, where increased PCDD/PCDF levels are linked to poor combustion. Differences in the chemical composition of the species may be responsible for this destruction.

357 *3.4.4 Laboratory versus Field Sampling*

The OBTF forest burn PCDD/PCDF emission factor was much lower (0.092 ng TEQ/kg) than from the field (1.2 ng TEQ/kg and 0.80 ng TEQ/kg, Table 1), contrary to previous results (Aurell and Gullett, 2013) from this area. The limited number of samples show little MCE differences; differences may be due to burn intensity but this remains to be explored further.

362

363 4. CONCLUSION

Sampling of forest and grass prescribed burn emissions was accomplished with both aerialand ground-based measurements. Field-gathered forest biomass was collected, burned, and sampled for emissions in a laboratory test facility.

367 Emission factors for a comprehensive array of pollutants were determined. $PM_{2.5}$ emission 368 factors from the forest burns were higher than previous values from the same site, higher than at

least two other reported wildfires, and in the upper range of published field and laboratory tests.
PM_{2.5} results for the grass field burns were consistent with measurements of other species.

While optically-measured BC values correlated well with thermally/optically-measured EC, the former were about 1.4 times higher, suggesting caution when using historical EC values to represent BC. TC and OC measurements appeared to be lower in aerial measurements than ground measurements, suggesting further reactions above ground-level samplers or a bias in the gases measured at ground vs air levels.

Laboratory burns of forest biomass in the OBTF showed some emission factor differences 376 377 from the field measurements which are not necessarily explained by differences in the MCE values. High BC values, associated with flaming combustion, are noted for the OBTF, consistent 378 with the higher MCE values. BrC concentrations, too, were associated with flaming combustion 379 380 and CO₂ evolution. As the combustion intensity declines and CO rises, BC and BrC decrease, followed by a lagging peak of PM_{2.5}. Laboratory measurements yielded lower PM_{2.5} and OC 381 values than field measurements, whereas the reverse was true for EC. Lower VOC concentrations 382 383 were noted in the laboratory burns; concentrations between field and lab measurements indicated good inverse correlations (e.g., $r^2 = 0.7 - 0.9$) with MCE. 384

Comparison of aerial- versus ground-based sampling showed differences in carbon measurements; aerostat-based sampling yielded about twice the BC and EC concentrations of ground sampling. This was true only for the forest burns and not the grass burns, however. Whether this is due to a mechanistic difference or a sampling bias is not clear and will require additional measurements.

Little difference was apparent in the PAH data between forest and grass, aerial and ground,
and field and laboratory burns. Naphthalene was the most prevalent PAH observed. PCDD/PCDF

emission factors were similar for grass and forest burns and aerial and ground sampling but
significantly lower for the OBTF. This is likely related to the higher MCEs observed in the OBTF,
as PCDDs/PCDFs are linked to marginal combustion conditions.

395

396 **5. ACKNOWLEDGMENT**

This work was funded by the U.S. EPA's Office of Research and Development. We are grateful 397 398 to Roger Ottmar, U.S. Forest Service, for his professional coordination and helpful assistance. Special thanks to Kevin Hiers and Brett Williams at Eglin Air Force Base (FL) for coordinating 399 400 the on-site effort. Contributing EPA personnel included Chris Pressley, Dennis Tabor, and Robert 401 Black (ORISE post-doctoral fellow). Aerostat operations were handled by Rob Gribble (ISSI, Inc.). Special thanks to Mike Hayes for EC/OC/TC analyses. Jeff Blair of AethLabs donated use 402 of the AE-52. This research was performed while Johanna Aurell held a National Research Council 403 Research Associateship Award at the U.S. EPA/NRMRL. This publication has been subjected to 404 the U.S. EPA's peer and administrative review and has been approved for publication as a U.S. 405 406 EPA document.

407

408 **6. REFERENCES**

409	Ahmed, T.; Dutkiewicz,	V.A.; Shareef, A.; T	Suncel, G.; Tuncel, S	S.; Husain, L.	Measurement of
-----	------------------------	----------------------	-----------------------	----------------	----------------

- black carbon (BC) by an optical method and a thermal-optical method: Intercomparison
 for four sites. Atmos. Environ. 43:6305-6311; 2009.
- Alexander, D.T.L.; Crozier, P.A.; Anderson, J.R. Brown carbon spheres in East Asian outflow and
 their optical properties. Science. 321:833-836; 2008.

414	Alves, C.A.; Vicente, A.; Monteiro, C.; Gonçalves, C.; Evtyugina, M.; Pio, C. Emission of trace
415	gases and organic components in smoke particles from a wildfire in a mixed-evergreen
416	forest in Portugal. Sci. Total Environ. 409:1466-1475; 2011.

- 417 Andreae, M.O.; Andreae, T.W.; Annegarn, H.; Beer, J.; Cachier, H.; le Canut, P.; Elbert, W.;
- 418 Maenhaut, W.; Salma, I.; Wienhold, F.G.; Zenker, T. Airborne studies of aerosol emissions
- 419 from savanna fires in southern Africa: 2. Aerosol chemical composition. J. Geophys. Res420 Atmos. 103:32119-32128; 1998.
- Andreae, M.O.; Gelencser, A. Black carbon or brown carbon? The nature of light-absorbing
 carbonaceous aerosols. Atmos. Chem. Phys. 6:3131-3148; 2006.
- 423 Aurell, J.; Gullett, B.K. Emission Factors from Aerial and Ground Measurements of Field and
- 424 Laboratory Forest Burns in the Southeastern US: PM2.5, Black and Brown Carbon, VOC,
 425 and PCDD/PCDF. Environ. Sci. Technol. 47:8443-8452; 2013.
- Aurell, J.; Gullett, B.K.; Pressley, C.; Tabor, D.; Gribble, R. Aerostat-lofted instrument and
 sampling method for determination of emissions from open area sources. Chemosphere.
 85:806-811; 2011.
- Aurell, J.; Gullett, B.K.; Tabor, D.; Touati, A.; Oudejans, L. Semivolatile and Volatile Organic
 Compound Emissions from Wood-Fired Hydronic Heaters. Environ. Sci. Technol.
 431 46:7898-7904; 2012a.
- Aurell, J.; Gullett, B.K.; Yamamoto, D. Emissions from Open Burning of Simulated Military
 Waste from Forward Operating Bases. Environ. Sci. Technol. 46:11004-11012; 2012b.
- 434 Burling, I.R.; Yokelson, R.J.; Akagi, S.K.; Urbanski, S.P.; Wold, C.E.; Griffith, D.W.T.; Johnson,
- 435 T.J.; Reardon, J.; Weise, D.R. Airborne and ground-based measurements of the trace gases

- and particles emitted by prescribed fires in the United States. Atmos. Chem. Phys.
 11:12197-12216; 2011.
- 438 Christian, T.J.; Kleiss, B.; Yokelson, R.J.; Holzinger, R.; Crutzen, P.J.; Hao, W.M.; Saharjo, B.H.;
- Ward, D.E. Comprehensive laboratory measurements of biomass-burning emissions: 1.
 Emissions from Indonesian, African, and other fuels. J. Geophys. Res-Atmos. 108:4719;
 2003.
- 442 Cubison, M.J.; Ortega, A.M.; Hayes, P.L.; Farmer, D.K.; Day, D.; Lechner, M.J.; Brune, W.H.;
- 443 Apel, E.; Diskin, G.S.; Fisher, J.A.; Fuelberg, H.E.; Hecobian, A.; Knapp, D.J.; Mikoviny,
- 444 T.; Riemer, D.; Sachse, G.W.; Sessions, W.; Weber, R.J.; Weinheimer, A.J.; Wisthaler, A.;
- Jimenez, J.L. Effects of aging on organic aerosol from open biomass burning smoke in
 aircraft and laboratory studies. Atmos. Chem. Phys. 11:12049-12064; 2011.
- Dhammapala, R.; Claiborn, C.; Corkill, J.; Gullett, B. Particulate emissions from wheat and
 Kentucky bluegrass stubble burning in eastern Washington and northern Idaho. Atmos.
 Environ. 40:1007-1015; 2006.
- 450 Dhammapala, R.; Claiborn, C.; Jimenez, J.; Corkill, J.; Gullett, B.; Simpson, C.; Paulsen, M.
- Emission factors of PAHs, methoxyphenols, levoglucosan, elemental carbon and organic
 carbon from simulated wheat and Kentucky bluegrass stubble burns. Atmos. Environ.
 41:2660-2669; 2007.
- Ferek, R.J.; Reid, J.S.; Hobbs, P.V.; Blake, D.R.; Liousse, C. Emission factors of hydrocarbons,
 halocarbons, trace gases and particles from biomass burning in Brazil. J. Geophys. ResAtmos. 103:32107-32118; 1998

457	Grandesso, E.; Gullett, B.; Touati, A.; Tabor, D. Effect of Moisture, Charge Size, and Chlorine
458	Concentration on PCDD/F Emissions from Simulated Open Burning of Forest Biomass.
459	Environ. Sci. Technol. 45:3887-3894; 2011.
460	Hays, M.D.; Geron, C.D.; Linna, K.J.; Smith, N.D.; Schauer, J.J. Speciation of gas-phase and fine
461	particle emissions from burning of foliar fuels. Environ. Sci. Technol. 36:2281-2295; 2002.
462	Hosseini, S.; Urbanski, S.P.; Dixit, P.; Qi, L.; Burling, I.; Yokelson, R.; Shrivastava, M.; Jung, H.;
463	Weise, D.R.; Miller, W.; Cocker, D. Laboratory characterization of PM emissions from
464	combustion of wildland biomass fuels. J. Geophys. Res-Atmos. 118:9914-9929; 2013.
465	Jenkins, B.M.; Jones, A.D.; Turn, S.Q.; Williams, R.B. Emission Factors for Polycyclic Aromatic
466	Hydrocarbons from Biomass Burning. Environ. Sci. Technol. 30:2462-2469; 1996.
467	Jeong, CH.; Hopke, P.K.; Kim, E.; Lee, DW. The comparison between thermal-optical
468	transmittance elemental carbon and Aethalometer black carbon measured at multiple
469	monitoring sites. Atmos. Environ. 38:5193-5204; 2004.
470	Kamboures, M.A.; Hu, S.S.; Yu, Y.; Sandoval, J.; Rieger, P.; Huang, S.M.; Zhang, S.; Dzhema,
471	I.; Huo, D.; Ayala, A.; Chang, M.C.O. Black carbon emissions in gasoline vehicle exhaust:
472	A measurement and instrument comparison. J. Air Waste Manage. 63:886-901; 2013.
473	Khan, B.; Hays, M.D.; Geron, C.; Jetter, J. Differences in the OC/EC Ratios that Characterize
474	Ambient and Source Aerosols due to Thermal-Optical Analysis. Aerosol Sci. Technol.
475	46:127-137; 2012.
476	Kirchstetter, T.W.; Novakov, T.; Hobbs, P.V. Evidence that the spectral dependence of light
477	absorption by aerosols is affected by organic carbon. J. Geophys. Res-Atmos. 109; 2004.
478	Kondo, Y.; Matsui, H.; Moteki, N.; Sahu, L.; Takegawa, N.; Kajino, M.; Zhao, Y.; Cubison, M.J.;
479	Jimenez, J.L.; Vay, S.; Diskin, G.S.; Anderson, B.; Wisthaler, A.; Mikoviny, T.; Fuelberg,

480	H.E.; Blake, D.R.; Huey, G.; Weinheimer, A.J.; Knapp, D.J.; Brune, W.H. Emissions of
481	black carbon, organic, and inorganic aerosols from biomass burning in North America and
482	Asia in 2008. J. Geophys. Res-Atmos. 116; 2011.
483	Larsen, J.C.; Larsen, P.B. Chemical carcinogens. in: Hester R.E., Harrison, R.M., ed. Air Pollution
484	and Health. The Royal Society of Chemistry, Cambridge, UK; 1998.
485	Laursen, K.K.; Ferek, R.; Hobbs, P.; Rasmussen, R.A. Emission factors for particles, elemental
486	carbon, and trace gases from the Kuwait oil fires. J. Geophys. Res. 97:14491-14497; 1992.
487	Lee, S.; Baumann, K.; Schauer, J.J.; Sheesley, R.J.; Naeher, L.P.; Meinardi, S.; Blake, D.R.;
488	Edgerton, E.S.; Russell, A.G.; Clements, M. Gaseous and Particulate Emissions from
489	Prescribed Burning in Georgia. Environ. Sci. Technol. 39:9049-9056; 2005.
490	Ottmar, R.D.; Hudak, A.T.; Wright, C.S.; Vihnanek, R.E.; Restaino, A.T.; Restaino, J. Pre- and
491	post-fire surface fuel and cover measurements from experimental and operational
492	prescribed fires in long-leaf pine ecosystems of the southeastern United States-
493	RxCADRE. Int. J. Wildland Fire. Submitted; 2014.
494	Ottmar, R.D.; Vihnanek, R.E.; Wright, C.S.; Hudak, A.T. Ground measurements of fuel and fuel
495	consumption from experimental and operational prescribed fires at Eglin Air Force Base,
496	Florida. Proceedings of the 4th Fire Behavior and Fuels Conference. Raleigh, North
497	Carolina; 2013.
498	Ram, K.; Sarin, M.M.; Tripathi, S.N. Inter-comparison of thermal and optical methods for
499	determination of atmospheric black carbon and attenuation coefficient from an urban
500	location in northern India. Atmos. Res. 97:335-342; 2010.
501	Ramanathan, V.; Carmichael, G. Global and regional climate changes due to black carbon. Nat.

502 Geosci. 1:221-227; 2008.

503	Singh, D.P.; Gadi, R.; Mandal, T.K.; Saud, T.; Saxena, M.; Sharma, S.K. Emissions estimates of
504	PAH from biomass fuels used in rural sector of Indo-Gangetic Plains of India. Atmos.
505	Environ. 68:120-126; 2013.

- 506 Strand, T.; Gullett, B.K.; Urbanski, S.; Potter, B.; O'Neill, S.; Aurell, J.; Holder, A.; Larkin, N.;
- 507 Moore, M.; Rorig, M. Smoke and emissions measurements during RxCADRE 2012. Int. J.
 508 Wildland Fire. Submitted; 2014.
- 509 U.S. EPA Method 3540C. Soxhlet extraction. 1996.
 510 <u>http://www.epa.gov/osw/hazard/testmethods/sw846/pdfs/3540c.pdf</u> Accessed September
 511 9, 2014.
- U.S. EPA Compendium Method TO-9A. Determination of polychlorinated, polybrominated and
 brominated/chlorinated dibenzo-p-dioxins and dibenzofurans in ambient air. 1999a.
 <u>http://www.epa.gov/ttnamti1/files/ambient/airtox/to-9arr.pdf</u> Accessed November 21,
 2012.
- 516U.S. EPA Method TO-13A. Determination of polycyclic aromatic hydrocarbons (PAHs) in517ambient air using gas chromatography/mass spectrometry (GC/MS). 1999b.

518 <u>http://www.epa.gov/ttnamti1/files/ambient/airtox/to-13arr.pdf</u> Accessed May 5, 2014.

- 519 NIOSH Method 5040. Elemental Carbon (Diesel Particulate). Issue 3. NIOSH Manual of
 520 Analytical Methods, 4th Ed., 30 September. 1999c.
- 521 U.S. EPA. 1970-2012 Average annual emissions, all criteria pollutants.
 522 http://www.epa.gov/ttnchie1/trends/ October 17, 2012.
- 523 Urbanski, S.P.; Hao, W.M.; Baker, S. Chemical composition of wildland fire emissions. in:
 524 Developments in Environmental Science; Volume 8, Chapter 4, 79-107, 2009.

525	Van den Berg, M.; Birnbaum, L.S.; Denison, M.; De Vito, M.; Farland, W.; Feeley, M.; Fiedler,
526	H.; Hakansson, H.; Hanberg, A.; Haws, L.; Rose, M.; Safe, S.; Schrenk, D.; Tohyama, C.;
527	Tritscher, A.; Tuomisto, J.; Tysklind, M.; Walker, N.; Peterson, R.E. The 2005 World
528	Health Organization reevaluation of human and mammalian toxic equivalency factors for
529	dioxins and dioxin-like compounds. Toxicol. Sci. 93:223-241; 2006.
530	Yelverton, T.L.B.; Hays, M.D.; Gullett, B.K.; Linak, W.P. Black carbon measurements of flame-
531	generated soot as determined by optical, thermal-optical, direct absorption, and laser
532	incandescence methods. Environ. Eng. Sci. 31:209-215; 2014.
533	Yokelson, R.J.; Burling, I.R.; Gilman, J.B.; Warneke, C.; Stockwell, C.E.; de Gouw, J.; Akagi,
534	S.K.; Urbanski, S.P.; Veres, P.; Roberts, J.M.; Kuster, W.C.; Reardon, J.; Griffith, D.W.T.;
535	Johnson, T.J.; Hosseini, S.; Miller, J.W.; Cocker Iii, D.R.; Jung, H.; Weise, D.R. Coupling
536	field and laboratory measurements to estimate the emission factors of identified and
537	unidentified trace gases for prescribed fires. Atmos. Chem. Phys. Discuss. 12:21517-
538	21578; 2012.
539	Yokelson, R.J.; Burling, I.R.; Urbanski, S.P.; Atlas, E.L.; Adachi, K.; Buseck, P.R.; Wiedinmyer,
540	C.; Akagi, S.K.; Toohey, D.W.; Wold, C.E. Trace gas and particle emissions from open
541	biomass burning in Mexico. Atmos. Chem. Phys. 11:6787-6808; 2011.
542	Yokelson, R.J.; Karl, T.; Artaxo, P.; Blake, D.R.; Christian, T.J.; Griffith, D.W.T.; Guenther, A.;
543	Hao, W.M. The Tropical Forest and Fire Emissions Experiment: overview and airborne
544	fire emission factor measurements. Atmos. Chem. Phys. 7:5175-5196; 2007.