

# Trends in Atmospheric Reactive Nitrogen for the Eastern United States

Robert W. Pinder<sup>a</sup>, K. Wyatt Appel<sup>a</sup>, Robin L. Dennis<sup>a</sup>

<sup>a</sup>US Environmental Protection Agency, Office of Research and Development

---

## Abstract

Reactive nitrogen can travel far from emission sources and impact sensitive ecosystems. From 2002 – 2006, policy actions have led to decreases in  $\text{NO}_x$  emissions from power plants and motor vehicles. In this study, atmospheric chemical transport modeling demonstrates that these emissions reductions have led to a downward trend in ambient measurements of transported reactive nitrogen, especially atmospheric concentrations and wet deposition of nitrate. The trend in reduced nitrogen, namely ammonium, is ambiguous. As reduced nitrogen becomes a larger fraction of the reactive nitrogen budget, wide-spread  $\text{NH}_3$  measurements and improved  $\text{NH}_3$  emissions assessments are a critical need.

### Capsule:

Ambient observations and simulation results indicate decreasing oxidized nitrogen levels over the Eastern US, but no trend in reduced nitrogen.

**Keywords:** reactive nitrogen, trends, atmospheric modeling

---

## 1. Introduction

In contrast to inert, diatomic  $\text{N}_2$ , atmospheric reactive nitrogen includes the forms of N that are chemically, biologically, or radiatively active. Due to the advent of the Haber-Bosch process and fossil fuel combustion, human production of reactive nitrogen has increased by an order of magnitude (Galloway et al., 2008). Oxidized nitrogen, such as  $\text{NO}_x$ , is largely emitted from combustion processes. When further oxidized to  $\text{HNO}_3$ , it rapidly deposits to surfaces. Reduced nitrogen, such as ammonia ( $\text{NH}_3$ ), is emitted from agriculture and can combine with  $\text{NO}_3^-$  or other anions to form fine particles. These particles can be transported far from their source and deposit into

11 sensitive ecosystems. This can lead to soil acidification, loss of plant diver-  
12 sity, eutrophication, algal blooms, and loss of fish and shellfish (Paerl, 1997;  
13 Alexander et al., 2001).

14 Over the past decade, regulatory policies have been put in place to sub-  
15 stantially reduce  $\text{NO}_x$  emissions. Accordingly, the US National Emission  
16 Inventory reports a steady decrease in  $\text{NO}_x$  emissions over the past decade  
17 (USEPA, 2008). However, emission trends are uncertain and should be crit-  
18 ically examined using measurements (Parrish, 2006; Dallmann and Harley,  
19 2010).

20 Several efforts have used measurements in North America to quantify the  
21 trend in  $\text{NO}_x$  concentration and have found large reductions that are most  
22 likely due to emissions. Recent work has attributed the decrease in urban  
23  $\text{NO}_x$  concentrations in the US (Godowitch et al., 2010) and Canada (Geddes  
24 et al., 2009) over the past decade to reductions in vehicle  $\text{NO}_x$  emissions.  
25 Space-based measurements have observed large  $\text{NO}_2$  reductions over the Ohio  
26 River Valley from 1999-2005 (Kim et al., 2006) as well as over urban areas  
27 in California from 2005-2008 (Russell et al., 2010).

28 In contrast to  $\text{NO}_x$ , there are few regulations relevant to  $\text{NH}_3$  emissions.  
29 Atmospheric  $\text{NH}_3$  concentration measurements in the Eastern US have been  
30 infrequent. Prior European assessments have demonstrated that interpret-  
31 ing the trend in reduced nitrogen without  $\text{NH}_3$  measurements is challenging  
32 (Fowler et al., 2005; Fagerli and Aas, 2008).

33 Despite these challenges, our goal is to understand the trend in reactive  
34 nitrogen that is transported from emission sources to sensitive ecosystems.  
35 While prior work in the US has characterized trends near emission sources  
36 and urban areas, we will quantify the trends found in the ambient mea-  
37 surements of particles and wet deposition fluxes from regional monitoring  
38 networks. We will use an atmospheric chemistry and transport model to  
39 simulate the gas and particle phase transformations and ultimately, under-  
40 stand if the trends in the ambient measurements can be explained by the  
41 estimated trend in emissions.

## 42 2. Methods

### 43 2.1. Observations

44 To understand the trend in reactive nitrogen, we analyze monitoring net-  
45 works designed to capture the gradient of concentration and deposition that

spans from emission sources to natural areas. This includes the Clean Air Status Trends Network (CASTNet) and the National Trends Network (NTN), part of the National Atmospheric Deposition Program. The location of the monitoring sites for both networks is shown in Figure 1.

CASTNet includes 69 sites in our modeling domain. Gas-phase nitric acid ( $\text{HNO}_3$ ), particle nitrate ( $\text{NO}_3^-$ ) and particle ammonium ( $\text{NH}_4^+$ ) are measured using a three-stage filter pack that is collected and replaced every seven days. Because the seven-day sampling period can cause measurement artifacts, we compare our simulations to the total nitric acid, defined as  $\text{TNO}_3 = \text{HNO}_3 + \text{NO}_3^-$ . We do not compare  $\text{NO}_3^-$  during the warm season because the concentrations are low.

NTN measures weekly precipitation amount and the ammonium and nitrate precipitation concentration at 211 sites in our modeling domain. Our calculations are based on the deposition flux ( $\text{kg ha}^{-1}$ ), defined as the product of the precipitation rate and the ammonium or nitrate concentration.

## 2.2. Atmospheric model

To understand if the trend in the observations is due to a trend in emissions, we use the Community Multi-scale Air Quality (CMAQ) model. The inputs to CMAQ are the anthropogenic emissions and meteorological conditions; outputs are concentrations and deposition fluxes for each hour of the simulation. CMAQ employs an Eulerian grid structure to explicitly simulate biogenic emissions, gas-phase, aqueous, and mixed-phase chemistry, advection and dispersion, aerosol thermodynamics and physics, and wet and dry deposition. A description and evaluation of the CMAQ processes is available in Foley et al. (2010) and the emissions and meteorological inputs are described in Appel et al. (2010). CMAQ has a high level of skill in modeling the inorganic aerosol system (Nolte et al., 2008) and capturing the inter-annual variability in wet deposition flux of nitrate and ammonium (Appel et al., 2010).

The simulations include two spatial domains – a 36 km horizontal resolution domain that includes all of the continental United States, and a 12 km horizontal resolution domain that spans from Maine to west Texas, as shown in Figure 1. We restrict our analysis to the more spatially-resolved 12 km domain and use the results of the 36 km domain as boundary conditions. The simulations use 24 vertical layers from the surface to 100 hPa.



### 81 2.3. Emissions databases

82 Quantifying the emission trends for our CMAQ simulation requires a dif-  
83 ferent approach for each sector. At many power plants,  $\text{NO}_x$  emissions are  
84 directly measured. For vehicle emissions, we use the MOBILE6 model, which  
85 includes year specific estimates of vehicle miles traveled, fleet characteristics,  
86 and meteorological conditions. These two sectors are two-thirds of the to-  
87 tal  $\text{NO}_x$  emissions, and have had large emission reductions over this period  
88 (Figure 2). Wildfire and soil  $\text{NO}_x$  emissions are derived from year-specific  
89 satellite fire counts and precipitation events, respectively. For other sources  
90 of  $\text{NO}_x$ , the emissions factors and activity data are not as well characterized  
91 (NARSTO, 2005), and we assume no change over this period.

92 Because  $\text{NH}_3$  emissions are largely from livestock and fertilized fields (Pin-  
93 der et al., 2006), and depend on specific farming practices and meteorological  
94 conditions, these emissions are even more uncertain (Beusen et al., 2008).  
95 Unlike reported  $\text{NO}_x$  emissions, indicators of  $\text{NH}_3$  emissions, such as fertil-  
96 izer use (<http://www.ers.usda.gov/Data/FertilizerUse/>), do not have  
97 the same downward trend. Our CMAQ simulations assume no change in  
98  $\text{NH}_3$  emissions over this time, except motor vehicle  $\text{NH}_3$  as estimated by  
99 MOBILE6.

### 100 2.4. Defining trends

101 To calculate the trend in the observations and the simulations, we first  
102 pair the model and the observations in space and time. The data are averaged  
103 across all measurement sites for either the warm season (May – September)  
104 or the cold season (all other months). This distinction reflects the larger  $\text{NO}_x$   
105 emission reductions from power plants during the warm season designed to  
106 mitigate ozone (USEPA, 2005). Linear regression is used to estimate the  
107 trend in the observations and model results as percent change per year. We  
108 also calculate  $p$ , the likelihood that the trend is not statistically different  
109 than zero. If  $p$  is greater than 0.1, there is at least a 10% chance that the  
110 trend is due to chance alone, and we report “no trend”.

111 A confounding factor is that there are year-to-year differences in the  
112 weather that could obscure the trends. Using the criteria above, we do not  
113 find statistically significant trend in precipitation nor is there a trend in the  
114 CMAQ precipitation bias (Appel et al., 2010). Likewise, Godowitch et al.  
115 (2010) examined the year-to-year bias over this period in wind direction and  
116 mixing rates and concluded that year-to-year differences in the meteorology  
117 could not explain the trend in near-source  $\text{NO}_x$  concentrations.

### 118 3. Results and Discussion

119 The observed and modeled trends in concentrations and deposition of  
120 reactive nitrogen are shown in Table 1. The observations show statistically  
121 significant downward trends for oxidized nitrogen. However, for reduced  
122 nitrogen, CASTNet  $\text{NH}_4^+$  concentrations have a decreasing trend, yet wet  
123 deposition flux measured at NTN sites do not show a trend. These trends  
124 and the extent to which they are captured by the CMAQ simulations is  
125 discussed below.

#### 126 3.1. Trends in oxidized nitrogen

127 Both atmospheric concentration and wet deposition flux measurements of  
128 oxidized nitrogen show a statistically significant downward trend (Table 1).  
129 The warm season trends are larger than the cold season trends, due to the  
130 larger reduction in warm season emissions.

131 The warm season  $\text{TNO}_3$  and wet deposition show a high level of cor-  
132 respondence between the CMAQ simulation and the measured trend. To  
133 explore this in more detail, Figure 3 shows the difference between 2006 and  
134 2002 for CMAQ  $\text{TNO}_3$  as measured at the CASTNet monitoring stations.  
135 The change in CMAQ matches the CASTNet observations at most locations,  
136 suggesting that the emission inputs and CMAQ simulation are capturing the  
137 spatial variability in the emission change and the resulting impact on at-  
138 mospheric nitrate. Figure 4 shows the monthly averaged CMAQ  $\text{TNO}_3$  and  
139 measurements at CASTNet. During winter, cooler temperatures cause more  
140 of the  $\text{HNO}_3$  to be particle  $\text{NO}_3^-$  while the warm season has higher levels of  
141 gas-phase  $\text{HNO}_3$ . Because particle  $\text{NO}_3^-$  has a much lower rate of dry depo-  
142 sition and winter precipitation is less frequent, the concentration of  $\text{TNO}_3$   
143 increases in the winter. CMAQ is able to capture these seasonal differences,  
144 although the error increases in the fall and winter.

145 Compared to observations, the CMAQ-estimated trend has lower signif-  
146 icance for cold season measurements of aerosol  $\text{NO}_3^-$ . This is likely because  
147 (1) the trend in emissions and measured concentrations is lower in the winter  
148 and (2) the partitioning between gas-phase and particle-phase is sensitive to  
149 larger uncertainties in the winter compared to the summer. The inorganic  
150 aerosol system is sensitive to  $\text{NH}_3$  emissions and the role of heterogeneous  
151 chemistry in the rate of  $\text{HNO}_3$  formation (Dennis et al., 2008). Both of these  
152 quantities are especially uncertain during wintertime (Gilliland et al., 2006;  
153 Davis et al., 2008).



154 Because precipitation efficiently scavenges both particle  $\text{NO}_3^-$  and gas-  
155 phase  $\text{HNO}_3$ , uncertainties in the aerosol partitioning have less of an impact  
156 on the CMAQ simulated trend. CMAQ accurately captures the trend in both  
157 warm and cold season  $\text{NO}_3^-$  wet deposition flux.

### 158 3.2. Trends in reduced nitrogen

159 Unlike oxidized nitrogen, the trends in reduced nitrogen are more am-  
160 biguous. The NTN measurements of  $\text{NH}_4^+$  wet deposition flux and CMAQ  
161 simulation show no trend, yet the atmospheric  $\text{NH}_4^+$  concentrations at CAST-  
162 Net show a cold season downward trend, which is not replicated by CMAQ.

163 The decrease in CASTNet  $\text{NH}_4^+$  concentration is likely due to decreases in  
164 emissions of other pollutants that influence the partitioning between  $\text{NH}_3$  and  
165  $\text{NH}_4^+$ . Previous studies of  $\text{NH}_4^+$  trends in the US (Civerolo et al., 2010)  
166 and Europe (Fagerli and Aas, 2008) have found substantial dependence on  
167 sulfate ( $\text{SO}_4^{2-}$ ). CASTNet measurements of  $\text{SO}_4^{2-}$  and  $\text{NH}_4^+$  have a correlation  
168 of 0.85. However, from 2002 – 2006, there is no statistically significant trend  
169 in CASTNet  $\text{SO}_4^{2-}$  for either season, nor a trend for  $\text{SO}_2$  emissions. There is a  
170 small downward trend in cold season NTN  $\text{SO}_4^{2-}$  deposition ( $-1.5\% \text{ yr}^{-1}$ ). For  
171 the period 2002 – 2006, it is unlikely that sulfate changes alone are causing  
172 the  $\text{NH}_4^+$  trend.

173 Because there has been no change in the  $\text{NH}_4^+$  wet deposition flux and  
174 a substantial reduction in the  $\text{NO}_3^-$ , the simplest explanation is that the  
175 reduction in atmospheric  $\text{NH}_4^+$  concentrations is due to a reduction in the  
176 availability of  $\text{HNO}_3$ , rather than a reduction in  $\text{NH}_3$  emissions. This expla-  
177 nation is consistent with the CMAQ simulations, that have higher  $\text{NH}_3$  in the  
178 gas-phase, increased  $\text{NH}_3$  dry deposition flux, and less  $\text{NH}_4^+$  in the particle  
179 phase.

## 180 4. Conclusions

181 Over the Eastern US, ambient measurements have found a substantial  
182 downward trend in indicators of oxidized nitrogen. CMAQ simulations show  
183 that this downward trend in  $\text{TNO}_3$  and  $\text{NO}_3^-$  wet deposition can be explained  
184 by decreases in  $\text{NO}_x$  emissions.

185 Unlike oxidized nitrogen, the trend in reduced nitrogen is difficult to  
186 discern from ongoing measurements and CMAQ simulations. However, as  
187 reduced nitrogen becomes a larger fraction of the reactive nitrogen budget,

188 wide-spread  $\text{NH}_3$  measurements and improved  $\text{NH}_3$  emissions assessments  
189 are a critical need.

## 190 5. Acknowledgments

191 The authors thank Donna Schwede and two anonymous reviewers for  
192 helpful feedback. *Disclaimer:* Although this article has been reviewed by  
193 the EPA and approved for publication, it does not necessarily reflect EPA  
194 policies or views.

## 195 6. References

### 196 References

- 197 Alexander, R., Smith, R., Schwarz, G., Preston, S., Brakebill, J., Srinivasan,  
198 R., Pacheco, P., 2001. Nitrogen Loading in Coastal Water Bodies: An  
199 Atmospheric Perspective Coastal and Estuarine Studies. AGU press, Ch.  
200 Atmospheric Nitrogen Flux from the Watersheds of Major Estuaries of the  
201 United States: An Application of the SPARROW Watershed Model, pp.  
202 119–132.
- 203 Appel, K. W., Foley, K. M., Bash, J. O., Pinder, R. W., Dennis, R. L.,  
204 Allen, D. J., Pickering, K., 2010. A multi-resolution assessment of the  
205 Community Multiscale Air Quality (CMAQ) Model v4.7 wet deposition  
206 estimates for 2002 - 2006. Geoscientific Model Development Discussions  
207 3 (4), 2315–2360.
- 208 Beusen, A., Bouwmana, A., Heuberger, P., Van Drecht, G., Van Der Hock,  
209 K., 2008. Bottom-up uncertainty estimates of global ammonia emissions  
210 from global agricultural production systems. Atmospheric Environment 42,  
211 6067–6077.
- 212 Civerolo, K., Hogrefe, C., Zalewsky, E., Hao, W., Sistla, G., Lynn, B., Rosen-  
213 zweig, C., Kinney, P. L., 2010. Evaluation of an 18-year CMAQ simulation:  
214 Seasonal variations and long-term temporal changes in sulfate and nitrate.  
215 Atmospheric Environment 44 (31), 3745 – 3752.
- 216 Dallmann, T. R., Harley, R. A., 2010. Evaluation of mobile source emission  
217 trends in the United States. J. Geophys. Res. 115 (D14), D14305.

- 218 Davis, J. M., Bhawe, P. V., Foley, K. M., 2008. Parameterization of  $\text{N}_2\text{O}_5$   
219 reaction probabilities on the surface of particles containing ammonium,  
220 sulfate, and nitrate. *Atmospheric Chemistry and Physics* 8 (17), 5295–  
221 5311.
- 222 Dennis, R. L., Bhawe, P. V., Pinder, R. W., 2008. Observable indicators of  
223 the sensitivity of  $\text{PM}_{2.5}$  nitrate to emission reductions—Part II: Sensitivity  
224 to errors in total ammonia and total nitrate of the CMAQ-predicted non-  
225 linear effect of  $\text{SO}_2$  emission reductions. *Atmospheric Environment* 42 (6),  
226 1287 – 1300.
- 227 Fagerli, H., Aas, W., 2008. Trends of nitrogen in air and precipitation: Model  
228 results and observations at EMEP sites in Europe, 1980-2003. *Environmental Pollution* 154 (3), 448 – 461, reduced Nitrogen in Ecology and the  
229 Environment.  
230
- 231 Foley, K. M., Roselle, S. J., Appel, K. W., Bhawe, P. V., Pleim, J. E., Otte,  
232 T. L., Mathur, R., Sarwar, G., Young, J. O., Gilliam, R. C., Nolte, C. G.,  
233 Kelly, J. T., Gilliland, A. B., Bash, J. O., 2010. Incremental testing of the  
234 Community Multiscale Air Quality (CMAQ) modeling system version 4.7.  
235 *Geoscientific Model Development* 3 (1), 205–226.
- 236 Fowler, D., Smith, R., Muller, J., Hayman, G., Vincent, K., 2005. Changes  
237 in the atmospheric deposition of acidifying compounds in the UK between  
238 1986 and 2001. *Environmental Pollution* 137 (1), 15 – 25, recovery from  
239 acidification in the UK: Evidence from 15 years of acid waters monitoring.
- 240 Galloway, J. N., Townsend, A. R., Erisman, J. W., Bekunda, M., Cai, Z.,  
241 Freney, J. R., Martinelli, L. A., Seitzinger, S. P., Sutton, M. A., 2008.  
242 Transformation of the Nitrogen Cycle: Recent Trends, Questions, and  
243 Potential Solutions. *Science* 320 (5878), 889–892.
- 244 Geddes, J. A., Murphy, J. G., Wang, D. K., 2009. Long term changes in  
245 nitrogen oxides and volatile organic compounds in Toronto and the chal-  
246 lenges facing local ozone control. *Atmospheric Environment* 43 (21), 3407  
247 – 3415.
- 248 Gilliland, A. B., Appel, K. W., Pinder, R. W., Dennis, R. L., 2006. Seasonal  
249  $\text{NH}_3$  emissions for the continental United States: Inverse model estimation  
250 and evaluation. *Atmospheric Environment* 40 (26), 4986 – 4998, special



- 251 issue on Model Evaluation: Evaluation of Urban and Regional Eulerian  
252 Air Quality Models.
- 253 Godowitch, J. M., Pouliot, G. A., Rao, S. T., 2010. Assessing multi-year  
254 changes in modeled and observed urban  $\text{NO}_x$  concentrations from a dy-  
255 namic model evaluation perspective. *Atmospheric Environment* 44 (24),  
256 2894 – 2901.
- 257 Kim, S. W., Heckel, A., McKeen, S. A., Frost, G. J., Hsie, E. Y., Trainer,  
258 M. K., Richter, A., Burrows, J. P., Peckham, S. E., Grell, G. A., 2006.  
259 Satellite-observed US power plant  $\text{NO}_x$  emission reductions and their im-  
260 pact on air quality. *Geophysical Research Letters* 33 (22).
- 261 NARSTO, August 2005. Improving Emission Inventories for Effective Air  
262 Quality Management Across North America. Tech. Rep. NARSTO 05-001,  
263 NARSTO Emission Inventory Assessment Team.
- 264 Nolte, C. G., Bhawe, P. V., Arnold, J. R., Dennis, R. L., Zhang, K. M.,  
265 Wexler, A. S., 2008. Modeling urban and regional aerosols—Application of  
266 the CMAQ-UCD Aerosol Model to Tampa, a coastal urban site. *Atmo-  
267 spheric Environment* 42 (13), 3179 – 3191.
- 268 Paerl, H. W., 1997. Coastal eutrophication and harmful algal blooms: Impor-  
269 tance of atmospheric deposition and groundwater as “new” nitrogen and  
270 other nutrient sources. *Limnology and Oceanography* 42 (5), 1154–1165.
- 271 Parrish, D. D., 2006. Critical evaluation of US on-road vehicle emission in-  
272 ventories. *Atmospheric Environment* 40 (13), 2288 – 2300.
- 273 Pinder, R. W., Adams, P. J., Pandis, S. N., Gilliland, A. B., 2006. Tempo-  
274 rally resolved ammonia emission inventories: Current estimates, evaluation  
275 tools, and measurement needs. *J. Geophys. Res.* 111 (D16), D16310.
- 276 Russell, A. R., Valin, L. C., Bucsela, E. J., Wenig, M. O., Cohen, R. C.,  
277 2010. Space-based Constraints on Spatial and Temporal Patterns of  $\text{NO}_x$   
278 Emissions in California, 2005–2008. *Environmental Science and Technology*  
279 44 (9), 3608–3615.
- 280 USEPA, 2005. Evaluating Ozone Control Programs in the Eastern United  
281 States: Focus on the  $\text{NO}_x$  Budget Trading Program, 2004. Tech. Rep.  
282 EPA-454-K-05-001, US Environmental Protection Agency.

Figure 1: The CMAQ 36 km modeling domain (outer map), the 12 km domain used in this analysis (grey) and CASTNet (black) and NTN (white) sites used for this analysis. Note that some CASTNet and NTN sites are co-located.

Figure 2: The National Emission Inventory reports a decrease in  $\text{NO}_x$  emissions from power plants and vehicles (two-thirds of the total), but indicators of  $\text{NH}_3$  emissions (fertilizer use) do not have a downward trend.

283 USEPA, January 2008. Latest findings on national air quality, status and  
284 trends through 2006. Tech. Rep. EPA-454/R-07-007, U.S. Environmental  
285 Protection Agency.

Figure 3: Warm season averaged CMAQ and CASTNet (circles) change in total nitrate from 2002 to 2006.

Figure 4: Monthly averaged CMAQ (blue) and CASTNet (black) total nitrate concentration. Grey vertical lines denote January.



Table 1: Summary of trends in reactive nitrogen, as observed and as modeled by CMAQ. If there is a least a 10% chance that the trend is due to chance alone, we report “no trend” ( $p > 0.1$ ).

	Observations Trend Significance Level ( <i>p</i> )	Observations % change per year	CMAQ % change per year	CMAQ Trend Significance Level ( <i>p</i> )	
<i>Air Concentration</i>					
CASTNet TNO <sub>3</sub>					
warm season		-6.4%	0.001	-4.8%	0.001
cold season		-1.7%	0.1	-1.1%	0.1
CASTNet NO <sub>3</sub> <sup>-</sup>					
cold season		-2.3%	0.01	-1.5%	0.1
CASTNet NH <sub>4</sub> <sup>+</sup>					
warm season		no trend	—	no trend	—
cold season		-1.9%	0.1	no trend	—
<i>Wet Deposition</i>					
NTN NO <sub>3</sub> <sup>-</sup>					
warm season		-3.2%	0.001	-3.6%	0.01
cold season		-2.7%	0.001	-2.7%	0.01
NTN NH <sub>4</sub> <sup>+</sup>					
warm season		no trend	—	no trend	—
cold season		no trend	—	no trend	—
<i>Dry deposition</i>					
Oxidized N					
warm season		direct		-2.0%	0.001
cold season		measurements		-1.0%	0.01
Reduced N					
warm season		are		1.1%	0.01
cold season		not available		1.1%	0.01

Figure 1  
[Click here to download Figure: Figure1.eps](#)

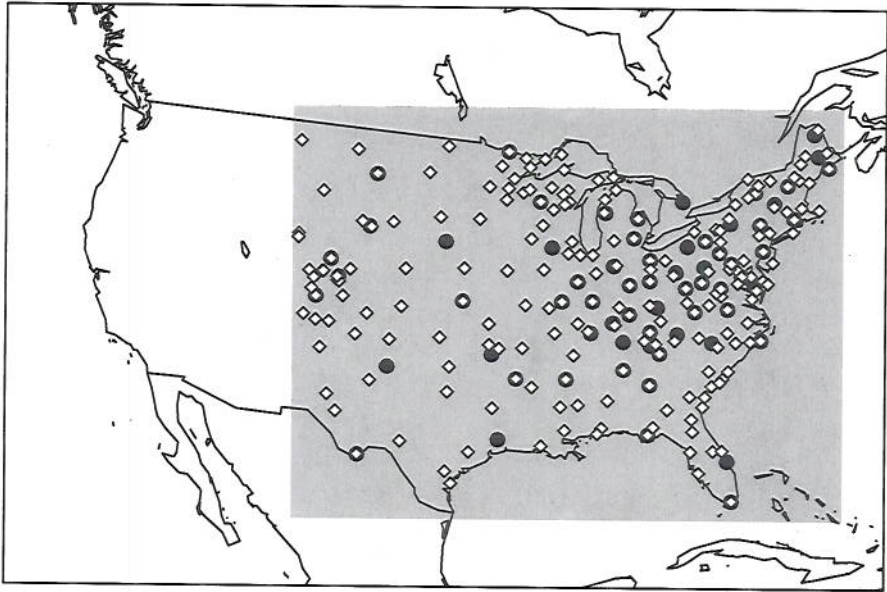


Figure 2

[Click here to download Figure: Figure2.eps](#)

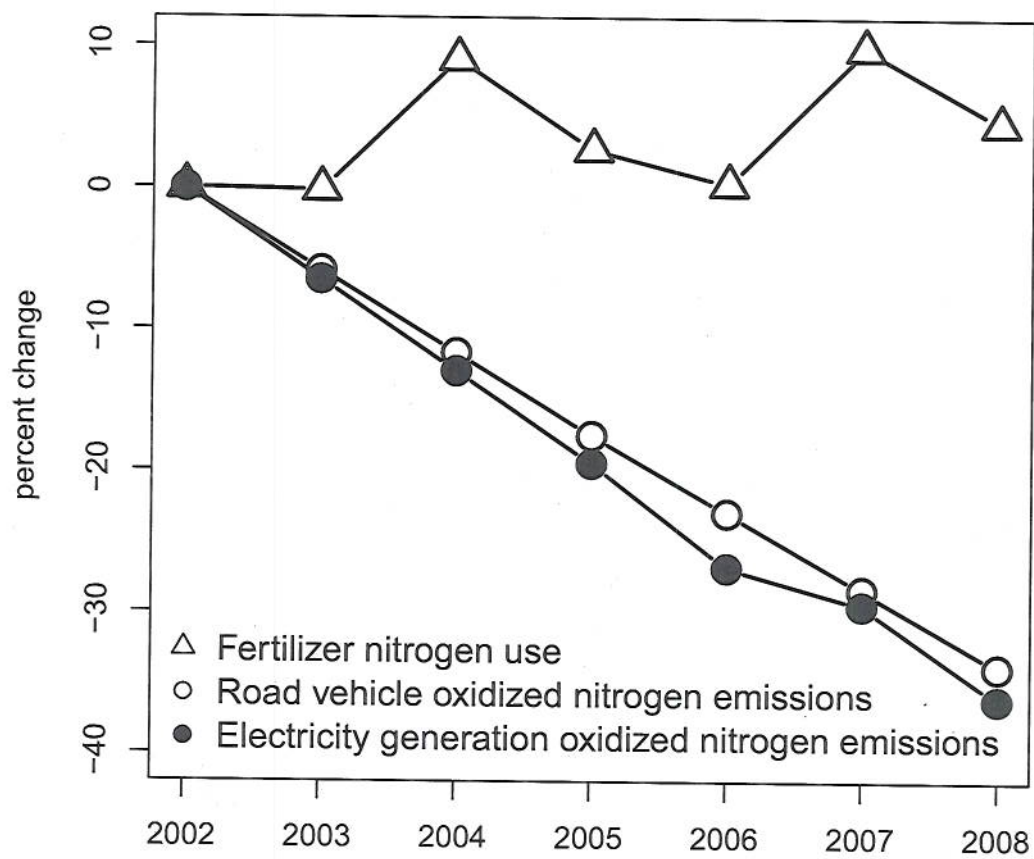




Figure3

[Click here to download high resolution image](#)

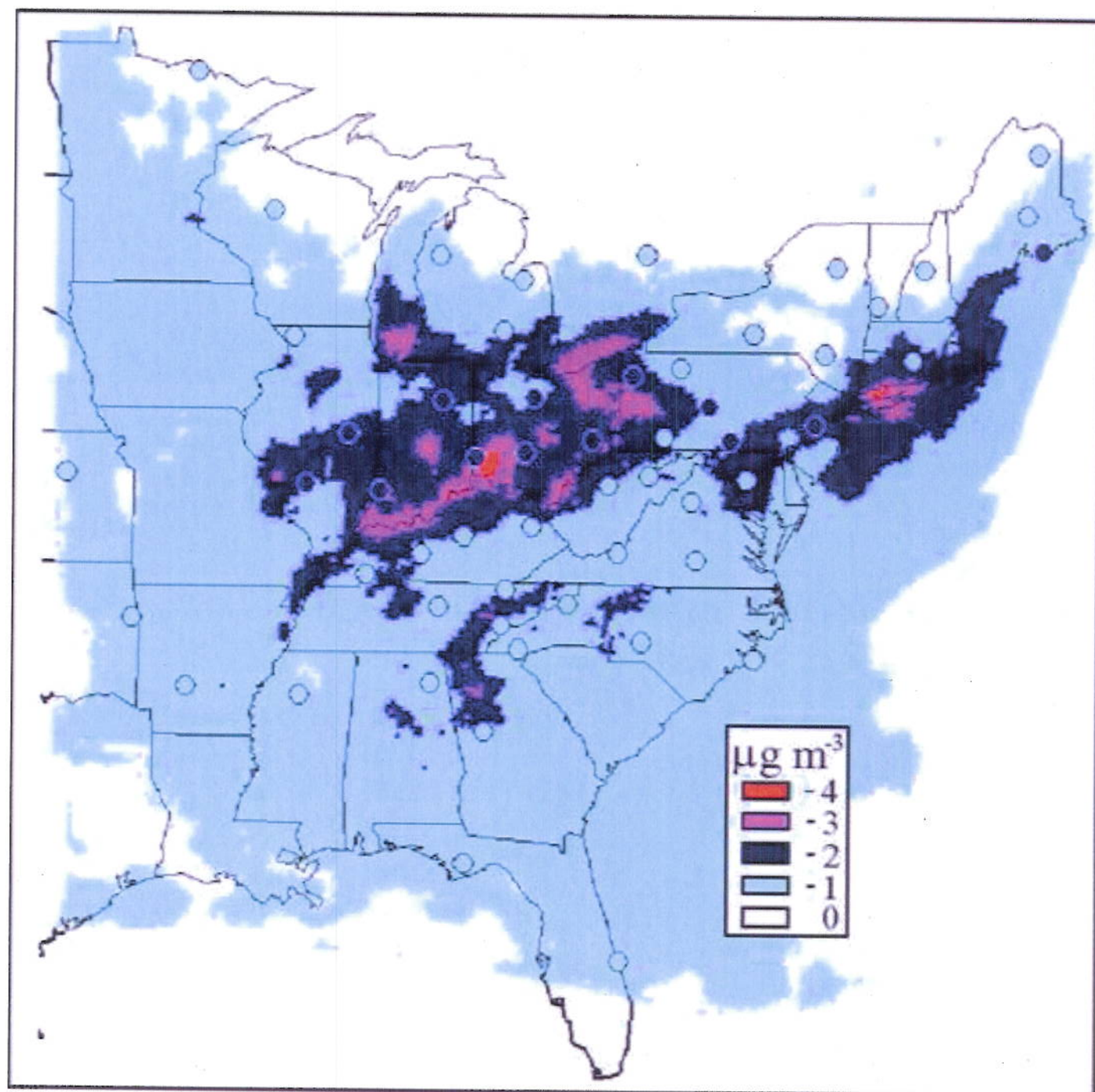


Figure 4

[Click here to download Figure: Figure4.eps](#)

