The triggering of myocardial infarction by fine particles is enhanced when particles are enriched in secondary species

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ABSTRACT

Previous studies have reported an increased risk of myocardial infarction (MI) associated with acute increases in PM concentration. Recently, we reported that MI/fine particle ($PM_{2.5}$) associations may be limited to transmural infarctions. In this study, we retained data on hospital discharges with a primary diagnosis of acute myocardial infarction (using International Classification of Diseases 9th Revision [ICD-9] codes), for those admitted January 1, 2004 to December 31, 2006, who were ≥18 years of age, and were residents of New Jersey at the time of their MI. We excluded MI with a diagnosis of a previous MI and MI coded as a subendocardial infarction, leaving n=1563 transmural infarctions available for analysis. We coupled these health data with $PM_{2.5}$ species concentrations predicted by the Community Multiscale Air Quality chemical transport model, ambient $PM_{2.5}$ concentrations, and used the same case-crossover methods to evaluate whether the relative odds of transmural MI associated with increased $PM_{2.5}$ concentration is modified by the $PM_{2.5}$ composition/mixture (i.e. mass fractions of sulfate, nitrate, elemental carbon, organic carbon, and ammonium). We found the largest relative odds estimates on the days with the highest tertile of sulfate mass fraction (OR=1.13; 95% CI = 1.00, 1.27), nitrate mass fraction (OR=1.18; 95% CI = 0.98, 1.35), and ammonium mass fraction (OR=1.13; 95% CI = 1.00 1.28), and the lowest tertile of EC mass fraction (OR=1.17; 95% CI = 1.03, 1.34). Air pollution mixtures on these days were enhanced in pollutants formed through atmospheric chemistry (i.e., secondary $PM_{2.5}$) and depleted in primary pollutants (e.g., EC). When mixtures were laden with secondary PM species (sulfate, nitrate, and/or organics) we observed larger relative odds of myocardial infarction associated with increased $PM_{2.5}$ concentrations. Further work is needed to confirm these findings and examine which secondary $PM_{2.5}$ component(s) is/are responsible for an acute MI response.

INTRODUCTION

Previous studies investigating triggering of myocardial infarction (MI) by particulate air pollution (PM) concentrations in the hours and days before MI onset have, in most cases, reported an increased risk of MI associated with increases in PM concentration on the same and previous day.[1-9] Recently, we reported that these MI/fine particle ($PM_{2.5}$) associations may be limited to full wall infarctions (i.e. transmural infarctions), and not subendocardial infarctions (i.e. non-transmural infarctions). Further, this association was independent of increases in nitrogen dioxide, sulfur dioxide, carbon monoxide, and ozone concentrations.[7] Whether the $PM_{2.5}$ composition/mixture (i.e. relative proportion of $PM_{2.5}$ mass that is sulfate, nitrate, ammonium, elemental carbon, or organic carbon) modifies this association has not been examined.

National studies have reported regional differences in PM mediated cardiovascular health effects, with larger relative risks of mortality or morbidity observed in the eastern US than in the western US.[10-14] These differences may be due to exposure error resulting from regional differences in the efficiency with which ambient $PM_{2.5}$ penetrates into and persists indoors.[15-18] Alternatively, others have argued that differences in effect estimates across regions of the United States occur because of differential effects of $PM_{2.5}$ species/components. These studies of health effects associated with individual PM components (e.g. sulfates, nitrates, elemental carbon, etc.) have been summarized previously.[19] In some cases, secondary species (i.e., formed through atmospheric photochemistry, e.g., sulfate) have been implicated and others have implicated primary species. Studies have reported greater mortality rates on days with elevated primary nickel, vanadium, and elemental carbon,[10] primary nickel, vanadium, and secondary sulfate concentrations across 60 US cities,[20] primary silicon, aluminum, and arsenic across 25 US

cities,[12] and primary bromine, chromium, and sodium ion across 27 US cities.[21] Others have directly estimated the change in mortality or morbidity associated with specific PM species and reported associations between elemental carbon and cardiovascular (CV) mortality in Phoenix, Arizona,[22] elemental carbon, organic carbon (primary plus secondary), iron, and potassium and CVD mortality in California,[23] and elemental carbon and CV hospitalizations across 119 US communities.[24] Recently, Kim et al (2012) reported that increased EC and OC, but not nitrate and sulfate, were associated with increased ischemic heart disease admissions in Denver, Colorado.[25] Although one or more PM component(s) does not clearly stand out as driving these air pollution mediated cardiovascular effects, these studies suggest that mixtures enhanced with specific particle components may be associated with a larger biologic response than others.

Our primary analysis[7] took advantage of continuous $PM_{2.5}$ monitoring at 7 monitoring sites across New Jersey, but limited monitoring of $PM_{2.5}$ species (i.e. only every 3rd day) did not permit us to examine how acute CV responses are affected by variations in $PM_{2.5}$ composition. Therefore, we used $PM_{2.5}$ species concentrations predicted by the Community Multiscale Air Quality chemical transport model,[26] ambient $PM_{2.5}$ mass concentrations measured at 7 continuous monitoring sites across the state, and the same case-crossover design and dataset as in our previous analysis,[7] to estimate the relative odds of transmural MI associated with increased $PM_{2.5}$ concentration on days with varying PM compositions. In this work, we did not estimate the relative odds of transmural MI associated with increases in the concentration of individual components, but rather focused on the whether the relative odds of MI associated with increased $PM_{2.5}$ concentration was different on days with differing particle composition (i.e., different proportions of $PM_{2.5}$ mass that were sulfate, nitrate, ammonium, elemental carbon, and organic carbon). We hypothesized that the relative odds of transmural MI associated with increased

103 $PM_{2.5}$ concentration would be modified by the composition of $PM_{2.5}$ (i.e. mass fractions of
104 sulfate, nitrate, ammonium, elemental carbon, organic carbon).
105
106 **METHODS**
107 ***Study population.*** The study population and definition of transmural infarction used in
108 this study have been described previously.[7] Briefly, we used the Myocardial Infarction Data
109 Acquisition System (MIDAS), a New Jersey statewide database that combines hospital discharge
110 data and death certificate registration data,[27] and extracted all records with a primary diagnosis of
111 acute myocardial infarction (International Classification of Diseases 9th Revision [ICD-9] code
112 410.01, 410.11, 410.21, 410.31, 410.41, 410.51, 410.61, 410.71, 410.81, 410.91), for patients
113 who were admitted between January 1, 2004 and December 31, 2006, were ≥18 years of age, and
114 were residents of NJ at the time of their MI. We excluded those MI with a diagnosis of a
115 previous MI, and those MI coded as a subendocardial infarction (410.7), leaving n=1563
116 transmural infarctions available for analysis. This study was approved by the University of
117 Medicine and Dentistry of New Jersey Institutional Review Board and the University of
118 Rochester Research Subjects Review Board. MIDAS was also approved by the New Jersey
119 Department of Health and Senior Services Institutional Review Board.
120 ***Ambient $PM_{2.5}$ and weather data.*** We used the same $PM_{2.5}$ mass concentrations and
121 weather data for each subject as in the previous analysis of these data.[7] In summary, we used
122 ambient hourly $PM_{2.5}$ mass concentrations, measured with a tapered element oscillating
123 microbalance (TEOM), retrieved from a United States Environmental Protection Agency
124 website,[28] for 7 monitoring stations for the study period (January 1, 2004 to December 31, 2006).
125 For each patient, we assigned TEOM $PM_{2.5}$ measurements from the closest monitor to their

126 residence, with those living greater than 10 km from a $PM_{2.5}$ monitoring station excluded from
127 all analyses (i.e. the same study subjects as in our previous work[7]). We assigned hourly
128 temperature and dew point measurements from the weather station (Newark, Caldwell, Somerset,
129 and Trenton airports) closest to each patient's residence, and then calculated the mean apparent
130 temperature[8, 29] in the 24 hours before the MI, as a measure of each patient's perceived air
131 temperature given the humidity, and used these values in all analyses.

132 ***Modeled $PM_{2.5}$ Species***. Ambient $PM_{2.5}$ species concentrations used in this study (sulfate
133 [SO_4^{2-}], nitrate [NO_3^-], ammonium [NH_3^+], elemental carbon [EC], organic carbon [OC] and
134 remaining other $PM_{2.5}$ mass) were simulated with CMAQ model Version 4.7. This model used
135 MM5 Version 3.7.4 meteorology (34 vertical layers) and gridded emissions of primary $PM_{2.5}$ and
136 precursors to secondary $PM_{2.5}$.[26, 30] The National Emissions Inventory (NEI) was the primary
137 basis for emissions. Hour-specific continuous emission monitoring systems data were used for
138 electric generating units. Hour-specific updates to mobile emissions were performed using the
139 MOBILE6 model, and daily estimates of fire emissions based on satellite detection of fires were
140 included. Monthly NH_3 emissions from livestock were by inverse modeling.[36] The AERO5
141 aerosol module,[31] Carbon-Bond 05 (CB05) chemical mechanism with chlorine chemistry
142 extensions,[32] and the ACM2 PBL scheme[33, 34] were also used. CMAQ simulations were
143 performed with 36 km continental horizontal grid spacing and 12 km grid resolution for the
144 eastern two-thirds of the U.S. Chemical boundary conditions were obtained from GEOS-
145 Chem.[35] Aerosol transport, atmospheric chemistry and secondary $PM_{2.5}$ formation were
146 simulated to provide hourly CMAQ $PM_{2.5}$ species concentrations.

147 CMAQ is known to exhibit seasonal biases in its $PM_{2.5}$ mass and $PM_{2.5}$ species outputs.[37]
148 To address this, the CMAQ output used in this study was adjusted using a statistical space/time

149 bias-correction model.[38] Briefly, EPA's $PM_{2.5}$ mass and species data from New Jersey and
150 surrounding states were used to correct the CMAQ bias on spatial scales of 50 km and temporal
151 scales of one month. The model regressed each monitor observation (PM mass or species) on
152 the CMAQ output for the appropriate $PM_{2.5}$ component, grid cell, and day and used these
153 relationships to adjust the CMAQ concentrations. Biases were allowed to vary in space and time
154 through the use of quadratic splines, and the model was constrained by requiring predicted
155 concentrations to be non-negative and requiring mass closure (sum of species equal total $PM_{2.5}$
156 mass). Bias-correction substantially attenuated the seasonal bias trends for all species. For
157 example, after bias correction, the maximum absolute monthly mean bias in the $PM_{2.5}$ mass
158 concentration (model minus monitor) decreased from 6 $\mu g/m^3$ to 1 $\mu g/m^3$. The maximum
159 absolute monthly mean bias for nitrate and organic carbon was reduced from 2.1 $\mu g/m^3$ to 0.8
160 $\mu g/m^3$ and from 3.3 $\mu g/m^3$ to 1.0 $\mu g/m^3$, respectively.

161 Hourly CMAQ concentrations were averaged over each day of the study period. Subjects
162 were assigned these daily surface-level (~0-36 m) $PM_{2.5}$ mass and $PM_{2.5}$ species concentrations
163 for the CMAQ grid-cell containing the ambient monitor nearest to their residences. We then used
164 the CMAQ daily concentrations to calculate the daily CMAQ $PM_{2.5}$ species mass fractions used
165 in the statistical analyses.

166 ***Study Design.*** We used the same time-stratified case-crossover design,[39, 40] as in our
167 previous analysis,[7] to estimate the risk of a transmural infarction associated with increased
168 TEOM $PM_{2.5}$ concentrations in the24 hours before emergency department arrival In this design,
169 each patient contributes information as a case during the period immediately before the MI, and
170 as a matched control during times when a MI did not occur. The case-crossover design is
171 analogous to a matched case-control study, but instead of estimating the relative risk of MI

172 contrasting ambient TEOM $PM_{2.5}$ concentrations between persons (i.e. cases versus controls), we
173 estimate the relative risk of MI contrasting TEOM $PM_{2.5}$ concentrations during different time
174 periods within the follow-up time of each case of MI. Because case periods and their matched
175 control periods are derived from the same person and a conditional analysis is conducted, non-
176 time varying confounders such as age, co-morbidities, and long term smoking history are
177 controlled by design. However, variables that may be related to both air pollution and the
178 incidence of MI that vary over short time periods (e.g., weather conditions) are possible
179 confounders that must be included in our analytic models. Case periods were defined as the 24
180 hour period before ER admission for MI, while control periods (3-4 per case depending on the
181 number of days in the calendar month) were matched to the case period by day of the week, time
182 of the day, year, and month. Pollutant concentrations corresponding to these case and control
183 periods are then contrasted in the statistical analyses.

184 ***Statistical Analyses - Main Analyses.*** First, for each day during the study period, we
185 calculated the daily sulfate mass fraction as the CMAQ sulfate concentration ($\mu g/m^3$) divided by
186 the CMAQ $PM_{2.5}$ concentration ($\mu g/m^3$) resulting in a proportion between 0 and 1. We then
187 ranked the sulfate mass fractions for all case and control periods into tertiles (i.e. looking at the
188 proportion of CMAQ $PM_{2.5}$ mass that is sulfate for each day [sulfate mass fraction with a value
189 from 0 to 1], the HIGH TERTILE equaled the days with the highest third of sulfate mass
190 fractions, the MIDDLE TERTILE equaled the days with the middle third of sulfate mass
191 fractions, and the LOW TERTILE equaled the days with the lowest third of sulfate mass
192 fractions). We repeated this mass fraction calculation and ranking procedure for the nitrate,
193 elemental carbon, organic carbon, and ammonium mass fractions. We then calculated descriptive
194 statistics for these species concentrations and species mass fractions. We also calculated Pearson

correlation coefficients for each pair of: TEOM $PM_{2.5}$; CMAQ sulfate, nitrate, ammonium, elemental carbon, an organic carbon mass fractions; and nitrogen dioxide, carbon monoxide, sulfur dioxide, and ozone concentrations. We repeated this for the summer months (June-August), and winter months (December-February).

Second, we used the same conditional logistic regression model as in our previous analysis[7] stratified on each MI, to regress case–control status (i.e., case period = 1, control period = 0) against the mean TEOM $PM_{2.5}$ concentration in the 24 hr before ED arrival, including a natural spline (3 degrees of freedom) of the mean apparent temperature in the 48 hr before ED arrival in the model. Next, we estimated the risk of a transmural infarction associated with each interquartile range increase (10.8 $\mu g/m^3$) in TEOM $PM_{2.5}$ concentration on days with the highest third, middle third, and lowest third of daily sulfate mass fractions. To the same model described above, we added indicator variables for sulfate tertile (MEDIUM_TERTILE + HIGH_TERTILE) and two interaction terms (TEOM $PM_{2.5}$*MIDDLE_TERTILE _+ TEOM $PM_{2.5}$*HIGH TERTILE). From this model, we estimated the relative odds of a transmural infarction associated with each 10.8 $\mu g/m^3$ increase in TEOM $PM_{2.5}$ concentration when the sulfate mass fraction is in the highest tertile, when it is in the middle tertile, and the lowest tertile. We repeated this analysis for the nitrate, elemental carbon, organic carbon, and ammonium mass fractions.

Third, within each tertile of sulfate mass fraction, we tabulated the mean CMAQ $PM_{2.5}$ species mass balance (i.e., percent of CMAQ $PM_{2.5}$ mass that is sulfate, nitrate, elemental carbon, organic carbon, other), and descriptive statistics of gaseous pollutant concentrations, temperature, and dew point. We repeated this for each nitrate, elemental carbon, organic carbon, and ammonium tertile.

Last, we examined the appropriateness of pooling the data from the 7 monitoring sites to estimate the relative odds of a transmural infarction associated with each IQR increase in TEOM $PM_{2.5}$ in a single statistical model (i.e. our main analysis described above). Alternatively, we could estimate 7 relative odds estimates separately, and then combine them via a meta-analysis technique used previously in a case-crossover study of $PM_{2.5}$ and mortality in 27 cities.[13] Therefore, we added 6 interaction terms to the model described above for the 7 TEOM $PM_{2.5}$ monitoring sites (e.g. TEOM $PM_{2.5}$ *Monitoring site #1; TEOM $PM_{2.5}$*Monitoring site #2, etc.)). We then tested whether the relative odds of transmural MI associated with each IQR increase in TEOM $PM_{2.5}$ was different for subjects residing near different monitoring locations using a Likelihood Ratio Test. If a Likelihood Ratio Test indicated significant modification of the $PM_{2.5}$ effect by monitoring site, then we would estimate 7 relative odds estimates separately, combine them using meta-analysis techniques, and compare this estimate to that from our main analysis described above. We used SAS (version 9.1.3; SAS Institute Inc., Cary, NC) and R software (version 2.6.1; R Foundation for Statistical Computing, Vienna, Austria) for all statistical analyses.

RESULTS

The characteristics of the patients with a transmural MI included in the study are shown in Table 1. Subjects were predominantly male (63%), white (69%), with 45% 65 years of age and older, and 26% 75 years of age and older. Fifty five percent had hypertension (55%), 27% had diabetes, and 67% had a history of ischemic heart disease (67%).

Daily TEOM $PM_{2.5}$ mass concentrations, bias-adjusted CMAQ $PM_{2.5}$ and CMAQ species concentrations, and CMAQ species mass fractions (bias-adjusted CMAQ) are summarized in

Table 2. We deleted one case from our analysis as the CMAQ $PM_{2.5}$ and species concentrations were very large due to a large fire in that grid, leaving n=1562 MI for analyses. On average, these five species represented 81% of the total bias-adjusted CMAQ estimated $PM_{2.5}$ mass concentration. The remainder consisted of minor ions such as sodium and chloride, metal oxides, non-carbon organic mass, and some unspeciated material from primary emission sources (e.g. soil, combustion).[41] At the median, sulfate and organic carbon comprised larger proportions of total bias-adjusted CMAQ $PM_{2.5}$ mass (23% each) than ammonium and nitrate (13%), with elemental carbon being the smallest contributor (9%). Pearson correlation coefficients, provided for each pair of TEOM $PM_{2.5}$, CMAQ $PM_{2.5}$ mass fractions, and gaseous pollutant concentrations in Table 3 (entire study period) and Table 4 (summer and winter only), align with expectations based on over 20 years of speciated air quality observations.[42] Mass fractions of sulfate, which peak in the summer because of sulfate's photochemical formation, were moderately, but negatively, correlated with those of nitrate, which is more volatile and enhanced at low temperature (r= - 0.61) and elemental carbon (r= -0.47) which is primary and peaks in the winter. The sulfate mass fraction was positively correlated with ammonium (r=0.51; Table 3). Ammonium was negatively correlated with elemental and organic carbon (r= -0.60 and -0.65 respectively)mass fractions which were weakly, but positively correlated with each other (r=0.35).

NO_2, SO_2 and CO were moderately positively correlated with each other (r>0.5) and weakly but negatively correlated with O_3 (r< -0.33). The correlation between O_3 and EC mass fraction was weak and negative (r= -0.32). Somewhat stronger associations were observed between O_3 and the nitrate (negative; r= -0.43) and sulfate (positive; r=0.45) mass fractions. Similar features were found when correlations were computed by season (Table 4). However,

the correlations of sulfate mass fraction with EC (negative; r= -0.77) and ammonium (positive; r=0.62) mass fractions were stronger in the summer-only data. The negative correlation of sulfate with nitrate mass fraction was stronger in the winter (r= -0.71), whereas there was no correlation between EC and sulfate mass fractions in the winter (r=0.10). In the winter, nitrate mass fraction was negatively correlated with EC (r= -0.49), OC (r= -0.43), and sulfate (r= -0.71) mass fractions. The negative correlation between O_3 and EC mass fraction was somewhat stronger in the summer-only analysis (r= -0.40) and was not observed in winter. A wintertime positive correlation between O_3 and OC mass fraction (r=0.42) is apparent in the seasonally-segregated analysis.

Next, we estimated the relative odds of a transmural infarction associated with each interquartile range increase (10.8 $\mu g/m^3$) in ambient TEOM $PM_{2.5}$ concentration in the previous 24 hours within tertiles of sulfate, ammonium, nitrate, elemental carbon, and organic carbon mass fractions on the day of the MI (Table 5). Effect estimates across tertiles were generally similar for sulfate and nitrate, with the largest and only statistically significant ($p<0.05$) or marginally significant ($p<0.10$) increased relative odds of a transmural infarction associated with increased TEOM $PM_{2.5}$ concentration within the highest sulfate and nitrate tertiles. Similarly, the highest and only statistically significant increased relative odds estimate for the ammonium mass fraction was within the highest tertile, with no increased relative odds within the lowest tertile. In contrast, the highest relative odds estimate for the elemental carbon mass fraction was within the lowest tertile. Organic carbon exhibited more complex behavior, which is not surprising considering the vast array of compounds that it contains. The highest relative odds estimate was within the middle tertile of organic carbon mass fraction (Table 5).

Next, we examined the composition of the pollutant mixture (both $PM_{2.5}$ composition and gaseous pollutant concentrations) within each tertile. Days in the high sulfate, high ammonium, and low elemental carbon tertiles (Table 6, Figure 1) had very similar compositions. For example, on days in the high sulfate tertile, CMAQ $PM_{2.5}$ was 33% sulfate, 22% OC, 14% ammonium, 10% nitrate, and 7% EC, on average, and the median temperature and relative humidity were 21.1°C and ~72%, respectively. On these days, median 8 hour maximum NO_2, SO_2, CO, and O_3 concentrations were 24.5 ppb, 5.0 ppb, 0.638 ppm, and 43.6 ppb, respectively (Table 6). Days in the low elemental carbon tertile, had similar CMAQ $PM_{2.5}$ composition with 29% sulfate, 21% OC, 14% ammonium, 14% nitrate, and 6% EC, on average, and the median temperature and relative humidity were also typical of summertime (18.4°C and ~67% RH). On these days, 8 hour maximum NO_2, SO_2, CO, and O_3 concentrations were also similar to the days with high sulfate (Table 6). In contrast, days in the low sulfate, low ammonium, and high elemental carbon tertiles generally had lower average sulfate mass fractions (17% to 22%), and median 8 hour maximum O_3 concentrations (25.7 ppb to 29.3 ppb), but higher average elemental carbon (10% to 11%) and organic carbon mass fractions (24% to 28%);Figure 2 and Table 6). Median 8 hour maximum NO_2, SO_2, and CO concentrations were similar to the high sulfate, high ammonium, and low elemental carbon tertile days (Table 6). Temperature and median relative humidity were lower and more typical of wintertime.

Comparison of high and low nitrate and organic carbon tertiles are presented in Figure 3. In comparison to the high sulfate tertile (Figure 1), the high nitrate tertile was depleted in sulfate and enriched in nitrate, with a substantially colder median temperature (5°C). Note that sulfate (summertime maximum) and nitrate (wintertime maximum) were both associated with ammonium, as these species are frequently present as ammonium salts.[42,43]

309 Days in the low organic carbon tertile had modestly larger contributions of EC and
310 nitrate, a slightly smaller contribution of OC, and a lower median temperature (13.6°C; Figure 3
311 and Table 3) compared to the high sulfate, high ammonium, and low elemental carbon tertile
312 days (Figure 1). Days in the high organic carbon tertile had larger contributions of elemental and
313 organic carbon, lower contributions of sulfate, nitrate, ammonium, and higher median NO_2, CO,
314 and SO_2 concentrations then the low organic carbon tertile (Figure 3 and Table 3). The
315 composition of the high organic carbon tertile was different from that of both the high and low
316 sulfate tertiles, presumably either because of differences in sources source regions, or formation
317 chemistry.
318 Last, we found that the relative odds of a transmural infarction associated with each IQR
319 increase in TEOM $PM_{2.5}$ concentration in the previous 24 hours was not significantly different
320 across the 7 monitoring sites (all $p>0.84$). Thus, this meta-analysis approach would not give
321 substantially different results from those described above, supporting our main analysis
322 methodology.
323
324 **DISCUSSION**
325 Using a combination of daily $PM_{2.5}$ species mass fractions estimated by the seasonally
326 bias-adjusted CMAQ model, ambient TEOM $PM_{2.5}$ concentrations from 7 continuous monitoring
327 sites, and MI hospital admissions data across New Jersey from 2004 to 2006, we evaluated
328 whether the relative odds of a transmural infarction associated with each 10.8 $\mu g/m^3$ increase in
329 TEOM $PM_{2.5}$ concentration in the previous 24 hours was modified by the $PM_{2.5}$ composition or
330 mixture (i.e. whether effect of TEOM $PM_{2.5}$ on MI was different when $PM_{2.5}$ was composed of a
331 high fraction of sulfate, nitrate, ammonium, EC or OC vs. days when the $PM_{2.5}$ composition

(mixture) contained a low fraction of each of these species). It should be noted that $PM_{2.5}$ composition simulated on a 12x12 km grid captures the *urban* mix of primary and secondary $PM_{2.5}$ species but does not capture the enhanced contribution of primary emissions in close proximity to sources. For example, the contributions to $PM_{2.5}$ of several carbonaceous species are substantially higher within 100 m of a major roadway (e.g., Polidori et al., 2010).[45] The analyses herein are restricted to people who live within 10 km of an urban $PM_{2.5}$ monitor and include predicted primary and secondary $PM_{2.5}$. Thus, this paper addresses modification of PM effect estimates by differences in community-level $PM_{2.5}$ composition, and does not address the enhanced contribution of primary emissions to PM composition in close proximity to primary sources. We found the largest relative odds estimates on the days with the highest tertile of sulfate, nitrate, and ammonium, and the lowest tertile of EC. The air pollution mixtures on the days in these tertiles were all enhanced by $PM_{2.5}$ pollutant species that are formed through atmospheric chemistry (i.e., secondary $PM_{2.5}$ formed through gas and/or aqueous photochemistry)and depleted in primary $PM_{2.5}$ pollutants (in a relative sense, e.g., low $EC/PM_{2.5}$).

OC exhibited complex behavior, with the highest relative odds estimate occurring for the middle tertile. The complex behavior of OC is not surprising. OC is comprised of thousands of compounds with a wide range of physical and chemical properties. It is both emitted directly (primary)[46,47] and also formed in the atmosphere (secondary).[48,49] Moreover, EC is a good tracer for primary OC. Sulfate exhibits strong correlations with low volatility oxygenated organic aerosol (a major component of secondary organic aerosol) and oxalate (a tracer for secondary organic aerosol formed through gas followed by aqueous chemistry, known as "aqueous SOA"), probably because all three are formed through atmospheric aqueous chemistry.[45,50-53]

355 Additionally, particulate organosulfates are known to form in wet aerosols that contain acidic
356 sulfate.[54] Thus a variable portion of OC behaves like nitrate or sulfate, while another portion
357 behaves like EC and total particulate OC is not highly correlated with any other mass fraction
358 (Table 4). Because secondary "aqueous" OC is enriched on days with low EC and high sulfate,
359 this greater acute MI response could be associated with sulfate, nitrate and/or secondary organic
360 aerosol, including SOA formed through aqueous chemistry.

361 Thus, we found that the relative odds of a transmural MI associated with $PM_{2.5}$ is greater
362 during times of greater secondary aerosol formation. This is consistent with the hypothesis that
363 mixtures laden with secondary PM species are associated with increased incidence of myocardial
364 infarctions and perhaps with other acute cardiovascular outcomes. Previous studies, done across
365 the United States, have reported increased risk of cardiovascular mortality and morbidity
366 associated with increased sulfate or nitrate concentrations.[23, 24, 55,56] Our findings are consistent
367 with these earlier results that have reported greater response to secondary PM species (sulfate,
368 nitrate, and/or organic matter). Note that we did not estimate the relative odds of transmural
369 infarction associated with increases in individual $PM_{2.5}$ species, and instead assessed whether the
370 MI/$PM_{2.5}$ association was modified by $PM_{2.5}$ species mass fractions. Many studies have reported
371 increased relative risks associated with primary PM, traffic sources, or markers of traffic
372 pollution.[23-25, 55,57-58] For example, Kim et al (2012) found increased ischemic heart disease
373 admissions were associated with increased EC and OC concentrations in the previous day, but
374 not with increased sulfate and nitrate concentrations.[25] The one component that both primary and
375 secondary PM have in common is organic matter.

376 Our study had several limitations that should be noted. First, we were only able to use
377 central site $PM_{2.5}$ mass concentrations, likely resulting in both Berkson and classical error,[60,61]

resulting in a bias towards the null, underestimating the risk of transmural MI associated with increased $PM_{2.5}$ concentration. Second, we used daily $PM_{2.5}$ species mass fractions averaged over 12 km by 12 km grids, and assigned them to each subject by residential location. This could have resulted in underestimates of concentrations of primary species (e.g., EC) for residences in close proximity to sources (e.g., < 200 m from a major roadway),[45] which may have resulted in some error in placing subjects in high, middle, and low EC tertiles. However, this error is unlikely to cause a large number of subjects to be incorrectly placed in the 'high' EC tertile when in fact they should have been in the 'low' EC tertile, and vice-versa. Third, we were unable to examine trace elements (e.g. nickel, vanadium, aluminum, etc.) that have been either associated with increased risk of CV events directly, or shown to modify PM/CV associations.[10, 12, 20, 21]

We evaluated whether the relative odds of a transmural infarction associated with each 10.8 $\mu g/m^3$ increase in $PM_{2.5}$ concentration in the previous 24 hours was different when the mass fractions of sulfate, nitrate, ammonium, EC and OC on that day were high versus low. We found the largest relative odds estimates on the days with the highest tertile of sulfate, nitrate, and ammonium, and the lowest tertile of EC, suggesting these effects are greatest on days when the mixture is enhanced with secondary PM. Note, secondary species typically make up the bulk of $PM_{2.5}$ in New Jersey.[43] Further work is needed to investigate which secondary species (i.e. sulfate, nitrate, ammonium, secondary organic species formed through gas and/or aqueous chemistry, organosulfates, reactive species carried in aerosol water such as peroxides) is/are responsible for this finding.

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REFERENCES

1. Berglind N, Ljungman P, Moller J, Hallqvist J, Nyberg F, Rosenqvist M, Pershagen G, Bellander T. Air pollution exposure--a trigger for myocardial infarction? *Int J Environ Res Public Health*. 2010;7:1486-1499.
2. D'Ippoliti D, Forastiere F, Ancona C, Agabiti N, Fusco D, Michelozzi P, Perucci CA. Air pollution and myocardial infarction in rome: A case-crossover analysis. *Epidemiology*. 2003;14:528-535.
3. Mustafic H, Jabre P, Caussin C, Murad MH, Escolano S, Tafflet M, Perier MC, Marijon E, Vernerey D, Empana JP, Jouven X. Main air pollutants and myocardial infarction: A systematic review and meta-analysis. *JAMA*. 2012;307:713-721.
4. Peters A, Dockery DW, Muller JE, Mittleman MA. Increased particulate air pollution and the triggering of myocardial infarction. *Circulation*. 2001;103:2810-2815.
5. Peters A, von Klot S, Heier M, Trentinaglia I, Cyrys J, Hormann A, Hauptmann M, Wichmann HE, Lowel H. Particulate air pollution and nonfatal cardiac events. Part I. Air pollution, personal activities, and onset of myocardial infarction in a case-crossover study. *Res Rep Health Eff Inst*. 2005:1-66; discussion 67-82, 141-148.
6. Pope CA, 3rd, Muhlestein JB, May HT, Renlund DG, Anderson JL, Horne BD. Ischemic heart disease events triggered by short-term exposure to fine particulate air pollution. *Circulation*. 2006;114:2443-2448.
7. Rich DQ, Kipen HM, Zhang J, Kamat L, Wilson AC, Kostis JB. Triggering of transmural infarctions, but not nontransmural infarctions, by ambient fine particles. *Environ Health Perspect*. 2010;118:1229-1234.

8. Zanobetti A, Schwartz J. The effect of particulate air pollution on emergency admissions for myocardial infarction: A multicity case-crossover analysis. *Environ Health Perspect*. 2005;113:978-982.

9. Sullivan J, Sheppard L, Schreuder A, Ishikawa N, Siscovick D, Kaufman J. Relation between short-term fine-particulate matter exposure and onset of myocardial infarction. *Epidemiology*. 2005;16:41-48.

10. Bell ML, Ebisu K, Peng RD, Samet JM, Dominici F. Hospital admissions and chemical composition of fine particle air pollution. *Am J Respir Crit Care Med*. 2009;179:1115-1120.

11. Bell ML, Ebisu K, Peng RD, Walker J, Samet JM, Zeger SL, Dominici F. Seasonal and regional short-term effects of fine particles on hospital admissions in 202 US counties, 1999-2005. *Am J Epidemiol*. 2008;168:1301-1310.

12. Franklin M, Koutrakis P, Schwartz P. The role of particle composition on the association between $PM_{2.5}$ and mortality. *Epidemiology*. 2008;19:680-689.

13. Franklin M, Zeka A, Schwartz J. Association between $PM_{2.5}$ and all-cause and specific-cause mortality in 27 US communities. *J Expo Sci Environ Epidemiol*. 2007;17:279-287.

14. Dominici F, Peng RD, Bell ML, Pham L, McDermott A, Zeger SL, Samet JM. Fine particulate air pollution and hospital admission for cardiovascular and respiratory diseases. *JAMA*. 2006;295:1127-1134.

15. Long CM, Suh HH, Catalano PJ, Koutrakis P. Using time- and size-resolved particulate data to quantify indoor penetration and deposition behavior. *Environ Sci Technol*. 2001;35:2089-2099.

16. Meng QY, Turpin BJ, Polidori A, Lee JH, Weisel C, Morandi M, Colome S, Stock T, Winer A, Zhang J. $PM_{2.5}$ of ambient origin: Estimates and exposure errors relevant to PM epidemiology. *Environ Sci Technol*. 2005;39:5105-5112.

17. Baxter LK, Ozkaynak H, Franklin M, Schultz BD, Neas LM. The use of improved exposure factors in the interpretation of fine particulate matter epidemiological results. *Air Quality and Atmospheric Health*. 2012 (in press).

18. Hodas N, Meng Q, Lunden MM, Rich DQ, Ozkaynak H, Baxter LK, Zhang Q, Turpin BJ. Variability in the fraction of ambient fine particulate matter found indoors and observed heterogeneity in health effect estimates. *J Expo Sci Environ Epidemiol*. 2012;22:448-454.

19. Rohr AC, Wyzga RE. Attributing health effects to individual particulate matter constituents. *Atmospheric Environment*. 2012;62:130-152.

20. Lippmann M, Ito K, Hwang JS, Maciejczyk P, Chen LC. Cardiovascular effects of nickel in ambient air. *Environ Health Perspect*. 2006;114:1662-1669.

21. Zanobetti A, Franklin M, Koutrakis P, Schwartz J. Fine particulate air pollution and its components in association with cause-specific emergency admissions. *Environ Health*. 2009;8:58.

22. Mar TF, Norris GA, Koenig JQ, Larson TV. Associations between air pollution and mortality in phoenix, 1995-1997. *Environ Health Perspect*. 2000;108:347-353.

23. Ostro B, Feng WY, Broadwin R, Green S, Lipsett M. The effects of components of fine particulate air pollution on mortality in california: Results from calfine. *Environ Health Perspect*. 2007;115:13-19.

24. Peng RD, Bell ML, Geyh AS, McDermott A, Zeger SL, Samet JM, Dominici F. Emergency admissions for cardiovascular and respiratory diseases and the chemical composition of fine particle air pollution. *Environ Health Perspect*. 2009;117:957-963.

25. Kim SY, Peel JL, Hannigan MP, Dutton SJ, Sheppard L, Clark ML, Vedal S. The temporal lag structure of short-term associations of fine particulate matter chemical constituents and cardiovascular and respiratory hospitalizations. *Environ Health Perspect*. 2012;120:1094-1099.

26. Byun DS, Schere KL Review of the governing equations, computational algorithms, and other components of the models-3 community multiscale air quality (CMAQ) modeling system *Applied Mechanics Reviews* 2006;59 51-77.

27. Kostis JB, Wilson AC, Lacy CR, Cosgrove NM, Ranjan R, Lawrence-Nelson J. Time trends in the occurrence and outcome of acute myocardial infarction and coronary heart disease death between 1986 and 1996 (a New Jersey statewide study). *Am J Cardiol*. 2001;88:837-841.

28. United States Environmental Protection Agency . Technology transfer network - AQS Datamart. Available: http://www.epa.gov/ttn/airs/aqsdatamart/index.html.

29. Steadman RG. The assessment of sultriness. Part II: Effects of wind, extra radiation and barometric pressure on apparent temperature. *Journal of Applied Meteorology*. 1979;18:874-885.

30. United States Environmental Protection Agency. Community Multiscale Air Quality Model. Available: http://www.epa.gov/AMD/CMAQ/cmaq_model.html.

31. Foley, KM, Roselle SJ, Appel KW, Bhave PV, Pleim JE, Otte TL, Mathur R, Sarwar G, Young JO, Gilliam CG, Kelly JT, Gilliland AB, Bash JO. Incremental testing of the Community Multiscale Air Quality (CMAQ) modeling system version 4.7. www.geosci-model-dev.net/3/205/2010/.

32. Yarwood G, Roa S, Yocke M, Whitten G. Updates to the carbon bond chemical mechanism:. Final report to the US EPA, rt-0400675. 2005

33. Pleim JE. A combined local and nonlocal closure model for the atmospheric boundary layer.Part i: Model description and testing, . *Journal of Applied Meteorology and Climatology*. 2007;46:1383-1395.

34. Pleim JE. A combined local and nonlocal closure model for the atmospheric boundary layer. Part ii: Application and evaluation in a mesoscale meteorological model. *J. Appl. Meteor. Clim.,* . 2007;46:1396-1907.

35. Bey I, Jacob, D.J., Yantosca, R.M., Logan, J.A., Field, B.D., Fiore, A.M., Li, Q., Liu, H.Y., Mickley, L.J., and Schultz, M.G. Global modeling of tropospheric chemistry with assimilated meteorology: Model description and evaluation. *Journal of Geophysical Research*. 2001;106:23,073-023,009.

36. Gilliland AB, Appel, K.W., Pinder, R., Dennis, R.L. . Seasonal nh3 emissions for the continental united states: Inverse model estimation and evaluation. *Atmospheric Environment*. 2006;40:4986-4998

37. Appel K, Bhave, P., Gilliland, A., Sarwar, G., and Roselle, S. . Evaluation of the community multiscale air quality (cmaq) model version 4.5: Sensitivities impacting model performance; part ii - particulate matter. . *Atmospheric Environment* 2008:6054-6066.

523 38. Crooks JL, and Özkaynak, H. Simultaneous statistical bias correction of multiple $PM_{2.5}$
524 species from a regional photochemical grid model. *J Environmental and Ecological*
525 *Statistics (in press),* 2012.
526 39. Levy D, Lumley T, Sheppard L, Kaufman J, Checkoway H. Referent selection in case-
527 crossover analyses of acute health effects of air pollution. *Epidemiology*. 2001;12:186-
528 192.
529 40. Maclure M. The case-crossover design: A method for studying transient effects on the
530 risk of acute events. *Am J Epidemiol*. 1991;133:144-153.
531 41. Reff A, Bhave PV, Simon H, Pace TG, Pouliot GA, Mobley JD, Houyoux M. Emissions
532 inventory of $PM_{2.5}$ trace elements across the United States. *Environ Sci Technol*
533 2009;43:5790-5796.
534 42. NARSTO (2004) Particulate Matter Science for Policy Makers: A NARSTO Assessment.
535 P. McMurry, M. Shepherd, and J. Vickery, eds. Cambridge University Press, Cambridge,
536 England. ISBN 0 52 184287 5.
537 43. U.S. EPA (U.S. Environmental Protection Agency). 2009. Integrated Science Assessment
538 for Particulate Matter. Research Triangle Park, NC: National Center for Environmental
539 Assessment, Office of Research and Development.
540 44 Seinfeld JH and Pandis SN. 2006 Atmospheric Chemistry and Physics - From Air
541 Pollution to Climate Change (2nd Edition). John Wiley & Sons.
542 (http://www.knovel.com/web/portal/browse/display?_EXT_KNOVEL_DISPLAY_booki
543 d=2126&VerticalID=0).

45. Polidori A, Kwon J, Turpin BJ, Weisel C. Source proximity and residential outdoor concentrations of PM(2.5), OC, EC, and PAHs. *J Expo Sci Environ Epidemiol*. 2010;20:457-468

46. Turpin BJ and Huntzicker JJ. Identification of secondary organic aerosol episodes and quantitation of primary and secondary organic aerosol concentrations during SCAQS. Atmospheric Environment 1995, 29, 3527–3544.

47. Turpin BJ, Saxena, P, Andrews E. Measuring and simulating particulate organics in the atmosphere: problems and prospects. Atmospheric Environment 2000, 34, 2983–3013.

48. Hallquist M WJ, Baltensperger U. The formation, properties and impact of secondary organic aerosol: Current and emerging issues. *Atmospheric Chemistry and Physics* 2009:3555-3762

49. Turpin BJ, Saxena P, Andrews E. Measuring and simulating particulate organics in the atmosphere: Problems and prospects. *Atmospheric Environment*. 2000;34:2983-3013.

50. DeCarlo PF UI, Crounse J, de Foy B, Dunlea EJ, Aiken AC, Knapp D, Weinheimer AJ, Campos T, Wennberg PO, Jimenez JL. Investigation of the sources and processing of organic aerosol over the central mexican plateau from aircraft measurements during milagro. *Atmos. Chem. Phys.* 2010:5257-5280

51. Ervens B TBWR. Secondary organic aerosol formation in cloud droplets and aqueous particles (AQSOA): A review of laboratory, field and model studies. *Atmos. Chem. Phys. Discuss.* 2012;11:11069-11102

52. Lanz VA AM, Baltensperger U, Buchmann B, Hueglin C, Prevot ASH. Source apportionment of submicron organic aerosols at an urban site by factor analysis modeling of apportionment of submicron organic aerosols at an urban site by factor analysis modeling of aerosol mass spectra *Atmos Chem Phys* 2007:1503-1522

53. Yu JZ, Huang XF, Xu J, Hu M. When aerosol sulfate goes up, so does oxalate: Implication for the formation mechanisms of oxalate. *Environ Sci Technol*. 2005;39:128-133

54. Surratt JD, Gomez-Gonzalez Y, Chan AW, Vermeylen R, Shahgholi M, Kleindienst TE, Edney EO, Offenberg JH, Lewandowski M, Jaoui M, Maenhaut W, Claeys M, Flagan RC, Seinfeld JH. Organosulfate formation in biogenic secondary organic aerosol. *J Phys Chem A*. 2008;112:8345-8378

54. Zhou J, Ito K, Lall R, Lippmann M, Thurston G. Time-series analysis of mortality effects of fine particulate matter components in Detroit and Seattle. *Environ Health Perspect*. 2011;119:461-466

56. Dockery DW, Pope CA, 3rd, Xu X, Spengler JD, Ware JH, Fay ME, Ferris BG, Jr., Speizer FE. An association between air pollution and mortality in six U.S. Cities. *N Engl J Med*. 1993;329:1753-1759

57. Institute HE. Traffic-related air pollution: A critical review of the literature on emissions, exposure, and health effects 2010:1-386

58. Ito K, Mathes R, Ross Z, Nadas A, Thurston G, Matte T. Fine particulate matter constituents associated with cardiovascular hospitalizations and mortality in New York City. *Environ Health Perspect*. 2011;119:467-473

59. Lall R, Ito K, Thurston G. Distributed lag analyses of daily hospital admissions and source-apportioned fine particle air pollution. *Environ Health Perspect*. 2011;119:455-460

60. Bateson TF, Coull BA, Hubbell B, Ito K, Jerrett M, Lumley T, Thomas D, Vedal S, Ross M. Panel discussion review: Session three--issues involved in interpretation of epidemiologic analyses--statistical modeling. *J Expo Sci Environ Epidemiol*. 2007;17 Suppl 2:S90-96

61. Zeger SL, Thomas D, Dominici F, Samet JM, Schwartz J, Dockery D, Cohen A. Exposure measurement error in time-series studies of air pollution: Concepts and consequences. *Environ Health Perspect*. 2000;108:419-426

Table 1. Frequency and percentage of characteristics of study analysis population (cases matched to $PM_{2.5}$ monitors at ≤ 10km distance).

CHARACTERISTIC	Transmural MI (n=1,562)	
	N	%
Age (years)		
18-44	136	9
45-54	326	20
55-64	404	26
65-74	297	19
75-84	277	18
≥85	122	8
Sex		
Male	979	63
Female	583	37
Race		
White	1,078	69
Black	180	12
Other	304	19
Year		
2004	452	29
2005	396	25
2006	714	46
Co-morbidities		
Hypertension	863	55
Diabetes Mellitus	423	27
COPD	164	10
Pneumonia	60	4
Heart Diseases	1,326	85
Ischemic Heart Disease	1,052	67
CHF	299	19
Atrial Fibrillation	174	11
Arrhythmia	466	30
Ventricular Tachycardia	129	8

Table 2. Distribution of PM species mass concentrations (CMAQ estimates; seasonally bias adjusted), CMAQ $PM_{2.5}$ mass; seasonally bias adjusted and TEOM $PM_{2.5}$ mass concentrations, and PM species mass fractions (CMAQ estimates; Mass Fraction = $PM_{2.5}$ species conc. / CMAQ $PM_{2.5}$ conc.) during the study period.

CMAQ Mass Concentrations (µg/m^3)	**Mean**	**Standard Deviation**	**5th %tile**	**25th %tile**	**50th %tile**	**75th %tile**	**95th %tile**
Sulfate (SO_4^{2-})	3.1	1.9	1.4	1.9	2.5	3.6	7.6
Ammonium (NH_4)	1.7	0.9	0.7	1.1	1.5	2.1	3.3
Nitrate (NO_3)	1.8	1.2	0.5	0.9	1.4	2.3	4.3
Elemental Carbon (EC)	1.0	0.4	0.6	0.8	1.0	1.3	1.8
Organic Carbon (OC)	2.8	0.9	1.6	2.2	2.7	3.3	4.4
CMAQ $PM_{2.5}$	12.5	5.2	6.1	8.5	11.3	15.2	22.3
TEOM $PM_{2.5}$	13.0	8.4	3.2	6.9	11.0	17.3	29.7
CMAQ Mass fractions							
Sulfate (SO_4^{2-})	0.25	0.07	0.15	0.19	0.23	0.29	0.39
Ammonium (NH_4^+)	0.13	0.02	0.10	0.12	0.13	0.15	0.16
Nitrate (NO_3^-)	0.14	0.06	0.06	0.09	0.13	0.18	0.24
Elemental Carbon (EC)	0.09	0.02	0.05	0.07	0.09	0.10	0.13
Organic Carbon (OC)	0.24	0.05	0.16	0.20	0.23	0.27	0.34

Table 3. Pearson correlation coefficients for pairs of pollutant concentrations (TEOM $PM_{2.5}$, NO_2, SO_2, CO, O_3) and CMAQ $PM_{2.5}$ component mass fractions.

Pollutant	TEOM $PM_{2.5}$	CMAQ Elemental Carbon mass fraction	CMAQ Ammonium mass fraction	CMAQ Nitrate mass fraction	CMAQ Organic Carbon mass fraction	CMAQ Sulfate mass fraction	NO_2	SO_2	CO
TEOM $PM_{2.5}$	---								
CMAQ EC mass fraction	-0.43	---							
CMAQ Ammonium mass fraction	0.41	-0.60	---						
CMAQ Nitrate mass fraction	-0.01	-0.12	0.15	---					
CMAQ OC mass fraction	-0.49	0.35	-0.65	-0.42	---				
CMAQ Sulfate mass fraction	0.33	-0.47	0.51	-0.61	-0.22	---			
NO_2	0.46	0.00	0.18	0.39	-0.43	-0.17	---		
SO_2	0.44	-0.14	0.11	0.43	-0.43	-0.23	0.57	---	
CO	0.35	0.09	0.07	0.23	-0.32	-0.07	0.64	0.49	---
O_3	0.19	-0.32	0.22	-0.43	0.11	0.45	-0.45	-0.33	-0.39

Table 4. Pearson correlation coefficients for pairs of pollutant concentrations and mass fractions, separately for SUMMER (June, July, August) and WINTER (December, January, February).

		WINTER									
	Pollutant	**TEOM $PM_{2.5}$**	**CMAQ Elemental Carbon mass fraction**	**CMAQ Ammonium mass fraction**	**CMAQ Nitrate mass fraction**	**CMAQ Organic Carbon mass fraction**	**CMAQ Sulfate mass fraction**	**NO_2**	**SO_2**	**CO**	**O_3**
SUMMER	**TEOM $PM_{2.5}$**	---	*-0.20*	*0.19*	*0.23*	*-0.50*	*-0.08*	*0.71*	*0.66*	*0.62*	*-0.54*
	CMAQ EC mass fraction	-0.47	---	*-0.36*	*-0.49*	*0.21*	*0.10*	*0.05*	*-0.22*	*0.09*	*-0.07*
	CMAQ Ammonium mass fraction	0.47	-0.61	---	*0.12*	*-0.51*	*0.47*	*0.11*	*0.12*	*0.13*	*-0.07*
	CMAQ Nitrate mass fraction	0.02	-0.01	0.19	---	*-0.43*	*-0.71*	*0.18*	*0.38*	*0.01*	*-0.10*
	CMAQ OC mass fraction	-0.57	0.41	-0.68	-0.19	---	*0.09*	*-0.39*	*-0.50*	*-0.28*	*0.42*
	CMAQ Sulfate mass fraction	0.50	-0.77	0.62	-0.25	-0.68	---	*-0.14*	*-0.29*	*0.08*	*0.10*
	NO_2	0.33	0.01	0.20	0.31	-0.32	0.06	---	*0.62*	*0.71*	*-0.63*
	SO_2	0.44	-0.17	0.15	0.14	-0.31	0.14	0.37	---	*0.48*	*-0.42*
	CO	0.11	0.10	-0.02	0.13	-0.22	0.06	0.43	0.28	---	*-0.43*
	O_3	0.59	-0.40	0.34	-0.06	-0.22	0.24	-0.13	0.19	-0.30	---

Table 5. Risk (and 95% confidence interval) of a transmural infarction associated with each interquartile range increase in TEOM $PM_{2.5}$, within each tertile of $PM_{2.5}$ species mass fraction.

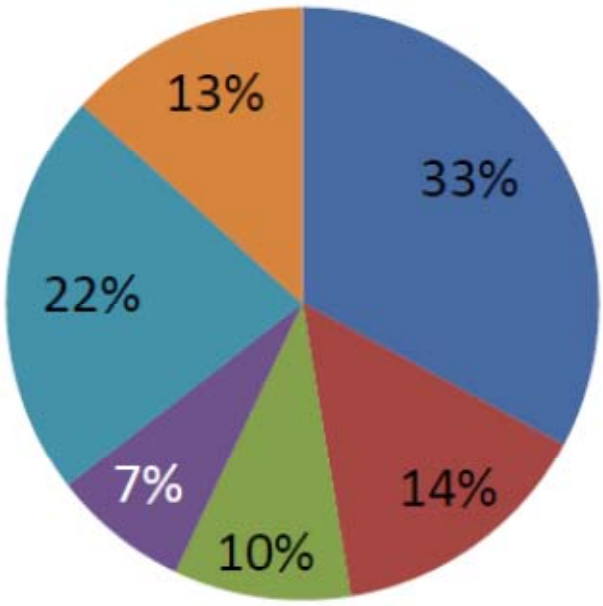
PM species tertile		Mass fraction Min.	Mass fraction Max.	N Total	N Cases	IQR ($\mu g/m^3$)	OR	95% CI	p-value
	Low	0.094	0.206	2250	494	10.8	1.08	0.92, 1.28	0.35
Sulfate	Middle	0.206	0.265	2335	549	10.8	1.11	0.95, 1.30	0.20
	High	0.266	0.531	2321	519	10.8	1.13	1.00, 1.27	0.05
	Low	0.075	0.125	2334	521	10.8	1.02	0.85, 1.23	0.83
Ammonium	Middle	0.125	0.141	2297	519	10.8	1.11	0.95, 1.29	0.18
	High	0.141	0.204	2275	522	10.8	1.13	1.00, 1.28	0.05
	Low	0.026	0.104	2427	529	10.8	1.08	0.94, 1.23	0.28
Nitrate	Middle	0.104	0.167	2268	541	10.8	1.11	0.97, 1.27	0.12
	High	0.167	0.343	2211	492	10.8	1.15	0.98, 1.35	0.08
	Low	0.027	0.078	2341	535	10.8	1.17	1.03, 1.34	0.01
EC	Middle	0.078	0.096	2253	508	10.8	1.06	0.92, 1.24	0.42
	High	0.096	0.192	2312	519	10.8	1.07	0.90, 1.28	0.43
	Low	0.110	0.211	2252	516	10.8	1.14	1.00, 1.30	0.04
OC	Middle	0.211	0.253	2258	519	10.8	1.21	1.03, 1.42	0.02
	High	0.253	0.493	2396	527	10.8	0.91	0.74, 1.13	0.39

NOTE: We regressed case–control status (i.e., case period = 1, control period = 0) against the mean $PM_{2.5}$ concentration in the 24 hr before ED arrival (calculated from continuous TEOM $PM_{2.5}$ measurements), including a natural spline (3 degrees of freedom) of the mean apparent temperature in the 48 hr before ED arrival in the model, indicator variables for sulfate tertile (MEDIUM_TERTILE + HIGH_TERTILE) and two interaction terms ($PM_{2.5}$*MIDDLE_TERTILE _+ $PM_{2.5}$*HIGH TERTILE). From this model, we estimated the risk of a transmural infarction associated with each 10.8 $\mu g/m^3$ increase in $PM_{2.5}$ (TEOM) concentration when the sulfate mass fraction is in the highest tertile, when it is in the middle tertile, and when in the lowest tertile. We repeated this for ammonium, nitrate, EC, and OC.

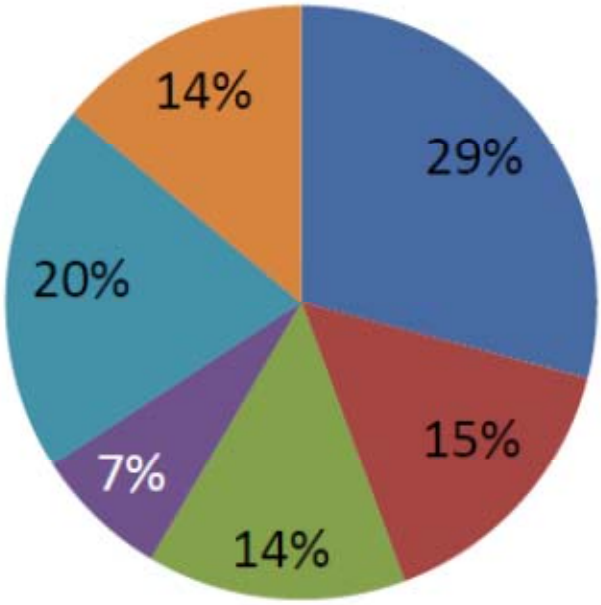
Table 6. Median daily gaseous pollutant concentration and weather characteristics, by $PM_{2.5}$ species tertile

Figure #	Tertile	NO2 8 hour maximum (ppb)	SO2 8 hour maximum (ppb)	CO 8 hour maximum (ppm)	O3 8 hour maximum (ppb)	Temperature (°C)	Relative Humidity (%)
1	High Sulfate	24.5	5.0	0.638	43.6	21.1	72.1
	High Ammonium	28.6	5.9	0.688	40.3	17.4	67.8
	Low Elemental Carbon	26.6	6.1	0.650	43.6	18.4	67.4
2	Low Sulfate	32.8	8.1	0.800	25.7	7.8	59.0
	Low Ammonium	27.1	5.1	0.750	29.3	14.4	63.0
	High Elemental Carbon	29.6	5.3	0.838	27.9	14.1	63.0
3	High Organic Carbon	22.4	3.8	0.600	35.0	16.4	64.2
	Low Organic Carbon	34.4	8.6	0.838	32.6	13.6	65.9
	High Nitrate	34.4	8.7	0.813	26.1	5.1	58.8
	Low Nitrate	22.3	4.4	0.613	43.5	22.3	69.1

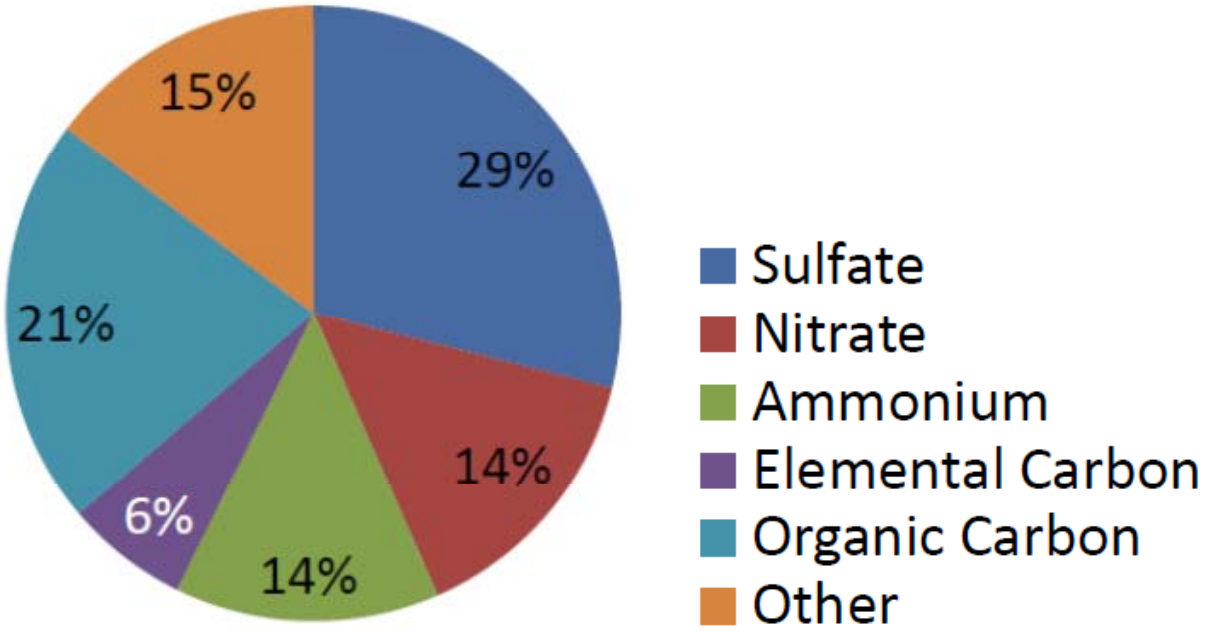
Figure 1.



a. High Sulfate

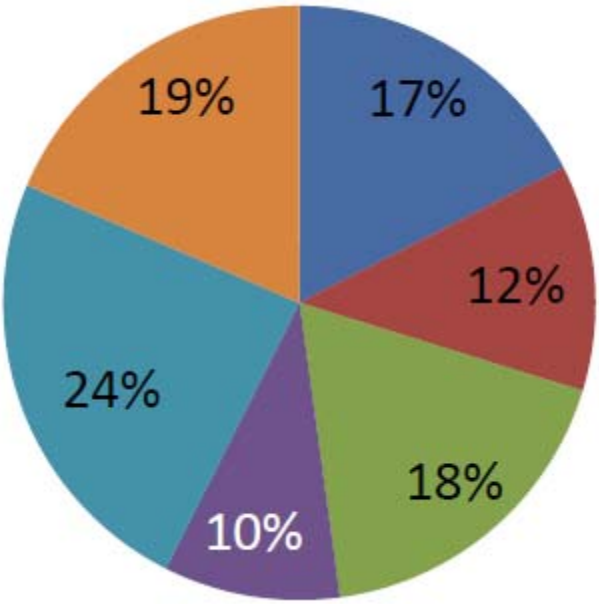


b. High Ammonium

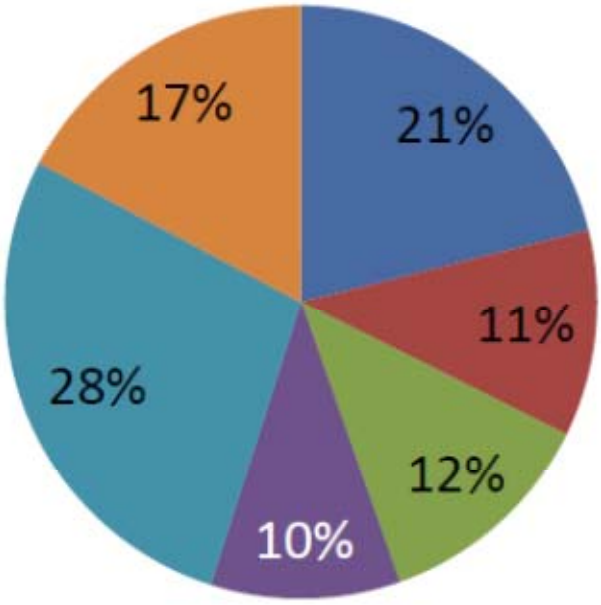


b. Low Elemental Carbon

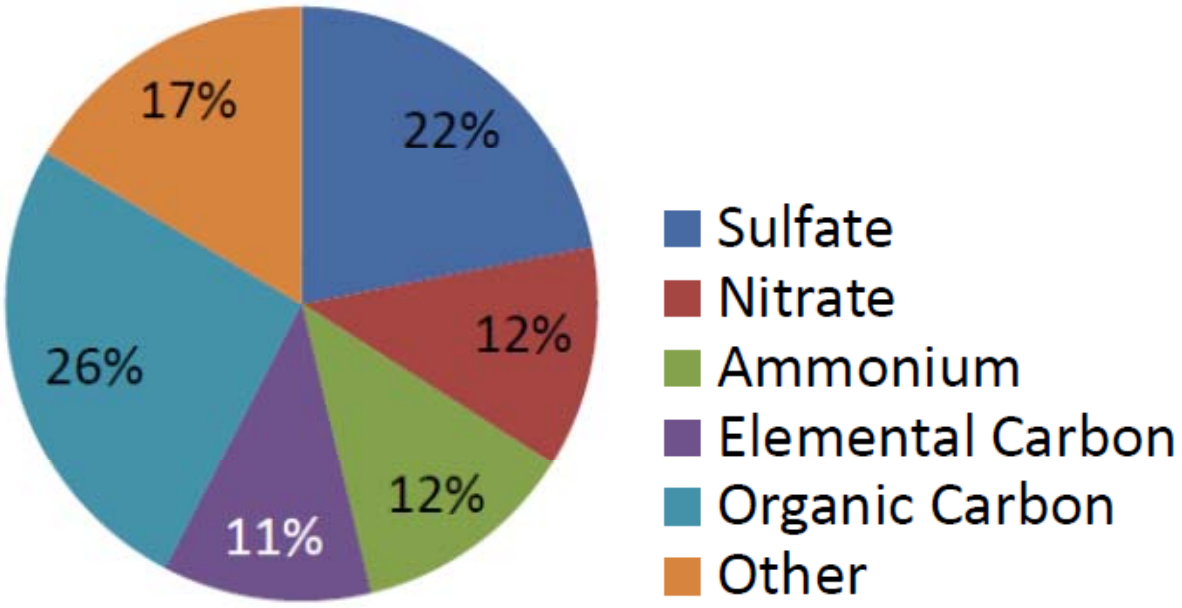
Figure 2.



a. Low Sulfate

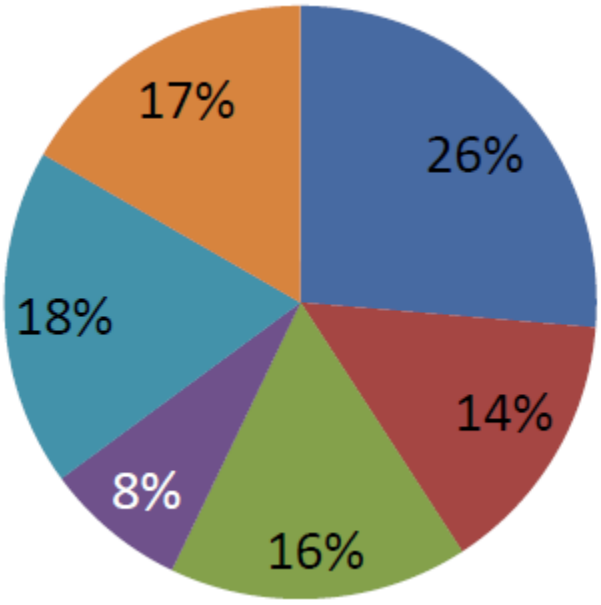


b. Low Ammonium

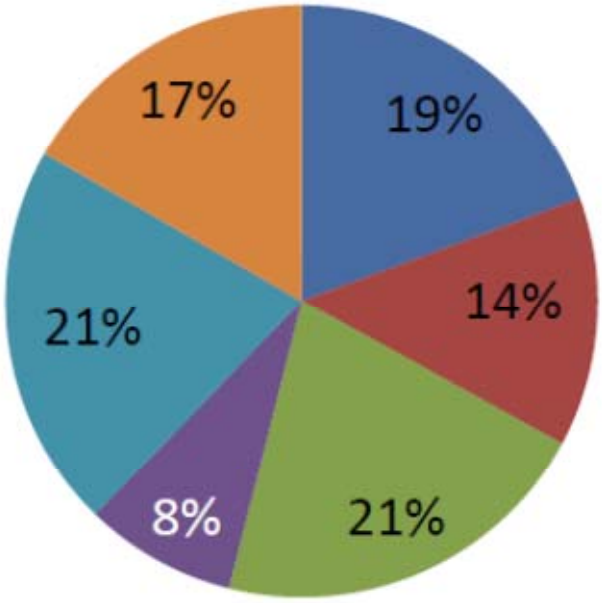


c. High Elemental Carbon

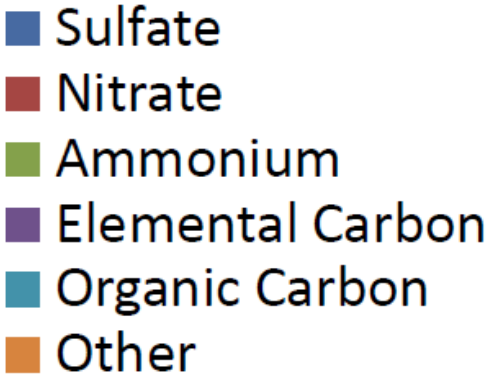
Figure 3.

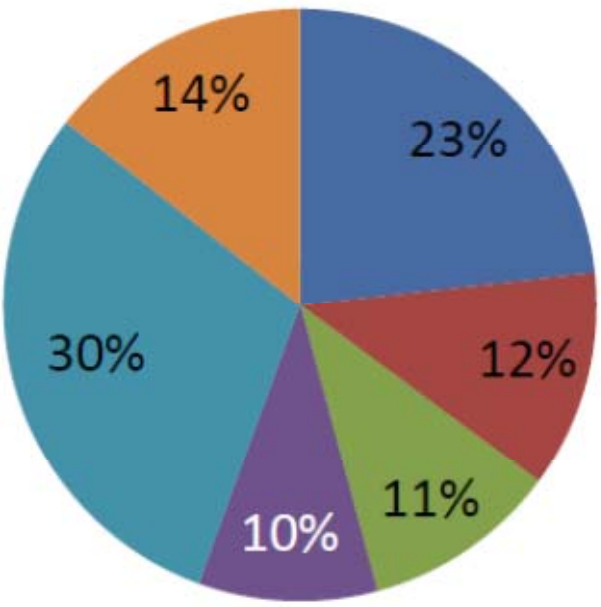


a. Low Organic Carbon

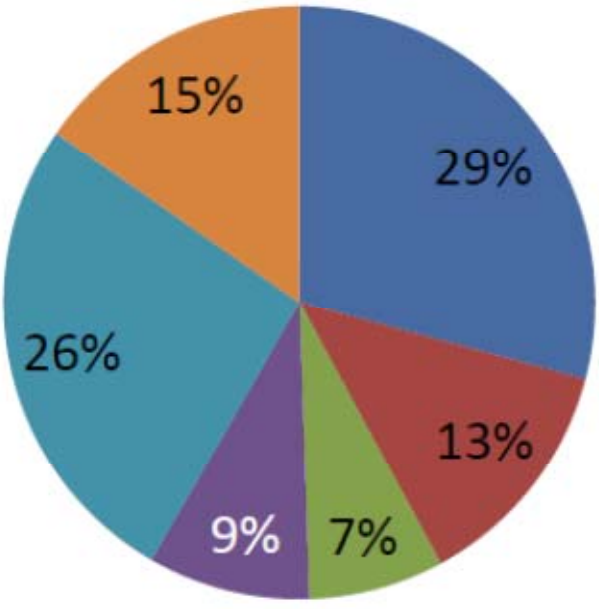


b. High Nitrate





c. High Organic Carbon



d. Low Nitrate

Figure Legend

Figure 1. Composition of fine particle mass, by $PM_{2.5}$ species tertile: a. High Sulfate; b. High Ammonium; and c. Low Elemental Carbon

Figure 2. Composition of fine particle mass, by $PM_{2.5}$ species tertile: a. Low Sulfate; b. Low Ammonium; and c. High Elemental Carbon

Figure 3. Composition of fine particle mass, by $PM_{2.5}$ species tertile: a. Low Organic Carbon; b. High Nitrate; c. High Organic Carbon; and d. Low Nitrate