

# Macroscopic to microscopic scales of particle dosimetry: from source to fate in the body

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## Abstract

Additional perspective with regards to particle dosimetry is achieved by exploring dosimetry across a range of scales from macroscopic-to-microscopic in scope. Typically, one thinks of dosimetry as what happens when a particle is inhaled and where it is deposited and how it is cleared from the body. However, this paper shows a much more complicated picture starting with emissions sources, showing how the source-to-intake fraction ( $iF$ ) can be used to estimate changes in the inhaled dose due to changes in emissions and then ending with particle-liquid, particle-cellular and subcellular interactions and movement of ultrafine particles across the lung-blood barrier. These latter issues begin to suggest mechanisms that can lead to adverse health effects the former can provide guidance to policy decisions designed to reduce the health impact of atmospheric particles. The importance of ultrafine particles, their ability to translocate to other parts of the body, and the potential impact of these particles has advanced significantly over the last decade, including studies that show the movement of ultrafine particles along the olfactory nerves in the nose with direct transport to the brain; the neurological effects of which are still unknown. Incremental advancements continue with regards to understanding particle deposition, including regional and local deposition (including hot spots), and clearance and the factors that affect these variables, in part due to the development and implementation of computational fluid dynamics (CFD) models and digital imaging of the lungs. CFD modeling will continue to provide new information for reducing uncertainty in dosimetric calculations. We better understand today how a number of diseases may develop based on the fate of particles after deposition in the lung and how changes in source

emissions might impact that dose. However, a number of uncertainties remain some of which can be reduced by addressing the research needs stated in this paper.

## Introduction

Leonardo da Vinci (1452-1519) warned readers in his Anatomical Atlas that “dust is harmful” (Gehr et al. 2010), and Paracelsus (1493—1541) added “The right dose differentiates a poison from a remedy” (Gallo 2008). This paper on particle dosimetry integrates and updates these concepts in the interest of reducing the adverse health effects of modern air pollution. This paper also addresses, in part, one of the policy-relevant Science Questions that formed the basis of the 2010 international conference “Air Pollution and Health: Bridging the Gap between Sources and Health Outcomes”<sup>1</sup> (Solomon et al. 2011). Specifically this paper addresses Science Question 4 “*What advances have been made in understanding the relationships between exposure, both spatially and temporally, and estimates of dose that tie to health outcomes?*”

Current research continues to show significant associations of fine particles with adverse health effects (Brook et al. 2010, He et al. 2010, Pope et al. 1995, 2009, Puett et al. 2009). Reducing the adverse health effects from air pollution requires knowledge across the source-to-health effects paradigm: source – atmospheric sciences – exposure – dose – health effect (NRC 1998; Mauderly et al. 2011, this issue; Solomon et al. 2011). Linking sources to adverse health effects is complicated due to the many factors, such as, source variability, the complex and changing nature of pollutants in air (e.g., see Seinfeld and Pandis 1998; Finlayson-Pitts and Pitts 2000) as well as component toxicity, anatomical and physiological factors, susceptibility and vulnerabilities of the target to pollutants, and confounding factors (2009a (Table 8-1); Sheppard et al. 2011, this issue; Mauderly et al. 2011, this issue; O’Neil et al. 2011, this issue).

Dosimetry research involves a broad understanding of source-to-dose relationships, essentially the first half of the source to health effects continuum. Dosimetric research encompasses a variety of *scales* ranging from the *macroscopic* to the *microscopic*. As well, the dose of the causal agent(s) received and retained are key parameters regarding the impact of an air pollutant that results in an adverse health effect.

At the macro-scale, the *intake fraction*, *iF*, is a parameter that provides an estimate of the amount of a pollutant that is inhaled (not deposited nor retained), relative to the amount emitted into the environment from a specified source or source category. The *iF* **begins to provide** an understanding and quantification of pollutant source-to-intake relationships. The *iF* precedes the deposition in the body of an inhaled air

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<sup>1</sup> American Association for Aerosol Research held in San Diego, CA March 22-26, 2010, <http://aaar.2010specialty.org/> (Solomon et al. 2011).

pollutant, which is an important parameter for dosimetry considerations, since the respiratory tract (RT) is the dominant exposure route.

Moving from the macro- towards the micro-scale, this paper next considers the fate of a pollutant during and just after inhalation. The inhalability, more specifically the *inhalable fraction* ( $\eta_i$ ), describes the sampling efficiency of the nose and/or mouth. The  $\eta_i$  is the ratio of pollutant concentration in the breathing zone (the immediate vicinity of the face) to the concentration that is inspired and enters the *extrathoracic* (ET) airways. The inhalation of particles into the nose or mouth follows the airflow where deposition may occur in any of the major regions of the RT shown schematically in Figure 1a and with micrographs for the gas exchange region of the lungs in Figure 1b. The three main regions include: the ET, *tracheobronchial* (TB), and the *alveolar-interstitial* (AI) regions. The particle deposition efficiencies in each of the major RT regions have been modeled for adults and children breathing at various levels of exertion (e.g. ICRP 1994; EPA 2009a) (e.g., Figure 2).

Following deposition within the RT, particles are subject to a variety of fates in the body. They may be retained in the RT for times ranging from minutes (or less) to a lifetime depending on a number of factors described later. Particles not retained in the lungs are removed or cleared by *mechanical processes* (e.g., mucociliary activity, coughing, swallowing), by *phagocytosis* (engulfed by cells and then sometimes digested), and by *translocation*, in which particles or their products move from the RT to other parts of the body. Deposition and clearance phenomena also are individual- and species-dependent (EPA 2009a).

At the micro-scale, deposited particles interact with fluids and cells throughout the body. The interactions are varied, and are the subject of ongoing research (Gehr et al. 2010). Interactions with fluids and cells in the body include dissolution, transport within the fluids, uptake by cells and tissue elements, and other physiologic phenomena. These interactions will largely determine the health-related consequences that follow pollutant deposition and again are dependent on a number of factors. Recent research has identified *ultrafine particles* (UF particles; generated by ambient air pollution sources and formed from gas phase precursors in the atmosphere) with dimensions  $\leq 0.1 \mu\text{m}$ , and their intentionally engineered counterparts, *nanoparticles* (NP), as having unique fates in the body and the potential to cause adverse health effects (Wichmann et al. 2000; Renwick et al. 2004; Geiser et al. 2005; Schultz et al. 2005; Oberdörster et al. 2005; de Haar et al. 2006; Donaldson et al. 2006; Kreyling et al. 2002, 2006a, 2006b). Throughout this summary, the term UF particles will be used to refer to both UF particles resulting from air pollution and nanoparticles that are intentionally engineered. UF particles can have access to the entire RT and other organs of the body including the brain (see section, *Particle Translocation*), they likely have the potential for producing greater adverse effects than would be predicted from their mass alone. Thus particle number, composition, or surface area may be more important for dosimetric purposes (Oberdörster 2010).

The dose *metric* (or indicator) refers to key pollutant properties that must be measurable, and have a causal relationship to one or more biological responses (Phalen et al. 2010). The metric may have

physical and/or temporal properties. Also, the metric need not be causal for every exposed individual (Phalen et al. 2010). Table 1 lists some known or proposed particle-related metrics relevant to the health effects of inhaled air-pollutant particles. In the body, dose metrics differ for soluble versus insoluble particles. For example, common metrics for soluble particles may include particle mass, composition, and aerodynamic diameter (dae). For insoluble particles the metrics also might include surface area, particle count, or other physical properties. Composition is usually a key metric for all air pollutants. All metrics are likely important and should be considered in evaluating pollutants for potential adverse health effects.

Relevant biological targets include anatomical (e.g. a cell type, tissue, organ, or organ system) or physiological effects, such as an essential biochemical process (e.g., lung surfactant synthesis) or a system function (e.g., learning or memory).

This paper describes the current understanding of selected macro- and micro-scale phenomena related to the adverse health effects of inhaled particles in human subjects, emphasizing current research in dosimetry. *In-vitro* studies have their own set of challenges, and one of the greatest is relating or extrapolating results from *in-vitro* experiments to humans (Gerde 2008), however a detailed discussion of this is outside the scope of this paper. For brevity, details and examples of many other topics discussed can be found in the references, which themselves provide only examples due to the vastness of the applicable literature in this growing field.

#### **Source-to-Intake Relationships: Intake Fraction**

Determining source-to-intake relationships for air pollutants allows for an estimate of the relative intake of a pollutant(s) to a population from a particular source category. One advantage is it allows for an estimate of the relative decreases in exposure resulting from decreases in emissions from the subject source category. A source-to-intake relationship takes into consideration the resulting concentrations from a pollutant released into the environment, determines the exposure concentrations, and considering breathing rate, estimates resulting intake to members of the population. A useful measure to estimate source-to-intake relationships is the intake fraction (*iF*) (Bennett et al. 2002a). The *iF* is defined as the integrated incremental intake of a pollutant, summed over all exposed individuals, and occurring over a given exposure time, released from a specified source or source class, per unit of pollutant emitted. The definition is expressed in the equation below:

$$iF = \frac{\sum_{\text{people, time}} \text{intake of pollutant by an individual (mass)}}{\text{mass released into the environment (mass)}} \quad \text{eq. 1}$$

There are two dimensions over which the pollutant intake is summed, population and time. In actuality, when a pollutant is released into the environment there is a distribution of individual exposures within the exposed population. For transient release scenarios, *iF* is made dimensionless by dividing the time-integrated intake by the total quantity released. For steady-state release and exposure conditions, *iF* is

the rate of intake divided by the rate of release, both using units expressed as mass per time. If the compound is persistent in the environment, then the exposure that occurs over the duration of the compound being present must be considered.

The *iF* allows different pollutant release scenarios to be considered, such as releases into the indoor environment, from outdoor mobile sources, or from industrial facilities (Humbert et al. 2011). Various environments to which the pollutant may be released also can be considered, such that one can differentiate between urban and rural regions or indoor and outdoor. In the case of some chemical pollutants, one may need to sum across multiple exposure pathways (i.e., considering inhalation, ingestion, and dermal pathways) (Bennett et al. 2002b). For air pollutant regulation, *iF* primarily (or exclusively) considers inhalation.

There are four attributes of the *iF* that further increases its usefulness. First, *iF* considers both the release scenarios and fates of the pollutant in the environment (Bennett et al. 2002a;)). Second, *iF* can be disaggregated to consider various populations (e.g., susceptible groups) that can then be summed to equal the total *iF* (Tainio et al. 2009; Greco et al. 2007a; Zhou and Levy 2008; Ries et al. 2009; Marshall and Behrentz 2005). Third, *iF* is compatible with dose-response calculations (Levy et al. 2009). Forth, *iF* values can be calculated from models or from measured data (Bennett et al. 2002b, Tainio et al. 2009, Greco et al. 2007a, 2007b, Zhou and Levy 2008, Levy et al. 2009, Hao et al. 2007, Heath et al. 2006, Klepeis and Nazaroff 2006, Li and Hao 2003, Luo et al. 2010, Stevens et al. 2007, Wang et al. 2006; Ries et al. 2009). While the *iF* value itself is a single, simple value, considerable complexity goes into the calculation. When calculating *iF* values using a model, the concentration profile in the surrounding area first needs to be determined. Dispersion models with varying levels of complexity have frequently been used in calculations. The concentration profile is then overlaid with the population density to determine the number of individuals exposed to the pollutant. This brings in the population density of the surrounding area. Breathing rates then determine the actual intake, such that a total mass of particulate matter inhaled can be determined.

In limited cases there is complete information on the total emissions of the compound into the environment and sufficient measurements of resulting concentrations have been made such that one can calculate an *iF* value solely from the measured data, without the use of models (Bennett 2002b and Ries 2009).

The resulting values are therefore subject to uncertainties resulting from model calculations and full quantification of these uncertainties is beyond the scope of this paper. A recent review quantified model uncertainties and found that variability in population density exceeded model uncertainties (Humbert 2011). The inclusion of the exposed population in the calculation makes *iF* values very useful for comparing release scenarios, minimizes the impact of model uncertainties, and points out the need for improved understanding of the exposed population.

Research communities can benefit from the numerous applications and models available in which the *iF* can be obtained from and incorporated. Within the modeling community, *iF* can facilitate the evaluation of models, both by comparing *iF* values calculated by models and through measured data (Bennett et al. 2002b) and by the comparing results between or among models. Another application is to generate modeling results in one location and use those results in another location after making necessary adjustments, such as for differences in population density (Stevens et al. 2007). This is useful when there are limited available data, or limited resources, such that it would otherwise be difficult to conduct an extensive model evaluation. Although *iF* values cannot be directly used in health assessments, values can be combined with dose-response functions, allowing for comparative risk assessment or evaluation of health-related damages from a release (Levy et al. 2009). The *iF* also is useful with a variety of tools used in decision-making, such as risk management, life cycle assessment, emissions trading, sustainable development, and policy development. Thus, *iF* modeling and the application of *iF* in models can assist in evaluating new sources and for determining the most effective way to reduce sources (Heath et al. 2006; Humbert et al. 2011).

When using models to calculate *iF* values, a wide range of models with varying levels of complexity have been used by various researchers. The resulting values are therefore subject to uncertainties resulting from model calculations and full quantification of these uncertainties is beyond the scope of this paper. A recent review quantified model uncertainties and found that variability in population density exceeded model uncertainties (Humbert 2011). The inclusion of the exposed population in the calculation makes *iF* values very useful for comparing release scenarios, minimizes the impact of model uncertainties, and points out the need for improved understanding of the exposed population.

#### Intake fraction values for PM emissions

A series of *iF* values calculated for PM<sub>2.5</sub> emissions derived from power plants (Levy et al. 2009; Heath et al. 2006; Wang et al. 2006; Hao et al. 2007; Li and Hao 2003), mobile sources (Greco et al. 2007a; Zhou and Levy 2008; Stevens et al. 2007), area sources (Ries et al. 2009) and indoor sources; Klepeis and Nazaroff 2006) are illustrated in Figure 3. In this figure, *iF* values span over three orders of magnitude among emission sources, ranging from  $0.1 \times 10^{-6}$  to  $3 \times 10^{-3}$ , which is similar to *iF* values calculated for various volatile and semi-volatile organic (VOC and SVOC, respectively) chemicals (Bennett et al. 2002b), also plotted in Figure 3.

This extreme range highlights the importance of targeting emission reduction strategies on sources with the highest intake fraction values. When *iF* is taken into account along with the magnitude of the source and any potential differences in the relative toxicity between particle emissions from various sources, regulators could likely develop strategies that could result in health improvements.

*iF* values resulting from power plants in Europe, the U.S., and China are shown towards the left side of Figure 3. U.S. values were based on a comprehensive study of PM<sub>2.5</sub> emissions from 407 power plants,

with exposures estimated using a source-receptor matrix (Levy et al. 2009). *iF* values from power plants in Europe were obtained based on the average *iF* values calculated for different countries within Europe (Tainio et al. 2009). *iF* values for China were calculated from power plants in Beijing (Hao et al. 2007) and the Hunan Province (Li and Hao 2003). Similar *iF* values were observed between the US and Europe but were slightly higher for China due to its higher population density (e.g., 84 per sq mile in the US versus 365 in China; <http://www.infoplease.com/ipa/A0934666.html>). A value from a hypothetical small distributed energy power plant (DE) in an urban area also is included (Heath et al. 2006).

For mobile sources, *iF* values were calculated for urban areas of Boston, MA (Greco et al. 2007a), urban street canyons in New York City, NY (Zhou and Levy 2008), Hong Kong (Luo et al. 2010) and estimated for Mexico City from these previous studies (Stevens et al. 2007). Additional *iF* values from mobile sources are available for each county in the U.S. (Greco et al. 2007b), although these values may be an underestimate due to lack of sufficient detail and modeling of near-road exposures. The variability in values resulting from mobile source emissions among locations largely results from differences in population density (e.g. Hong Kong) and/or high exposure levels (e.g. NY street canyons). An *iF* value for school buses also has been calculated, including exposures to both people inside and outside the bus, thereby resulting in a higher intake fraction value than for other mobile sources (Marshall and Behrentz 2005).

There are limited examples in the literature of *iF* values for specific area sources. *For example*, *iF* values for woodsmoke emissions in Vancouver were calculated for a large wildfire (Ries et al. 2009). In this example, levoglucosan, a unique marker for wood smoke was used to determine the portion of the particulate matter that resulted from the wildfire, allowing the authors to determine the relationship between measured concentrations and the source rate.

For indoor tobacco smoke, *iF* values were calculated for different scenarios in which individuals were in the same room as the smoker, different room than the smoker, and also by varying the HVAC system (Klepeis and Nazaroff 2006). The resulting *iF* values in the tobacco study were dependent on air exchange rates and the location of people in relation to the source.

The *iF* is a simple, transparent, and comprehensive measure of the source-to-intake relationship. As shown in Figure 3, the non-dimensionality of *iF* facilitates comparisons of values calculated from different emission sources and it can be compared among investigators and modeling scenarios. It also is a metric that can be calculated for any scenario and can be reported in addition to plots of concentration. This highlights the utility of *iF* calculations in understanding source-to-intake relationships and in decision-making policies for targeting emission reduction strategies that may have the greatest impact in protecting public health.

## **The Fate of Inhaled Particles in the Respiratory Tract**

Moving from the macroscopic view, and having an estimate of the fraction of air pollutants that can enter the body (*iF*) this section considers the fate of inhaled atmospheric particles inside the body. The fate of inhaled particles involves four general steps: deposition; clearance; translocation; and retention (EPA 2009a). *Deposition* occurs when inhaled particles contact the wall of an airway and can occur during inhalation or exhalation. *Clearance* commonly refers to either removal of the particles from the respiratory tract or from the entire body. *Translocation* refers to the extrapulmonary movement of particles from the lungs to other locations in the body. *Retention* is simply what remains in the body at some post exposure time, whether within the lung or elsewhere in the body. Chemical transformation of pollutants also occurs in the body. Each step is dependent on a range of physical, chemical, and biological factors that include: particle properties; exposure location; breathing patterns; and anatomical and physiological characteristics of the exposed subjects such as disease states, body size, age, race, and gender.

### Particle Deposition

An inhaled particle can deposit anywhere within the RT upon contacting a surface within the RT; it is assumed that both bounce and re-entrainment are negligible. Within the RT, the AI region of the lungs where gas exchange occurs is of interest due to its large internal surface area of about 150 m<sup>2</sup> and the very thin air-blood tissue barrier of <1 µm in thickness over a large part of the lung's surface, which is primarily located in the AI region (see Figure 1b) (Gehr *et al.* 1978).

Particle deposition occurs by several mechanisms in the RT as described below and depends strongly on the size of the inhaled particle, the route of inhalation (nose versus mouth), the tidal volume ( $V_T$ ), the breathing frequency, RT disease, and the RT morphology (ICRP 1994; EPA 2009a). Physical exertion and reflex-mediated bronchial caliber may be important in some exposure scenarios, for example rescue operations. These factors can influence which deposition mechanisms are dominant within the different regions of the RT (Figure 1). Biological factors depend on the mammalian species, size, status of development and growth and health status, since they will affect lung structure and function, and thus, particle deposition and subsequent fate.

Much is known about the deposition of inhaled “ideal” particles, i.e. smooth uncharged solid spheres with unit density (1 gm/cm<sup>3</sup>). In the diameter range from 0.01 to 100 µm aerodynamic diameter ( $d_{ae}$ ), the particle mass, surface area, the number of particles of a given size in 1 µg of ideal particles as well as how far an ideal particle of a given size will travel in a second in still air based on gravitational settling and diffusion are well known Phalen *et al.* (2010). Such particle characteristics are used for dose calculations. The major mechanisms for the deposition of particles in the RT include inertial impaction, gravitational sedimentation, and Brownian diffusion (ICRP 1994; Brown *et al.* 2005; EPA 2009a). The first two deposition mechanisms depend on the  $d_{ae}$ , which is the physical diameter of a sphere of standard density that has the same gravitational terminal settling velocity as the particle in question. Aerodynamic diameter is most applicable to particles larger than about 0.3 µm (EPA 2009a). Diffusion is



thermodynamically controlled, and it dominates the motion of particles less than a few tenths of micrometer in physical diameter, where gravitational and inertial effects are usually negligible.

Within the RT, impaction is an efficient deposition mechanism for larger particles, occurring primarily in areas with obstructions or sharp transitions, such as in the nasal turbinates, the larynx, and at airway bifurcations, thus primarily in the ET and upper TB. Sedimentation probability is inversely proportional to the airflow rate and is most important in the lower TB and AI regions, where airflow velocities are low and distances to the walls of airways are small. Both impaction and sedimentation are important for particles in the size range above  $1\text{ }\mu\text{m d}_{ae}$  (EPA 2009a). Deposition due to diffusion is proportional to residence times and proximity to airway walls, so it occurs effectively at low velocities and with turbulent flow. Thus particles with diameters under  $0.1\text{ }\mu\text{m d}_{ae}$  deposit most efficiently in the lower TB and AI regions. Particles in this size range also tend to deposit due to diffusion in the ET and upper TB regions. Figures 2a and 2b illustrate typical deposition curves, including total deposition and deposition in the major RT regions; the curves are for an adult male at rest and at a light exercise. The competing mechanisms of impaction, diffusion, and sedimentation results in the typical U-shaped total RT deposition curves, with a minimum in the range between  $0.1$  and  $1\text{ }\mu\text{m d}_{ae}$ , since particles in this size range are too small to be efficiently impacted or undergo significant gravitational settling, and are too large to be strongly influenced by diffusion. Total deposition approaches 100% for particles larger than about  $10\text{ }\mu\text{m d}_{ae}$  and less than  $0.01\text{ }\mu\text{m}$  in diameter, due to the high efficiency of impaction and sedimentation, and diffusion, respectively. Average curves are shown in Figure 2, but each individual has a unique particle deposition pattern at any given time because of the complexity of the deposition process. Thus, caution must be applied when using average curves for population or individual risk assessment purposes.

Respiratory tract disease, from the typical acute infections that adults get 1-3 times per year, to advanced chronic obstructive pulmonary disease (COPD), can significantly influence particle deposition by altering airway structure and ventilatory parameters (see EPA 1996, 2004 for original references). Changes in disease states include differences in regional lung deposition, more heterogeneous deposition patterns, greater deposition in the TB region, and a decrease in the AI deposition due to less particles reaching the AI region. Total RT deposition is generally increased with increasing airway obstruction. Slower clearance in COPD patients was observed by Scheuch et al. (2008), which may be limited to the larger bronchial airways. However, Smaldone et al. (1993) and Brown et al. (2002) found normal clearance in peripheral airways in COPD patients. Relative to healthy adults, individuals with COPD, in which a more limited lung volume receives the total airflow, had a similar resting  $V_T$  and an increased breathing frequency; thus a higher than normal tidal peak flow, minute ventilation, and average deposition rate of particles, all suggesting higher doses for individuals with COPD (Bennett et al. 1997; Phalen et al. 2010; EPA 1996, 2004, 2009a). Additional discussion on this topic can be found in Phalen et al. (2010) and EPA (2009a) and references within these publications.

The complex morphology of the RT (e.g., bifurcations, bends, and other obstructions to airflow), in conjunction with the several deposition mechanisms results in a strongly non-uniform distribution of deposited particles. Particles greater than about  $1\ \mu\text{m}$   $d_{ae}$  and UF particles deposit preferentially at bifurcations, mainly on the carinal ridges, due to impaction and secondary flow patterns. These areas are referred to as *hot spots* that result from high particle deposition in small areas (EPA 2009a; Kleinstreuer and Zhang 2010; Phalen et al. 2006). The size of the hot spot area is referred to as the *patch size* by dosimetry modelers. Hot spots are illustrated in Figure 4. In this figure, computational fluid dynamic (CFD) simulations are used to model the deposition of UF (NP) particles and gases and micrometer sized particles in the ET and upper TB (mouth to large bronchi).

Deposition enhancement factors (EFs) are used to quantify hot spot deposition dose enrichments. The EF is defined as the ratio of particles deposited at the hot spot per unit surface area of epithelial cells (e.g., a patch size of  $0.1\ \text{mm} \times 0.1\ \text{mm}$ ) to the average deposition on the surrounding tissue (Kleinstreuer and Zhang 2010; EPA 2009a; Balásházy et al. 1999, 2003). EFs, estimated from CFD modeling, are shown as a function of patch size and particle diameter in Table 2, and as a function of ventilation rate (5 and  $30\ \text{L/min}$ ) in Table 3 (Balásházy et al. 1999, 2003). In these examples, EFs range from less than 10 to about 400 times that of the surrounding epithelial surface. EFs up to 1,200 have been reported (Farkas and Balásházy 2008). Hot spots also bring up a unique challenge for cell culture studies with respect to air pollutant dosing of the *in-vitro* system (Phalen et al. 2006). Should an average dose be applied, the enhanced dose applied to the entire culture, or the enhanced dose just applied to a reasonable patch sizes surrounded by cells in the culture with no dose?

### Particle clearance

Clearance usually refers to the removal of deposited particles from the RT, whether removed from the body or translocated to other parts of the body (e.g., through the respiratory epithelium) where they may be subsequently removed from the body by other mechanisms. Clearance rate data for inhaled particles from the RT are required in dosimetric calculations. The rates and mechanisms of particle clearance depend on the size of the particle, where it has deposited, whether it is *soluble* or *insoluble*, health status of the individual, particle loading, and other factors (Phalen and Méndez 2009).

Although particles are often referred to as insoluble or soluble, it is more appropriate to refer to their dissolution rate in lung fluids, both on RT surfaces and within cells (Phalen et al. 2010). When the dissolution rate is slow compared to the mucus flow or other transport rate, the particles are considered to be *slowly dissolving* or *poorly soluble*. When particles dissolve rapidly, they are considered to be *rapidly dissolving*. For simplicity, soluble and insoluble are used herein.

Insoluble particles deposited in the ET are typically cleared by bulk physical processes, such as sneezing, coughing, expectoration, or by mucociliary transport followed by swallowing. The latter mechanism is more important for particles deposited in the posterior portions of the nasal region, with the former

mechanisms being more important in the anterior nasal region. UF particles also may clear from the ET via the olfactory nerves with direct transport to the brain (to be described later). Both insoluble and soluble particles deposited in the mouth are removed by swallowing or expectoration.

Clearance in the TB region of insoluble particles is primarily by mucociliary movement toward the epiglottis where particles are swallowed (ICRP 1994; EPA 2009a). Mucus movement is faster in larger TB airways than in the smaller ones (EPA 2009a; ICRP 1994). This leads to longer residence times for particles deposited deeper in the TB region. Previously, it was assumed that insoluble particles deposited in the TB region were completely cleared by mucociliary action within 24 hr. Recently, this assumption has been shown to be inaccurate, as a fraction of the deposited particles may be retained for much longer time periods (ICRP 1994; Kreyling et al. 2006a, 2006b; Smith et al. 2008; EPA 2009a; Phalen et al. 2010). The mechanisms for such slow bronchial clearance include: mucus stasis, mucus retrograde flow, macrophage/epithelial cell uptake, and particle displacement into the subphase of the mucus (EPA 1996, 2004, 2009a; Gehr et al. 1990; Schürch et al. 1990). Results summarized in EPA (2009a and references within), indicate that particle clearance from the TB region also is particle size dependent (therefore, deposition location dependent) with probable slower clearance of UF particles (up to several months for complete clearance) and more rapid clearance of particles in the 6-10  $\mu\text{m}$  diameter range. Bolus inhalation studies used to support slow bronchial clearance in humans have been challenged, since it may have been assumed that particles deposited in AI region were actually deposited in the TB airways (Phalen et al. 2010) and the AI region has slower particle clearance than TB region. This assumption would overestimate the fraction of particles that are cleared slowly. Therefore, TB clearance rates are probably more variable, and uncertain than is usually assumed.

The primary mechanisms for insoluble particle clearance from the AI region are believed to include: (1) macrophage phagocytosis with subsequent migration to interstitial spaces, i.e., the lymphatics or terminal bronchioles, where the engulfed particles are cleared by mucociliary action; (2) transport in alveolar surface fluids to ciliated airway; or (3) uptake by non-mobile alveolar cells. Macrophage uptake is more efficient for particles larger than 1  $\mu\text{m}$  (Oberdörster 1988), allowing insoluble UF particles to have longer residence times in the AI region, and thus, more time to be taken up by epithelial cells and/or translocate intact across the alveolar epithelium. Translocation, discussed below, is a forth mechanism for particle clearance from the AI region, and noted separately here since the particles remain intact the body.

Respiratory tract disease (e.g., acute respiratory infections or COPD) can slow the clearance of insoluble particles from the TB region of the lungs, although there is conflicting information in the literature on this issue as summarized in EPA (2009a), Phalen et al. (2010), Scheuch et al. (2008) and Brown et al. (2002). No difference related to disease was observed in the AI region (EPA 2009a) possibly because RT disease influences where particles deposit, and when there are narrowed RT airways (AI region), particles tend to have more proximal deposition in the lungs. Also, RT infections can slow TB clearance rates, and hence stimulate life-saving coughing fits.

Soluble particles dissolve in the lung fluid into their constituents that are cleared by mechanisms that depend on the site of deposition, particle surface area, chemical composition of the particle, particle surface coatings, and molecular weights of the dissolved constituents (EPA 2009a). Soluble particles and constituents may be absorbed by, or diffuse through, the epithelial layers of the TB and AI regions. The rate of this mechanism would be inversely proportional to molecular weight. In the TB region, Lay et al. (2003) observed that mucociliary transport may be slower for soluble than insoluble particles. Water soluble metals may transport through the lung via transepithelial absorption more rapidly than low water or low acid soluble metals and may lead to translocation to other organs within 24 hours (Wallenborn et al. 2007). This is likely important for redox active metals (e.g., often Ni and V as well as Zn, Al, Cd, Fe(II), Pb) (Chen and Lippmann 2009), since they can generate reactive oxygen species (ROS), which can lead to adverse health outcomes.

### Particle translocation

The process of extrapulmonary translocation primarily refers to the migration of particles across the lung fluid and tissue lining to the circulatory systems (Kreyling et al. 2002; Geiser et al. 2005). The mechanisms of translocation are varied including blood and lymph systems, and particle movement through membranes. Once in the circulatory system, particles can translocate to other organs in the body with the potential to cause adverse health effects in the impacted organs (Ferin et al. 1992; Geiser et al. 2005; Semmler et al. 2004; EPA 2009a; Nemmar et al. 2002; Mühlfeld et al. 2007; Brown et al. 2002; Oberdörster et al. 2004; Renwick et al. 2004). Under normal physiological conditions, the translocation rate of insoluble particles with diameters larger than about 200 nm is assumed to be negligible (Geiser and Kreyling 2010). After deposition in the AI region, UF particles, which have a longer residence time in the AI region than larger particles, can enter circulation through the air-blood tissue barrier in the alveoli (Geiser et al. 2005). Results indicate that small but detectable amounts of circulating UF particles can accumulate in secondary target organs such as the liver, spleen, heart, kidney, and brain (Kreyling et al. 2002; Semmler et al. 2004). The most studied UF particle types included particles composed of carbon, TiO<sub>2</sub>, Au, and Ir. For human lungs, however, only one study exists that describes a rapid and significant translocation of inhaled and deposited carbonaceous UF particles to systemic circulation and, thus, to secondary organs (Nemmar et al. 2002). In most other studies translocation for iridium (Kreyling et al. 2002) or carbonaceous UF particles was minimal (Mills et al. 2006, Wiebert et al. 2006). It is accepted nowadays that the degree to which inhaled UF particles translocate into the circulatory system is rather small but it is significant (Mühlfeld et al. 2007; Figure 5). However, information on cumulative effects of this translocation process is lacking.

Another pathway of UF particle translocation, that does not involve systemic circulation, is transport via the olfactory nerve in the nose to the brain (EPA 2009a; Oberdörster 2010; Phalen et al. 2010 and references within). This latter mechanism was reported in 1934 (Brodie and Elvidge 1934) and recently brought to attention by Oberdörster et al. (2004), and Elder et al. (2006). Translocation of UF particles to

the brain from the nose has been observed in several species including humans (EPA 2009a; Oberdörster 2010; Phalen et al. 2010 and references within) and likely occurs in less than an hour. Comparison between rodents and humans may be difficult since rats and mice have much larger olfactory regions of epithelial mucosa (about 50% of the nasal epithelium) than people (about 5%), since humans are more visual than olfactory (Aschner et al. 2005). However the potential health impact of this mechanism is still not understood for air-pollutant particles (Doty 2008, 2009; EPA 2009a; Phalen et al. 2010; Oberdörster 2010).

#### *Subject characteristics*

Human characteristics that influence particle dose include body size, gender, race, age, and RT disease (the later noted above). Phalen et al. (2010) and EPA (2009a) summarized the effects these parameters have on RT deposition of particles. Dosimetric calculations are often normalized to body weight (or body mass) since airway size, lung volume, and minute ventilation vary with body size. Deposition in average men and average women vary, in part because average women have smaller ET and TB airways, which can shift deposition proximally. This effect can cause greater total deposition in the ET and TB regions; more rapid clearance; and reduced deposition in the AI region. Biological variables, primarily in the ET region influence deposition across race (Phalen et al. 2010, EPA 2009a). In healthy adults, age per-se does not seem to alter deposition; although differences may occur between young children and adults due to body size differences and differences in RT anatomy and ventilation parameters (Bennett and Zeman 2004; Ginsberg et al. 2005; Foos et al. 2008; EPA 2009a).

Young children are generally assumed to represent a susceptible population in comparison to adults (O'Neil et al. 2011, this issue). Evidence is limited, but potential effects in children also may include adverse birth outcomes, infant mortality, respiratory effects, such as cough, bronchitis, and asthma attacks, and possibly incomplete lung development. However, such effects are not yet firmly established.

Multiple factors affecting particle deposition in children include breathing pattern (healthy young children presumably breathe more through the nose than the mouth), possibly lower efficiency of particle deposition in the nose for children, smaller  $V_T$ , faster breathing rate, and differences in airway dimensions and shapes. Modeled and experimental deposition data for children were reviewed and compared by Isaacs and Martonen (2005). Results indicated that the deposited dose per unit surface area of the RT is greater in small children than adults at given similar levels of exertion. Bennett and Zeman (1998) performed clinical studies that indicated that the deposition rate of fine particles normalized to lung surface area might be greater in children than adults. They also noted a correlation of an increased deposition rate in children with higher body mass index, with heavier children having higher  $V_T$  and minute ventilation, both which led to a higher deposition rate.

Interspecies differences in the clearance of insoluble particles from the TB and AI regions have been noted (Snipes, et al. 1989; Phalen and Méndez 2009; EPA 2009a and references within). Mice and rats,

commonly used laboratory subjects, have faster clearance of insoluble particles than people. In the AI and TB region this may be due to shorter distances for removal (e.g., from the AI region, as particles captured by macrophages are transported to the TB region). Also, the lack of respiratory bronchioles in rodents, results in easier access to TB mucus, and thus, may lead to faster clearance.

## **The Fate of Deposited Particles in the Lung**

After having described the fraction which enters into the RT and the mechanisms for deposition and clearance within the RT, the fate of the deposited particles and their effects at the microscopic scale are presented. Specifically, the interactions of UF particles with the internal surface of the airways and the subsequent interaction with cells and cellular sub-structures are described.

The focus on UF particles is because research indicates that respiratory and cardiovascular diseases related to inhaled UF particles are frequent and increasing (Schultz *et al.* 2005). UF particles that enter cells are of particular risk because they can cause oxidative stress through the generation of ROS (MacNee 2001). Oxidative stress can cause a reduction in cell metabolic competence via a reduction in mitochondrial respiration as well as an increase in pro-inflammatory cytokine production and other cellular inflammatory reactions and toxicity that can lead to apoptosis, pulmonary and cardiovascular disease, or cancer (Donaldson *et al.* 2003, 2006; Poland *et al.* 2008). The generation of oxidants is likely due to the high organic content and pro-oxidative potential of UF particles (Oberdörster *et al.* 2005). UF particles deposited in the AI region have longer residence times, and thus, more time to interact with cells or translocate through lung tissue and into capillary blood vessels as noted earlier. Epidemiological studies also have convincingly shown the association between UF particles and adverse health effects (Pope *et al.* 1995; Peters *et al.* 1997, 2001; Wichmann *et al.* 2000; Künzli *et al.* 2005).

Understanding particle interactions at the tissue and cellular level allows for the development of hypotheses regarding possible adverse health effects, appropriate particle metrics, and causal relationships. An important goal of the microscale research is to define the properties of air pollutants and mechanisms of interaction that produce adverse health effects.

### Particle interactions with pulmonary fluids

The fate of inhaled and deposited particles depends on their physicochemical characteristics, deposition location, on the exposed cell types, and on other properties of their microenvironment. Upon deposition, particles contact the internal surface of the RT which is assumed to consist of a continuous extracellular fluid lining from the larynx to the most distal alveoli, including the surfactant (Gehr *et al.* 1990, Schürch *et al.* 1990). In the ET and TB airways the fluid layer is thought to consist of two phases, a *sol*/ phase, in which the cilia beat, and a viscous *gel*/ phase, the mucus layer (Kilburn 1968). Particle removal is by mechanisms discussed earlier. Throughout the RT, the surfactant system represents a first line of defense (Gehr *et al.* 1996). The composition and structure of the surfactant is complex and varies

between airways and alveoli as well as in the AI region during the breathing cycle. In general, it consists of a continuous phospholipid layer containing four different surfactant proteins (Gehr et al. 1990, Schürch et al. 1990) that reduce the surface tension at the air-lung tissue interface to help maintain a particle free surface. This is important in the AI region for gas exchange.

Inhaled particles deposited in the RT are displaced into the aqueous subphase below the surfactant film, by surface tensions and possibly other forces exerted on them by the surfactant film (Gehr et al. 1990, Schürch et al. 1990). Particles may be modified by surfactant components or coated with surfactant or surfactant components during the displacement process (Gil and Weiber 1971; Gehr et al. 1990, 1996; Schürch et al. 1990). As a result of the displacement, particles come first into contact with surfactant components in the aqueous subphase and then with the RT epithelium where they may interact with pulmonary cells, such as alveolar or airway epithelial cells and cells of the immune system (e.g., macrophages and dendritic cells) and be effectively removed from causing potential harm (Geiser et al. 2005; Holt and Stumbles 2000; Peters et al. 2004; Vermaelen and Pauwels 2005). In the AI region, UF particles may also translocate into the capillary blood and be transported to elsewhere in the body, as noted earlier. If particles are not removed by these processes, then they can cause pulmonary inflammation by the interaction with cells leading to a range of adverse health effects (de Haar et al. 2006).

#### Particle-cell interactions

After displacement into the hypophase, inhaled and deposited particles come into close contact with cell membranes, such as those associated with epithelial cells and cells that are part of the immune system and can enter these cells by a range of processes (Brandenberger et al. 2010).

All mammalian cells have a common membrane structure that allows or limits what can enter a cell, under most conditions. Cell membranes consist of a very thin film of lipid and numerous protein molecules, mainly held in place by noncovalent interactions (Singer and Nicolson 1972; Kendall 2007). The lipid molecules are arranged as a continuous double layer about 5 nm thick in cell membranes. This lipid bilayer serves as a relatively impermeable barrier to the passage of most water-soluble molecules. The protein molecules are interspersed within, and pass through the lipid bilayer and are referred to as transmembrane proteins. These proteins mediate specific functions, such as transporting molecules across the bilayer or catalyzing membrane-associated reactions. Some proteins serve as structural links that connect the cytoskeleton through the lipid bilayer to the extracellular matrix or an adjacent cell by integrins and cadherins, while others serve as receptors to detect and transduce chemical signals into the cells' environment (Eisenberg et al. 1984).

Small molecules can traverse the cellular plasma membrane through the action of protein pumps or channels, while macromolecules must be carried into cells in membrane-bound vesicles derived from the invagination and pinching-off of pieces of the plasma membrane to form endocytic vesicles. Micron-sized

particles, which cannot directly penetrate the cellular plasma membrane, enter cells via phagocytosis largely mediated by macrophages, granulocytes and dendritic cells. UF particles can enter the cells via a variety of endocytic pathways or by another, yet to be defined mechanism (Rothen-Rutishauser et al. 2007a). Apart from these mechanisms, it has been proposed that UF particles can passively pass through the cellular membrane with subsequent access to subcellular organelles such as the mitochondria and the nucleus (Gehr et al. 2010).

Depending on the entry mechanism, particles may be found in vesicles or free in the cellular cytoplasm (Rothen-Rutishauser et al. 2007b; Geiser et al. 2005) (Figure 6). While micron-sized particles are usually found in membrane-bound vesicles, in-vitro studies have shown membrane-free UF particles present in the cytoplasm where they can have direct access to cytoplasmic proteins, and important biochemical molecules in organelles (e.g., the respiratory chain in the mitochondria and the DNA in the nucleus), which may greatly enhance their toxic potential.

Once inside the cells, particles transported via endocytic pathways are thought to be distributed non-randomly due to intracellular trafficking pathways. However this might not be the case for UF particles, which can be found free in the cytoplasm. The intracellular trafficking and distribution of particles within the cell is of great interest since it can help identify the relationship among cellular responses and specific intracellular targets. Intracellular trafficking as well as the intracellular location of the particles will have an influence on their effects. However, many knowledge gaps remain in this area due to the technical difficulties of quantifying the intracellular distribution of UF particles, their allocation to specific subcellular structures, and their trafficking.

## **Dosimetry Modeling of Inhaled Particles**

Deposition in the RT is often estimated mathematically based on: (1) semi-empirical models; (2) traditional mechanistic models; or (3) newer sophisticated CFD modeling (Rostami 2009). Studies consider the total (entire RT) and/or regional (ET, TB, AI) deposition, or even more specific subsections within regions. The Association for Inhalation Toxicologists (Alexander et al. 2008) reviewed a number of studies and determined that the total inhaled delivered dose (DD, mg/Kg) can be estimated as follows:

$$DD = (C \times RMV \times D \times \eta_i) / BW \quad (\text{Eq. 2})$$

Where C is the pollutant concentration in air (mg/L); RVM is respiratory minute volume (L/min); D is the exposure duration (min);  $\eta_i$  is the inhalable fraction; and BW is body weight (kg). This is the dose to which the RT is *exposed*, but not the *deposited* dose. Including a RT deposition term, the deposition fraction, allows for an estimate of the initial total dose delivered and deposited (Alexander et al. 2008; Ginsberg et al. 2005; Méndez et al. 2010). Finlay and Martin (2008) estimated total and regional deposition fractions empirically, based on a review of a number of studies. They proposed separate equations for impaction, sedimentation, and diffusion deposition mechanisms. Separate equations were



given for total RT deposition during mouth breathing and nasal breathing, and regional deposition efficiencies.

Equation 2 and several semi-empirical models partially based on laboratory data, typically fit the existing clinical data well, but have limited predictive value for situations that were not tested. More complex traditional mechanistic mathematical models apply simplifying assumptions regarding airway structure to predict deposition in major regions of the RT based on particle deposition mechanisms (e.g. ICRP 1994, 1995). Traditional models include the ICRP (International Commission on Radiation Protection) and the multiple path particle dosimetry (MPPD, [http://www.ara.com/products/mppd\\_capabilities.htm](http://www.ara.com/products/mppd_capabilities.htm)) models. These models have been used to compare inhalation deposition in rats and humans Brown et al. (2005) and for adults versus children (3-mo old) by RT region Ginsberg et al. (2005). A comparison of these models is given in Figure 2 for both nose and mouth breathing, for rest (Figure 2a) and light exercise (Figure 2b), and for total and major RT regions (ET, TB, and AI) (EPA (2009a; ICRP 1995).

CFD models are even more sophisticated, being based on fundamental equations of airflow structure using iterative numerical techniques (Rostami 2009). In CFD models, particles are introduced into the predicated flow fields and deposit where they cross an airway-wall boundary. Many assumptions are used in CFD modeling, such as isothermal airflow, rigid airways, and spherical non-interacting particles. Challenges in validating CFD models are discussed in Oldham (2006). More recently, lung structure is being obtained by medical imaging techniques, such as magnetic resonance imaging, positron emission tomography, computed tomography, and ultrasound. Digitized data obtained from these methods potentially allow for more detailed and reliable CFD simulations of particle deposition using more realistic lung morphology. Results are dependent on numerous selectable inputs that still require validation. Also, due to the complexity of the pulmonary region, CFD models currently model only limited airway regions, e.g., from the nose and mouth to the 15<sup>th</sup> airway generation (Kleinstreuer and Zhang 2010). An example of CFD modeling is given in Figure 4 for UF (NP) particles and micrometer sized particles. From this figure, it is easy to see hot spots that develop for both UF particles and micrometer sized particles as well as the more uniform deposition for UF particles throughout the upper RT, as noted earlier.

## **Areas of Future Research**

Significant advancements have occurred over the last 5-7 years with regards to particle dosimetry as touched upon in this paper. However, as understanding has improved, many new questions have arisen that once addressed should further reduce uncertainty in estimating particle dose that will provide important information for health researchers examining mechanisms and the adverse health effects of PM pollution. Listed here is a subset of suggested research that should further reduce uncertainties across the source-to-dose-to-health effects continuum.

### *Source to Intake:*

- *iF* values should be obtained for scenarios in more developing countries, where urban population densities greatly exceed those in the developed world, where there are significantly more people on the streets receiving high levels of exposure, and where buildings likely allow for greater infiltration of particles.
- Further work should be done differentiating the *iF* resulting from exposures to low socio-economic status individuals and those of higher socio-economic status, as those with lower socio-economic status may be more vulnerable to exposures to PM as a result of living and working close to pollution sources.
- Continued communication with policy makers is needed to determine how to provide *iF* values with the needed attributes to best incorporate this tool into policy decisions, enabling them to determine the most efficient way for reducing PM exposures to improve health.
- Chemical processes in the atmosphere, including the formation of secondary pollutants and movement of pollutants between the gas and solid/liquid phases should be more clearly elucidated for incorporation into *iF* calculations.

#### *Fate in the Respiratory Tract*

- Identification of the appropriate dose metrics for PM, given the great variability of particles, exposed populations, and potential health effects, represents a major important challenge. Specifically, each adverse health-effect scenario may require an evaluation of the proper metric, or metrics.
- Understanding differences in the delivered dose among species, within strains, and animal models that are created by pretreatment to produce diseases will improve our understanding of the impact of these diseases in humans.
- Quantifying differences in deposition patterns, clearance phenomena, and air-pollutant effects in human subpopulations due to differences in anatomy and physiology, (e.g., for children, the elderly and the diseased) is needed to reduce uncertainty in dosimetric calculations, and the impact of changes in lung morphology due to differences among humans.
- Translating results between animal studies to *in-vitro* studies and from animal studies and from *in-vitro* studies to humans, particularly as far as dose is concerned is critical since many needed studies cannot be directly performed using human subjects, due to practical and ethical constraints.
- Computational fluid dynamic (CFD) models of PM deposition hold great promise for calculating local and regional deposition doses. Improvements are needed in computational capabilities to allow for a greater extent of the lung to be modeled and models should be more thoroughly evaluated. In addition, advances in medical imaging approaches should allow for more detailed airway morphometric data assisting computational studies to model beyond the 16<sup>th</sup> TB branch.

- Most traditional mechanistic and CFD dosimetric models are validated for ideal particles (e.g. smooth, uncharged, spheres). Realistic environmental particles are, as yet, to be modeled leaving significant uncertainty with regards to actual dose for ambient PM..
- Measuring and modeling particle deposition hot spots, quantifying enhancement factors associated with hot spots, and verifying modeling results are needed to understand the toxicological significance of hot spots.
- The influence of particle size and other properties on the mechanisms driving slow-bronchial clearance still need to be elaborated since this process may drive the fraction of particles and components retained in the lungs and allows UF particles to translocate across the air-blood tissue barrier to other parts of the body .

#### *Tissue, Cellular, and Sub-cellular Interactions*

- UF particles, their translocation, micro dosimetry, and health effects, are still in an early phase of understanding, although significant progress has occurred over the last decade. Engineered NPs represent a vast area of research that is just beginning to be explored as these particles become widely used in commercial applications.
- A major research frontier involves elucidating the interactions of PM, in its great variability, with the varied cells of the body. The complexities of particle fates in cells and the resulting potential toxicities (e.g. cytotoxicity, genotoxicity, immunotoxicity) and cascades of biological responses are still at the beginning of being understood.
- The coating of airborne particles (outside the body) with potentially toxic substances and the coating of the particles upon interaction with body fluids, tissues, and cells is just beginning to be investigated and is poorly understood. Coating with fluids of any kind, i.e., coating most likely with proteins, may be the key to better understand toxicity and health effects.
- The mechanism of entry into cells and specifically, interactions of UF particles with subcellular structures and possible enhancements in particle toxicity due to these interactions is recent area of research that should be pursued to better understand the mechanisms that drive PM health effects.

#### **Summary**

This paper provides a brief summary of the current science associated with air pollution dosimetry and begins to address the forth policy-relevant Science Question addressed in part at 2010 International Air Pollution and Health conference<sup>2</sup>. The paper begins at the macroscale describing the definition and

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<sup>2</sup> Air Pollution and Health, 2010 (Solomon et al. 2011) SQ4: "What advances have been made in understanding the relationships between exposure, both spatially and temporally, and estimates of dose that tie to health outcomes?"

attributes of the intake fraction ( $iF$ ), a parameter that provides an estimate of the amount of pollutant inhaled (not deposited or retained) relative to the mass emitted into the environment and allows for a better understanding and quantification of source-to-intake (inhaled) relationships. Inhalation of air pollution is the first step in dosimetry since the respiratory tract (RT) is the primary route for air pollutants to enter the body. The dimensionless parameter,  $iF$ , also allows for evaluation of how the amount inhaled of a pollutant might vary as a function of emissions control strategies or other interventions (van Erp et al. 2011, this issue) that reduce pollutant concentrations from a source(s). The next section describes what happens once a pollutant is inhaled and provides updates on recent advances in dosimetry and factors that relate to quantifying dose. A description of the processes and mechanisms of how pollutants, specifically particles, are deposited and cleared from the RT is given. Translocation, the movement of particles from the lungs to other parts of the body, also is discussed.

Looking at the microscale, the next section examines how UF particles interact at the cellular and sub-cellular levels. UF particles constitute a unique size range that recently has been identified as a key pollutant of interest, since they can penetrate deep into the lungs and have the ability to cross the air-blood barrier of the lung and translocate to other systems and organs. UF particles also have been shown to translocate to the brain directly from inside the nose.

Considerable uncertainty remains with regards to UF particle interactions within the RT, in part because most of the experimental data, except for the epidemiological data, have been obtained from animal and in vitro experiments. Data received from these experiments are providing a significant basic idea of how adverse health effects develop, but the extrapolation to humans is still not well defined and additional research is needed to better understand the adverse health effects of ultrafine particles in people.

The last section describes dosimetric modeling from semi-empirical to CFD modeling. Continued advances in application of medical imaging techniques to provide digital data bases for CFD models will allow CFD models to simulate the lower TB and AI regions of the lung.

A series of research needs also is presented

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Table 1. Some particle-related that may apply to the effects of inhaled air-pollutant particles. Note that these properties are not mutually exclusive.

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Mass related to aerodynamic size intervals (e.g. PM <sub>10</sub> , PM <sub>2.5</sub> , and coarse particles)
Size distribution properties (e.g. geometric standard deviation)
Number per unit volume of air
Surface area (e.g. BET surface), or projected area
Surface reactivity in the body
Chemical composition (e.g., metals (e.g., V, Ni), acids, OC, EC (or BC), organic species etc.)
Oxidative properties (e.g., reactive oxygen potential)
Mobility within the body
Fractal properties
Electrical properties (e.g. net charge and/or zeta potential)
Dissolution rates in biological fluids and other media
Volatility of particle components
Shape (e.g. aspect ratio of fibers)
Infectivity, irritancy, allergenicity, and odor (odor may be important for dogs and rodents)
Biochemical interactions and metabolic products
Presence of gaseous co-pollutants
Changes in exposure level (e.g. spikes in air-pollutant concentrations)
History of exposure (e.g. producing sensitization and/or adaptation)

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Table 2. Computed particle deposition enhancement factors (DEFs) vs. square patch dimension and particle diameter for unit specific gravity spheres.

DEF			
Patch size (mm)			
Particle Diam. ( $\mu\text{m}$ )	0.1	1.0	3.0
0.01	52	18	6.7
0.1	58	18	7.1
1.0	81	20	6.9
10.0	113	42	22

Data from Balásházy et al. (1999).

Table 3. Computational fluid dynamics modeled hot spot deposition enhancement factors (DEF) for an adult male at resting (5 L/min) and exercising (30 L/min) ventilation states. Source: Phalen et al. (2010).

Particle aerodynamic diameter ( $\mu\text{m}$ )	Ventilation (l/min)	DEF
1	5	107
2	5	110
5	5	115
1	30	80
2	30	190
5	30	380

Data from Balashazy et al. (2003).

Figure Captions.

Figure 1a. Diagrammatic representation of respiratory tract regions in humans. Structures are anterior nasal passages, ET1; oral airway and posterior nasal passages, ET2; bronchial airways, BB; bronchioles, bb; and alveolar interstitial, AI. Reproduced from EPA (2009)

Figure 1b. Human lung structure, tissue barrier structure. (a) Low magnification of human lung tissue, showing gas exchange parenchyma (GP), blood vessels (BV), and airways (AW); magnification x15. Courtesy of the Institute for Anatomy, University of Bern, Bern, Switzerland. (b) Low magnification scanning electron micrograph of gas exchange parenchyma, clearly showing alveoli (A); magnification x60. (c) Higher magnification scanning electron micrograph of alveolar duct (AD) with concentrically arranged alveoli (A) around it; magnification x180. Modified from Gehr et al. (1978). (d) Scanning electron micrograph of broken up interalveolar septum showing erythrocytes (EC) in a capillary and the air–blood tissue barrier (AT); A, alveoli; magnification x550. Modified from Gehr et al. (1978). (e) Low magnification transmission electron micrograph of interalveolar septa, with capillaries containing erythrocytes (black) meandering around a connective tissue frame; A, alveolar air; magnification x500. Modified from Gehr et al. (1978). (f) Higher magnification transmission electron micrograph of a capillary (C) with erythrocytes (black) in an interalveolar septum; the three layers of the air–blood tissue barrier (AT) can be recognized; A, alveolar air; magnification x4500. Picture sequence a-f from Chang et al. (2010), original source, Zhang et al. (2005). Reprinted from Journal of Aerosol Science, 36(2), Zhang et al. 2005, Copyright (2005), with permission from Elsevier.

Figure 2a. Comparison of total and regional deposition results from the ICRP and MPPD models for a resting breathing pattern ( $VT = 625 \text{ mL}$ ,  $f = 12 \text{ min}^{-1}$ ) and corrected for particle inhalability. Regions are extrathoracic, ET; tracheobronchial, TB; and alveolar, A. Panels a-b are for nose breathing; panels c-d are for mouth breathing. (Reproduced from EPA 2009a, Chapter 4)

Figure 2b. Comparison of total and regional deposition results from the ICRP and MPPD models for a light exercise breathing pattern ( $VT = 1250 \text{ mL}$ ,  $f = 20 \text{ min}^{-1}$ ) and corrected for particle inhalability. Regions are extrathoracic, ET; tracheobronchial, TB; and alveolar, A. Panels a-b are for nose breathing; panels c-d are for mouth breathing. (Reproduced from EPA 2009a, Chapter 4)

Figure 3. Intake Fraction ranges for various release scenarios. Ranges for PM<sub>2.5</sub> from power plants and industrial sources for various regions and countries are listed, with median values indicated, as well as from a hypothetical distributed energy power plant (DE) (Tainio et al. 2009; Levy et al. 2009; Hao et al. 2007; Heath et al. 2006; Li and Hao 2003; Wang et al. 2006). Ranges for mobile source emissions of PM<sub>2.5</sub> in the US and Mexico are included, as well as point values for NY street canyons and Hong Kong (Greco et al. 2007a; Zhou and Levy 2008; Greco et al. 2007b; Luo et al. 2010; Stevens et al. 2007). A range of indoor  $iF$  values for PM<sub>2.5</sub> from smoking



sources is included (Klepeis and Nazaroff 2006). Values resulting from school buses and wood smoke are also included (Marshall and Behrentz 2005; Ries et al. 2009), as well as for a range of VOCs and SVOCs from outdoor emissions (Bennett et al. 2002b).

Figure 4. Comparisons of micron- and nanoparticle deposition in an idealized upper airway model (Kleinstreuer and Zhang 2010). DEF, deposition enhancement factor. Reprinted from Annual Review Of Fluid Mechanics, 42, Kleinstreuer and Zhang 2010), Copyright (2010), with permission from Annual Reviews.

Figure 5. Quantitative analyses of translocation of deposited UF particles from the air into the capillary blood in alveoli. (a) Illustration of the observed (white columns) and the expected (black columns) UF particles within the four tissue compartments which were analyzed at 1 h after exposure, (b) 24 hours after exposure. The total chi-squared showed that the distributions of observed and expected particles differed significantly. There is only one compartment with an RDI > 1 and a substantial contribution to the total chi squared: the connective tissue is the only compartment that is preferentially targeted by the UF particles at 1 h after exposure and the capillary lumen is the only compartment that is preferentially targeted at 24 h after exposure respectively. Figures reproduced from Mühlfeld et al. (2007). Reproduced with permission from Particle and Fibre Toxicology 2007 per open access license.

Figure 6. A 549 epithelial cells (cell line) with gold UF particles (A, plain gold) and (B, PEG-coated gold). In (A), the particles (arrows) are localized within vesicles of different sizes, in (B) within a lysosome (arrows I) and in the cytosol (arrow II); n: nucleus, m: mitochondria; scale bars: 500 nm. Pictures from Brandenberger et al. (2010). Reprinted from Small, 6(15), Brandenberger et al. (2010), John Wiley and Sons, Copyright © 2010 Wiley-VCH Verlag GmbH & Co.

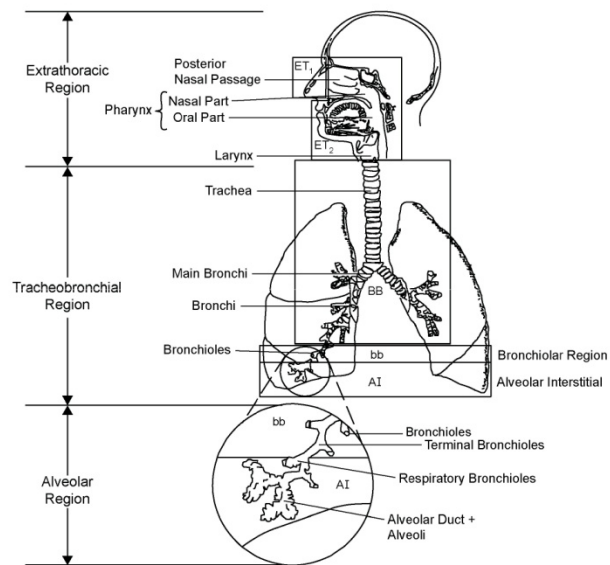


Figure 1a.

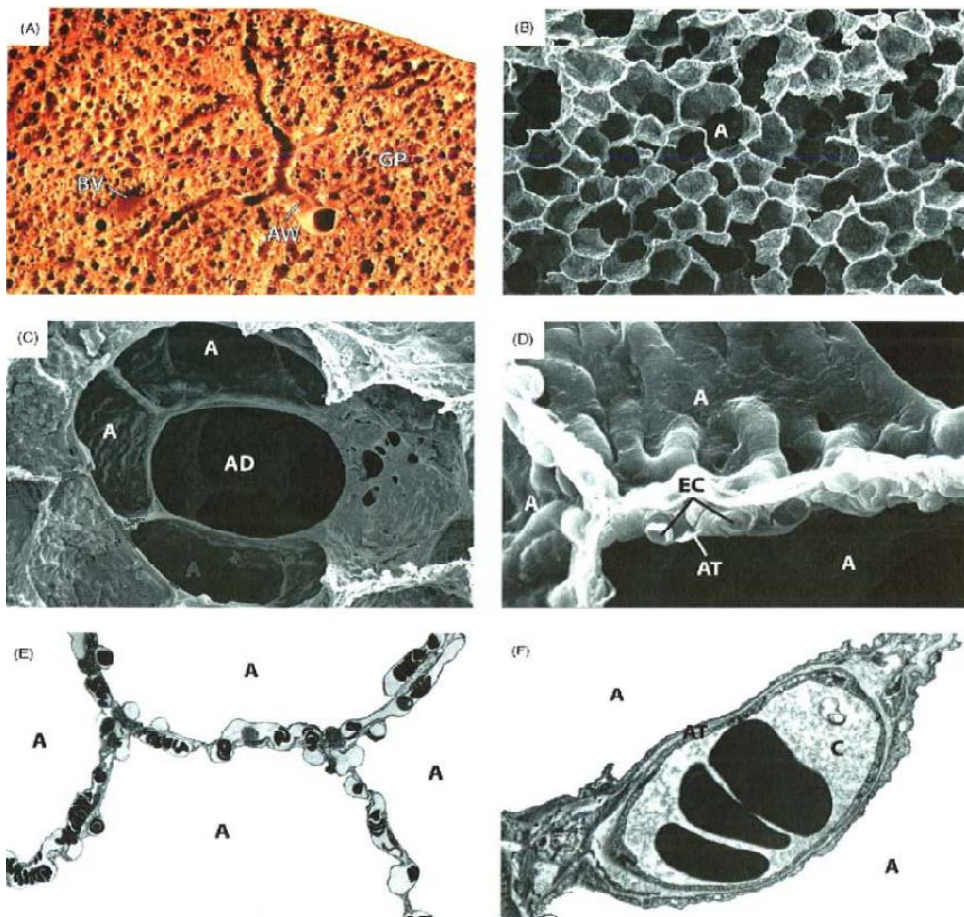


Figure 1b.

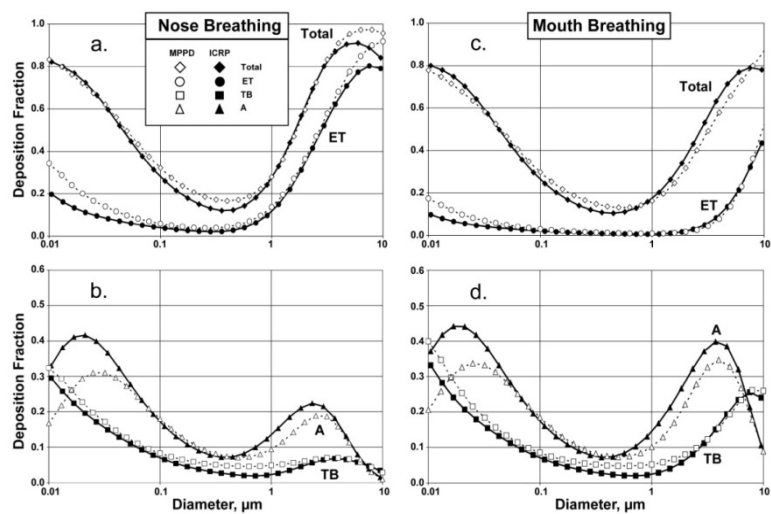


Figure 2a.

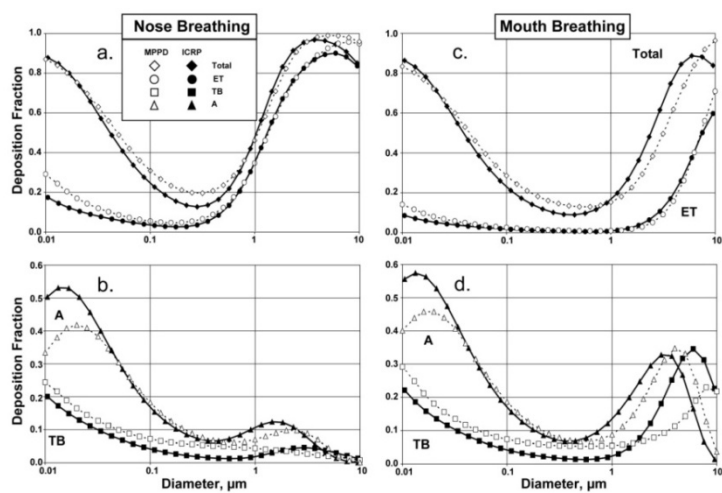


Figure 2b

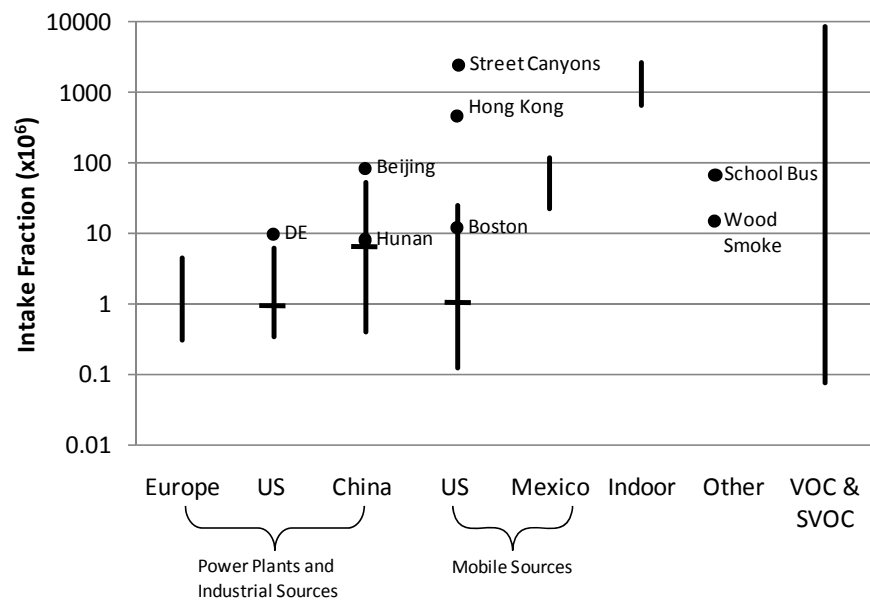


Figure 3.

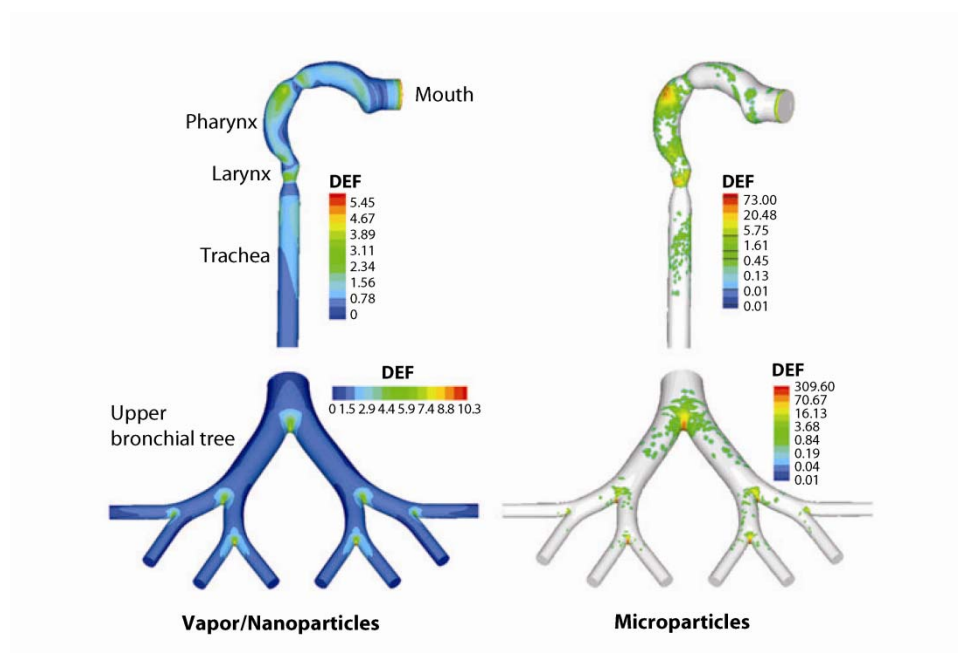
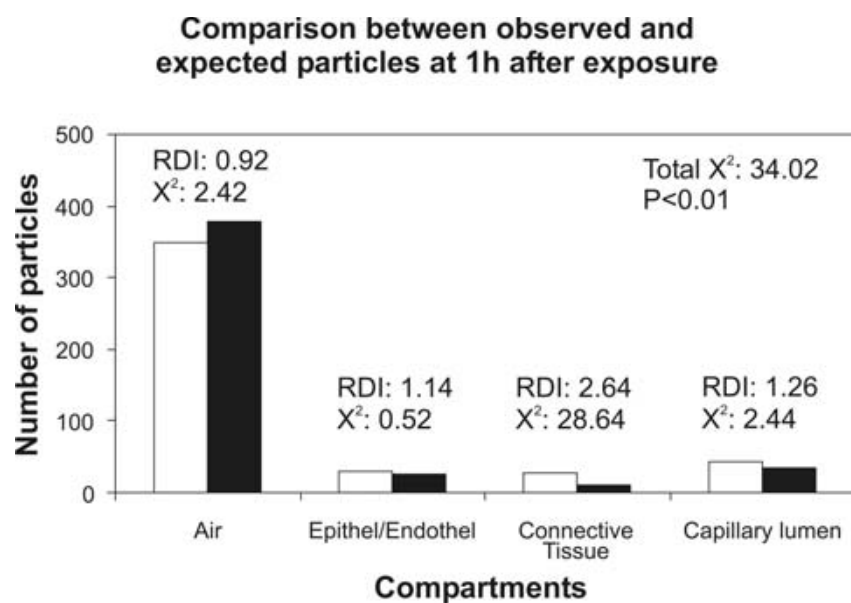


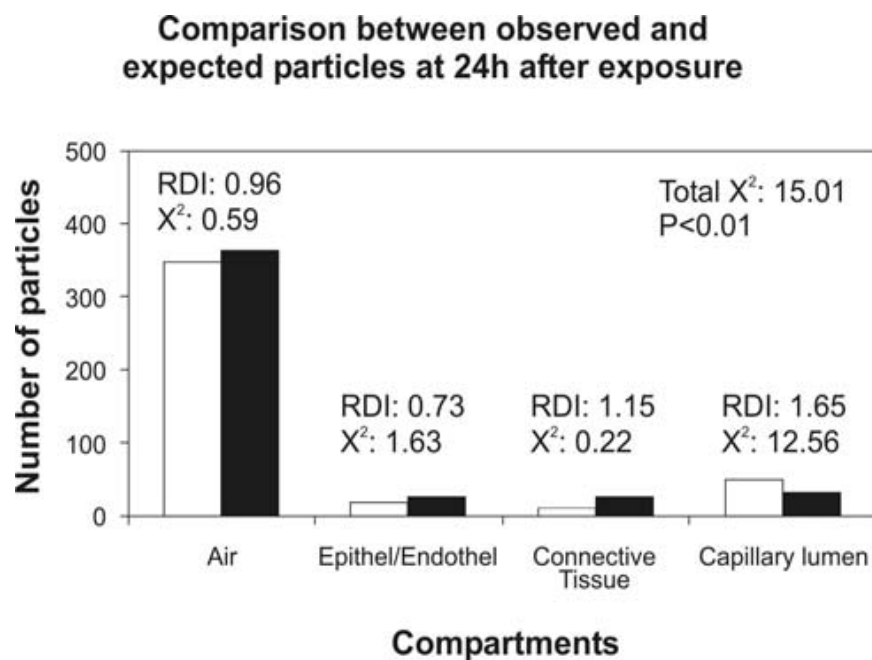
Figure 4.

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Figure 5a.



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Figure 5b.

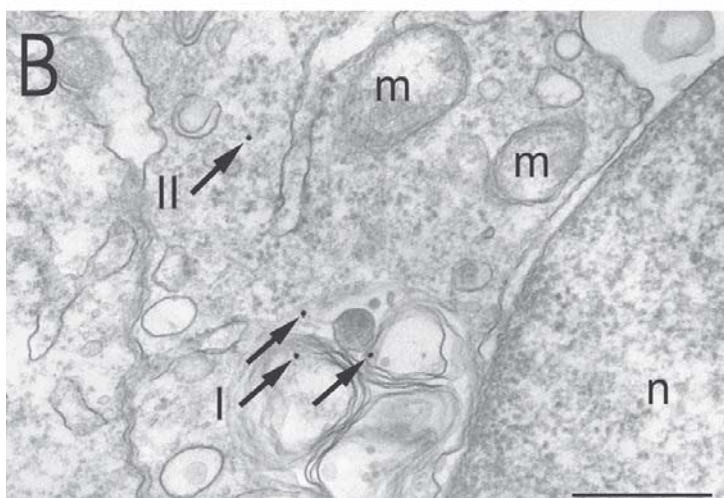
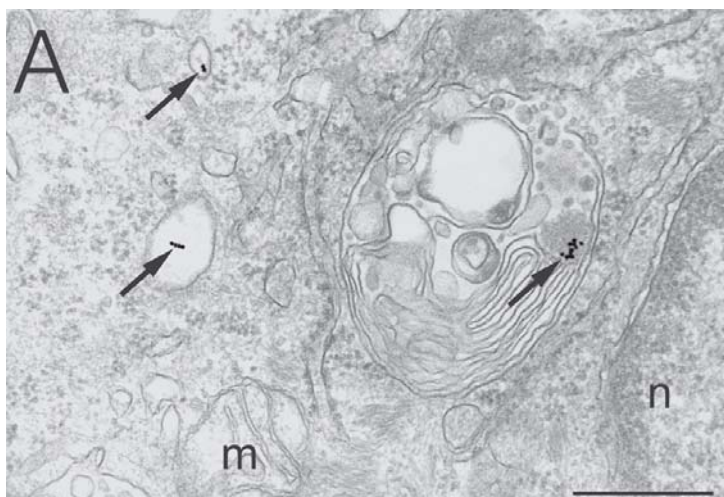


Figure 6.